

Candidate Forename						Candidate Surname				
Centre Number						Candidate Number				

**OXFORD CAMBRIDGE AND RSA EXAMINATIONS
GENERAL CERTIFICATE OF SECONDARY EDUCATION**

A217/02

TWENTY FIRST CENTURY SCIENCE

ADDITIONAL SCIENCE A

**Unit 3: Modules B6 C6 P6
(Higher Tier)**

TUESDAY 22 JUNE 2010: Morning

DURATION: 40 minutes

SUITABLE FOR VISUALLY IMPAIRED CANDIDATES

Candidates answer on the Question Paper

A calculator may be used for this paper

OCR SUPPLIED MATERIALS:

None

OTHER MATERIALS REQUIRED:

Pencil

Ruler (cm/mm)

READ INSTRUCTIONS OVERLEAF

INSTRUCTIONS TO CANDIDATES

- Write your name clearly in capital letters, your Centre Number and Candidate Number in the boxes on the first page.
- Use black ink. Pencil may be used for graphs and diagrams only.
- Read each question carefully and make sure that you know what you have to do before starting your answer.
- Answer ALL the questions.
- Write your answer to each question in the space provided. Additional paper may be used if necessary but you must clearly show your Candidate Number, Centre Number and question number(s).

INFORMATION FOR CANDIDATES

- The number of marks is given in brackets [] at the end of each question or part question.
- The total number of marks for this paper is 42.
- A list of physics equations is printed on pages 4–5.
- The Periodic Table is printed on the back page.

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TWENTY FIRST CENTURY SCIENCE EQUATIONS

USEFUL RELATIONSHIPS

EXPLAINING MOTION

$$\text{speed} = \frac{\text{distance travelled}}{\text{time taken}}$$

$$\text{momentum} = \text{mass} \times \text{velocity}$$

$$\begin{aligned}\text{change of momentum} \\ = \text{resultant force} \times \text{time for which it acts}\end{aligned}$$

$$\begin{aligned}\text{work done by a force} \\ = \text{force} \times \text{distance moved by the force}\end{aligned}$$

$$\text{change in energy} = \text{work done}$$

$$\text{change in GPE} = \text{weight} \times \text{vertical height difference}$$

$$\text{kinetic energy} = \frac{1}{2} \times \text{mass} \times [\text{velocity}]^2$$

ELECTRIC CIRCUITS

$$\text{resistance} = \frac{\text{voltage}}{\text{current}}$$

$$\frac{\text{voltage across primary coil}}{\text{voltage across secondary coil}} = \frac{\text{number of turns in primary coil}}{\text{number of turns in secondary coil}}$$

$$\text{energy transferred} = \text{power} \times \text{time}$$

$$\text{power} = \text{potential difference} \times \text{current}$$

$$\text{efficiency} = \frac{\text{energy usefully transferred}}{\text{total energy supplied}} \times 100\%$$

THE WAVE MODEL OF RADIATION

$$\text{wave speed} = \text{frequency} \times \text{wavelength}$$

Answer ALL the questions.

- 1 Ted shouts, making a sound wave of frequency 1500 Hz with a wavelength of 0.2 m.

- (a) What is the speed of the wave?

Put a **ring** around the correct answer.

30 m/s 300 m/s 3000 m/s 30 000 m/s

[1]

- (b) Draw straight lines to link each WAVE PROPERTY to its MEANING.

WAVE PROPERTY

intensity

MEANING

waves per second

amplitude

energy per second

frequency

size of disturbance

wavelength

distance between crests

[2]

(c) Complete the sentences below using words from this list.

INCREASES **DECREASES** **STAYS THE SAME**

The sound wave travels away from Ted.

Its amplitude _____ as it travels away.

Its speed _____ as it travels away.

[1]

[Total: 4]

2 Jane uses her mobile phone to talk to Mike.

(a) Complete the sentences. Choose words from the list.

AMPLIFY ANALOGUE DIGITAL MORSE

MODULATE NOISE PULSES SOUND TRANSMIT

Jane speaks into the microphone of her phone.

