

Candidate forename						Candidate surname				
Centre number						Candidate number				

**OXFORD CAMBRIDGE AND RSA EXAMINATIONS
GENERAL CERTIFICATE OF SECONDARY EDUCATION**

A216/01

**TWENTY FIRST CENTURY SCIENCE
ADDITIONAL SCIENCE A**

Unit 2: Modules B5 C5 P5 (Foundation Tier)

MONDAY 24 JANUARY 2011: Afternoon

DURATION: 40 minutes

SUITABLE FOR VISUALLY IMPAIRED CANDIDATES

Candidates answer on the question paper.

A calculator may be used for this paper.

OCR SUPPLIED MATERIALS:

None

OTHER MATERIALS REQUIRED:

Pencil

Ruler (cm/mm)

READ INSTRUCTIONS OVERLEAF

INSTRUCTIONS TO CANDIDATES

- Write your name, centre number and candidate number in the boxes on the first page. Please write clearly and in capital letters.
- Use black ink. Pencil may be used for graphs and diagrams only.
- Read each question carefully. Make sure you know what you have to do before starting your answer.
- Write your answer to each question in the space provided. Additional paper may be used if necessary but you must clearly show your candidate number, centre number and question number(s).
- Answer **ALL** the questions.

INFORMATION FOR CANDIDATES

- The number of marks is given in brackets [] at the end of each question or part question.
- The total number of marks for this paper is **42**.
- A list of physics equations is printed on pages 4–5.
- The Periodic Table is provided.

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TWENTY FIRST CENTURY SCIENCE EQUATIONS

USEFUL RELATIONSHIPS

EXPLAINING MOTION

$$\text{speed} = \frac{\text{distance travelled}}{\text{time taken}}$$

$$\text{momentum} = \text{mass} \times \text{velocity}$$

$$\begin{aligned}\text{change of momentum} \\ = \text{resultant force} \times \text{time for which it acts}\end{aligned}$$

$$\begin{aligned}\text{work done by a force} \\ = \text{force} \times \text{distance moved by the force}\end{aligned}$$

$$\text{change in energy} = \text{work done}$$

$$\text{change in GPE} = \text{weight} \times \text{vertical height difference}$$

$$\text{kinetic energy} = \frac{1}{2} \times \text{mass} \times [\text{velocity}]^2$$

ELECTRIC CIRCUITS

$$\text{resistance} = \frac{\text{voltage}}{\text{current}}$$

$$\frac{\text{voltage across primary coil}}{\text{voltage across secondary coil}} = \frac{\text{number of turns in primary coil}}{\text{number of turns in secondary coil}}$$

$$\text{energy transferred} = \text{power} \times \text{time}$$

$$\text{power} = \text{potential difference} \times \text{current}$$

$$\text{efficiency} = \frac{\text{energy usefully transferred}}{\text{total energy supplied}} \times 100\%$$

THE WAVE MODEL OF RADIATION

$$\text{wave speed} = \text{frequency} \times \text{wavelength}$$

Answer ALL the questions.

1 Air is a mixture of different gases.

Each year we extract thousands of tonnes of gases from the air.

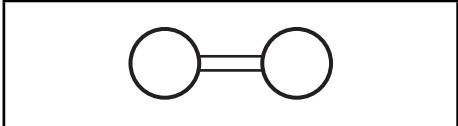
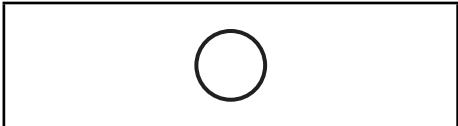
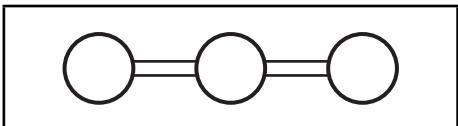
Many of these gases are very useful.

(a) Draw a straight line from the NAME of each gas to its FORMULA.

name	formula
argon	H_2O
carbon dioxide	Ar
oxygen	CO_2
water vapour	O_2

[2]

(b) Draw a straight line from the FORMULA of each gas to its STRUCTURE.

formula	structure
Ar	
CO ₂	
O ₂	

[2]

(c) Oxygen is a gas at room temperature.

