MARK SCHEME for the October/November 2014 series

5129 COMBINED SCIENCE

5129/22

Paper 2 (Theory), maximum raw mark 100

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge will not enter into discussions about these mark schemes.

Cambridge is publishing the mark schemes for the October/November 2014 series for most Cambridge IGCSE[®], Cambridge International A and AS Level components and some Cambridge O Level components.

® IGCSE is the registered trademark of Cambridge International Examinations.



Pa	ige 2	2	Mark Scheme	Syllabus	Paper
			Cambridge O Level – October/November 2014	5129	22
1	Ene Aer Wa Car Lac Mus	ergy obi iter rboi ctic scle	<pre>/; c; in dioxide; } either order (accept correct formulae) acid; ;; ;; ;; ;; ;; ;; ;; ;; ;; ;; ;; ;; ;</pre>		[6]
2	(a)	Fo St	our points correctly drawn ($\frac{1}{2}$ square tolerance) ; raight line through the points ;		[2]
	(b)	(i)	240 ; (ecf from graph)		
		(ii)	24000 ; (ecf from (b)(i))		[2]
	(c)	Lii Ge	newater ; (incorrect test = 0 mark) bes milky/cloudy/white precipitate ;		[2]
3	(a)	(i)	Not a straight line / line is a curve ;		[1]
		(ii)	0;		[1]
		(iii)	Time 0.4 s and $y = 2.0 \text{ m/s}^{\circ}$		
		(,	$s = v \times t \text{ or } 0.4 \times 2.0 ;$ 0.8 ;		[3]
	(b)	(i)	<i>F</i> = <i>m</i> x <i>a</i> or 0.03 × 8 ; 0.24 ;		[2]
		(ii)	$D = m/v \text{ or } 0.030/8 \times 10^{-6};$		
			kg/m ³ ;		[3]
			(Accept a numerical answer with consistent unit -3.75 g/cm ³ = 3 n	narks)	
4	(a)	Uı Gl	ea/ <u>other</u> amino acids ; ycogen/fats ;		[2]
	(b)	Al Ho	cohol/ethanol ; ormones/a named hormone ;		
		Dı Ha	ugs/any named drug ; any 2 any 2		
		Тс	oxic/poisonous chemicals ;		[2]

Ρ	age :	3	Mark Scheme	Syllabus	Paper
			Cambridge O Level – October/November 2014	5129	22
5	(a)	Ha	logens ;		[1]
	(b)	Sc Bla	lid ; ack ;		[2]
	(c)	7 e	electrons in <u>outer</u> / <u>valency</u> shell ;		[1]
	(d)	All	zenes ;		[1]
6	(a)	Le Ex	ngth = 14.4 ; tension. 1.2, 2.4, 3.6, 4.8 ;		[2]
	(b)	Sp We	ring force = 0.30 ; eight = 0.96 ;		[2]
7	(a)	(i)	Osmosis (ignore diffusion) ;		[1]
		(ii)	Sugar solution more concentrated than cell contents <u>Or</u> water concentration in cell is more than in sugar concentration ; Water (molecules) move out of the cells ; Through the partially permeable membrane ; To make concentrations equal ;	} any 3	[3]
		(iii)	Concentration of cell contents equal to concentration of sugar solut <u>Or</u> no further loss of water (from the tissue/sample) <u>Or</u> no further osmosis ;	ion	[1]
	(b)	27	(cm ³) ;		[1]
	(c)	(i)	Has same volume (of cells as first piece) ; (ignore same mass)		[1]
		(ii)	Has smaller surface area (than first piece) ; A calculation of the surface areas (78 cm ³ and 54 cm ³) ; } any 1		[1]
8	(a)	10	14 ; 2,8,1 ;		
		١ð	, 20 ; 2,8,8 ;		[5]
	(b)	(i)	Na ₂ S ;		[1]
		(ii)	lonic ;		[1]

Page 4		4	Mark Scheme		Paper
			Cambridge O Level – October/November 2014	5129	22
9	(a)	X = Y = Z = (ac	e potassium hydroxide ; sulfuric acid ; ammonia ; cept correct formulae for all)		[3]
	(b)	Ne	utralisation/exothermic;		[1]
	(c)	Fue	el (for rockets)/ making margarine/ammonia/Haber process		[1]
10	(a)	(i)	Alpha/α;		
		(ii)	Gamma/γ;		[2]
	(b)	24	100 ;		[2]
11	(a)	Noi	rmal correct ;		[1]
	(b)	Firs Seo	st reflection vertical ; cond reflection horizontal ;		[2]
12	(a)	Ext (tha	ernally administered substance ; at) modifies / affects chemical reactions (in the body) ;		[2]
	(b)	(Pc Ado Wit Vei	werful) depressant ; diction / dependency ; hdrawal symptoms ; n collapse ;		[3]
13	(a)	(i)	Fractional distillation ;		[1]
		(ii)	Bitumen ; Kerosene / paraffin ;		[2]
	(b)	(i)	Alkanes ;		[1]
		(ii)	13 12 14 ;		[1]

Pa	age 5	5	Mark Scheme	Syllabus	Paper
			Cambridge O Level – October/November 2014	5129	22
14	(a)	Co In	prrect symbol on the circuit ; parallel with variable resistor ;		[2]
	(b)	I = 0.3	<i>V/R</i> or 0.6/2 ; 3 ;		[2]
	(c)	(i)	Decreases/less;		[1]
		(ii)	Decreases/less;		[1]
	((iii)	Increases/more;		[1]
15	(a)	(i)	A – Aorta ; B – Pulmonary vein ; C – Semi-lunar valve / aortic valve ;		[3]
		(ii)	<pre>(D) Pumps/contracts more strongly/with more force ; Raises/higher blood pressure ; Propels blood greater distance/further ; (Than E) (Or stated converse)</pre>	ny 1	[1]
	(b)	(i)	Platelet/thrombocyte ;		[1]
		(ii)	<pre>(Platelet) produces thrombin ; (Soluble) fibrinogen ; Converted to (insoluble) fibrin ; Form mesh across wound/seals wound ;</pre>		[2]
	((iii)	Take more exercise ; Reduce stress levels ; Do not smoke (cigarettes) ; Eat less meat/food rich in fat/less fat ; Eat more fruit/vegetables/fibre ;		
			Reduce weight ; Reduce (blood) cholesterol level ; Note: the question asks for <u>a change</u> in lifestyle.		[3]
16	(a)	0	ygen ;		[1]
	(b)	Ni	rogen ;		[1]
	(c)	He	lium ;		[1]
	(d)	Ca	irbon monoxide ;		[1]
	(e)	Sı	lfur dioxide/carbon dioxide ;		[1]

Page 6		Mark Scheme	Syllabus	Paper
		Cambridge O Level – October/November 2014	5129	22
17	(a) I	Higher/increase/ <u>more</u> positive/more to the right ;		[1]
	(b) 	Reverse polarity (of magnet)/turn magnet round ; Reverse motion (of magnet)/pull magnet out ; Reverse connection to ammeter ;		[2]
18	(a) I	Best/good absorber/absorbs heat faster ;		[1]
	(b)	i) Increases/more ;		[1]
	(i) Decreases/less;		[1]