

Then, for the proton,

$$(a_p)_x = e\mathcal{E}_x/m_p$$

Using

$$(s_p)_x = (u_p)_x t + \frac{1}{2}(a_p)_x t^2,$$

the proton's displacement when it passes the electron is

$$d = \frac{1}{2}(a_p)_x t^2, \quad \text{since } (u_p)_x = 0.$$

For the electron,

$$(a_e)_x = -e\mathcal{E}_x/m_e,$$

and its displacement as it passes the proton is

$$d - S = \frac{1}{2}(a_e)_x t^2.$$

Then

$$\frac{d - S}{d} = \frac{\frac{1}{2}t^2(-e\mathcal{E}_x/m_e)}{\frac{1}{2}t^2(e\mathcal{E}_x/m_p)} = -\frac{m_p}{m_e}. \quad (i)$$

Therefore

$$1 - \frac{S}{d} = -\frac{1.67 \times 10^{-27} \text{ kg}}{9.11 \times 10^{-31} \text{ kg}} = -1.83 \times 10^3.$$

Hence

$$d = \frac{3.7 \times 10^{-2} \text{ m}}{1.83 \times 10^3} = 2 \times 10^{-5} \text{ m},$$

i.e. the particles cross just $20 \mu\text{m}$ from the positive plate.

Marks for working:

10

Checking

Equation (i) shows that if the two particles were of equal mass (and magnitude of charge), they would cross at $d = S/2$, exactly halfway across the gap, which is what one would expect.

Marks for checking:

1

Total for Q19:

16

Question 20

Preparation

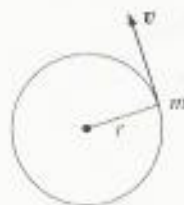


Figure 6

Relevant equations and principles:

$$L = mvr = n\hbar \quad \text{with } \hbar = \frac{h}{2\pi}$$

$$E_{\text{trans}} = \frac{1}{2}mv^2 \quad \text{and} \quad E_{\text{pot}} = -\frac{e^2}{4\pi\epsilon_0 r_n} \quad 2$$

Marks for preparation:

3

Working

In an orbit with quantum number n , the speed of the electron is

$$v = \frac{n\hbar}{m_e r_n} = \frac{n\hbar}{m_e n^2 a_0} = \frac{\hbar}{m_e n a_0} \quad 2$$

The kinetic energy of the electron is

$$E_{\text{trans}} = \frac{1}{2}m_e v^2 = \frac{1}{2}m_e \left(\frac{\hbar}{m_e n a_0} \right)^2 = \frac{\hbar^2}{8\pi^2 m_e n^2 a_0^2} \quad 2$$

The potential energy of the electron is

$$E_{\text{pot}} = -\frac{e^2}{4\pi\epsilon_0 n^2 a_0} \quad 2$$