What does this tell you about

- the size of the molecules
- the forces between the molecules?

[2]

[Total: 6]

2 We have extracted iron since the Iron Age.

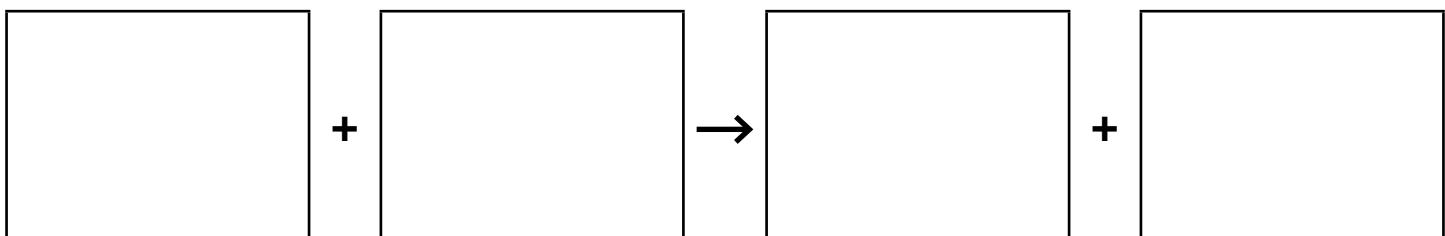
We still use the same method.

We extract the iron by heating iron oxide with carbon in a furnace.

Different reactions take place in the furnace.

In one reaction, carbon takes the oxygen away from iron oxide.

(a) Fill in the boxes to write a word equation for this reaction.



(b) Use words from this list to complete the sentences below.

combined

electrolysed

melted

oxidised

precipitated

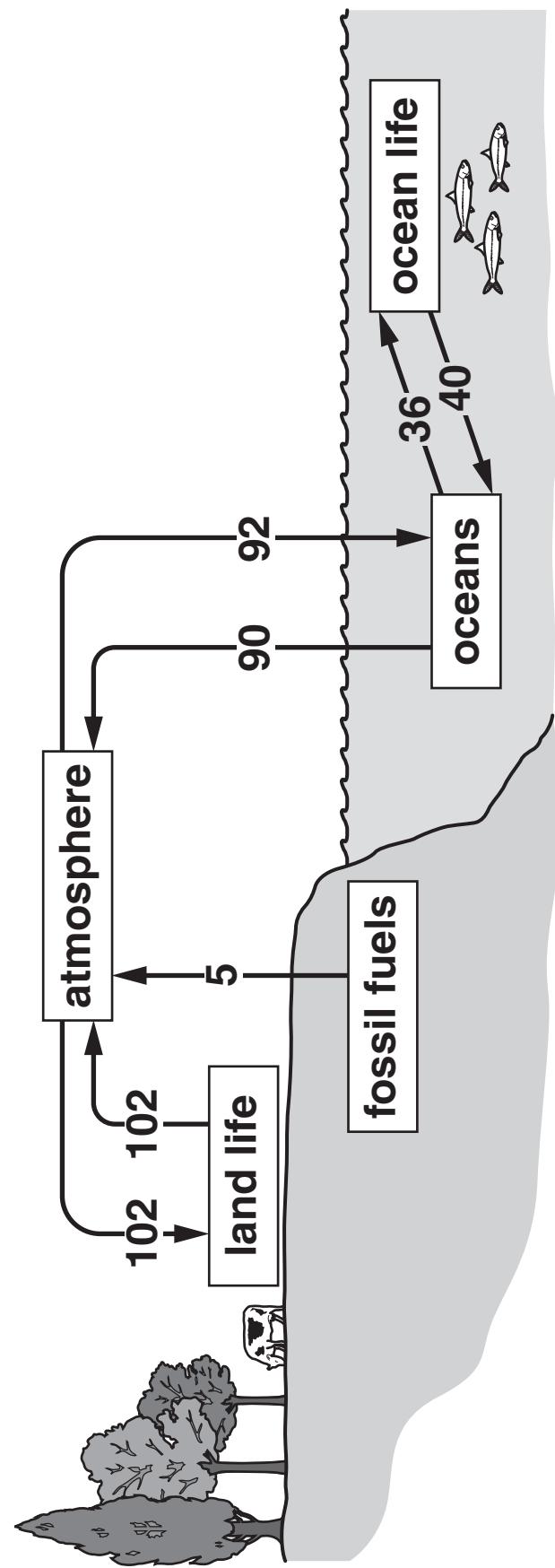
reduced

When carbon gains oxygen we say that the carbon has been _____.

