

Junior Certificate Examination 2003

Science

Higher Level

Marking Scheme

GUIDELINES TO EXAMINERS ON CANCELLED, REPEATED OR EXCESS ANSWERS

CANCELLED ANSWERS

SECTION A If an answer is cancelled and a second answer given you should accept the cancellation and award marks for the uncancelled answer. If neither is cancelled then give zero except in the case where both answers are correct.

SECTION B, C, D and E If candidates answer a question or part of a question only once and then cancel, you should ignore the cancelling and mark in the usual way. It candidates answer a question or part of a question more than once and then cancel one attempt, you should ignore the cancelling and mark all the answers whether cancelled or not, however count only the marks gained in respect to the highest scoring answer. The disallowed marks should be enclosed in square brackets.

REPEATED ANSWERS

SECTIONS B, C, D AND E If candidates repeat an answer (answer the same question twice) you should mark both answers and allow marks for the highest scoring answer. The disallowed marks should be enclosed in square brackets.

EXCESS ANSWERS

<u>SECTION A</u> Mark all parts but count only the marks for the eight highest scoring parts. Disallowed marks should be enclosed in square brackets.

SECTION B, C AND D Mark all questions but count only the marks awarded to the highest scoring question in each section. Disallowed marks should be enclosed in square brackets.

SECTION E Mark all questions but count only the marks awarded to the two highest scoring questions. Disallowed marks should be enclosed in square brackets. Extra care should be taken with Q.10 (Earth Science), Q.11 (Horticulture) and Q. 13 (Food): count only the marks awarded to the two highest scoring parts (a), (b) or (c). Care should also be taken with options in Q.12 (Materials Science).

DEDUCTION OF MARKS FOR OMITTED DIAGRAM

Assign marks in the usual way. Then use square brackets to deduct the marks.

Junior Certificate Examination

SCIENCE

Higher Level Paper

Structure

Five sections A, B, C, D, E.

Section A.

Section A:		3 question (attempt all questions)
		10 parts in each question (attempt any 8 parts)
Section B:	Physics	2 questions (attempt any 1 question)
Section C:	Chemistry	2 questions (attempt any 1 question)
Section D:	Biology	2 questions (attempt any 1 question)
Section E:	Applied Sc.	6 questions (attempt any 2 questions)

Marking

Without Local Studies:	$(6 \times 48) + (2 \times 36) =$	288 + 72 = 360 marks
With Local Studies:	(6×48)	= 288 marks

Grades

Grade	Ma	irks
	Without LS	With LS
Α	306 - 360	245 - 288
В	252 - 305	202 - 244
С	198 - 251	158 - 201
D	144 - 197	115 - 157
Ε	90 - 143	72 - 114
F	36 - 89	28 - 71
NG	0 - 35	0 - 27

