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B622/02

GENERAL CERTIFICATE OF SECONDARY EDUCATION

GATEWAY SCIENCE

SCIENCE B

Unit 2 Modules B2 C2 P2 (Higher Tier)

TUESDAY 17 JUNE 2008

Morning Time: 1 hour

Candidates answer on the question paper.

Additional materials (enclosed):

None

Calculators may be used.

Additional materials: Pencil

Ruler (cm/mm)



Candidate Forename	1			Candidate Surname						
Centre Number							Candidate Number			

INSTRUCTIONS TO CANDIDATES

- Write your name in capital letters, your Centre Number and Candidate Number in the boxes above.
- Use blue or black ink. Pencil may be used for graphs and diagrams only.
- Read each question carefully and make sure that you know what you have to do before starting your answer.
- Answer all the questions.
- Do not write in the bar codes.
- Write your answer to each question in the space provided.

INFORMATION FOR CANDIDATES

- The number of marks for each question is given in brackets [] at the end of each question or part question.
- The total number of marks for this paper is **60**.
- A list of physics equations is printed on page two.
- The Periodic Table is printed on the back page.

FOR EXAMINER'S USE		
Section	Max.	Mark
Α	20	
В	20	
С	20	
TOTAL	60	

		printed pages.	
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This document consists of 20 printed pages



2

EQUATIONS

$$efficiency = \frac{useful\ energy\ output}{total\ energy\ input}$$

$$energy = mass \times specific\ heat\ capacity \times temperature\ change$$

$$energy = mass \times specific\ latent\ heat$$

$$fuel\ energy\ input = waste\ energy\ output\ +\ electrical\ energy\ output$$

$$power = voltage \times current$$

$$energy\ supplied\ =\ power \times time$$

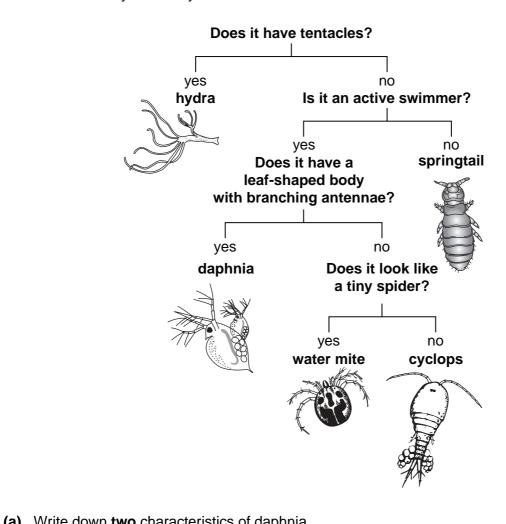
$$energy\ (kilowatt\ hours)\ =\ power\ (kW) \times time\ (h)$$

$$wave\ speed\ =\ frequency \times wavelength$$

Answer **all** the questions.

Section A - Module B2

John investigates a pond water habitat. He collects a sample of water and looks at the animals in it. He then uses a key to identify the animals.



(a)	write down two characteristics of daprillia.	
	Use the information in the key.	
	1	
	2	[2
(b)	The animals in the key are invertebrates.	
	What does invertebrate mean?	
		[1

((c)	John	finds	out	about	the	plants	in	the	pond	١.
١,		, 001111	mias	Out	about	uic	piarito	11 1	uic	POLIC	١.

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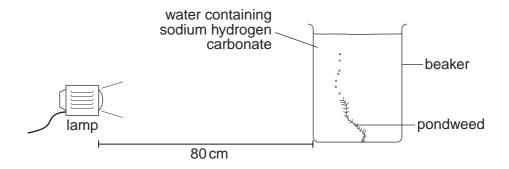
Explain why.			

.....[1]

(d) John investigates how quickly pondweed produces oxygen.

Look at the diagram.

It shows the apparatus he uses.



John adds different amounts of sodium hydrogen carbonate to the water in the beaker.

This provides the pondweed with different concentrations of carbon dioxide to use for photosynthesis.

He counts the number of bubbles given off in five minutes.

The table shows his results.

mass of sodium hydrogen carbonate in grams	number of bubbles given off in five minutes
0.00	6
0.02	25
0.04	39
0.06	45
0.08	45

(i)	Describe the pattern in the results.
	[2

(ii) Explain the pattern in the results.

		Use ideas about limiting factors.
		rol
		[2]
		[Total: 8]
Loc	k at	the picture of the bear.
		and the state of t
(a)		wn bears are not adapted to live in cold conditions in the Arctic. ar bears have adaptations so they can live in very cold conditions.
	Des	scribe one of these adaptations.
	Exp	lain how the adaptation you have chosen helps the polar bear.
		[2]
(b)		April 2006, a bear with both polar bear and brown bear features was found in Canada. A showed the bear had a polar bear mother and a brown bear father.
	(i)	What is the name given to the offspring of two different species?
		[1]
	(ii)	Past crosses between polar bears and brown bears have resulted in fertile offspring. Some scientists suggest that polar bears and brown bears should be classified as the same species.
		Explain why.
		[1]