When a metal oxide loses oxygen we say that the metal has been _____.

[1]

[Total: 3]



- 3 Scientists are worried about the increasing amounts of carbon in our atmosphere.**

Look at the diagram of the carbon cycle, on the facing page.

The numbers show how many gigatonnes of carbon move in each direction every year.

- (a) Put a ring around the BEST term to complete these sentences.**

**Carbon enters the atmosphere when fossil fuels
are BURNED / DISCOVERED / SOLD.**

**Most carbon in the atmosphere is in the form of
CARBON DIOXIDE / OZONE / SOOT.**

[2]

- (b) The amount of carbon in the atmosphere is increasing.**

By how many gigatonnes does it increase every year?

Put a ring around the correct answer.

2

3

5

90

102

[1]

- (c) Land life does not change the overall amount of carbon in the atmosphere.**

Explain how you can tell this from the diagram.

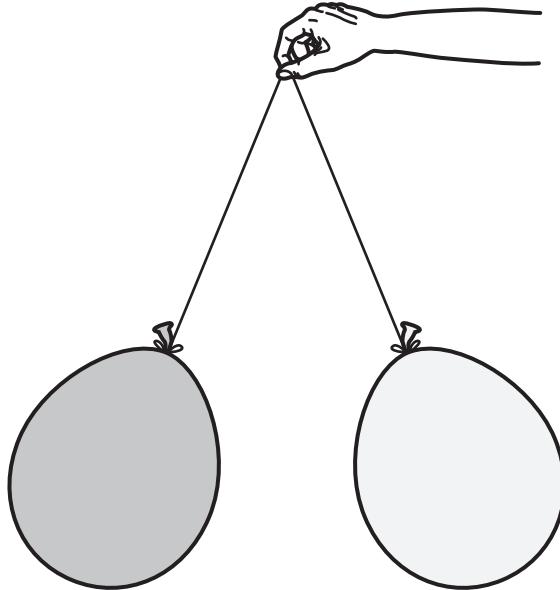
[2]

[Total: 5]

- 4 Zara does an experiment with a pair of balloons on strings.**

She rubs each balloon against her clothing.

Zara then holds the balloons up by their strings.

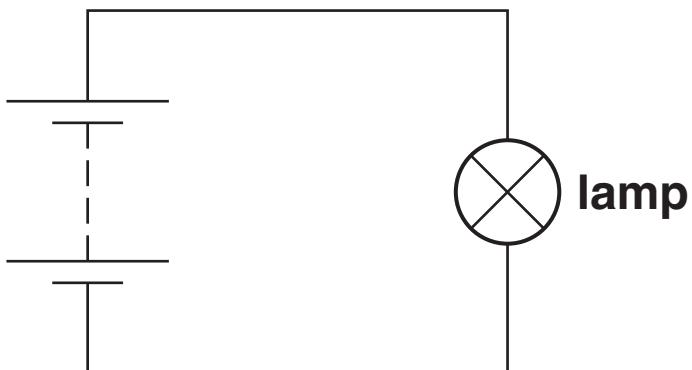


Explain why the balloons do not touch each other.

[3]

[Total: 3]

5 Charles puts this circuit together.



- (a) There is not enough current in the circuit to make the lamp glow brightly.

How can Charles make the lamp glow more brightly?

Put a tick (\checkmark) in the box next to the change that INCREASES the current.

Put a switch after the lamp.

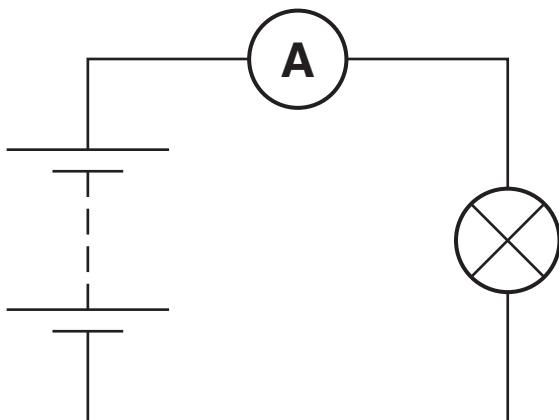
Use longer connecting wires.

