## MARK SCHEME for the October/November 2010 question paper

## for the guidance of teachers

## 0654 CO-ORDINATED SCIENCES

0654/32 Paper 3 (Extended Theory), maximum raw mark 100

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

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	<u> </u>		IGCS	E – October/Novembe	er 2010	0654	32	
1	(a) (i	) pin	pink / orange / brown / copper (layer) ;					
	<ul> <li>(ii) 2+ ; two negative charges from chloride must balance the charge on the ion / owtte ;</li> </ul>					charge on the co	pper [2]	
	(iii	<ul> <li>(iii) (L) it is a negative ion / has a negative charge / has more electrons than protor reference to attraction between opposite charges ;</li> </ul>					ons ; [2]	
	(iv	×						
			e shared pair ; other electrons	correctly shown ;			[2]	
	(b) (i	) <u>car</u>	<u>bon dioxide</u> ;				[1]	
	(ii	(ii) $2PbO + C \rightarrow 2Pb + CO_2$ ;; (correct formula then look for balance)						
	(iii	<ul> <li>(iii) (no reaction)         idea that carbon is less reactive than potassium ;         and so cannot remove / combine with the oxygen ;         (allow 1 mark for saying potassium is too reactive)</li> </ul>					[2]	
							[Total: 12]	
2	(a)							
2	(a)	a	mmeter	current / amps	]			
			A <sub>1</sub>	0.7	-			
			A <sub>2</sub>	0.3	-			
			A <sub>3</sub>	0.4				
			A <sub>4</sub>	0.3				
					-		[2]	
	(b) (i		s – no mark) aight line on gr	aph so current is direct	y proportional	to voltage ;	[1]	
	(ii		mps ; planation e.g. 1	3 × 0.15 A <b>or</b> 2 × 1 A ;			[2]	

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			IGCSE – October/November 2010	0654	32		
	(c) (i)	only	e magnetic field stronger ; magnetic while the current is on ; nat magnetic field can be reversed ;		[max 2]		
	(ii)	V <sub>s</sub> =	$V_p \times N_s / N_p = 200 \times 1000 / 10000 = 20 V$ ;		[1]		
					[Total: 8]		
3	trai cor wa gas	nspira ndens ter va s char	<u>pour</u> lost from plant's leaves ; ition ; ation ; pour cooled ; nged to liquid / water vapour changed to water (drop articles and (kinetic) energy ;	lets) ;	[max 4]		
		·					
	(b) (i)		of turgor (in leaf cells)/cells become flaccid ; ause water lost from the cells ;		[2]		
	(ii)	(sup	ported by) xylem / lignin ; (reject if reason is that xyle	em contains water)	[1]		
	(iii)	oute	roximately similar shaped cell, with all parts shown ; er cell wall slightly caved in ;				
			uole much smaller ; plasm pulled away from cell wall ;		[max 3]		
					[Total: 10]		
4	(a) (i)	sour	nd/ultrasound;		[1]		
	(ii)	infra	i-red ;				
	(iii)	gam	ma ;		[1]		
	(b) (i)	num	ber of, waves / oscillations, per, second / unit time ;		[1]		
	(ii)	•	– no mark) imum human frequency about 20 000 Hz ;		[1]		
	(iii) v = f × λ ; wavelength = 330 / 50 000 ;						
		= 0.0	0066 m ;		[3]		
					[Total: 8]		