The microphone output is converted into a series of pulses, called a _____ code.

These pulses _____ the amplitude of the waves which leave the phone.

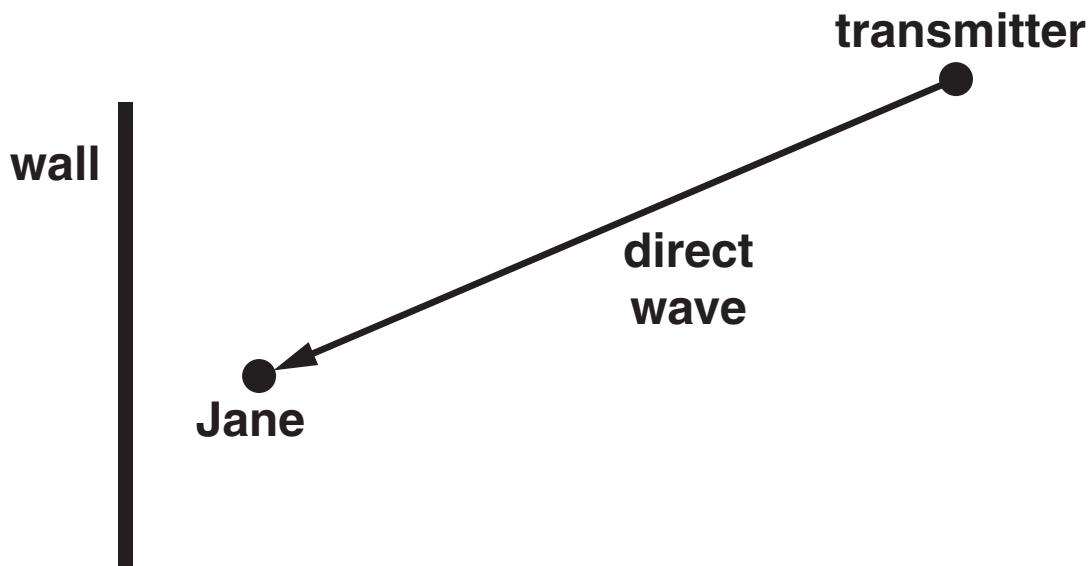
**Waves that her phone receives pick up
_____ on the way from the transmitter.**

This can be removed to restore the correct pattern of wave _____ . [2]

- (b) Jane finds that mobile phone reception is very bad when she stands close to a wall.**

She thinks that this is because waves from the transmitter reflect off the wall.

- (i) The diagram shows the direction of a wave travelling directly from the transmitter to Jane.**



Another wave reaches Jane from the transmitter.

It reflects off the wall before it reaches her.

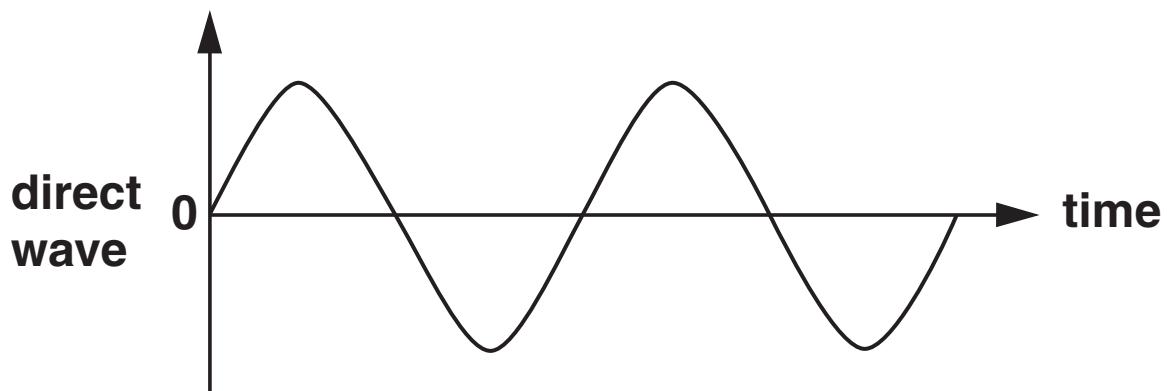
Draw this wave on the diagram.

