## UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS

International General Certificate of Secondary Education

## MARK SCHEME for the May/June 2006 question paper

## **0653 COMBINED SCIENCE**

0653/03

Paper 3, maximum raw mark 80

These mark schemes are published as an aid to teachers and students, to indicate the requirements of the examination. They show the basis on which Examiners were initially instructed to award marks. They do not indicate the details of the discussions that took place at an Examiners' meeting before marking began. Any substantial changes to the mark scheme that arose from these discussions will be recorded in the published *Report on the Examination*.

All Examiners are instructed that alternative correct answers and unexpected approaches in candidates' scripts must be given marks that fairly reflect the relevant knowledge and skills demonstrated.

Mark schemes must be read in conjunction with the question papers and the *Report on the Examination*.

The minimum marks in these components needed for various grades were previously published with these mark schemes, but are now instead included in the Report on the Examination for this session.

• CIE will not enter into discussion or correspondence in connection with these mark schemes.

CIE is publishing the mark schemes for the May/June 2006 question papers for most IGCSE and GCE Advanced Level and Advanced Subsidiary Level syllabuses and some Ordinary Level syllabuses.



| Page 1     | Mark Scheme<br>IGCSE – May/June 2006  | Syllabus Pap<br>0653 03 |           |  |
|------------|---|-------------------------|-----------|--|
| 4 (c)      |   | 0000                    |           |  |
| 1 (a)      | takes up the shape of its<br>container and has a<br>constant volume   |                         |           |  |
|            | expands the most when heated  | gas                     |           |  |
|            | particles are only very   | liquid                  |           |  |
|            | weakly attracted to each<br>other   | solid                   |           |  |
|            | particles have very strong<br>forces of attraction<br>between them  |                         | [3]       |  |
| <i>4</i> . | <i></i>   |                         | [3]       |  |
| paı<br>col | rticles moving randomly;<br>rticles collide with the walls of container;<br>llide more often;<br>greater force exerted on walls of container; |                         |           |  |
|            | rease in pressure;  |                         | [3]       |  |
|            |   | г                       | Total: 6] |  |

[Total: 6]

| Page | e 2    | Mark Scheme  | Syllabus | Paper     |
|------|--------|--|----------|-----------|
|      |        | IGCSE – May/June 2006  | 0653     | 03        |
| 2 (a |        | iety of a gene ;<br>t only has an effect when dominant one not present ;   |          | [2]       |
| (b   | o) (i) | aa ;   |          | [1]       |
|      | (ii)   | parents are Aa and Aa ;<br>gametes from each parent are A and a ;<br>offspring are AA, Aa (twice) and aa ;                 |          | [3]       |
| (c   | me     | ods containing) proteins ;<br>at / fish / cheese / other e.g. of high protein food ;<br>teins are made up of amino acids ; |          | [2 max]   |
|      |        |  | [        | Fotal: 8] |

| Page 3                                    | Mark Scheme  | Syllabus | Paper     |
|---|--|----------|-----------|
|   | IGCSE – May/June 2006  | 0653     | 03        |
| <b>3 (a)</b> (C <i>l</i> )<br>(Br)<br>(I) | gas<br>liquid<br>solid;  |          | [1]       |
|   | our shared pairs;<br>symbols correctly shown for each atom;                  |          | [2]       |
| (ii) 4                                    | $\mathbf{4Cl}_2 + \mathbf{CH}_4 \rightarrow \mathbf{CCl}_4 + \mathbf{4HCl};$ |          | [1]       |
| • • •                                     | fluorine)<br>reactivity decreases down Group 7 / owtte;                      |          | [1]       |
|   | (nucleus of) $Cl - 37$ contains more neutrons than $Cl - 3$<br>2 more;       | 5;       | [2]       |
|   | (12 x 1) + (4 x 35.5);<br>= 154;   |          | [2]       |
|   |  | I        | Total: 9] |

| Page 4    | Mark Scheme                                   | Syllabus | Paper    |
|-----------|---|----------|----------|
|           | IGCSE – May/June 2006                         | 0653     | 03       |
| 4 (a) (i) |   |          |          |
|           | reference to products;                        |          | [2]      |
| (ii)      | relatively short half life but not too short; |          | [1]      |
| (iii)     | 3 half lives;                                 |          |          |
|           | so 0.2 g;                                     |          | [2]      |
| (b) (i)   |   |          |          |
|           | this reduces energy losses;                   |          | [2]      |
| (ii)      | resistance = voltage/current;                 |          |          |
|           | = 22 ohms                                     |          | [2]      |
|           |   | ר]       | otal: 9] |

| Page  | 5            | Mark Scheme  | Syllabus  | Paper   |
|-------|--------------|--|-----------|---------|
|       |              | IGCSE – May/June 2006  | 0653      | 03      |
| 5 (a) | cell<br>chlo | leus A<br>wall C<br>proplast none (allow A)<br>surface membrane B  |           |         |
|       | thre         | correct two marks<br>e correct one mark<br>or two correct no marks   |           | [2]     |
| (b)   | ) (i)        | ref to water molecules ;<br>water passes from beaker through ppm ;<br>because more water outside than inside / correct ref to g<br>starch (molecules) cannot pass through the membrane ; | radient ; | [3 max] |
|       | (ii)         | add iodine (solution) ;<br>orange / brown / yellow ;   |           | [2]     |
| (c)   |              | root hair ;<br>oss cells in root ;   |           | [2]     |

