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CO-ORDINATED SCIENCES

0654/33

Paper 3 Extended Theory

May/June 2016

MARK SCHEME
Maximum Mark: 120

Published

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-	age z	-	Cambridge IGCSE – May/June 2016	0654	33
1	(a)	(i)	malleability ;		[1]
		(ii)	resistant to corrosion;		[1]
	(b)	(i)	alloy;		[1]
		(ii)	(alloy is) stronger; so can withstand the increased pressure inside the can;		[2]
	(c)	(i)	electrolyte must be kept liquid/molten/aluminium oxide has a high point; reference to the need for ionic mobility;	melting	[2]
		(ii)			[2]
	((iii)	C + $O_2 \rightarrow CO_2/2C$ + $O_2 \rightarrow 2CO$ formulae ; balancing ;		[2] [Total: 11]
2	(a)	(i)	red blood cell;		[1]
		(ii)	engulfs/surrounds foreign particles; digests them;		[2]
	((iii)	produce <u>antibodies</u> ; rejection;		[2]
	(b)	(i)	(artery) carries blood <u>away from the heart</u> / vein carries blood <u>towards the heart</u> ;		[1]
		(ii)	stronger wall/prevents bursting; (because) blood pressure is high;		[2]
	((iii)	stretch/recoil/expand; smooths out (variation in rate of) blood flow/ (variations in) pressure/pulses;		[2]
					[Total: 10]
3	(a)	(i)	volume = $37.5 \text{ (m}^3)$;		[1]
		(ii)	(mass =) density \times volume or 880 \times 37.5 ; = 33 000 (kg) ;		[2]
	(b)		al and natural gas the rest renewable ;		[1]

Syllabus

Paper

P	age :	3	Mark Scheme	Syllabus	Paper
			Cambridge IGCSE – May/June 2016	0654	33
	(c)	(i)	from 20 (Hz) (allow 10 Hz) to 20 000 (Hz) (allow 25 000 Hz);		[1]
		(ii)	(distance =) speed \times time or 1500 \times 1.2 (= 1800) ; (divide by 2) 900 (m) ;		[2]
		(iii)	compressions further apart and distance between two compressions/rarefactions identified;		[1]
					[Total: 8]
4	(a)	fou	ss → zebra → lion → flea r organisms in correct order ; rect arrows ;		[2]
	(b)	(i)	ecosystem;		[1]
		(ii)	trophic level;		[1]
		(iii)	decomposer;		[1]
	(c)	gra ene	ss ; ergy losses at each stage ;		[2] [Total: 7]
5	(a)	(i)	lamps in parallel/all correct symbols; all else correct;		[2]
		(ii)	$(Q =)$ It or = 1.5 \times 300; = 450; C;		[3]
	(b)	(i)	speed = wavelength \times frequency or $4.8 \times 10^{-7} \times 6.25 \times 10^{14}$; = 3.0×10^{8} (m/s);		[2]
		(ii)	all travel at same speed; are transverse waves; have electric and magnetic field components;		[max 1]
	(c)		droplets have same charge ; s repel each other ;		[2]
	(d)	late	erally inverted/upright/virtual ;		[1]

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	(e)	incı	reased airflow ; reased temperature ; ger surface area ;		
		low	er humidity ;		[max 2]
					[Total: 13]
6	(a)	(i)	D on any cell after fertilisation and		
			H on either sperm or egg;		[1]
		(ii)	mitosis ;		[1]
		(iii)	23 ; 46 ;		[2]
	(b)	(i)	mutation ;		[1]
		(ii)	some individuals more tolerant of drought/AW; some bacteria more resistant to the antibiotic;		[2]
	(c)	(i)	A;		[1]
		(ii)	discontinuous/discrete;		[1]
		(iii)	(different) genes/alleles/genotypes;		[1]
					[Total: 10]
7	(a)	оху	gen ;		[1]
	(b)	(i)	(A) amount of gas produced in a given time is greatest/the gradient is great	est;	[1]
		(ii)	59 ± 1 seconds;		[1]
		(iii)	volume stops increasing/no more gas being produced (after 59s); graph becomes horizontal, flattens, gradient = 0		[1]
	(c)	2.0	(g);		
	(0)		alysts are not consumed/permanently changed;		[2]
	(d)	the	e of reaction would be greater; idea that molecules of R must collide with the surface of Q ;		
		_	her concentration of R means larger number of molecules (per unit volumerence to increased frequency of (reactive) collisions;	e);	[max 3]
					[Total: 9]