Junior Certificate 2003

Science – Higher level

Marking Scheme

Section A	Q.1 Q.2 Q.3		$ \begin{array}{l} 8 \times 6 \\ 8 \times 6 \\ 8 \times 6 \end{array} $
Section B	Q.4 Q.5	(a) (b) (a)	$2 \times 3, 1 \times 3, 2 \times 3, 3 \times 3,$ $1 \times 3, 4 \times 3, 1 \times 3, 2 \times 3$ $1 \times 3, 2 \times 3, 1 \times 3, 2 \times 3$ $1 \times 3, 2 \times 3, 1 \times 3, 1 \times 3, 2 \times 3, 1 \times 3,$
Section C	Q.6 Q.7	 (b) (a) (b) (a) (b) 	$1 \times 3, 1 \times 3, 1 \times 3, 1 \times 3, 2 \times 3, 1 \times 3, 2 \times 3, 1 \times 3, 2 \times 3, 2 \times 3, 2 \times 3, 2 \times 3, 1 \times 3, 1 \times 3, 6 + 3, 1 \times 3, 3 \times 3, 2 \times$
~		(b) (c)	$2 \times 3, 2 \times 3, 6 + 3, 1 \times 3,$
Section D	Q.8 Q.9	(a) (b) (a) (b)	$4 \times 3, 4 \times 3, 2 \times 3, 1 \times 3, 2 \times 3, 1 \times 3, 2 \times 3, 3 \times 3, 2 \times 3, 3 \times 3, 1 \times 3, 3 \times 3, 2 \times 3, 2 \times 3, $
Section E	Q.10	(a) (b) (c)	$2 \times 3, 4 \times 3,$ $1 \times 3, 3 \times 3, 2 \times 3,$ $2 \times 3, 2 \times 3, 1 \times 6,$ Any two parts
	Q.11	(a) (b) (c)	$5 \times 3, 1 \times 3,$ $2 \times 3, 4 \times 3,$ $2 \times 3, 4 \times 3,$ Any two parts
	Q.12	(a) (b)	$2 \times 3, 2 \times 3, 2 \times 3,$ $2 \times 3, 4 \times 3,$ Any one of four (i) - (iv)
	Q.13	(a) (b) (c)	$3 \times 3, 3 \times 3,$ $2 \times 3, 2 \times 3, 2 \times 3,$ $2 \times 3, 4 \times 3,$
	Q.14	(a) (b)	$2 \times 6, 2 \times 3, \\1 \times 3, 4 \times 3, 1 \times 3$
	Q.15	(a) (b)	$2 \times 3, 2 \times 3, 2 \times 3, 2 \times 3, 3 \times 3, 1 \times 3$

SECTION A (144 MARKS)

Answer <u>each</u> of the questions 1, 2 and 3.

Question 1. Any eight items, (a), (b), (c), etc. (8 X 6 marks)

(a)	energy that is replaced / will not run out Biomass / geothermal / hydropower / solar (sun) / tidal / wind /	(3) (3)	[6]
(b)	$2 \times 8 / v = at / v = u + at$ 16 (16 only - allow 6 marks)	(3) (3)	[6]
(c)	any <i>one</i> of the four levers in the diagram marked Point (line) about which a lever rotates (pivots) (turns)	(3) (3)	[6]
(d)	Tack A is stable Tack B is unstable	(3) (3)	[6]
(e)	Sponge cake and whipped egg white are insulators/ air in sponge (egg)	(6)	[6]
(f)	changes to a gas / vapour (do not accept evaporate) ammonium chloride / dry ice (CO ₂) (Carbon dioxide) / iodine/ camphor etc	(3) (3)	[6]
(g)	Frequency / wavelength	(6)	[6]
(h)	A is blue B is brown	(3) (3)	[6]
(i)	$2 \times 3 \times 10$ 60 / \notin 0.6 (60 only - allow 6 marks)	(3) (3)	[6]
(j)	converges / forms an image /refracted / will focus changes direction (bends) - allow 3 marks only	(6)	[6]

Question 2. Any eight items, (a), (b), (c), etc. (8 X 6 marks)

(a)	hydrogen peroxide / H ₂ O ₂ manganese dioxide/ MnO ₂	(3) (3)	[6]
(b)	any <i>one</i> from: sulphur dioxide (SO ₂) / nitrogen dioxide (NO ₂) (nitrogen oxides) (NO _x)/ sulphur trioxide (SO ₃) any <i>one</i> from: kills plants / kills fish / damages buildings (stone)	(3)	
	reduces crop yields / releases toxic elements (Al) (Cd) from soil / corrosion etc.	(3)	[6]
(c)	any <i>two</i> from: wear eye protection (goggles) (safety glasses) / do not point the test-tube at anyone / never look into the open end of a test-tube / use test-tube holder (peg)/ hold at an angle / use a small amount of substance / heat gently_etc	(2×2)	តេ
	/use a small amount of substance / neat gently etc.	(2×3)	ĮνJ
(d)	method: magnetic / filtration / evaporation / distillation / separating funnel / chromatography etc. example <i>matched</i> mixture: iron and sand / sand and water /	(3)	
	salt and water / alcohol and water / oil and water/ dyes etc.	(3)	[6]
(e)	(Accept suitable mixture for 3 marks only) Graduated (measuring) cylinder	(3)	
	pipette / burette / volumetric flask / syringe	(3)	[6]
(f)	Prevents/ protect (hardens) (strengthens)	(3)	
	tooth decay/ teeth (enamel)	(3)	[6]
(g)	A: toxic / poisonous	(3)	
	B: flammable	(3)	[6]
(h)	remove hardness / softens	(3)	
	from water	(3)	[6]
(i)	Capillarity / capillary action	(3)	
	absorbency of fabrics / transport in plants / rising damp/ chromatography etc.	(3)	[6]
(j)	any one from: it rusted (corroded) / combined (reacted) with		
	oxygen /	(3)	10
	why?: oxygen used up (removed)	(3)	[0]