Put a resistor before the lamp.

Increase the voltage of the battery.

[1]

(b) Charles adds an ammeter to measure the current in the lamp.



Complete each sentence by putting a **ring** around the correct option.

The ammeter reads 0.69

AMPS / VOLTS / WATTS.

The current in the lamp is due to a flow of

ATOMS / ELECTRONS / WATER.

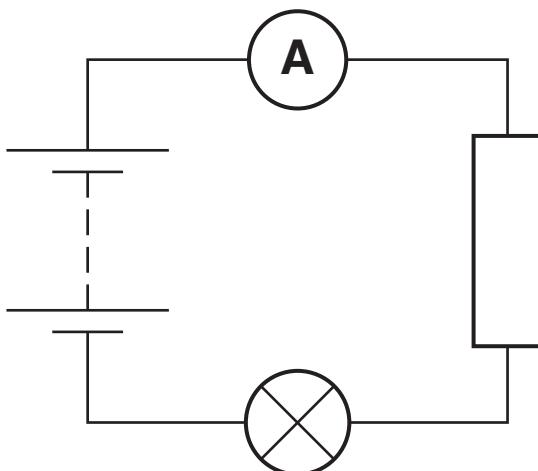
The lamp glows because its filament

COOLS DOWN / HEATS UP / REACTS.

[2]

[Total: 3]

6 Fleur assembles this circuit.



(a) The resistor, ammeter and lamp are in series.

All three components have the same current.

Put ticks () in the boxes next to the TWO correct statements.

More charge flows into a component than flows out of it.

Less charge flows into a component than flows out of it.

All parts of the circuit contain charges that are free to move.

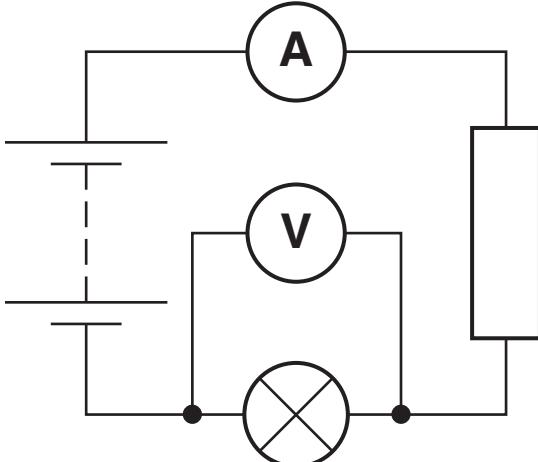
Only the connecting wires contain charges that are free to move.

The same amount of charge flows into a component as flows out of it.

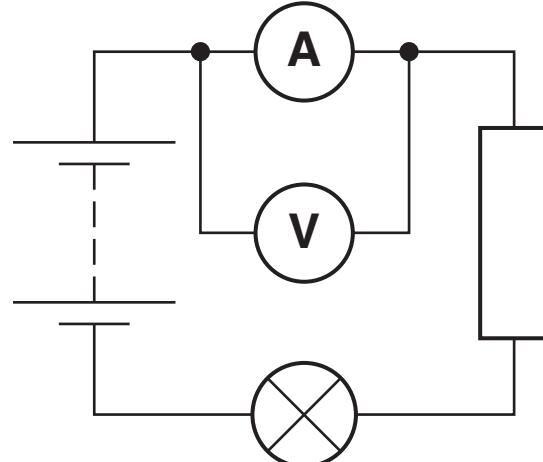
[2]

(b) Fleur wants to add a voltmeter to measure the potential difference across the resistor.

