

# UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS General Certificate of Education Ordinary Level

CHEMISTRY 5070/03

Paper 3 Practical Test May/June 2008

CONFIDENTIAL INSTRUCTIONS

Great care should be taken to ensure that any confidential information given does not reach the candidates either directly or indirectly



The Supervisor's attention is drawn to the form on page 7 which must be completed and returned with the scripts.

If you have any problems or queries regarding these Instructions, please contact CIE

by e-mail: International@cie.org.uk,

by phone: +44 1223 553554, by fax: +44 1223 553558,

stating the nature of the query and the syllabus number quoted above.

UNIVERSITY of CAMBRIDGE International Examinations

#### Safety

Supervisors are advised to remind candidates that **all** substances in the examination should be treated with caution. Only those tests described in the question paper should be attempted. Please also see under 'Apparatus' on the use of pipette fillers and safety goggles.

In accordance with COSHH (Control of Substances Hazardous to Health) Regulations, operative in the UK, a hazard appraisal of the examination has been carried out.

Attention is drawn, in particular, to certain materials used in the examination. The following codes are used where relevant.

C = corrosive substance F = highly flammable substance

**H** = harmful or irritating substance **O** = oxidising substance

T = toxic substance N = dangerous for the environment

The attention of Supervisors is drawn to any local regulations relating to safety, first-aid and disposal of chemicals.

'Hazard Data Sheets', relating to materials used in this examination, should be available from your chemical supplier.

# **Preparing the Examination**

1 Access to the question paper is NOT permitted in advance of the examination.

#### 2 Preparation of materials

Where quantities are specified for each candidate, they are sufficient for the experiments described in the question paper to be completed.

In preparing materials, the bulk quantity for each substance should be increased by 25% as spare material should be available to cover accidental loss. More material may be supplied if requested by candidates, without penalty.

All solutions should be bulked and mixed thoroughly before use to ensure uniformity.

Every effort should be made to keep the concentrations accurate to within one part in 50 of those specified.

Supervisors are asked to carry out any confirmatory tests given on page 4 to ensure the materials supplied are appropriate.

# 3 Labelling of materials

Materials must be labelled as specified in these instructions. Materials with a letter code (e.g. **P**, **Q**) should be so labelled, **without** the identities being included on the label – where appropriate, the identity of a letter-coded chemical is given in the question paper itself.

#### 4 Identity of materials

It should also be noted that descriptions of solutions given in the question paper may not correspond exactly with the specifications in these Instructions. **The candidates must assume the descriptions given in the question paper.** 

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# 5 Size of group

In view of the difficulty of the preparation of large quantities of solution of uniform concentration, it is recommended that the maximum number of candidates per group be 30 and that separate supplies of solutions be prepared for each group.

### **Apparatus**

- 1 In addition to the fittings ordinarily contained in a chemical laboratory, the apparatus and materials specified below will be necessary.
- 2 Pipette fillers (or equivalent safety devices) and safety goggles should be used where necessary.
- 3 For each candidate
  - $1 \times 50 \text{ cm}^3 \text{ burette}$
  - 1 × burette clamp
  - $1 \times stand$
  - 1 × funnel for filling burette
  - 1 × white tile
  - $1 \times 20 \text{ cm}^3 \text{ or } 25 \text{ cm}^3 \text{ pipette}$

(It is essential that all candidates at a Centre have pipettes of the same capacity.)

- 1 × pipette filler (or equivalent safety device)
- 1 × flask or other suitable vessel for titration
- a supply of test-tubes
- 1 × test-tube rack
- 1 × stirring rod

# **Chemicals Required**

It is especially important that great care is taken that the confidential information given below does not reach the candidates either directly or indirectly.

Particular requirements

ard	hazard label	per candidate	identity	notes
	Д	150 cm <sup>3</sup>	0.05 mol/dm <sup>3</sup> sulphuric acid	
[H]	Ø	150 cm <sup>3</sup>	0.10 mol/dm³ sodium hydroxide	$4.0\mathrm{g/dm^3}$ sodium hydroxide <b>[C]</b>
			methyl orange or screened methyl orange indicator	

Supervisors are asked to carry out a standard acid/base titration between solutions  $\mathbf{P}$  and  $\mathbf{Q}$  to ensure that the concentrations of the two solutions fall within the given range. It is **essential** that 25.0 cm<sup>3</sup> of  $\mathbf{Q}$  reacts with between 23.0 cm<sup>3</sup> and 27.0 cm<sup>3</sup> of  $\mathbf{P}$  (or 20.0 cm<sup>3</sup> of  $\mathbf{Q}$ ).

