## UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS

**International General Certificate of Secondary Education** 

## MARK SCHEME for the May/June 2010 question paper for the guidance of teachers

## 0654 CO-ORDINATED SCIENCES

0654/21

Paper 21 (Core Theory), maximum raw mark 100

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

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Page 2		Mark Scheme: Teach	Syllabus	Paper	
		IGCSE – May/Jui	ne 2010	0654	21
	) brain lab	pelled ;			[
	receptor nerves;				
	effectors				[-
(c	:)				
•	,		sexual	asexual	
			reproduction	reproduction	
		lves gametes.	✓	,	
		only one parent.		<b>√</b>	
	The offsp	oring are genetically identical.		✓	[1
					;;; [
					[Total:
(a	) (i) B –	metamorphic;			
•		igneous;			[3
	(ii) refe	rence to the heating (of rock A)	) (by rock <b>C)</b> ;		[
/h	) (i) oorb	on diavide and produced :			
(b	sho	oon dioxide gas produced ; ws the soil contains a carbonat			
	lime	estone is (mainly) composed of	(calcium) carbonate;		[max :
	(ii) amr	monia ;			[
	(iii) amr	monium ;			[
					[Total:
(a	) (nower=	=) work/time ;			
(-		,			_
	= 12000	/60 = 200 (W) ;			[.
(b		e) distance/time ;			<u>-</u>
	= 600/2	= 300 (m/s);			[:
(c		symbols correct;			
		cells displayed ;			Г

[3]

[1]

all symbols connected in series;

(ii) 6(V);

	Pa	ge 3		Mark Scheme: Teachers' version	Syllabus	Paper		
				IGCSE – May/June 2010	0654	21		
	(d)	(de = 5	nsity = / 10 =	=) mass/volume ; : 0.5 (kg/dm³) ;		[2]		
	(e)	two	strai		[1]			
						[Total: 11]		
4	(a)	(i)	prote	eins ;		[1]		
		(ii)	mon	omer(s);		[1]		
		(iii)	gluc	ose;		[1]		
	(b)	(i)	furni fuel	ling materials ; ture ;		[max 2]		
		(ii)	loss threa	of habitat ; at to biodiversity ; at to (new) chemical resources ; r reasonable ;		[max 1]		
	(c)	сус	le rep	oftens/melts, then hardens (on cooling); beats on further heating; e resin does not soften/it chars;		[3] <b>[Total: 9]</b>		
5	(a)	(i)	<b>C</b> ar	nd <b>D</b> ;		[1]		
		(ii)	<b>A</b> ar	nd <b>D</b> ;		[1]		
	(b)			I/tricuspid/atrioventricular) valve is (pushed) shut ; pressure of blood causes this ;		[2]		
	(c)	(i)	haer	moglobin ;		[1]		
		(ii)	iron	;		[1]		
		(iii)		espiration/to combine with glucose; lease energy/to provide energy;		[2]		
	(d)	d) white blood cells fight disease ; bacteria/viruses/pathogens ;						

	Page 4			Mark Scheme: Teachers' version	Paper					
				IGCSE – May/June 2010	0654	21				
	(e)	(i)	[max 2]							
		(ii) liver;								
						Total: 13				
6	(a)	(i)	[1]							
		(ii)	ed [2]							
		reference to pH 7;  (iii) tell what ph it is / ability to tell how acidic a solution is rather than simply acidic;								
	(b)	(i) magnesium chloride ; calcium sulfate ;								
		(ii) boiling/ion exchange/sodium carbonate/bath salts/washing soda;								
	(c)	element contains only one type of atom; compound contains different atoms (bonded); reference to diagram e.g. $H_2$ and $O_2$ diagrams show only one size of circle;								
						[Total: 9]				
7	(a)			r, is a poor conductor/gap prevents conduction;						
		foam, stops <u>convection</u> of air/traps air; <u>radiation</u> reflected by shiny surfaces/foil/metal;								
	(b)	water can conduct electricity/danger of electrocution/electric shock;								
	(c)	(i) 60W;								
		(ii)	[1]							
		(iii) input – electrical; output – light and heat;;								
		(iv) named part of spectrum ; use ;								

	Pa	ige 5	j	Mark Scheme: Teachers' version Syllabus						
				IGCSE – May/June 2010	0654	21				
	(d)	(i)	curre prod this	[3]						
		(ii)	(no – no mark) aluminium is not magnetic/not attracted to electromagnet;							
		(iii)	(yes still a	[1]						
		(iv)	v) more coils/bigger voltage/bigger core ;							
8	(a)	Gei		[1]						
	(b)	(i)	can	remove electrons from atoms/can form ions;		[1]				
		(ii) alpha radiation is more ionising than gamma;								
		more likely to be absorbed by body/cells; will cause more damage internally;								
	(c)	nuc	nuclei split ;							
	(d)	) protective clothing described / radiation badges to monitor exposure / lead shielding to stop radiation ;								
9	(a)	(i)	root	/root hair ;		[1]				
		(ii)	(ii) nitrogen gas is, unreactive/inert;							
		(iii)	to m	ake protein/amino acids;		[1]				
		(iv)	more deta	tage of something in the soil ; e proteins can be made (so more growth) ; iil, e.g. more cells/more cytoplasm ; ect ref. to function of P or K ;		[max2]				
		(v) manure contains plant and animal waste e.g. proteins/urea; which needs to be, broken down decomposed; to produce, ammonia/nitrates/something that can be used by plants; NPK has ions that can be absorbed immediately;								

Page 6				Mark Scheme: Teachers' version							Syllabus	Paper		
						IGC	SE –	May/J	une 20	)10			0654	21
	(b)	(i)	osm	osis	;									[1]
		(ii)	carb oxyg		lioxide	<b>;</b>								[2]
		(iii)	palis	ade	/mes	ophyll								[1]
		(iv)	by d	iffus	stoma ion ; ation ;	ta ;								
					tion ;									[max 2]
														[Total: 13]
10	(a)						_							
						\	/							
										<b>X</b> ;				
			Y	<b>'</b> ;										
												J		
			!									;;		[2]
	(b)	(i)					ensity /	lower	reactiv	vity / fo	rms co	oloured	compounds	
			act a	is a	cataly	st;								[1]
		(ii)		heated with carbon/carbon monoxide/other reducer which works e.g. hydrogen/ (named) more reactive metal;										
								vith/re	acts w	ith oxy	gen ;			[2]
	(c)	(i)				e/CO <sub>2</sub> )/H <sub>2</sub> O								[2]
		(ii)		produces hotter flame/reaches a higher temperature ; reasonable reference to air behaving as 'dilute' oxygen ;										
				reference to higher temperatures needed to melt metals; [max 1]										

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[Total: 8]