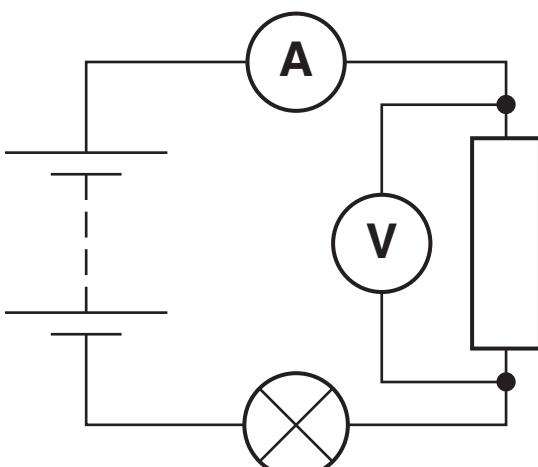
(i) Here are four ways of connecting the voltmeter.



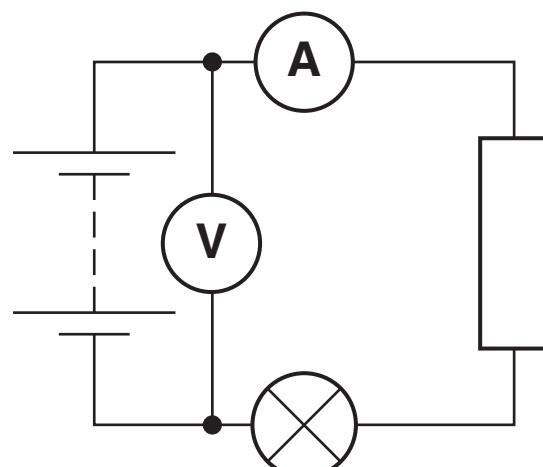
circuit A



circuit B



circuit C

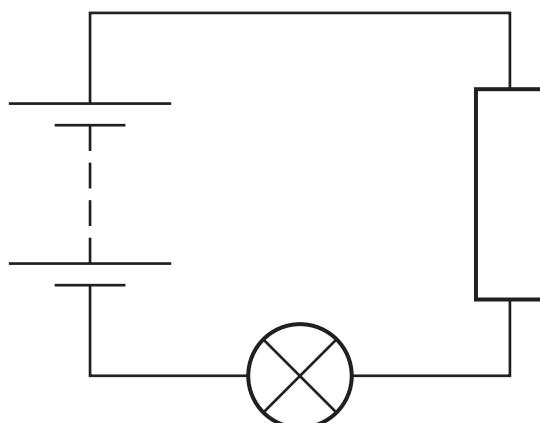


circuit D

Which circuit, A, B, C or D, shows the correct way of measuring the potential difference across the resistor?

answer _____ [1]

- (ii) Fleur finds that the potential difference across the resistor in the circuit below is 6V.



The potential difference across the battery is 9V.

What is the potential difference across the lamp?

Put a **ring** around the correct answer.