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(a)	particles move more slowly/have less energy; particles become closer together; gas becomes more dense;		[max 2]
(b)	B AND (most) particles are touching; and random arrangement;		[2]
(c)	energy required to overcome attractive forces between particles; red. to latent heat of fusion;		[max 1]
(d)	$1/R_T$ = $1/R_1$ + $1/R_2$ or $1/R_T$ = $1/5500$ + $1/5500$ = $2/5500$ or R_T = $R_1R_2/(R_1$ + $R_2)$ or R_T = $5500\times5500/(2\times5500)$ or effective resistance of two equal resistances in parallel is half one of the resistances or owtte or $5500/2$; R_T = $2750\;(\Omega)$;	е	[2]
(e)	(SHC =) energy/(mass × change in temperature) or		
(-)	3.03/0.20 × 15 ;		
	= 1.01 (J/kg°C);		[2]
(f)	current produces magnetic field around coil; magnetic field produced interacts with other magnetic field; force on current carrying conductor in magnetic field; force acts on side of coil; forces act in opposite direction on each side of coil; current reverses every half turn;		[m. a.v. 2]
	keeps coil turning in same direction ;		[max 3]
			[Total: 12]
(a)	(i) a carbon atom/nucleus contains 6 protons; (on average) an oxygen atom has a mass 16 times greater than a hydrogen atom/1/12 mass of C-12/ other valid forms of the definition;		[2]
	(ii) $19 \times 2 = 38$;		[1]
(iii) neon; all electron shells are full/outer shell is complete/has 8 electrons/ need to bond to complete outer shell;	does not	[2]
(b)	(i) 7 electrons/in shells/energy levels surrounding the nucleus; 2,5 configuration;		[2]
	(ii) three shared pairs;both lone pairs and no extra electrons;		[2]
			[Total: 9]

Syllabus

Paper

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10	(a)		lity to detect/sense changes in the environment ; d) to respond to them ;		[2]
	(b)	(i)	phototropism;		[1]
		(ii)	stem would grow upwards, with or without light/anyway; no control experiment; (could be) growing against/responding to gravity;		[max 2]
		(iii)	auxin accumulates on lower/darker side; so more growth/cell elongation on this side;		[2]
					[Total: 7]
11	(a)	(i)	area under graph or evidence in working ; distance = $(\frac{1}{2} \times 20 \times 6) + (20 \times 6) + (\frac{1}{2} \times 5 \times 6)$; 195 (m) ;		[3]
		(ii)	(KE =) $\frac{1}{2} m v^2$ or $\frac{1}{2} \times 80 \times 6 \times 6$; = 1440 (J);		[2]
	(b)		ow labelled A going downwards ; ow labelled B going to the left ;		[2]
					[Total: 7]
12	(a)	(i)	L is fractional distillation; M is (catalytic/thermal) cracking;		[2]
		(ii)	contain only carbon and hydrogen/is a hydrocarbon; relatively unreactive; are saturated/contain only single bonds; have the general formula C_nH_{2n+2} ;		[max 2]
	(b)		C and 4 × H ; rect bonding ;		[2]
	(c)	(i)	addition;		[1]
		(ii)	colourless;		[1]
		(iii)	calculate the M_{r} of dibromoethane = 188 ; calculate mass of 0.00625 moles = 188 \times 0.00625 = 1.175 ;		[2]
		(iv)	no reaction; (as alkane/ethane has no double bond to react/owtte);		[1]
					[Total: 11]

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13 (a) for food/energy/oxygen;

[1]

(b) (i) if further then less light/colder; (ORA) rate of photosynthesis depends on light (intensity)/temperature; needs liquid water;

[max 2]

(ii) CO₂ needed for photosynthesis;

[1]

(iii) more oxygen/less CO₂; due to photosynthesis;

[2]

[Total: 6]