Question 3. Any eight items, (a), (b), (c), etc (8 X 6)

(a)	any <i>one</i> from: gastric juice / digestive juice / hydrochloric acid (acid) / enzymes / correctly named enzyme in the stomach etc.	(3)	
	part A: large intestine / colon / bowel	(3)	[6]
(b)	heart 72 accept 69 to 75	(3) (3)	[6]
(c)	by (on) decaying (dead)(organic) matter any <i>one</i> from: recycle matter (elements) / dispose of dead plants (animals) / dispose of organic wastes (faeces) (fallen leaves)/ /fertiliser (humus) / prevents build up of organic waste etc.	(3)	[6]
(d)	any two from: excretion / protection / sensitivity / insulation / helps control temperature / perspiration / makes vitamin D	(2×3)	[6]
(e)	<pre>part A: enamel / crown item: artery / vein / capillary / nerve / blood / lymph/ bacteria</pre>	(3) (3)	[6]
(f)	any <i>one</i> plant: tomato / pea / water lily / dandelion etc any <i>one matched</i> way: animal / self / water / wind etc	(3) (3)	[6]
(g)	leaf, any <i>one</i> from: photosynthesis (makes food) / gaseous exchange (gives off oxygen) (takes in CO ₂) / absorbs sunlight/ transpiration (releases water vapour) / food storage / propagation root. any <i>one</i> from: anchors (supports) plant / absorbs water	(3)	
	(minerals) / stores food / propagation	(3)	[6]
(h)	growth (response) of plant to gravity	(3) (3)	[6]
(i)	diaphragm movement	(3) (3)	[6]
(j)	Petal Pollen / male gamete / male part	(3) (3)	[6]

SECTION B – PHYSICS (48 marks) Answer <u>either</u> question 4 <u>or</u> question 5.

Question 4. (48 marks)

(a)	Define	Force / F divided by area / over A	(3) (3)	[6]
	<u>Give</u>	Nm ⁻² / Ncm ⁻² / Pa / B / psi / atm / torr (symbols or words)	(3)	[3]
	Why?	Greater / increased pressure	(3) (3)	[6]
	<u>Describe</u>	Show or state: boil water in can seal can and cool / stop heating and seal	(3) (3)	
		can collapses	(3) or	
		tumbler of water covered with card	(3)	
		invert holding card in place	(3)	
		card does not fall, when hand is taken away accept equivalent experiments [no diagram – deduct 3 marks]	(3)	[9]
(b)	What?	Vibrates / moves	(3)	[3]
	Describe	Show or state:		
		vibrating fork /sound source	(3)	
		microphone	(3)	
		C.R.O	(3)	
		wave on display	(3)	[12]
		accept equivalent experiments	$\langle \mathbf{a} \rangle$	[2]
	<u>What?</u>	0.33	(3)	[3]
	<u>Calculate</u>	$f = \frac{v}{\lambda} / \frac{330}{0.33} / 330$ divided by	(3)	
		incorrect value	(3)	[6]
		IUUU / I KIIZ	(-)	1~1