3V

6V

9V

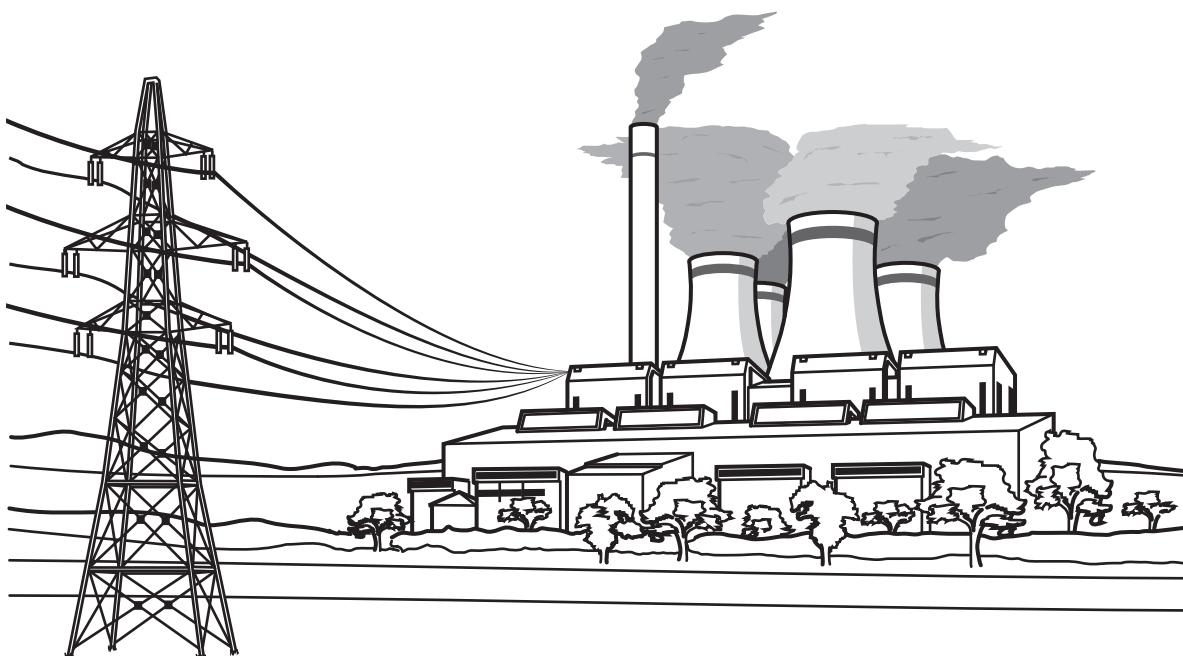
15V

[1]

[Total: 4]

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7 Mains electricity is produced by generators in power stations.



Each generator contains a magnet and a coil of wire.

(a) Describe how the magnet and the coil of wire are used to make electricity.

Include the name of the process.

[2]

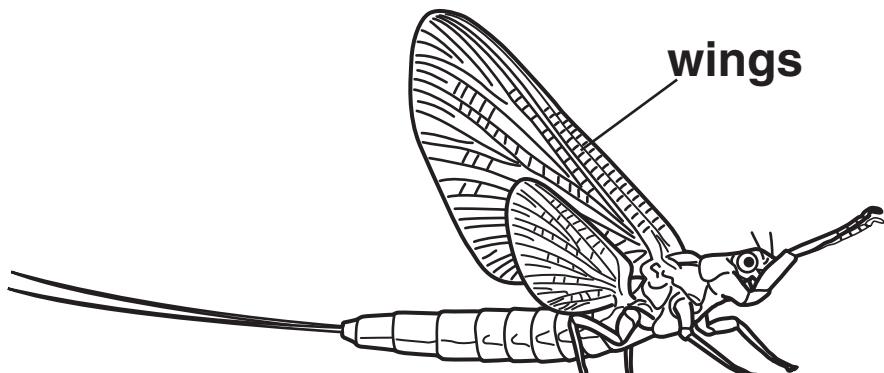
(b) State TWO ways that you could increase the voltage of the electricity produced by a generator.

[2]

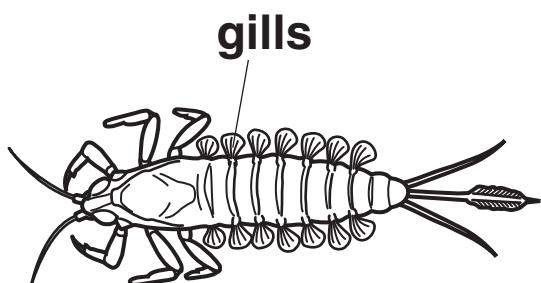
[Total: 4]

8 Martin is studying mayflies.

He looks at an adult mayfly and a mayfly nymph.



adult mayfly



mayfly nymph

(a) The adult mayfly lays eggs in the water.

The eggs hatch into nymphs.

The nymphs grow bigger by producing more cells.

Which type of cell division is responsible for the nymphs growing bigger?

[1]

(b) (i) Mayfly cells contain genes.

In which part of the cell are the genes?

Put a ring around the correct answer.

cell membrane

cytoplasm

nucleus

[1]

(ii) Genes carry the instructions for making a particular type of molecule.

Which type of molecule?

Put a ring around the correct answer.

carbohydrates

fats

minerals

proteins

[1]

- (c) The adult mayfly has wings but no gills, while the mayfly nymph has gills but no wings.

The adult mayfly and the mayfly nymph both have the same genes.

Explain how they can both have the same genes yet only the adult has wings and only the nymph has gills.

[3]

[Total: 6]

9 Susie sees a plant she likes in a friend's garden.

She asks if she can take a cutting.

- (a) Which part of her friend's plant should Susie use to grow an identical plant?**

Put a ring around the correct answer.

flower

fruit

seed

stem

[1]

- (b) Susie dips the bottom of her cutting in some rooting powder.**

She then puts the cutting into a pot of soil.