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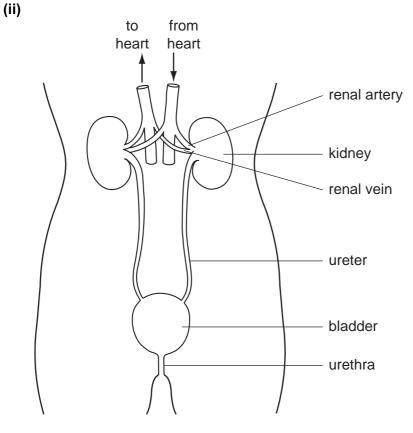
	Pa	ge 4	Mark Scheme: Teachers' version	Syllabus	Paper
5	(a)		anisms will pass through the filter / owtte ; ea that chlorination and distillation <u>kill</u> microorganis	0654 ms whereas filtra	32 [1] Ition
	(b)	by reflect transpare	s are scattered ; tion from dispersed material ; ency requires most rays to be undeviated / owtte ; points could come from diagram of scattered rays)		[max 2]
	(c)	<b>(i)</b> 0.05	;		[1]
			vant working e.g. 0.05 × 12.5 / 1000; 000625 ;		[2]
		acid	ence of the use of molar ratio. e.g. 2 mols of alkali n / acid to alkali = 1:2 / 0.000625 ÷ 2 ; 00031(25) ;	eutralise one mol	e of [2]
					[Total: 8]
6	(a)	A written	anywhere between 0 and 13 seconds ;		[1]
	(b)		ler graph / other working ; s × 8 = 51.2 m ;		[2]
	(c)	KE = ½ r	n speed = 16 m / s nv <sup>2</sup> ; 300 × 16 × 16 = 102 400 J ;		[3]
	(d)		um is directly proportional to $v / momentum = mv$ ; ectly proportional to $v^2 / explained$ using numbers;		[2]
					[Total: 8]
7	(a)	different	; y glands ; types of teeth ; ear flaps ;		[max 2]
	(b)	dilate / ge so more	g blood to skin surface ;		[max 3]

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 (c) (i) sensed by pancreas; pancreas secretes insulin; insulin affects <u>liver</u>; causes <u>liver</u> to take glucose from blood; (liver) converts glucose to glycogen;

[max 3]



one renal artery and vein drawn and labelled ; two renal arteries and veins drawn ; (at least one) ureter drawn and labelled ; urethra drawn and labelled ;

[4]

[Total: 12]

8	(a)		king ; ours ; (allow leeway if carefully shown on graph)	[2]
	(b)	(i)	causes, atoms/molecules, to lose electrons/to become ions ;	[1]
		(ii)	alpha is less penetrating and is stopped by, the air / clothes / skin ; alpha is more ionising and so causes more damage when close to cells ;	[2]

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	Page 6			Mark Scheme: Tea	achers' version		Syllabus	Paper
				GCSE – October/	November 2010		0654	32
	(c)	(i)	fusion is join of nuclei ;	ing and fission is s	plitting (of atoms /	nuclei) ;		[2]
		(ii)		ks / ref. Chernobyl ; ations in, local peop of waste ;				
			needs to be	stored safely for a	long period ;			[max 2]
								[Total: 9]
9	(a)							
	. ,			Table 9.1				
		elei	ment name	protons	neutrons			
				1		1		

 (oxygen)
 8
 8

 phosphorus
 (15)
 (16)
 ;; (1 mark per row)

 [2]

 (b) (i)
 silicon / Si ; periodic pattern refers to (repeating) patterns (of properties) across periods / every eight elements / owtte ; silicon is eight elements further on (in Periodic Table) from element No. 6 / carbon ;
 [max 2]

 (ii) carbon has a giant structure <u>and</u> nitrogen is simple molecular; much energy needed to, melt/break down, giants/converse for molecular; because strong bonds must be broken/converse for molecular; because many bonds must be broken/converse for molecular; [max 3]

(c) (i) Н Н = CC= Н H;; (2C and 4H bonded and double bond shown) [2] (ii) (catalytic/thermal) cracking; fractions are boiled / vaporised / heated ; passed over (hot) catalyst/subjected to very high temperature and pressure ; [3] (iii) double bonds become single ; single bonds form between molecules to form a long chain ; [2] (marks can be obtained by clear diagrams)

[Total: 14]

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	IGCSE – October/November 2010	0654	32
<b>10 (a) (i)</b> 23			[1]
<b>(ii)</b> 46			[1]
<b>(iii)</b> nuc	leus ;		[1]
( <b>b) (i)</b> ovid	luct / fallopian tube ;		[1]
(ii) ute	rus/womb;		[1]
	es / contains, amniotic fluid ; / supports, embryo ;		[2]
individu passing repeate most/a	al with the mutation is more likely to survive ; al with the mutation is more likely to reproduce ; mutation on to its offspring ; d over many generations ; l, of population have the mutation ; characteristic that the mutated gene produces ;		[max 4]
			[Total: 11]

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