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UNIVERSITY COLLEGE LONDON

University of London

EXAMINATION FOR INTERNAL STUDENTS

For The Following Qualification:-

M.Sc.

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M11: Fluid Particle Systems

COURSE CODE	: CENG0M11
DATE	: 06-MAY-03
TIME	: 10.00
TIME ALLOWED	: 3 Hours

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TURN OVER

Answer FOUR QUESTIONS. Answer at least ONE question from each section. Only the first FOUR answers will be marked. ALL questions carry a total of 20 MARKS each, distributed as shown []

Section A

- 1.
- (i) Derive an overall, steady-state force balance for a control volume of a fluidised bed and use it to obtain an expression for the fluid pressure drop Δp and the *unrecoverable pressure loss* ΔP across the control volume [4]

Use this to show that the unrecoverable pressure loss across the whole bed, ΔP_B , remains independent of the fluid velocity and hence sketch the theoretical relationship between ΔP_B and volumetric flux of the fluid, u, starting at u = 0. Describe how this compares with experimental observations giving reasons for any differences. [3]

(ii) Show that the effect of the fluid pressure gradient gives rise to the following expression for the effective weight w_e (weight minus buoyancy force) of a particle immersed in a fluidised bed:

$$w_e = V(\rho_p - \rho_f)g\varepsilon,$$

where V is the particle volume, ρ_p and ρ_f are the particle and fluid densities, and ε is the void fraction. [4]

(iii) Derive an expression for the expansion characteristics $\varepsilon(u)$, of a homogeneously fluidised bed on the basis of the following constitutive relation for the pressure drop through a bed of particles:

$$\Delta P = K u^a \, \varepsilon^b,$$

where K is a constant, u is the fluid flux, ε is the void fraction, and the parameters a and b depend on the particle Reynolds number. [7]

What do experimental measurements reveal about the values of a and b. [2]

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- 2.
- (i) Describe carefully, with the aid of sketches, how a homogeneously fluidised bed responds to both a sudden increase and a sudden decrease in the fluid velocity. Hence confirm that the bed surface velocity, following such a change, may be assumed to travel at velocity $U_2 - U_1$, the difference in fluid fluxes after and before the flow rate change. State any simplifying assumptions that are made, and discuss practical limitations to the simple theory. [10]
- Use the analysis of part (i) to evaluate the kinematic wave speed for a fluidised suspension in terms of the Richardson-Zaki parameter n. [4]
- (iii) Discuss briefly the significance of the kinematic wave speed for the stability of the state of homogeneous fluidisation. [6]

Section **B**

3.

(i) Experiments have to be carried out to study elutriation (or carryover) of small spherical mono-size particles from a gas-fluidized bed. In order to identify the operative conditions in the fluid-bed, the particle terminal fall velocity, u_t , has to be determined.

The force balance on a particle, written as a function of the Reynolds and Archimedes dimensionless numbers, may be applied to give u_t .

$$C_D R e_i^2 = \frac{4}{3} A r$$

where:

$$Ar = \frac{d_p^3 \rho_f(\rho_p - \rho_f) g}{\mu^2}$$

Assuming creeping flow regime, calculate the particle terminal fall velocity with:

 $d_p = \text{particle diameter} = 40 \ \mu m$ $\rho_p = \text{particle density} = 1400 \ \text{kg m}^{-3}$ $\rho_f = \text{gas density} = 1.22 \ \text{kg m}^{-3}$ $\mu = \text{gas viscosity} = 1.8 \times 10^{-5} \ \text{kg m}^{-1} \ \text{s}^{-1}$ $g = \text{acceleration gravity} = 9.81 \ \text{m s}^{-2}$

[10]

(ii) Discuss the effect of increasing gas temperature and pressure on the calculation of the particle terminal fall velocity. [6]

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(iii) To avoid or reduce elutriation of particles from the fluidized bed, the operative gas velocity has to be kept between the minimum fluidization velocity, u_{mf_i} and the particle terminal fall velocity, u_t . If the system of particles under investigation was not mono-size but contained a distribution of different sizes, which particle diameter should you use to calculate u_{mf} for the polydisperse system and which particle diameter should you use to calculate u_t so that elutriation is reduced.

[4]

4. Consider a fluidized bed of solids operated at constant temperature T_s. Hot (i) fluidizing gas at temperature T_{gi} enters the bed, with $T_{gi} > T_s$. The expression for the distance "L_n" at which the gas-to-particle temperature difference falls by a factor "n" from its initial value is shown in Eq. (1):

$$L_n = -5.5 \left[\frac{\ln n \,\mu^{1.3} \,\mathrm{d}_p^{0.7} \,\mathrm{C}_g}{\mathrm{u}_{\rm rel}^{0.3} \,\rho_g^{0.3} \,(1 - \varepsilon) \,\mathrm{k}_g} \right] \quad (1)$$

Write the heat balance equation across an element of fluid-bed of height dL and show the derivation of Eq.(1). [14]

To this end, use the following information:

Assume the following boundary conditions: $T_g = T_{gi}$ at L=0; and denote with T_{ge} the temperature of the gas leaving the fluid-bed element, where $T_{ge} < T_{gi}$.