[Total: 9]

| Page 6 |   | 6    | Mark Scheme Syllabus   |      | Paper   |
|--------|---|------|--|------|---------|
|        |   |      | IGCSE – May/June 2006 06   | 53   | 03      |
| 6      | 6 (a) carbon dioxide / carbon monoxide / carbon / soot / water; (any two) |      | )  | [1]  |         |
|        | (b)   | (i)  | ( <b>B</b> is methane) methane molecules have five atoms (bonded) / is $CH_4$ ;  |      | [1]     |
|        |   | (ii) | (addition) polymerisation;<br>many small molecules / monomers join to form a long chain;                                 |      | [2]     |
|        | (c)   | (i)  | heated / vaporised / boiled;<br>passed over catalyst;  |      | [2]     |
|        |   | (ii) | suggests that only single bonds between carbon atoms / satur if double bonds present bromine would have been decolourise | -    | [2]     |
|        |   |      |  | [Tot | tal: 8] |

| Р | Page 7   |       | Mark Scheme   | Syllabus       | Paper     |
|---|--|-------|---|----------------|-----------|
|   |  |       | IGCSE – May/June 2006   | 0653           | 03        |
| 7 |  | roots | s / plants, stop rain hitting the ground (hard) ;<br>hold soil in place ;<br>cing stops water running down slopes ;     |                | [2 max]   |
|   | (b)  | colou | rless / green / small / no petals / dangling anthers / dar  | ngling stigmas | s; [1]    |
|   | <ul> <li>(c) (i) by diffusion ;</li> <li>through wall of small intestine ;</li> <li>ref. to villi ;</li> </ul> |       |   |                | [2 max]   |
|   | (  | S     | eancreas ;<br>ecretes insulin ;<br>auses, cells / liver, to take glucose from the blood ;                               |                | [3]       |
|   | (i   | , b   | is level moves away from norm ;<br>process initiated to bring it back ;<br>ake these points from a specific description |                | [2]       |
|   |  |       |   | ר]             | otal: 10] |

| Pag  | Page 8   |       | Mark Scheme Sy  |      | Paper     |
|------|--|-------|---|------|-----------|
|      |  |       | IGCSE – May/June 2006   | 0653 | 03        |
| 8 (a | bonded together;<br>element cannot be simplified and a compound can be broken into its e |       |   |      |           |
|      | (i   | , v   | <sup>-</sup> e <sup>3+</sup> ;<br>vorking refers to charge balance;<br>reject vague criss cross answers)  |      | [2]       |
| (ł   | , p  | oreve | er of zinc covers the steel / provides a barrier;<br>ents reaction between steel and oxygen and water;<br>v correct references to sacrificial protection) |      | [2]       |
| (0   | c) (   | (i) ⊦ | <b>1</b> ⁺;   |      | [1]       |
|      | (i   | i) r  | no more gas evolved;  |      | [1]       |
|      |  |       | prey crystals appear / magnesium reacts and dissolves<br>a metal displacement reaction occurs / or equation;  | ;    |           |
|      |  |       | because magnesium more reactive than zinc;  |      | [3]       |
|      |  |       |   | [To  | otal: 10] |

| Page 9             | Mark Scheme                             | Syllabus | Paper     |
|--------------------|---|----------|-----------|
|                    | IGCSE – May/June 2006                   | 0653     | 03        |
| <b>9 (a) (i)</b> a | cceleration;                            |          | [1]       |
| (ii) c             | onstant speed;                          |          | [1]       |
| (b) area           | under curve = ; (or other suitable)     |          |           |
|                    | 25 = 175 m;                             |          | [2]       |
| <b>(c)</b> equal   | and opposite/ balanced                  |          | [1]       |
| ( ) ( )            | orce = mass x acceleration;<br>: 120 N; |          | [2]       |
|                    | ower = work/time;<br>600 W;             |          | [2]       |
| <b>(e) Q</b> – n   | o mark                                  |          |           |
|                    | st CoG;                                 |          |           |
| base               | wider than <b>P</b> ;                   |          | [2]       |
|                    |   | ד]       | otal: 11] |