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MARK SCHEME for the May/June 2013 series

5054 PHYSICS

5054/22

GCE Ordinary Level

Paper 2 (Theory), maximum raw mark 75

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge will not enter into discussions about these mark schemes.

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Page 2	Mark Scheme	Syllabus	Paper
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Section A

1	(a)	a) travels further in each second / in same time / between images			
	(b)) <i>d/t</i> in any form algebraic or numerical cm/s; 0.4(0) m/s	C1 A1	
	(c)		resistance increases ght constant	B1 B1	
	(d)	or I	ces balance /cancel no resultant/net force resultant of any two forces equal and opposite to third	B1	[6]
2	(a)		ce × distance pendicular distance	M1 A1	
	(b)	(i)	<i>T</i> × 8 or 2000 × 2 seen 500 N	C1 A1	
		(ii)	(two forces) equal (in magnitude) (two forces) opposite (in direction)	B1 B1	[6]
3	(a)	(i)	(<i>W</i> =) <i>Fd</i> or 90×0.3 or 90×30	C1	
			27 J	A1	
		(ii)	(P=) W/t or Fd/t or 27(× 20)/60 or 27/3	C1	
			9(.0) W	A1	
	(b)	(i)	800 × 30/180 or 800/6 or 6 seen or proportionality clearly used	C1	
			133 or 130 cm	A1	
		(ii)	extension more than 143 cm or (extra) extension > 10 cm or (some) extension permanent	B1	[7]

	Page 3		3	Mark Scheme	Syllabus	Paper	
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4	(a)	gar	nma r	rays, visible light, infra-red		B1	
	(b)	(i)	micr	owaves		B1	
		(ii)	sate	llite (receives and) sends/transmits/emits/boosts/am	plifies signal	B1	
		(iii)	or u	er a large area over the horizon / only one (transmitten naffected by tall buildings/hills o obstructions	er/station) needed	etc. B1	[4]
5	(a)	ele	ctrons	s move onto polythene / rod		B1	
		ele	ctrons	s/negative charge move off cloth		B1	
	(b)	(re	gion o	of space) where force is exerted on a charge		В1	
	(c)	(i)		ce charges attract rod) attracts +ve charge/ions/particles		B1	
				els like charge rod) repels –ve charge/ions/electrons/particles		B1	
		(ii)	(net)) positive charge on water near rod		B1	[6]
6	(a)	(i)		2.1 (V) to any value between 11 and 12 (V) bove 2/2.1(V)		B1	
		(ii)	temp	perature increases / gets hotter		B1	
	(b)	(i)	(rate	e of) flow of charge/electrons		B1	
		(ii)	0.35	A cao		B1	
		(iii)	or 6	V/R algebraic /20 /0.35		C1	
			or $1/2$	0) (A) $/R_T = 1/20 + 1/17.1$ $R_T = 9.2 (\Omega)$ seen		C1	
			0.65	A		A1	[7]

	Page 4	Mark Scheme Sy	llabus Paper	•
	<u> </u>	GCE O LEVEL – May/June 2013	5054 22	
7	(a) oval/ci clockw	B1 B1		
	(b) fields	(due to X and Y) cancel or X and Y fields equal and opposite	B1	
	` ' ` '	the left r towards X/A/B	B1	
		urrent (in wire Y) and (magnetic) field (caused by other wire) r two (magnetic) fields interact	B1	[5]
8	EITHER			
	(a) steel /	magnadur / alnico / magnetite	B1	
	(b) (i) m	nention of cutting (lines of) magnetic field / change in (magnet	tic) flux M1	
	OI	reat(est) rate of change r fast(est) cutting r other explanation involving time	A1	
	01	ertical/upright r turned through 90° r normal to (magnetic) field	В1	
	OR			
	(a) NOT (or inve	gate) erter	В1	
	(b) 1,0		B1	
	(c) (i) (v	voltage across R₁) <u>becomes</u> 0/low	B1	
	(ii) de	ecrease any of R _{1,} R ₂ , C _{1,} C ₂	B1	[4]
			[Total	l: 45]