Question 5. (48 marks)

(a)	<u>Name</u>	ammeter	(3)	[3]
	<u>Copy</u>	—(A)—	(3)	
		shown in series	(3)	[6]
	Are?	series	(3)	[3]
	What?	6	(3)	[3]
	<u>Calculate</u>	I = $\frac{V}{R} / \frac{6}{6} / 1/6$ divided by incorrect value above	(3)	
		Amps / A	(3)	ю
	What?	direct / d.c.	(3)	[6] [3]
(b)	What?	degree of hotness / how hot (cold)	(3)	[3]
	Name	mercury / alcohol	(3)	[3]
	<u>Give</u>	any <i>one</i> matched: mercury: easily seen / wide range / does not wet tube / measures higher temperatures or alcohol: cheaper / non-toxic / measures lower	(3) or	[3]
		temperatures	(3)	[3]
	What?	expands / rises	(3)	[3]
	Why?	human (body)(core) temperature is 37 °C / between given values	(3) (3)	[6]
	<u>What?</u>	keep (hold) / give sufficient time the reading / or stop	(3) (3) or (3)	
		liquid moving	(3)	[6]

SECTION C - CHEMISTRY (48 marks) Answer <u>either</u> question 6 <u>or</u> question 7.

Question 6. (48 marks)

(a)	<u>(i) Name</u>	oxygen O ₂	(3) (3)	[6]
	<u>(ii) Name</u>	hydrogen H ₂ (names only given but reversed - allow only 3 marks)	(3) (3)	[6]
		(formulae only given but reversed - allow only 3 marks) (names and formulae given but reversed - allow 3×3 marks)		
	(iii) What?	$H_2O / H:O = 2:1 / 2$ atoms of H to 1 atom of O	(3)	[3]
	(iv) Which?	X connected to positive terminal of battery / positive	(3)	[6]
	<u>(v) Name</u>	/ oxygen released graphite / platinum / carbon / nickel / nichrome	(3) (3)	[3]
(b)	<u>Draw</u>	two in orbit nearest nucleus / eight in second orbit seven in the third orbit shown [no diagram – deduct 3 marks]	(3) (3)	[6]
	<u>What?</u>	shared pair of electrons	(3) (3)	[6]
	<u>Describe</u>	two chlorine atoms one shared pair / two dots / dash [no diagram – deduct 3 marks]	(3) (3)	[6]
	<u>Give</u>	any <i>two</i> from: gases / liquids / low melting point / low boiling point/ poor heat conductors (good heat insulators) / poor electrical conductors (good electrical insulators) / soluble in hexane (non-polar solvent)/ insoluble in water etc.	(2×3)	[6]

Question 7. (48 marks)

(a)	Name	magnesium oxide / MgO	(3)	[3]
	How?	add indicator / Named acid - base indicator/ pH paper/ litmus paper	(6)	
	What?	Basic / matched colour in base	(3)	[9]
	What?	exothermic	(3)	[3]
	Write	Mg + $O_2 \longrightarrow MgO$		
		all formulae correct, in an equation - allow all marks	(3×3)	[9]
(b)	Define	(no equation - less 3 marks) oxidation is the loss reduction is the gain	(3) (3)	[6]
	Name	Oxygen / O ₂ /O	(3)	
	Give	gained electrons/ combines with a metal	(3)	[6]
(c)	What?	any <i>one</i> from: fizzes (gas released) / dissolves / heat given off / fast/ clean surface etc.	(6)	
	Name	hydrogen / magnesium sulphate (Epsom salts)	(3)	[9]
	<u>Name</u>	beryllium / calcium / strontium / barium / radium (names only)	(3)	[3]

SECTION D – BIOLOGY (48 marks) Answer <u>either</u> question 8 <u>or</u> 9.