[1]

- (ii) Jane's phone reception is bad because of the reflected wave.

The diagram shows the direct wave.

Complete the diagram to show the reflected wave



[1]

- (iii) Jane finds that she can improve the phone reception by choosing carefully where she stands near the wall.

Put ticks (\checkmark) in the boxes next to the TWO statements that explain this.

The wall attracts waves from the transmitter.

Waves which arrive in step add and reinforce each other.

The direct wave absorbs the reflected wave when they overlap.

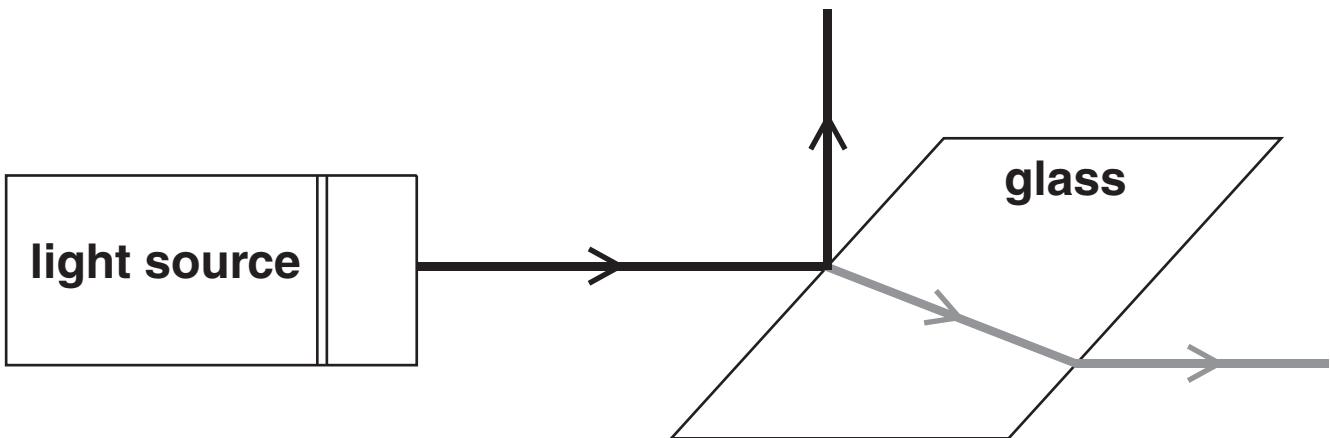
The amplitude of the reflected wave increases as it reflects off the wall.

There is an interference pattern where the direct and reflected waves overlap.

[1]

[Total: 5]

3 Fiona shines a beam of light onto a block of glass.



- (a) Draw straight lines to link the START of each sentence to its best ENDING.

START

The light does not
change speed ...

ENDING

... as it enters and
leaves the glass.

The light slows
down ...

... as it reflects off
the glass.

The light speeds
up ...