What does the rooting powder contain that helps the cutting grow roots?

[1]

(c) Susie's cutting grows into a new plant.

Which statements explain how this is possible?

Put ticks (✓) in the boxes next to the TWO correct answers.

Some unspecialised cells develop into other tissues.

Some unspecialised cells develop into organs.

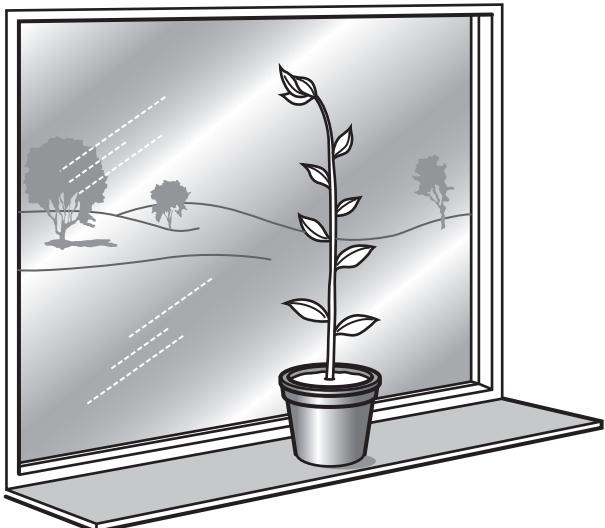
Some xylem cells become phloem cells.

Some plant cells become unspecialised.

Some leaf cells become root cells.

[2]

(d) Susie leaves her new plant next to a window.



The plant stem grows towards the light.

(i) What is the name for growth towards the light?

Put a ring around the correct answer.

photography

photosynthesis

phototropism

[1]

(ii) How does this response help the plant?

Put a tick (✓) in the box next to the correct answer.

the plant gets more carbon dioxide

the plant gets more oxygen

the plant gets more water

the plant gets more light

[1]

[Total: 6]

10 DNA is the molecule that carries genetic information.

(a) What shape is a DNA molecule?

Put a ring around the correct answer.

single helix

double helix

triple helix

[1]

(b) Here are some statements about DNA.

Use numbers from the list to complete the statements.

1 2 3 4 5

DNA is made up of _____ strand(s).

DNA is made up of _____ different bases. [1]

[Total: 2]

END OF QUESTION PAPER

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The Periodic Table of the Elements

	1	2	3	4	5	6	7	0
	7 Li lithium 3	9 Be beryllium 4	11 B boron 5	12 C carbon 6	14 N nitrogen 7	16 O oxygen 8	19 F fluorine 9	4 He helium 2
Key								
	relative atomic mass atomic symbol name atomic (proton) number							
39 K potassium 19	40 Ca calcium 20	45 Sc scandium 21	48 Ti titanium 22	51 V vanadium 23	52 Cr chromium 24	55 Mn manganese 25	56 Fe iron 26	59 Co cobalt 27
85 Rb rubidium 37	88 Sr strontium 38	89 Y yttrium 39	91 Zr zirconium 40	93 Nb niobium 41	96 Mo molybdenum 42	[98] Tc technetium 43	101 Ru ruthenium 44	103 Rh rhodium 45
133 Cs cesium 55	137 Ba barium 56	139 La* lanthanum 57	178 Hf hafnium 72	181 Ta tantalum 73	184 W tungsten 74	190 Os osmium 76	192 Ir iridium 77	195 Pt platinum 78
[223] Fr francium 87	[226] Ra radium 88	[227] Ac* actinium 89	[261] Rf rutherfordium 104	[262] Db dubnium 105	[266] Sg seaborgium 106	[264] Bh bohrium 107	[277] Hs hassium 108	[271] Ds darmstadtium 110
						[272] Rg roentgenium 111		[209] Po polonium 84
								[210] At astatine 85
								[222] Rn radon 86
								Elements with atomic numbers 112-116 have been reported but not fully authenticated

* The lanthanoids (atomic numbers 58-71) and the actinoids (atomic numbers 90-103) have been omitted.

The relative atomic masses of copper and chlorine have not been rounded to the nearest whole number.