

Answer EVERY question from section A and TWO questions from section B.

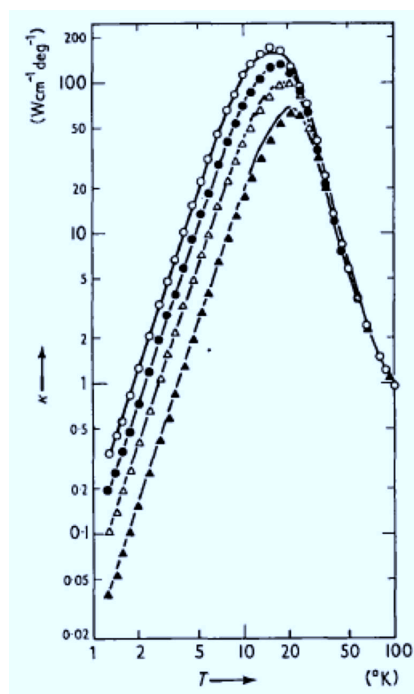
The numbers in square brackets in the right-hand margin indicate the provisional allocation of maximum marks per sub-section of a question.

Mass of the electron	$m_e$	=	$9.11 \times 10^{-31}$ kg
Charge on the electron	$e$	=	$-1.602 \times 10^{-19}$ C
Permittivity of free space	$\epsilon_0$	=	$8.854 \times 10^{-12}$ F m <sup>-1</sup>
Boltzmann's constant	$k_B$	=	$1.38 \times 10^{-23}$ J K <sup>-1</sup>
Planck's constant/ $2\pi$	$\hbar$	=	$1.05 \times 10^{-34}$ J s
Speed of light	$c$	=	$3 \times 10^8$ m s <sup>-1</sup>

SECTION A

[Part marks]

1. Sketch how a crystal lattice becomes distorted in the presence of an *edge dislocation*. [3]  
Describe how the strength of materials can be reduced by the presence of dislocations. [3]



2. The graph above shows the thermal conductivity of several samples of LiF measured as a function of temperature. What physical process is responsible for the conductivity increasing with temperature on the left side of the graph? [3]  
What physical process is responsible for the conductivity decreasing with temperature on the right side of the graph? [3]

3. Sketch the dispersion relation of the mechanical waves travelling in a one-dimensional crystal made of alternating atoms of different masses, connected with identical springs. Label the axes. Label the various branches with their usual names. [3]

Describe all the possible motions of the atoms in the allowed (longitudinal) waves in the long wavelength limit. [4]

4. An Aluminium ( $Z = 13$ ) atom substitutes for one of the silicon ( $Z = 14$ ) atoms in a crystalline block of silicon. Explain how its bonding into the lattice gives rise to a 'hole'. [4]

Show with the aid of a suitable diagram where the energy of this 'hole' lies with respect to the electronic levels of the silicon host. [3]

5. A hypothetical potential function is proposed for the interaction between two atoms of an inert gas,

$$U(r) = U_0 \left[ \left( \frac{r_0}{r} \right)^9 - \left( \frac{r_0}{r} \right)^6 \right]$$

Which of the two terms represents the *attractive* part of the potential? [3]

A molecule is composed of three atoms of the gas in a triangular configuration. What is its equilibrium bond length? [4]

6. What are the allowed energies of free electrons occupying a one-dimensional box of length  $L$ ? Sketch some typical wave functions. [4]

If  $N$  electrons are placed in the box in the minimum energy configuration (at  $T = 0$ ), what is the energy of the most energetic electron? [3]

SECTION B

7. A *monovalent* element forms a two-dimensional solid with a rectangular unit cell of lattice constants  $a$  and  $b$  with  $a > b$  and one atom per unit cell. Consider the allowed states of *free* electrons in a finite piece of the substance with dimensions  $L \times L$ .

(a) Is it generally expected that a monovalent solid would be a metal or an insulator? Why so? [5]

(b) Draw labelled diagrams of the direct space lattice and the corresponding reciprocal lattice showing the first Brillouin Zone and the free-electron Fermi surface. Indicate clearly which states are filled and which are empty at  $T = 0$ . [7]

(c) What condition(s) on  $a$  and  $b$  are required for the free electron Fermi surface to extend into the second Brillouin Zone? Illustrate how this transition takes place in your reciprocal space diagram of the material. [8]

(d) Now consider the nearly-free electron approximation where a band gap opens at the zone boundary. Show how the Fermi surface distorts instead of filling the second zone. [4]

(e) If you could continuously distort the lattice through this transition, how would you expect the electrical conductivity to vary along the  $a$  and  $b$  directions? [6]

8. In two dimensions, the Debye theory of the specific heat of a solid predicts the internal energy of a solid containing  $N$  atoms to be:

$$U = X(T) \int_0^{x_D} \frac{x^2 dx}{e^x - 1}$$

where  $x = \hbar\omega/k_B T$  is the dimensionless form of the vibration frequency. In this problem, you may find it helpful to use the symbol  $\zeta_2$  for following standard definite integral  $\zeta_2 = \int_0^\infty \frac{x^2 dx}{e^x - 1}$ .

(a) According to this theory, what is the physical origin of the internal energy? What are the standard approximations made in the Debye model? [8]

(b) Derive an expression for the quantity  $x_D$  which appears above in terms of these approximations. [6]

(c) By equating the *high-temperature* limit of the above expression to the classical 2D result  $U = 2Nk_B T$ , deduce the unknown function  $X(T)$ . [7]

(d) Hence derive an expression for the *low-temperature* specific heat of a 2D solid in the Debye theory. [9]

9. (a) Sketch the typical energy band diagram for electrons in a semiconductor, indicating which states are empty and which are filled at  $T = 0$ . Which part(s) of the diagram are responsible for the “effective masses”  $m_e^*$  and  $m_h^*$ ? Label the energies of the conduction and valence band edges as  $E_C$  and  $E_V$ . [8]

(b) Deduce an expression for the “density of states”, the number of electron states per unit energy in the conduction band in terms of  $E_C$  and  $m_e^*$ . [6]

(c) Hence derive the following expression for the total number of electrons per unit volume in the conduction band,

$$n = 2 \left( \frac{m_e^* k_B T}{2\pi\hbar^2} \right)^{3/2} e^{(\mu - E_C)/k_B T}.$$

You may wish to use the standard integral  $\int_0^\infty \sqrt{x} e^{-x} dx = \frac{1}{2}\sqrt{\pi}$ . [9]

(d) What physical significance is attributed to the symbol  $\mu$ ? By equating the expression above with an equivalent one for the number of holes in the valence band, deduce an equation for  $\mu$  for an intrinsic semiconductor. [7]

10. The diffraction angles of the first three peaks of the powder diffraction pattern, measured from a certain sample at room temperature, are listed below. X-rays of wavelength  $\lambda = 0.1542\text{nm}$  were used.

When the sample is cooled, some of the peaks are found to split into pairs of two angles given in the second column.

$2\theta$ angle Room temperature	$2\theta$ angle Low temperature	
22.80°	22.69°	22.86°
32.47°	32.43°	32.55°
40.05°	40.05°	

(a) Why is diffraction only seen at these angles shown and not in between? State the law that connects the Bragg angle,  $\theta$ , and the crystal lattice spacing. [6]

(b) Deduce the crystal symmetry and the lattice parameter of the room temperature sample from the observed angles. [10]

(c) At what  $2\theta$  angle would you expect the next peak to appear for the room temperature sample? [6]

(d) Explain with as much detail as possible what happens to the crystal structure at low temperature. [8]