Question 8. (48 marks)

(a)	Name	A: uterus (womb)	(3)	
		B: vagina (birth canal)	(3)	
		C: ovary	(3)	
		D: fallopian tube / oviduct	(3)	[12]
	Give	A: implantation / growth (development) (holds)		
		foetus (baby) / menstruation	(3)	
		B: holds penis / intercourse / receives semen (sperm)		
		/ birth canal (if not given above)/ discharge	(3)	
		C: produces eggs / hormones / named hormone / ovulation	(3)	
		D: fertilisation occurs here / transports zygote	(3)	[12]
		(ovum) (egg) to uterus (womb)	()	L J
(b)	Give	any two from: support / protection / shape / makes		
		blood cells	(2×3)	[6]
	Name	hinge / synovial / moveable	(3)	[3]
	Name	P: ligament / capsule	(3)	
		Q: cartilage / gristle	(3)	[6]
	Evoloin	any and from raduce friction (wear) / sheek	(2)	[2]
	LAplan	absorber	(3)	[3]
	What?	show or state		
		Pairs / biceps relax (contract)	(3)	
		move in opposite directions (against each other) /		
		triceps contract (relax)	(3)	[6]
		(named example - allow 3 marks only)		

Question 9. (48 marks)

(a) <u>Copy &</u> <u>Name</u>

(b)



(3×3)	[9]
(3) (3)	[6]
(2×3)	
(3)	[9]
(3)	[3]
(3) (3)	[9]
(3)	
(3) (3)	[6]
(3) (3)	
(3) (3)	[6]
	 (3×3) (3) (3) (2×3) (3) (3)<!--</td-->

SECTION E – APPLIED SCIENCE (72 marks) Answer <u>two</u> questions from this section.

Question 10 – Earth Science (36 marks). Answer any two of (a), (b), (c).

(a)	<u>Explain</u>	planet: orbits a star (sun)	(3)	
		moons: orbits (satellite of) a planet	(3)	[6]

<u>Explain</u>



		Sun's rays shown correctly / position of Sun Moon orbits Earth any two phases shown (named) correctly in the diagram [no diagram – deduct 3 marks]	(3) (3) (2×3)	[12]
(b)	Name	cumulus / cumulonimbus / cumulostratus	(3)	[3]
	<u>Explain</u>	moist air (water vapour) rises	(3)	
		cools	(3)	
		condenses	(3)	[9]
	What?	amount	(3)	
		of water vapour (moisture)	(3)	[6]
(c)	Give	Measurements: measure volume on ruler Measure temperature on	(3)	
		thermometer	(3)	[6]
		show or state	(3)	
		Graph: two correctly labelled axes	(3)	[6]
		Result / conclusion: straight-line graph volume depends absolute temperature		
		volume is proportional to temperature	(6)	[6]

(a)	Give	Aphid life cycle		
		eggs (3) wingless females (3)		
		males and females / mating (3)	(5×3)	
		females with wings (3) more wingless females (3)	or	
	<u>or</u>	Butterfly lifecycle		
		eggs (3)	(5×3)	[15]
		male imago (adult) (3) female imago (adult) (3)		
	What?	damaged plant gets diseases (fungus) (virus)	(3)	[3]
(b)	<u>Give</u>	any <i>two</i> from: respiration / drainage / root growth / room for soil organisms etc.	(2×3)	[6]
	Describe	Two graduated cylinders / measure volume of can Soil in one, water in the other / make hole in bottom of	(3)	
		can, push can into soil	(3)	
		Pour water into soil / tape hole, lift can filled with soil, fill can with water	(3)	
		Volume decrease is volume of air / volume of water added equals volume of air in soil	(3)	[12]
(c)	<u>Explain</u>	growth	(3)	
	Nama	into a new plant	(3)	[6]
	<u>Name</u>	cabbage (named brassica) / viola (spring bedding plant)		
		rye grass (amenity grass) etc.	(3)	
	<u>Outline</u>	A seed / mark out 10×10 spaces on compost in seed tray	(3)	
		In compost / sow one seed per space	(3) (2)	[14]
		that germinate	(3)	[12]

Question 11 – Horticulture (36 marks). Answer any two of (a), (b), (c).

