

Cambridge International Examinations Cambridge International General Certificate of Secondary Education

CO-ORDINATED SCIENCES

0654/63 May/June 2016

Paper 6 Alternative to Practical MARK SCHEME Maximum Mark: 60

Published

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge will not enter into discussions about these mark schemes.

Cambridge is publishing the mark schemes for the May/June 2016 series for most Cambridge IGCSE[®], Cambridge International A and AS Level components and some Cambridge O Level components.

 \circledast IGCSE is the registered trademark of Cambridge International Examinations.

International Examinations

Pa	ge 2	2	Mark Scheme	Syllabus	Paper
			Cambridge IGCSE – May/June 2016	0654	63
1	(a)	inc mc	isor (front) ; lar (back) ;		[2]
	(b)	(i)	dissolve sample of plaque in distilled water ; use of full range indicator/pH meter ;		[2]
		(ii)	below 7 ;		[1]
		(iii)	acid produced by bacteria/sugar forms acid ;		[1]
	(c)	2 g ove cor les sw	proups – brushing twice and brushing three times ; er several days/weeks ; mpare/measure amount of staining ; s staining means less plaque ; ap groups over as a control ;		[max 4]
					[Total: 10]
2	(a)	(i)	salt C label pointing to residue in filter paper AND salt B label pointing to filtrate in beaker ;		[1]
		(ii)	correct residue label AND correct filtrate label ;		[1]
	(b)	(i)	[]		

	conclusion		
(add HCl)	not carbonate/not CO ₃ ²⁻ ;		
(add HCl + BaCl)	sulfate/SO ₄ ²⁻ ;		
(add NaOH)	copper(II)/Cu ²⁺ ;		

[3]

- (ii) copper(II) sulfate ; [1]
- (c) (i) limewater goes milky/white ppt.; [1]
 - (ii) white ppt. ; ppt. dissolves ; [2]
 - (iii) ZnCO₃; [1] [Total: 10]

P	age (3	Mark Scheme Cambridge IGCSE – May/June 2016	Syllabus 0654	Paper 63
3	(a)	0.	19 (V) ;		[1]
	(b)	R co	values correct (should be: 0.79, 2.42, 4.00) ; nsistent significant figures ;		[2]
	(c)	ax su	tes labelled with units ; itable choice of scales ($\ge \frac{1}{2}$ the grid used) ;		
		plo go	ots correct to $\frac{1}{2}$ small square ; ood best-fit line judgement ;		[4]
	(d)	dir str	rectly proportional ; raight line through the origin ;		[2]
	(e)	sv	vitch off between readings/fan the wire/resistor in series with the wire	;	[max 1]
					[Total: 10]
4	(a)	go co	ood size drawing with clear lines ; rrect shape ;		[2]
	(b)	(i)	correct measurement (34 mm) ;		[1]
		(ii)	correct measurement (from their drawing);		[1]
		(iii)	correct calculation ;		[1]
	(c)	(i)	3 correct labels ;;;		[3]
		(ii)	(agree) cell wall and nucleus = 1 mark ; <i>any one from:</i> starch grain/vacuole/chloroplast ;		[max 2]
					[Total: 10]
5	(a)	(i)	71.8 ; 79.6 ;		[2]
		(ii)	20.3 ; 28.1 ;		[2]
		(iii)	48 ;		[1]
		(iv)	not all iron reacted/not hot enough ; etc.		[1]
	(b)	ch	lorine/gas is toxic ;		[1]

Page 4		4	Mark Scheme		Paper
			Cambridge IGCSE – May/June 2016	0654	63
	(c)	us m irc	e of sodium hydroxide ; ention of dissolving, water, solution or aqueous ; n(II) green ppt. AND iron(III) brown ppt.;		[3]
					[Total: 10]
6	(a)	(i)	118 ;		[1]
		(ii)	83 (only) ;		[1]
		(iii)	max use of paper e.g. vertical axis starts at 30 ; correct plotting ; smooth curves :		
			(at least) one curve labelled ;		[4]
		(iv)	<i>(similar)</i> both start at same temp/both go down, etc. ; <i>(different)</i> go down at different rates/end at different temps, etc. ;		[2]
	(b)	re	sult at <u>8 mins</u> is wrong/anomalous ;		[1]
	(c)	e. or	g. pour same volume of water into each container ; record initial temperatures in the beakers ;		[max 1]
					[Total: 10]