CAMBRIDGE

NOVEMBER 2001

INTERNATIONAL GCSE

MARK SCHEME

MAXIMUM MARK: 60

SYLLABUS/COMPONENT: 0620/6

CHEMSTRY (ALTERNATIVE TO PRACTICAL)



Page 1 of 2		Mark Scheme Syll	abus	Paper
		IGCSE Examinations – November 2001 06	620	6
1 (a)	Boxes completed to show beaker (1), pipette (1), burette (1)			
(b)	indicator (1), colour change (1)		[2]	
(c)	repeat the experiment / pH meter			[1] [Total 6]
2 (a)	cathode / negative			[1]
(b)	silver		[1]	
(c)	silver nitrate (1) solution (1) any silver salt not C <i>l</i> ⁻ , I ⁻			
(d) (i)	silver will not coat / stick or similar			[1]
(ii)	to give ev	ren coating / all of it gets coated		[1] [Total 6]
3 (a)	to increas	se ease of extraction / surface area, etc	[1]	
(b)	if hot yeast is killed		[1]	
(c)	spatula		[1]	
(d)	best temp	perature for yeast (1) too cool does not multiply / yeast is killed > 4	/ yeast is killed > 40°C(1)	
(e)	to preven to allow C	t air (oxygen) / bacteria entering CO ₂ to escape		[1] [1]
(f) (i)	3 - 4 days	s (1) + (1) for unit		[2]
(ii)	10 days			[1]
(iii)	yeast die: no sugar	s (1) / solution too concentrated re alcohol / orange juice all used up (1)	[2] (Total 12]

Page 2	of 2	Mark Scheme	Syllabus	Paper		
		IGCSE Examinations – November 2001	0620	6		
4 (a)	pops hydrogen			[1] [1]		
	Table of F All readin	Results gs correct (3) marks, (-1 for any incorrect) 25 28 26 33 26 37 25 39 25 42		[3]		
		25 45				
(b)	all points (-1 for any straight li	plotted correctly y incorrect) ne best fit		[3]		
(0)	tomporati	$r_{\rm r}$ from graph (+ 1%C)		[1]		
(C)	indication	on grid		[1]		
(d)	exotherm	ic		[1]		
(e) (i)	experime	nt 6		[1]		
(ii)	largest piece / greatest concentration Mg (1) \therefore more reaction / collisions with acid partic etc. (1)					
(f)	use a burette instead of m. cylinder / insulate / lag apparatus (1) more accurate / reduce heat losses (1) repeat (1), average (1)					
	same initi	al temperatures (1) :: easy comparison (1)		[max 2] [Total 17]		
5 (a) (i)	green (1)	precipitate (1)		[max 2]		
(b)	red / brow	vn (1) precipitate (1)		[2]		
(c)	green (1) brown (1)	precipitate (1)		[2] [1]		
(e)	ammonia			[1]		
(f)	ammoniu	m		[1]		
(g)	sulphate			[1]		
				[l otal 10]		
6 (a)	Universal Indicator solution / pH paper (1), read pH from chart (1) / use a pH meter (2)[max					
(b)	chromato apply cola	graphy (1) paper (1) a (1) separation with solvent (1)		[max 3]		
(c)	can open filled with	under water to collect gas in graduated tube / m. cylinder (water (1), syringe = 0 (would not work)	1)	[2]		
(d)	limewater			[1]		
	not lighte	d splint		[']		