

<b>Cambridge IGCSE : Physics : 0625</b>	<b>Page numbers in New Physics for You</b>
<i>italic = extended / supplement material</i>	
<b>1. General Physics</b>	
<b>1.1 Length and time</b>	
-use and describe the use of rules and measuring cylinders to determine a length or a volume	pages 8, 75, 365
<i>-use and describe the use of a mechanical method for the measurement of a small distance</i>	
-use and describe the use of clocks and devices for measuring an interval of time.	9
<i>-measure and describe how to measure a short interval of time (including the period of a pendulum)</i>	9
<b>1.2 Speed, velocity and acceleration</b>	
-define speed and calculate speed from $\frac{\text{total distance}}{\text{total time}}$	122
-plot and interpret a speed/time graph or a distance/time graph	123 – 126
-recognise from the shape of a speed/time graph when a body is (a) at rest, (b) moving with constant speed, (c) moving with changing speed	124
<i>-distinguish between speed and velocity</i>	122
<i>-recognise linear motion for which the acceleration is constant and calculate the acceleration</i>	124-5
<i>-recognise motion for which the acceleration is not constant</i>	124-5
-calculate the area under a speed/time graph to determine the distance travelled for motion with constant acceleration	125
-demonstrate some understanding that acceleration is related to changing speed	124
-state that the acceleration of free fall for a body near to the Earth is constant	128-9
<i>-describe qualitatively the motion of bodies falling in a uniform gravitational field with and without air resistance (including reference to terminal velocity)</i>	89, 128
<b>1.3 Mass and weight</b>	
-show familiarity with the idea of the mass of a body	9, 68

-state that weight is a force	page 65
<i>-demonstrate an understanding that mass is a property which 'resists' change in motion</i>	68, 130
-demonstrate understanding that weights (and hence masses) may be compared using a balance	67, 75
<i>-describe, and use the concept of, weight as the effect of a gravitational field on a mass</i>	65, 67, 131
<b>1.4 Density</b>	
-describe an experiment to determine the density of a liquid and of a regularly shaped solid and make the necessary calculation	74-76
<i>-describe the determination of the density of an irregularly shaped solid by the method of displacement and make the necessary calculation</i>	75
<b>1.5 Forces</b>	
<b>(a) Effects of forces</b>	
-state that a force may produce a change in size and shape of a body	65
-plot extension/load graphs and describe the associated experimental procedure	66
<i>-interpret extension/load graphs</i>	66
<i>-state Hooke's Law and recall and use the expression <math>F = kx</math></i>	66
<i>-recognise the significance of the term 'limit of proportionality' for an extension/load graph</i>	66
-describe the ways in which a force may change the motion of a body	69
-find the resultant of two or more forces acting along the same line	86
<i>-recall and use the relation between force, mass and acceleration (including the direction)</i>	130-131
<i>-describe, qualitatively, motion in a curved path due to a perpendicular force (<math>F = mv^2 / r</math> is not required)</i>	70-71
<b>(b) Turning effect</b>	
-describe the moment of a force as a measure of its turning effect and give everyday examples	90
-describe, qualitatively, the balancing of a beam about a pivot	91
<i>-perform and describe an experiment (involving vertical forces) to verify that there is no net moment on a body in equilibrium</i>	91
<i>-apply the idea of opposing moments to simple systems in equilibrium</i>	91