Question 12 – Materials Science (36 marks). Answer both parts, (a) and (b).

(a)	<u>Name</u>	any <i>two</i> from: aluminium / chromium (chrome) / steel / enamel / leather / plastic (named plastic) / rubber / glass	(2×3)	[6]
		/ paint etc.		
	Give	any two <u>matched</u> from: <u>for aluminium</u> light (low		
		density) / does not rust etc.; for chromium does not rust /		
		looks nice etc.; for steel strong / hard wearing etc.		
		for enamel prevents rust / hard wearing /		
		attractive finish etc.; for leather friction / heat insulation		
		/ hard wearing / looks nice etc.; for plastic friction / heat		
		insulation etc.; for rubber friction / shock absorption etc.	(2×3)	[6]
	Give	any <i>two</i> from: wash (clean) / polish / lubricate (oil)		
	<u></u>	(grease) / paint (touch-up) etc	(2×3)	[6]

(b) Answer any one of the following (i), (ii), (iii), (iv).

(i) <u>Plastics</u>

first stage: small molecules / monomers / fractionation /	(3)	
refining / separate		[6]
second stage: big molecules / polymers	(3)	
show or state		
wrap a metal can with plastic, leave a second can unwrapped	(3)	
fill each can with boiling (at same temperature) water	(3)	
record final temperatures	(3)	
smallest drop in temperature for can covered with plastic	(3)	[12]
accept equivalent experiments		
	first stage: small molecules / monomers / fractionation / refining / separate second stage: big molecules / polymers show or state wrap a metal can with plastic, leave a second can unwrapped fill each can with boiling (at same temperature) water record final temperatures smallest drop in temperature for can covered with plastic accept equivalent experiments	first stage: small molecules / monomers / fractionation /(3)refining / separate(3)second stage: big molecules / polymers(3)show or state(3)wrap a metal can with plastic, leave a second can unwrapped(3)fill each can with boiling (at same temperature) water(3)record final temperatures(3)smallest drop in temperature for can covered with plastic(3)accept equivalent experiments(3)

(ii) <u>Metals</u>

natural (mineral) (rock) (co	ompound) containing metal	(3)	
any one from: copper / lead / silver / zinc		(3)	[6]	
show or state	OP	Two motols	(2)	
uy to scratch metal A	UK	I WO IIIetais	(3)	
with metal B		Inall	(3)	
repeat for B on A		Scratch each	(3)	
harder metal less worn		harder scratches		
		least	(3)	[12]
accept equivalent experim	nents			
	natural (mineral) (rock) (co any <i>one</i> from: copper / lea show or state try to scratch metal A with metal B repeat for B on A harder metal less worn accept equivalent experim	natural (mineral) (rock) (compound any <i>one</i> from: copper / lead / silve show or state try to scratch metal A OR with metal B repeat for B on A harder metal less worn accept equivalent experiments	natural (mineral) (rock) (compound) containing metal any one from: copper / lead / silver / zinc show or state try to scratch metal A OR Two metals with metal B Nail repeat for B on A Scratch each harder metal less worn harder scratches least accept equivalent experiments	natural (mineral) (rock) (compound) containing metal(3)any one from: copper / lead / silver / zinc(3)show or state(3)try to scratch metal AORTwo metals(3)with metal BNailrepeat for B on AScratch eachharder metal less wornharder scratchesleast(3)accept equivalent experiments