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Section B

9	(a)	(air) molecules hit walls / liquid (surface)	B1	
	(air) molecules move fast(er) /great(er) kinetic energy) molecules move fast(er) /great(er) kinetic energy	B1	
) molecules hit <u>more</u> often/ <u>more</u> frequently/ <u>greater</u> rate / hard <u>er</u> / <u>more</u> force liquid) molecules evaporate	B1	[3]
	(b)	(i)	(flask) <u>in</u> (pure) <u>melting</u> ice (and water)	B1	
			(flask) in (pure) boiling water / above boiling water (at one atmosphere)	B1	
		(ii)	thin(ner) tube or large(r) flask or more air/less liquid or use liquid that expands more (1 mark for each)	B2	
			er dee liquid that expande more (1 mark for each)		
		(iii)	divisions not equally spaced or scale not uniform/not proportional	C1	
			different distance (along scale) for same temperature rise or different change in temperature for same distance (along scale)	A1	[6]
	(c)	(i)	(<i>M</i> =) $d \times V$ in any form or $1200 \times 5 \times 10^{-5} \times 0.15$	C1	
			$9(.0) \times 10^{-3} \text{ kg}; 0.009(0) \text{ kg}$	A1	
		(ii)	0.09(0) N ecf (i)	B1	
		(iii)	(P=) hdg in any form or (P=) F/A in any form	C1	
			1800 Pa	A1	[5]
	(d)		ids expand less (than air) great(er) forces between liquid molecules	B1	[1]
				[Total:	15]
10	(a)	cor	rect normal by eye rect angle of incidence between candidate's normal and incident ray rect angle of refraction marked between candidate's normal and BC	B1 B1 B1	[3]
	(b)	dec	rease / change in speed / wavelength	B1	[1]

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(c) $n = \sin i/s$		= sin <i>i</i> /s	sin <i>r</i> seen in any form		C1	
	•	,	sin 45°/1.5 (4) seen		C1	
	28(.1)°				C1	[3]
((d) ref	fracts l	ess at first face and on correct side of normal		B1	
		fractior erging	n at second face away from normal so that red ray a	and blue ray are	B1	[2]
((e) (i)	-	e of incidence is 0 ay along normal/perpendicular to glass		B1	
	(ii)		e of incidence/ θ is larger than critical angle internal reflection occurs		B1 B1	
	(iii)	refle	cted ray drawn correctly and emerging without refra	action from block	B1	
	(iv)	or lig	ntually) light emerges (into air at Q) ght refracts (out at Q) veak) refracted ray appears		B1	
		_	emerging at Q coloured in some way prrect description of movement of reflected ray (as	heta decreases)	B1	[6]
					[Total:	15]
11 (apply with ammeter and heater in series r in parallel with heater/ power supply		B1 B1	[2]
((b) (i)		VI in any form 2×12		C1	
		50(.4	4) W		A1	
	(ii)	(<i>E</i> =)	Pt i.e. any power \times any time e.g. $50(.4) \times 8$		C1	
			.13(3) seen vision by 1000 seen anywhere		C1	
		0.00	67(2) (kWh)		A1	[5]

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(c) (i) molecules escape (from surface/leave water) / become gas or vapour / break bonds

C1

fast(er) moving / high energy/ energetic molecules escape

Α1

(ii)

change M1	explanation A1
wind / draught / breeze	wind knocks molecules away
or larger surface area	more chance/possibility of escape/more space to escape or more molecules come to/near/at surface
or decrease humidity / drier air	fewer molecules return/from air
or decrease atmospheric pressure	fewer air molecules to hit during escape

(iii) evaporation occurs at surface and boiling inside liquid/bubbles evaporation occurs at any temperature (accept room temperature) and boiling occurs at boiling point/100°C/ fixed / specific temperature evaporation increased by draughts/higher temp/more area and boiling is not OR increase in pressure stops boiling but only reduces evaporation any two

B2 [6]

(d) water heats air (by conduction)

В1

or water loses heat/energy (to cup or air)
or air gains heat/energy (from water)

hot / heated air / particles rise

B1 [2]

- or cold air / particles sink
- or hot air is less dense
- or cold air is more dense

[Total: 15]