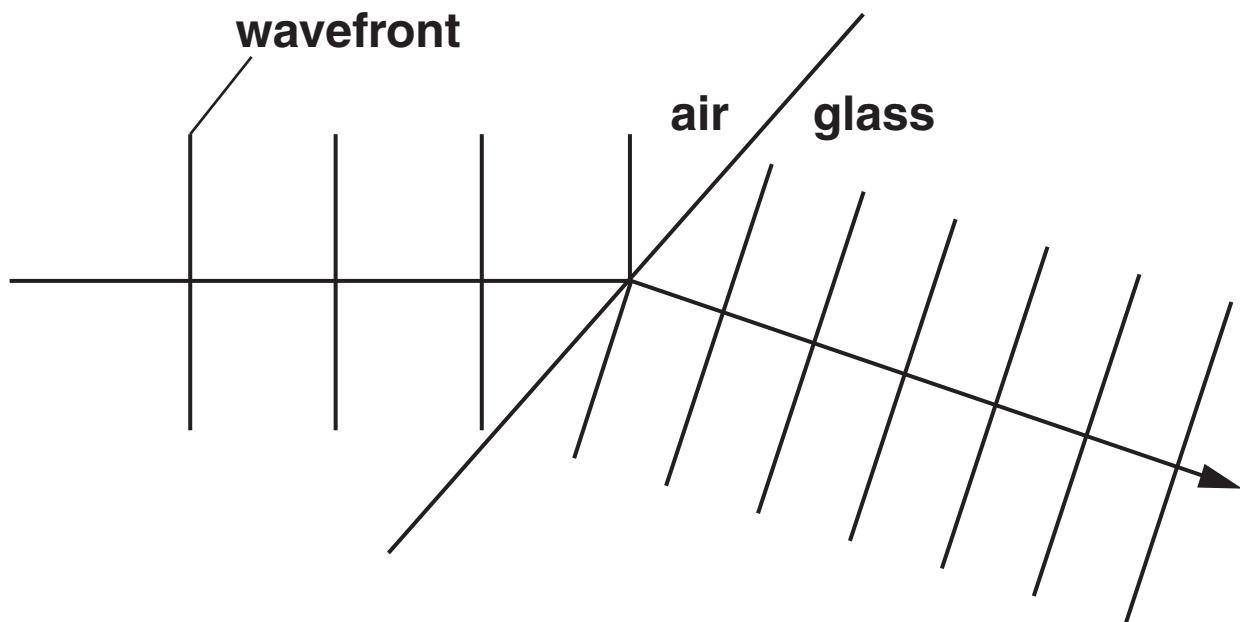
... as it enters the
glass.

The light refracts ...

... as it leaves the
glass.

[2]

(b) The diagram shows what happens to wavefronts of the light as it enters the glass.



Explain the change in spacing of the wavefronts as the light enters the glass.

Your answer should refer to the wavelength, frequency and speed of the wave.

[2]

(c) The light is an electromagnetic wave.

Here are some electromagnetic waves in the spectrum, in order of frequency.

Complete the table by writing in the missing parts of the spectrum shown opposite. [1]

[Total: 5]

	microwaves	infrared	ultraviolet	X-rays
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4 Daisy chooses an acid to clean metals.

Acids can be bought with different levels of purity.

Some are very pure. Others still contain impurities.

- (a) Suggest and explain why she chooses an impure acid to clean metals.**

[2]

- (b) Daisy wants to find the pH of the acid before she uses it.**

Give TWO ways that she could do this.

For each method

- state what she will use**
- describe what she will see.**

method one _____

method two _____

[2]

[Total: 4]

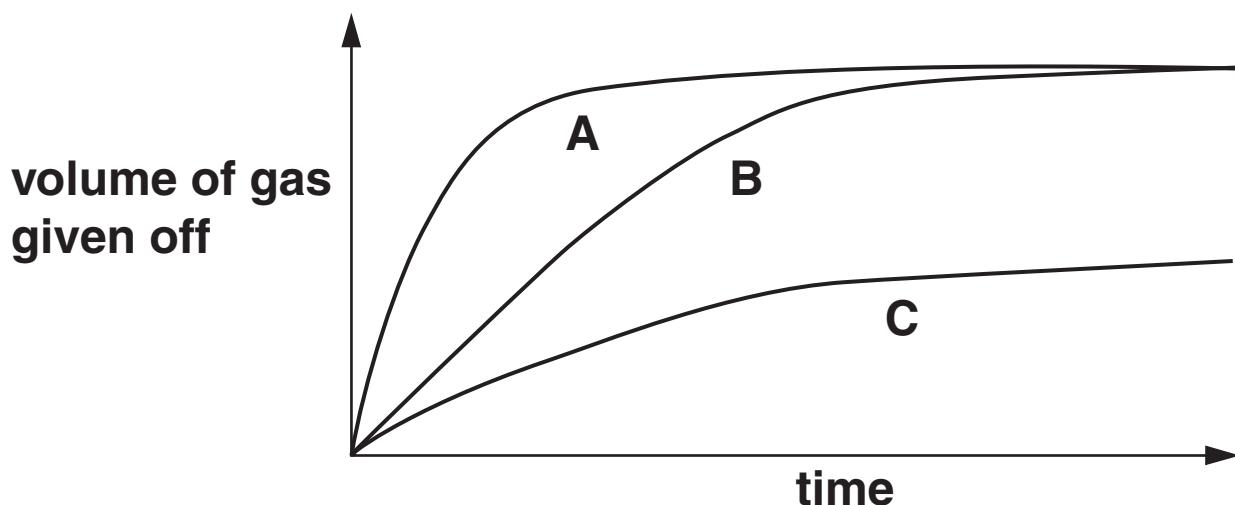
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- 5 Arthur, Beth and Chet measure how fast a gas is given off when they react acid with an excess of marble chips.