<p><b>(c) Conditions for equilibrium</b>          -state that, when there is no resultant force and no resultant turning effect, a system is in equilibrium.</p>	pages 86-87, 91
<p><b>(d) Centre of mass</b>          -perform and describe an experiment to determine the position of the centre of mass of a plane lamina.</p>	92-93
<p>-describe qualitatively the effect of the position of the centre of mass on the stability of simple objects.</p>	93
<p><b>(e) Scalars and vectors</b>  <i>-demonstrate an understanding of the difference between scalars and vectors and give common examples</i></p>	86
<p><i>-add vectors by graphical representation to determine a resultant</i></p>	86
<p><i>-determine graphically a resultant of two vectors</i></p>	86
<p><b>1.6 Energy, work and power</b></p>	
<p><b>(a) Energy</b>          -demonstrate an understanding that an object may have energy due to its motion or its position, and that energy may be transferred and stored</p>	10-11, 97-99
<p>-give examples of energy in different forms, including kinetic, gravitational, chemical, strain, nuclear, internal, electrical, light and sound</p>	10-11, 98
<p><i>-recall and use the expressions <math>k.e. = \frac{1}{2}mv^2</math> and <math>p.e. = mgh</math></i></p>	108-109
<p>-give examples of the conversion of energy from one form to another and of its transfer from one place to another</p>	11, 98, 100-101
<p>-apply the principle of energy conservation to simple examples</p>	98
<p><b>(b) Energy resources</b>          -describe how electricity or other useful forms of energy may be obtained from</p>	
<p>(i) chemical energy stored in fuel</p>	101, 104-106
<p>(ii) water, including the energy stored in waves, in tides, and in water behind hydroelectric dams</p>	15, 101, 106
<p>(iii) geothermal resources</p>	15
<p>(iv) nuclear fission</p>	349
<p>(v) heat and light from the Sun</p>	14, 103
<p><i>-show an understanding that energy is released by nuclear fusion in the Sun</i></p>	156
<p><i>-show a qualitative understanding of efficiency</i></p>	102
<p><b>(c) Work</b>          -relate, without calculation, work done to the magnitude of a force and the distance moved</p>	97
<p><i>-describe energy changes in terms of work done</i></p>	99
<p><i>-recall and use <math>\Delta W = Fd = \Delta E</math></i></p>	97-99

<p><b>(d) Power</b>          -relate, without calculation, power to work done and time taken, using appropriate examples</p> <p><i>-recall and use the equation <math>P = E/t</math> in simple systems</i></p> <p><b>1.7 Pressure</b></p> <p>-relate, without calculation, pressure to force and area, using appropriate examples.</p> <p><i>-recall and use the equation <math>p = F/A</math></i></p> <p>-describe the simple mercury barometer and its use in measuring atmospheric pressure</p> <p>-relate, without calculation, the pressure beneath a liquid surface to depth and to density, using appropriate examples</p> <p>-use and describe the use of a manometer</p> <p><i>-recall and use the equation <math>p = h\rho g</math></i></p>	<p>page 110</p> <p>110-111</p> <p>77</p> <p>77</p> <p>80</p> <p>78</p> <p>80</p> <p>78</p>
<p><b>2. Thermal Physics</b></p> <p><b>2.1 Simple kinetic molecular model of matter</b></p> <p><b>(a) States of matter</b>          -state the distinguishing properties of solids, liquids and gases</p> <p><b>(b) Molecular model</b>          -describe qualitatively the molecular structure of solids, liquids and gases</p> <p><i>-relate the properties of solids, liquids and gases to the forces and distances between molecules and to the motion of the molecules</i></p> <p>-interpret the temperature of a gas in terms of the motion of its molecules</p> <p>-describe qualitatively the pressure of a gas in terms of the motion of its molecules</p> <p>-describe qualitatively the effect of a change of temperature on the pressure of a gas at constant volume</p> <p>-show an understanding of the random motion of particles in a suspension as evidence for the kinetic molecular model of matter</p> <p>-describe this motion (sometimes known as Brownian motion) in terms of random molecular bombardment</p> <p><i>-show an appreciation that massive particles may be moved by light, fast moving molecules</i></p>	<p>17</p> <p>16-18</p> <p>17-18</p> <p>18, 20, 27, 34</p> <p>34</p> <p>32, 34</p> <p>18</p> <p>18</p> <p>18</p>

