

**MARK SCHEME for the October/November 2009 question paper
for the guidance of teachers**

<p>0652/02 0652 PHYSICAL SCIENCE Paper 2 (Core Theory), maximum raw mark 80</p>
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This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

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- 1 (a) covalent [1]
- (b) correct arrangement with shared electron pair, correct outer shells [1]
- (c) any **two** from:
 high melting point
 electrolyte when molten or aqueous
 crystalline
 soluble in water
 etc. 1 + 1 [2]
- [Total: 4]**
- 2 (a) $R = V / I$ or $6.0 / 2.4$
 $= 2.5 \Omega$ 1
 1 [2]
- (b) 5.0Ω (e.c.f.) [1]
- (c) $I = V / R$ or $= 6 / 5$ (e.c.f.)
 $= 1.2 \text{ A}$ 1
 1 [2]
- [Total: 5]**
- 3 (a) substance which (is burned to) release heat / energy [1]
- (b) (i) any **two** from:
 non-polluting / makes only water when burned
 easy to transport through pipes
 lights easily
 high heat output
 etc. 1 + 1 [2]
- (ii) has to be manufactured / etc. [1]
- (c) (i) fermentation [1]
- (ii) add to limewater 1
 turns cloudy / milky / white precipitate 1 [2]
- (iii) fractional distillation [1]
- [Total: 8]**

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4	(a) (i) greater amplitude			[1]
	(ii) more waves on screen / waves close together (accept higher frequency / shorter wavelength)			[1]
	(b) (i) 20 000 Hz (20 kHz) (accept 10 – 30 kHz)			[1]
	(ii) $v = \text{distance} / \text{time}$ or $\text{distance} = vt$ or 320×0.075	1		
	= 24 m	1		
	but $\frac{1}{2}$ this distance = 12 m from wall	1		[3]
				[Total: 6]
5	(a) (i) moment = 250×0.6 = 150 (Nm)	1		
		1		[2]
	(ii) $150 = F \times 2.4$ $F = 63$ (62.5)N	1		
	(if final force (62.5N) is correctly found and inserted into (i) score 3 out of 4 marks, ignore remainder in (ii)).	1		[2]
	(b) (i) horizontal line at 2.5 m diagonal line to time axis covering 8 s.	1		
		1		[2]
	(ii) attempt to find area under graph $(2.5 \times 12) + (\frac{1}{2} \times 2.5 \times 8)$ = 40 m	1		
		1		
		1		[3]
				[Total: 9]
6	(a) mixture of metals			[1]
	(b) e.g. brass ornaments / electrical terminals / etc.	1		
		1		[2]
	(c) (i) painting / chrome plating / etc.			[1]
	(ii) too dense / too expensive / not strong enough / etc.			[1]
				[Total: 5]

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7	(a) (i) radiation			[1]
	(ii) ray correctly drawn			[1]
	(iii) both angle of incidence and angle of reflection correctly drawn			[1]
	(iv) angle of incidence = angle of reflection			[1]
	(b) (i) conduction			[1]
	(ii) hot water less dense than cold therefore floats / rises to the top (mention of convection – C1)		1 1	[2]
	(c) (i) distillation			[1]
	(ii) idea of waste energy from turbine used			[1]
				[Total: 9]
8	(a) A turns red	no gas	1 + 1	
	B fizzes / dissolves	hydrogen	1 + 1	
	C fizzes / dissolves	carbon dioxide	1 + 1	[6]
	(b) no change		1	
	relevant explanation about acids e.g. all contain hydrogen ions, etc.		1	[2]
				[Total: 8]
9	(a) splitting of <u>nucleus</u> (into two more or less equal halves) with release of energy		1 1	[2]
	(b) advantage: no greenhouse gases released / chemical pollutants		1	
	disadvantage: danger of radioactive substances leaking / difficulty of dealing with long half-life waste (do not accept explosions, etc.)		1	[2]
				[Total: 4]

Page 5	Mark Scheme: Teachers' version	Syllabus	Paper
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- 10 (a)
- | | | |
|----------|---|--|
| hydrogen | 2 | |
| sulfur | 8 | |
| oxygen | 1 | |
| | 4 | |
- (3 correct names = 1 mark)
(4 correct numbers = 3 marks; 3 correct = 2 marks; 2 correct = 1 mark) [4]

- (b) 28 g
(allow one mark for '2 atoms nitrogen' with incorrect final answer)
(calculation of mass of one mole of ammonium hydroxide = (80) C1) [2]

[Total: 6]

- 11 (a) source (much) nearer to detector 1
because alphas short range or different type of detector 1 [2]

- (b) (i) mention of background count 1
subtracted from original count 1 [2]

- (ii) smooth curve going within 1 square of all points [1]

- (iii) clear working or 12.5 ± 1.0 s 1
 12.5 ± 0.5 s 1 [2]

[Total: 7]

- 12 (a) faster [1]

- (b) (i) unreactive / can withstand high temperature / etc. [1]

- (ii) only small amount needed / increases surface / etc. [1]

- (c) not used up by reactions [1]

- (d) $2\text{CO} + 2\text{NO} \rightarrow 2\text{CO}_2 + \text{N}_2$
(correct formulae – 1 mark correct balancing – 1 mark) [1]

[Total: 5]

13 (a)

particle	relative mass	relative charge
electron	0 / very small / 1/2000 etc.	-1
neutron	1	0
proton	1	+ 1

[3]

(b) number of protons in an atom / nucleus

[1]

[Total: 4]