The surface area of solids per unit volume of bed S (m^{-1}) is expressed as:

$$S = \frac{6(1-\varepsilon)}{d_p}$$

The gas-to-particle heat transfer coefficient hgp (W m⁻² K⁻¹) is expressed as:

$$h_{gp} = \frac{0.03 \, d_p^{0.3} \, \rho_g^{1.3} \, u_{rel}^{1.3} \, k_g}{\mu^{1.3}}$$

 d_p = particle diameter (µm)

$$\rho_{\rm g}$$
 = gas density (kg m⁻³)

 $\rho_{g} = gas density (kg m)$ $\mu = gas viscosity (N s m⁻²)$ $k_{g} = gas conductivity (W m⁻¹ K⁻¹)$

$$C_g$$
 = specific heat capacity of the gas (J kg⁻¹ K⁻¹)

$$u_{rel}$$
 = gas-particle relative velocity (m s⁻¹)

 ε = fluid-bed voidage

 ΔT_g = change in temperature of the gas flowing through the element of bed (K)

dL = element of bed dL deep and of unit cross section area (m)

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(ii) Using the equation derived for " L_n " (Eq.1) and the data given below, calculate the distance penetrated by the gas into a bed of constant temperature before the difference between its temperature and that of the bed solids is reduced by half. [6]

 $\begin{array}{rcl} d_{p} &=& 300 \ \mu m \\ \rho_{g} &=& 1.2 \ kg \ m^{-3} \\ \mu &=& 1.8 \times 10^{-5} \ N \ s \ m^{-2} \\ k_{g} &=& 0.0262 \ W \ m^{-1} \ K^{-1} \\ C_{g} &=& 1005 \ J \ kg^{-1} \ K^{-1} \\ \epsilon &=& 0.45 \\ u_{mf} &=& minimum \ fluidization \ velocity = 0.10 \ m \ s^{-1} \end{array}$

Section C

5.

- Sketch the product CSD (crystal size distribution) from a continuous MSMPR (mixed-suspension, mixed-product-removal) crystallizer in terms of *population density* and *mass fraction*. [2]
- (ii) State the *population balance* concept and show that the CSD from a continuous MSMPR crystallizer at steady-state may be expressed by:

$$n(L) = n^{\circ} \exp(-L/G\tau)$$

where L is the crystal size, G is the overall linear crystal growth rate and τ is the mean residence time in the vessel. State clearly any assumptions that you may make. [6]

(iii) The suspension density M_T in a continuous MSMPR crystallizer operated at steady-state is given by the equation:

$$M_T = 6f_v \rho_c n^{\circ} (G\tau)^4$$

where f_v and ρ_c are the crystal volume shape factor and density respectively, n^O is the nuclei population density, G is the linear crystal growth rate and τ is the mean residence time within the crystallizer.

The CSD from an MSMPR crystallizer with a working volume of 1 m^3 operated with a magma density of 100kg crystals/m³ slurry and a production rate of 1000kg crystals/hr has a Sauter mean size of 100 μ m. Calculate:

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- a) the linear crystal growth rate (m/s)
- b) the population density of nuclei $(\#/m^4)$ and,
- c) the nucleation rate $(\#/sm^3)$.

[Take the crystal volume shape factor to be 1.0 and the crystal density to be 2000 kg/m^3] [12]

- 6.
- (i) Define the moments about the origin of the CSD (crystal size distribution) from a continuous MSMPR (mixed-suspension, mixed-product-removal) crystallizer and explain their physical significance.
 [8]
- (ii) An MSMPR crystallizer is operated continuously at steady state. The suspension density M_T is given by:

$$M_{\tau} = 6f_{\nu}\rho_{c}n^{o}(G\tau)^{4}$$

where f_{ν} and ρ_c are the volume shape factor and crystal density respectively, n° is the nuclei population density, B° is the nucleation rate, G is the linear crystal growth rate and τ is the mean residence time within the crystallizer.

The crystal growth rate and nucleation rate B° are found to be related by:

$$B^o \propto M_{\tau} G^2$$

Estimate the effect of decreasing the throughput of the crystallizer by 50% on:

- a) the crystal growth rate (m/s)
- b) the nucleation rate $(\#/sm^3)$, and
- c) the dominant crystal size (μm)

You may assume that the suspension density is controlled to a constant value. [12]

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