<b>(c) Evaporation</b>	
-describe evaporation in terms of the escape of more-energetic molecules from the surface of a liquid	page 56
<i>-demonstrate an understanding of how temperature, surface area and draught over a surface influence evaporation</i>	56
-relate evaporation and the consequent cooling	56
<b>(d) Pressure changes</b>	
-relate the change in volume of a gas to change in pressure applied to the gas at constant temperature	29, 34
<i>-recall and use the equation <math>pV = \text{constant}</math> at constant temperature</i>	29
<b>2.2 Thermal properties</b>	
<b>(a) Thermal expansion of solids, liquids and gases</b>	
-describe qualitatively the thermal expansion of solids, liquids and gases	21
<i>-show an appreciation of the relative order of magnitude of the expansion of solids, liquids and gases</i>	24
-identify and explain some of the everyday applications and consequences of thermal expansion	22-23
-describe qualitatively the effect of a change of temperature on the volume of a gas at constant pressure	30-34
<b>(b) Measurement of temperature</b>	
-appreciate how a physical property which varies with temperature may be used for the measurement of temperature and state examples of such properties	26-27, 31-32
<i>-demonstrate understanding of sensitivity, range and linearity</i>	26, 362
-recognise the need for and identify fixed points	
-describe the structure and action of liquid in-glass thermometers	26-27
<i>-describe the structure of a thermocouple and show understanding of its use for measuring high temperatures and those which vary rapidly</i>	
<b>(c) Thermal capacity</b>	
-relate a rise in temperature of a body to an increase in internal energy	35-37
-show an understanding of the term thermal capacity	36, 38
<i>-describe an experiment to measure the specific heat capacity of a substance</i>	36-37
<b>(d) Melting and boiling</b>	
-describe melting and boiling in terms of energy input without a change in temperature	53-55

<p><i>-distinguish between boiling and evaporation</i></p> <p><i>-state the meaning of melting point and boiling point</i></p> <p><i>-describe condensation and solidification</i></p> <p><i>-use the terms latent heat of vaporization and latent heat of fusion and give a molecular interpretation of latent heat</i></p> <p><i>-describe an experiment to measure specific latent heats for steam and for ice</i></p> <p><b>2.3 Transfer of thermal energy</b></p> <p><b>(a) Conduction</b></p> <p><i>-describe experiments to demonstrate the properties of good and bad conductors of heat</i></p> <p><i>-give a simple molecular account of heat transfer in solids</i></p> <p><b>(b) Convection</b></p> <p><i>-relate convection in fluids to density changes and describe experiments to illustrate convection</i></p> <p><b>(c) Radiation</b></p> <p><i>-identify infra-red radiation as part of the electromagnetic spectrum</i></p> <p><i>-describe experiments to show the properties of good and bad emitters and good and bad absorbers of infra-red radiation</i></p> <p><b>(d) Consequences of energy transfer</b></p> <p><i>-identify and explain some of the everyday applications and consequences of conduction, convection and radiation</i></p>	<p>(pages 55, 56)</p> <p>(27, 53, 55)</p> <p>(56, 54)</p> <p>pages 53, 55</p> <p>54, 55</p> <p>40-42</p> <p>41</p> <p>44-45</p> <p>46-48, 209</p> <p>46-47</p> <p>40-51</p>
<p><b>3. Properties of waves, including light and sound</b></p> <p><b>3.1 General wave properties</b></p> <p><i>-describe what is meant by wave motion as illustrated by vibration in ropes, springs and by experiments using water waves</i></p> <p><i>-use the term wavefront</i></p> <p><i>-give the meaning of speed, frequency, wavelength and amplitude</i></p> <p><i>-recall and use the equation <math>v = f\lambda</math></i></p> <p><i>-distinguish between transverse and longitudinal waves and give suitable examples</i></p>	<p>166-169</p> <p>168</p> <p>167</p> <p>167</p> <p>166</p>