They each carry out two different experiments but they all use the same size of marble chips.

The experiments that they carry out are shown in the table opposite.

They plot their results.



Complete the table with A, B or C to show the shape of the graph that would be expected for each experiment.

STUDENT	EXPERIMENT 1	EXPERIMENT 2
Arthur		
Beth		
Chet		

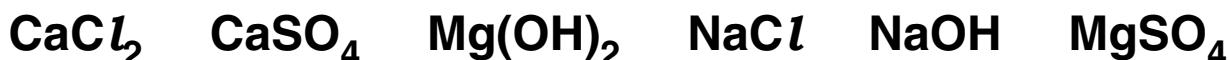
[4]

[Total: 4]

STUDENT	EXPERIMENT	VOLUME OF ACID IN cm ³	RELATIVE CONCENTRATION OF ACID	TEMPERATURE IN °C
Arthur	1	50	1.0	20
Arthur	2	50	1.0	30
Beth	1	50	1.0	20
Beth	2	25	1.0	20
Chet	1	50	1.0	20
Chet	2	25	2.0	20

6 Acids react with alkalis.

(a) Put **ring** around the **TWO** alkalis in this list.



[1]

(b) Nitric acid reacts with sodium carbonate.

What substances are formed?

[2]

(c) Acids also react with metal oxides.

When hydrochloric acid reacts with copper oxide, CuO , it makes copper chloride, CuCl_2 , and another substance.

Write a balanced symbol equation for this reaction.



[3]

[Total: 6]

- 7 Amelia is learning the words for her part in the school play.
- (a) What is happening to the information in Amelia's memory as she rehearses the play?
- Put a **ring** around each of the **TWO** words which best describe what is happening.
- DUPLICATION LOSS RETRIEVAL
STORAGE TRANSMISSION
- [1]
- (b) The sentences describe what happens in her brain.
- Complete the sentences below using the **BEST** words from this list.
- AXON BILLIONS EXPERIENCE HUNDREDS
RESPONSE THOUSANDS TRANSMIT
- Amelia's brain has _____ of neurons.
- They can connect together to make pathways.
- When she first reads her words, they are a new _____.
- This causes new pathways to form.
- When she repeats the words, these new pathways are more likely to _____ impulses.
- [2]

- (c) After a week of rehearsals, some of Amelia's words in the play are changed.

She has to learn new words.

How is this possible?

Put a tick (✓) in the box next to the BEST explanation.

Some neurons will stop transmitting impulses.

New pathways are created by moving neurons around.

New experiences remove existing pathways in the brain.

The variety of potential pathways makes it possible to adapt to new situations.

[1]

[Total: 4]

8 Keith is looking after a puppy.

(a) Keith buys frozen food for his puppy.

He heats the food in a microwave oven.

The puppy produces saliva when it smells the food heating up.

The microwave oven rings a bell when the heating is complete.

After some time the puppy produces saliva every time a bell rings.

(i) Draw a straight line from each PART OF THE REFLEX to the correct EVENT.

PART OF THE REFLEX

EVENT

puppy eats food

response

puppy barks

primary stimulus

puppy produces saliva

smell of food

secondary stimulus

sound of bell

- (ii) A CONDITIONED REFLEX has been created in the puppy.