[H] S 30 cm <sup>3</sup> approximately 0.2 mol/dm <sup>3</sup> 50g of hydrated iron [III) chloride containing 10 g/dm <sup>3</sup> [H] [N] T 30 cm <sup>3</sup> approximately 0.2 mol/dm <sup>3</sup> 50g of hydrated cop copper(II) sulphate A small volume of dilickel(II) sulphate A small volume of	50 g of hydrated iron(III) chloride, ${\rm FeC}_{l_3}$ .6 ${\rm H_2O}$ [H], in 1 dm $^3$ aqueous sodium chloride containing 10 g/dm $^3$ NaC $l$	50g of hydrated copper(II) sulphate, CuSO <sub>4</sub> .5H <sub>2</sub> O <b>[H] [N]</b> , dissolved in 1 dm <sup>3</sup> of distilled water	A small volume of dilute sulphuric acid should be added to the solution to prevent hydrolysis. (10 cm <sup>3</sup> of 0.5 mol/dm <sup>3</sup> sulphuric acid per 1.0 dm <sup>3</sup> of solution should be sufficient.)	100g of hydrated nickel(II) sulphate, NiSO <sub>4</sub> .7H <sub>2</sub> O <b>[H]</b> , dissolved in 1 dm <sup>3</sup> of distilled water A small volume of dilute sulphuric acid should be added to the solution to prevent	hydrolysis. (10 cm <sup>3</sup> of 0.5 mol/dm <sup>3</sup> sulphuric acid <b>[H]</b> per 1.0 dm <sup>3</sup> of solution should be sufficient.)
[H] S 30cm <sup>3</sup> (H) H (N) T 30cm <sup>3</sup> (H)	50g of hydrated iron(III) cl containing 10g/dm <sup>3</sup> NaCl	50g of hydrated cop water	A small volume of dill (10cm³ of 0.5mol/dn	100g of hydrated nic A small volume of dli	hydrolysis. (10cm³ o sufficient.)
S F Ε	approximately 0.2 mol/dm³ iron(III) chloride	approximately 0.2 mol/dm <sup>3</sup> copper(II) sulphate		approximately 0.35 mol/dm <sup>3</sup> nickel(II) sulphate	
E E E	30 cm <sup>3</sup>	30 cm <sup>3</sup>		30 cm <sub>3</sub>	
	S	<b>-</b>		<b>-</b>	
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The standard bench reagents specifically required are set out below. If necessary, they may be made available from a communal supply: however, the attention of the Invigilators should be drawn to the fact that such an arrangement may enhance the opportunity for malpractice between candidates. က

hazard	label	identity	notes
Z E	[T] [N] aqueous lead(II) nitrate	0.2 mol/dm <sup>3</sup> lead(II) nitrate	66 g/dm³ lead(II) nitrate [T] [N]
[C] [N]	[C] [N] aqueous silver nitrate	0.05 mol/dm³ silver nitrate	
Ε	aqueous barium nitrate	0.2 mol/dm³ barium nitrate	0.2 mol/dm³ barium chloride [T] (labelled barium nitrate) may be used as an alternative.
<u></u>	aqueous sodium hydroxide	1.0 mol/dm <sup>3</sup> sodium hydroxide	
[H]	aqueous ammonia	1.0 mol/dm³ ammonia	
	aqueous potassium iodide	$0.2\mathrm{mol/dm^3}$ potassium iodide	

The reagents, materials and apparatus to test the gases listed in the syllabus must be available to candidates. If necessary, they may be made available from a communal supply: however, the attention of the Invigilators should be drawn to the fact that such an arrangement may enhance the opportunity for malpractice between candidates.