-describe the use of water waves to show	
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(ii) refraction due to a change of speed	168
(iii) diffraction produced by wide and narrow gaps	169
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<b>3.2 Light</b>	
<b>(a) Reflection of light</b>	
-describe the formation, and give the characteristics, of an optical image by a plane mirror	176-179
-use the law angle of incidence = angle of reflection	177
<i>-perform simple constructions, measurements and calculations</i>	176-178
<b>(b) Refraction of light</b>	
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-use the terminology for the angle of incidence $i$ and angle of refraction $r$ and describe the passage of light through parallel-sided transparent material	184-185
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<i>-recall and use the equation <math>\sin i / \sin r = n</math></i>	185
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-describe internal and total internal reflection	187
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-state the approximate value of the speed of electro-magnetic waves	page 209
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-describe the longitudinal nature of sound waves	225
-describe compression and rarefaction	225
-state the approximate range of audible frequencies	230
-show an understanding that a medium is required in order to transmit sound waves	225
-describe an experiment to determine the speed of sound in air	227
-state the order of magnitude of the speed of sound in air, liquids and solids	227
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-describe how the reflection of sound may produce an echo	226
<b>4. Electricity and magnetism</b>	
<b>4.1 Simple phenomena of magnetism</b>	
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-give an account of induced magnetism	(281), 283
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-describe an experiment to identify the pattern of field lines round a bar magnet	282
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-state that charge is measured in coulombs	245



-state that unlike charges attract and that like charges repel	page 241
-describe an electric field as a region in which an electric charge experiences a force	244
<i>-state the direction of lines of force and describe simple field patterns</i>	244
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<i>-recall and use the simple electron model to distinguish between conductors and insulators</i>	242
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-use and describe the use of an ammeter	250
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<i>-show understanding that e.m.f. is defined in terms of energy supplied by a source in driving charge round a complete circuit</i>	252, 261
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-use and describe the use of a voltmeter	252
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-describe an experiment to determine resistance using a voltmeter and an ammeter	255
-relate (without calculation) the resistance of a wire to its length and to its diameter	254
-recall and use quantitatively the proportionality between resistance and the length and the inverse proportionality between resistance and cross-sectional area of a wire	254

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<b>4.3 Electric circuits</b>	
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<b>(b) Series and parallel circuits</b> -understand that the current at every point in a series circuit is the same	250
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-state that, for a parallel circuit, the current from the source is larger than the current in each branch	251
<i>-recall and use the fact that the current from the source is the sum of the currents in the separate branches of a parallel circuit</i>	251
-state that the combined resistance of two resistors in parallel is less than that of either resistor by itself	257
<i>-calculate the effective resistance of two resistors in parallel</i>	257
<i>-state the advantages of connecting lamps in parallel in a lighting circuit</i>	250-251
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-describe the action of thermistors and light dependent resistors and show understanding of their use as input transducers	259, 319, 323-4, 329
-describe the action of a capacitor as an energy store and show understanding of its use in time delay circuits	245, 325
-describe the action of a relay and show understanding of its use in switching circuits	320-321, 323, 329
<i>-describe the action of a diode and show understanding of its use as a rectifier</i>	316-317
<i>-describe the action of a transistor as an electrically operated switch and show understanding of its use in switching circuits</i>	322-324
<i>-recognise and show understanding of circuits operating as light sensitive switches and temperature operated alarms (using a relay or a transistor)</i>	323-324

<b>(d) Digital electronics</b>	
-explain and use the terms digital and analogue	page 218
- state that logic gates are circuits containing transistors and other components	326
-describe the action of NOT, AND, OR, NAND and NOR gates	326-327
-design and understand simple digital circuits combining several logic gates	328-327
-state and use the symbols for logic gates (the American ANSI#Y 32.14 symbols will be used)	326-327
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-state the hazards of	(268)
(i) damaged insulation	
(ii) overheating of cables	
(iii) damp conditions	
-show an understanding of the use of fuses and/or circuit-breakers	269
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<b>(a) Electromagnetic induction</b>	
-describe an experiment which shows that a changing magnetic field can induce an e.m.f. in a circuit	296-297
-state the factors affecting the magnitude of an induced e.m.f.	296
-show understanding that the direction of an induced e.m.f. opposes the change causing it	297
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-describe a rotating-coil generator and the use of slip rings	298
-sketch a graph of voltage output against time for a simple a.c. generator	299
<b>(c) Transformer</b>	
-describe the construction of a basic iron-cored transformer as used for voltage transformations	301-302
-describe the principle of operation of a transformer	302
-recall and use the equation $(V_p / V_s) = (N_p / N_s)$	302-303
-recall and use the equation $V_p I_p = V_s I_s$ (for 100% efficiency)	302
-describe the use of the transformer in high-voltage transmission of electricity	303
-give the advantages of high voltage transmission	303
-discuss energy losses in cables	303, 305