(iii) <u>Textiles</u>

<u>Name</u>	plant, any <i>one</i> from: coconut / cotton / flax / hemp / jute / sisal / nettle	(3)	
	animal, any <i>one</i> from: alpacas / camel / caterpillar (silkworm) / goat / llama / sheep / rabbit / horse	(3)	[6]
Describe	show or state		
	weigh two pieces of different fabrics	(3)	
	soak pieces in water	(3)	
	remove fabrics from water and allow to drip for a short time	(3)	
	reweigh one with greatest increase in weight has	(3)	
	greatest absorbency	(3)	[12]
	accept equivalent experiments	(-)	LJ
(iv) <u>Timb</u>	<u>er</u>		
What?	hardwoods any <i>one</i> from: broad / fall in autumn (deciduous)	(3)	
	softwoods any <i>one</i> from: needle / fall year-round (evergreen)	(3)	[6]
Describe	show or state		
	clamp (support) wood at one (both) ends	(3)	
	add weights to opposite end (middle)	(3)	
	repeat for opposite grain direction	(3)	
	cross-grain bends more (breaks more easily)	(3)	[12]
	accept equivalent experiments		

Question 13 – Food (36 marks). Answer any two of (a), (b), (c).

(a)	Which?	(i) carrots / cheese cake	(3)	
		(11) cheese cake / trout	(3)	
		(iii) carrots	(3)	[9]
	Describe	add sodium hydroxide solution	(3)	
		add copper sulphate solution	(3)	
		(accept Bluret Test for 6 marks)	(2)	[0]
		violet (purple) (lilac) colour	(3)	[9]
(b)	Name	any two from: distance / terrain / weather/		
		perishability/ cost/ war	(2×3)	[6]
		poor infrastructure / banditry etc.		
	Name	any two from: drought / erosion / war / AIDS /	(2×3)	[6]
		cash crops / deforestation / floods etc.	()	
	Give	any <i>two</i> from: death / disease / collapse of society		
		/ migration / poverty / illness etc.	(2×3)	[6]
(c)	What?	hacteria	(3)	
(0)	<u>vvnat:</u>	buctoffu	(5)	
	Give	any one from: increase in acidity (lower pH) /		
		new smell / new taste (sours)/ greater viscosity		
		(thicker) (creamier) etc.	(3)	[6]
	D 1			
	Describe	heat milk to 90° C (87 to 93 degrees Celsius)	(3)	
		cool	(3)	
		add culture (bacteria) (natural yoghurt)	(3)	
		keep at 40°C (37 to 43 degrees Celsius) / store in a thermos flask	(3)	[12]

Question 14 – Electronics (36 marks). Answer both parts (a) and (b).





<u>Draw</u> **either diagram**, marks for items shown ignore battery polarity and orientation of diagram



bulbs in parallel shown switch in position shown

switch in any position in series (3)

(b) <u>Give</u>

	(3)	[3]
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- What?(i) LED glows(3)(ii) LED does not glow(3)
- GiveLDR has low resistance in bright light(3)LDR has high resistance in darkness(3)[12]
- How? cathode wire shorter / flat on plastic lens at cathode/ (3) [3] LED glowing means the cathode is connected to negative of battery



(where (i) and (ii) are reversed allow 2×3 marks if the reasons are matched correctly)

Question 15 – Energy Conversions (36 marks). Answer both parts (a) and (b).

(a)	Write	electrical to motion (kinetic) electrical to heat	(3) (3)	[6]
	<u>Give</u>	 any two from: <u>electricity</u> to <i>any</i> of the following: chemical / microwaves / sound / light / infra red (IR) / ultra violet (UV) / radio waves etc. (if energy changes in a hairdryer are incorrect or omitted then accept them in this list) 	(2×3)	[6]
	<u>Name</u>	any two <u>matched</u> from: battery charger / microwave cooker (mobile 'phone) / radio (TV) (stereo) (speaker) (door bell) / lamp (bulb) (torch) (TV) / remote control / sun ray (tanning) lamp / cordless 'phone etc. (accept other domestic appliances e.g. gas with matching energy changes)	(2×3)	[6]
(b)	<u>Name</u>	A: magnet / magnetic (north) pole B: coil / conductor / copper wire	(3) (3)	[6]
	<u>Outline</u>	Current / conductor in a magnetic field has a force act on it	(3) (3) (3)	[9]
	<u>What?</u>	runs in the reverse direction	(3)	[3]