Here are some statements about conditioned reflexes.

Put a tick (✓) in the box next to the correct statement.

The final response of a conditioned reflex ...

... has no connection at all to the secondary stimulus.

... has a direct connection to the primary stimulus.

... has no direct connection to the primary stimulus.

... has a voluntary connection to the secondary stimulus.

[1]

(b) The puppy tries to catch a fly but the fly moves.

The fly moving is an example of a SIMPLE REFLEX action.

Describe the advantages and disadvantages to the fly of using simple reflexes.

[3]

[Total: 6]

- 9 Diane puts her hand into a sewing box and stabs her finger on a needle.**
She pulls her hand away quickly.
The reaction involves a reflex arc.
- Some of the statements below can be arranged in sequence to describe what happens in part of a reflex arc.**

In each pair of statements, only one is true for a reflex arc.

- A this reaches the myelin sheath**
- B this reaches the synapse**
- C causing release of a chemical transmitter**
- D causing release of an electrical spark**
- E an electrical impulse travels up the sensory neuron**
- F a chemical impulse travels up the sensory neuron**
- G and binds to receptor molecules on the membrane of the motor neuron**
- H and enters the motor neuron through the membrane**
- I this jumps to the next neuron**
- J this diffuses to the next neuron**

Put a ring around the letter of the correct statement in each pair.

Arrange the correct statements in the right order for the sequence of events in a reflex arc.

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[4]

[Total: 4]

END OF QUESTION PAPER



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The Periodic Table of the Elements

1	2								3	4	5	6	7	0	4		
7 Li lithium 3	9 Be beryllium 4								11 B boron 5	12 C carbon 6	14 N nitrogen 7	16 O oxygen 8	19 F fluorine 9	20 Ne neon 10	He helium 2		
23 Na sodium 11	24 Mg magnesium 12								27 Al aluminum 13	28 Si silicon 14	31 P phosphorus 15	32 S sulfur 16	35.5 Cl chlorine 17	40 Ar argon 18			
39 K potassium 19	40 Ca calcium 20	45 Sc scandium 21	48 Ti titanium 22	51 V vanadium 23	52 Cr chromium 24	55 Mn manganese 25	56 Fe iron 26	59 Co cobalt 27	59 Ni nickel 28	63.5 Cu copper 29	65 Zn zinc 30	70 Ga gallium 31	73 Ge germanium 32	75 As arsenic 33	79 Se selenium 34	80 Br bromine 35	
85 Rb rubidium 37	88 Sr strontium 38	89 Y yttrium 39	91 Zr zirconium 40	93 Nb niobium 41	96 Mo molybdenum 42	[98] Tc technetium 43	101 Ru ruthenium 44	103 Rh rhodium 45	106 Pd palladium 46	108 Ag silver 47	112 Cd cadmium 48	115 In indium 49	119 Sn tin 50	122 Sb antimony 51	128 Te tellurium 52	127 I iodine 53	131 Xe xenon 54
133 Cs caesium 55	137 Ba barium 56	139 La* lanthanum 57	178 Hf hafnium 72	181 Ta tantalum 73	184 W tungsten 74	186 Re rhenium 75	190 Os osmium 76	192 Ir iridium 77	195 Pt platinum 78	197 Au gold 79	201 Hg mercury 80	207 Pb lead 82	209 Bi bismuth 83	[209] Po polonium 84	[210] At astatine 85	[222] Rn radon 86	
[223] Fr francium 87	[226] Ra radium 88	[227] Ac* actinium 89	[261] Rf rutherfordium 104	[262] Db dubnium 105	[266] Sg seaborgium 106	[264] Bh bohrium 107	[277] Hs meitnerium 108	[268] Mt mendelevium 109	[271] Ds darmstadtium 110	[272] Rg roentgenium 111							

Elements with atomic numbers 112-116 have been reported but not fully authenticated

* The lanthanoids (atomic numbers 58-71) and the actinoids (atomic numbers 90-103) have been omitted.

The relative atomic masses of copper and chlorine have not been rounded to the nearest whole number.