<p><b>(d) The magnetic effect of a current</b>          - describe the pattern of the magnetic field due to currents in straight wires and in solenoids</p>	pages 286-287
<p><i>-state the qualitative variation of the strength of the magnetic field over salient parts of the pattern</i></p>	286-287
<p><i>-describe the effect on the magnetic field of changing the magnitude and direction of the current</i></p>	287
<p>-describe applications of the magnetic effect of current, including the action of a relay</p>	288-289, 320-321
<p><b>(e) Force on a current carrying conductor</b>          -describe an experiment to show that a force acts on a current-carrying conductor in a magnetic field, including the effect of reversing:          (i) the current          (ii) the direction of the field</p>	290-291
<p><i>-describe an experiment to show the corresponding force on beams of charged particles</i></p>	309
<p><i>-state and use the relative directions of force, field and current</i></p>	290
<p><b>(f) d.c. motor</b>          -state that a current-carrying coil in a magnetic field experiences a turning effect and that the effect is increased by increasing the number of turns on the coil</p>	291-293
<p>-relate this turning effect to the action of an electric motor</p>	291-293
<p><i>-describe the effect of increasing the current</i></p>	290-293
<p><b>4.6 Cathode ray oscilloscopes</b></p>	
<p><b>(a) Cathode rays</b>          -describe the production and detection of cathode rays</p>	308-311
<p>-describe their deflection in electric fields</p>	309, 310
<p>-state that the particles emitted in thermionic emission are electrons</p>	308
<p><b>(b) Simple treatment of cathode-ray oscilloscope</b>  <i>-describe in outline the basic structure and action of a cathode-ray oscilloscope (detailed circuits are not required)</i></p>	309-310
<p><i>-use and describe the use of a cathode-ray oscilloscope to display waveforms</i></p>	311

<b>5. Atomic Physics</b>	
<b>5.1 Radioactivity</b>	
<b>(a) Detection of radioactivity</b>	
-show awareness of the existence of background radiation	pages 340, 350
-describe the detection of $\alpha$ -particles, $\beta$ -particles and $\gamma$ -rays ( $\beta^+$ is not included: $\beta$ -particles will be taken to refer to $\beta^-$ .)	340-341
<b>(b) Characteristics of the three kinds of emission</b>	
-state that radioactive emissions occur randomly over space and time	page 339
-state, for radioactive emissions:	340-341
(i) their nature (ii) their relative ionising effects (iii) their relative penetrating abilities	
<i>-describe their deflection in electric fields and magnetic fields</i>	340-341
<i>-interpret their relative ionising effects</i>	340-341
<b>(c) Radioactive decay</b>	
-state the meaning of radioactive decay, using equations (involving words or symbols) to represent changes in the composition of the nucleus when particles are emitted	344-345
<b>(d) Half-life</b>	
-use the term half-life in simple calculations which might involve information in tables or decay curves	344, 352
<b>(e) Safety precautions</b>	
-describe how radioactive materials are handled, used, stored in a safe way	350
<b>5.2 The nuclear atom</b>	
<b>(a) Atomic model</b>	
-describe the structure of an atom in terms of a nucleus and electrons	342-343
<i>-describe how the scattering of <math>\alpha</math>-particles by thin metal foils provides evidence for the nuclear atom</i>	342
<b>(b) Nucleus</b>	
-describe the composition of the nucleus in terms of protons and neutrons	342-343
-use the term proton number $Z$	343
-use the term nucleon number $A$	343
-use the term nuclide and use the nuclide notation ${}^A_Z X$	343
<b>(c) Isotopes</b>	
<i>-use the term isotope</i>	343
<i>-give and explain examples of practical applications of isotopes</i>	346-347
end of specification content	