hazard	label	identity	notes
<u></u>	limewater	saturated aqueous calcium hydroxide, Ca(OH) <sub>2</sub>	Prepare fresh limewater by leaving distilled water to stand over solid calcium hydroxide [C] for several days, shaking occasionally. Decant or filter the solution.
N E	aqueous potassium dichromate(VI)	0.10mol/dm³ K <sub>2</sub> Cr <sub>2</sub> O <sub>7</sub>	Dissolve 29.5g of K <sub>2</sub> Cr <sub>2</sub> O <sub>7</sub> <b>[T] [N]</b> in each dm <sup>3</sup> of solution which should contain about 10% of dilute (1.0 mol/dm <sup>3</sup> ) sulphuric acid. The use of plastic gloves may be considered to prevent contact with skin.
		red and blue litmus paper or universal indicator paper	
		plain filter paper strips for use with aqueous potassium dichromate $\left( VI\right)$	
		wooden splints	
		the apparatus normally used in the Centre for use with limewater in testing for carbon dioxide	

#### **During the Examination**

1 The Supervisor, or other competent chemist must carry out the experiments in question 1 and question 2 and record the results on a spare copy of the question paper which should be labelled 'Supervisor's Results'.

This should be done for:

each session held and each laboratory used in that session, and each set of solutions supplied.

It is essential that each packet of scripts contains a copy of the applicable Supervisor's Results as the candidates' work cannot be assessed accurately without such information.

2 The Supervisor must complete the Report Form on page 7 to show which candidates attended each session. If all candidates took the examination in one session, please indicate this on the Report Form. A copy of the Report Form must accompany each copy of the Supervisor's Results in order for the candidates' work to be assessed accurately.

The Supervisor must give details on page 8 of any particular difficulties experienced by a candidate, especially if the Examiner would be unable to discover this from the written answers.

#### After the Examination

Each envelope returned to Cambridge must contain the following items.

- 1 The scripts of those candidates specified on the bar code label provided.
- 2 A copy of the Supervisor's Report relevant to the candidates in 1.
- **3** A copy of the Report Form, including details of any difficulties experienced by candidates (see pages 7 and 8).
- 4 The Attendance Register.
- 5 A Seating Plan for each session/laboratory.

Failure to provide appropriate documentation in each envelope may cause candidates to be penalised.

#### Colour Blindness

With regard to colour-blindness – a minor handicap, relatively common in males – it is permissible to advise candidates who request assistance on colours of, for example precipitates and solutions (especially titration end-points). Please include with the scripts a note of the candidate numbers of such candidates.

Experience suggests that candidates who are red/green colour-blind – the most common form – do not generally have significant difficulty. Reporting such cases with the scripts removes the need for a 'Special Consideration' application for this handicap.

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# **REPORT FORM**

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С	Centre Number	
1	1 Supervisor's Results	
	Supervisors are asked to use a spare copy of the question paper to report their <b>Q.2</b> and enclose this copy of the question paper with the candidate's answer question paper should be clearly labelled 'Supervisor's Results'. Failure to enand this report form may lead to candidates being unavoidably penalised.	s. This copy of the
	If candidates from more than one Centre are taking the examination, it is esse the 'Supervisor's Results' should be sent with the scripts from <b>each Centre</b> .	ential that a copy of

**2** The candidate numbers of candidates attending each session were:

First session	Second session

3		Supervisor is invited to report details of any difficulties experienced by particular candidates, ng names and candidate numbers. This report should include reference to:
	(a)	any general difficulties encountered in making preparation;
	(b)	difficulties due to faulty apparatus or materials;
	(c)	accidents to apparatus or materials;
	(d)	assistance with respect to colour-blindness.
		er cases of hardship, e.g. illness, temporary disability, should be reported direct to CIE on the mal 'Application for Special Consideration' form.
4	-	lan of work benches, giving details by candidate numbers of the places occupied by the ididates for each experiment for each session, must be enclosed with the scripts.
	oui	raidates for each experiment for each second, must be enclosed with the scripts.
NA	ME C	OF CENTRE
		SIGNED
		Supervisor
CE	NTR	E NUMBER
		ndidates' Centre number is different from the number of the Centre at which the examination en, the Supervisor should write <b>both Centre numbers in the space provided.</b>
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