[Total: 4] [Turn over

2

3 Look at the picture of a fin whale.

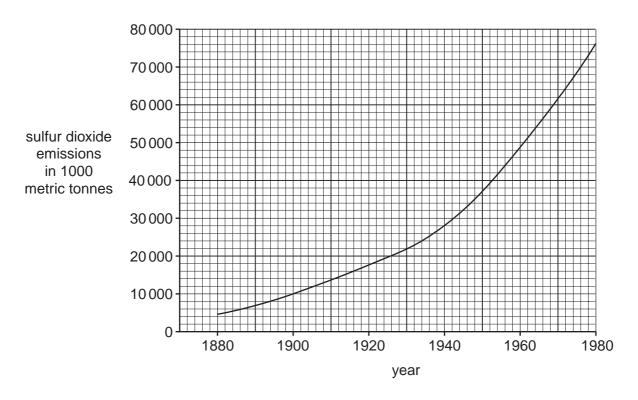


(a)	Some whales are kept in captivity.	
	Suggest one argument for and one argument against keeping whales in captivity.	
	for	
	against	
		.[2
(b)	In 1986, the International Whaling Commission agreed to stop hunting whales.	
	In October 2006, Iceland announced their intention to start hunting again.	
	Fin whales are an endangered species.	
	They intend to hunt fin whales within sustainable limits.	
	Sustainable development of fin whales could allow hunting without extinction.	
	Describe how.	
		.[2

[Total: 4]

4 Look at the graph.

It shows world sulfur dioxide emissions between 1880 and 1980.



(a)	The change in sulfur dioxide levels is causing problems in the environment.
	Write down one of these problems

•		.[1	J
Т	The human population is increasing rapidly		

Write down the name given to the increasing rate of human population growth.
--

(c)	The increase in the human population has led to an increase in sulfur dioxide pollution.

Describe **two other** consequences of the increase in human population.

1

2

.....[2

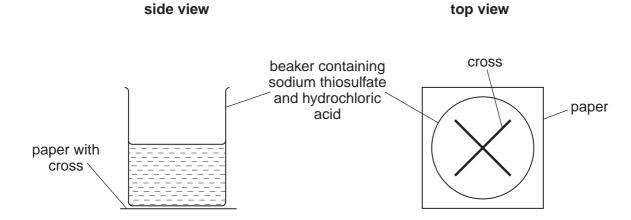
[Total: 4]

Section B - Module C2

5 Rob and Catherine investigate the reaction between sodium thiosulfate and hydrochloric acid.

A yellow solid is made during the reaction.

Look at the diagram. It shows the apparatus they use.



Rob and Catherine look down at the cross. A yellow solid is made. The liquid in the beaker gets cloudy.

After a time they cannot see the cross.

Rob and Catherine measure this reaction time.

They do the experiment four times.

They use four different concentrations of sodium thiosulfate solution, A, B, C and D.

They do all the experiments at 20 °C.

Look at their results.

concentration of sodium thiosulfate	reaction time in seconds	
Α	41	
В	74	
С	135	
D	67	

Choose from A, B, C or D.

answer [1]

(b)	The rate of the reaction can be increased by raising the temperature of the sodium thiosulfate to 40 °C.
	Explain why. Use ideas about collisions between particles.
	[3]
(c)	The reaction between sodium thiosulfate and hydrochloric acid happens quite slowly.
	Rusting is another slow reaction.
	Iron objects rust more quickly in areas near to the sea.
	Suggest why.
	[1]
(d)	Iron reacts with oxygen and water to make hydrated iron(III) oxide.
` ,	What is the name of this process?
	Choose from the list.
	alloying
	decomposition
	electrolysis
	oxidation
	answer[1]
	[Total: 6]

- **6** This question is about the air.
 - (a) Carbon monoxide is a pollutant found in air.

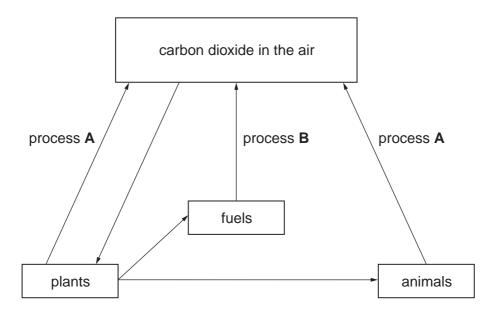
Carbon monoxide, CO, burns in oxygen, ${\rm O_2}$.

Carbon dioxide, CO_2 , is made.

Write a balanced **symbol** equation for this reaction.

.....[2]

(b) Look at the diagram of the carbon cycle.



(i) What is the name of process A?

Choose from the list.

combustion

degassing

photosynthesis

respiration

answer[1]

(ii)	What is the name of process B?
	Choose from the list.

combustion

degassing

photosynthesis

respiration

	answer[1]
(c)	The carbon cycle keeps the composition of the air constant.
	Cutting down rain forests is called deforestation .
	Deforestation changes the composition of the air.
	Suggest how the composition changes.
	Explain your answer.
	[2]
	[Total: 6]

7 This question is about construction materials.

Look at the photograph. It shows some buildings.



© iStockphoto.com / Malcolm Romain

(a)	Rocks such as granite, limestone and marble are used to construct buildings.					
	Limestone and marble are	both calcium carbonate.	oth calcium carbonate.			
Marble is much harder than limestone.						
	Explain why. Use ideas abo	out rock types.				
(b)	(b) Some construction materials are made from rocks in the Earth's crust.					
Iron, brick and glass are construction materials.						
Draw straight lines to link each construction material to the rock it is made from.						
(construction material		rock it is made from			
iron			haematite ore			
brick			sand			
glass			clay			

[2]

[Total: 4]

T	nis question is about paints.					
P	ints are made up of three materials.					
T	ey are a solvent , a binding medium and a pigment .					
(a	Pigments give the paint its colour.					
	Some pigments are thermochromic.					
	Thermochromic pigments change colour when they are heated.					
	Write down one use of thermochromic pigments.					
	[1]				
(k) In oil paints, the pigment is dispersed in an oil.					
	Oil paints dry slowly.					
	Explain how oil paints dry.					
	[1]				
(c) Paints are colloids.					
	Look at the sentences about colloids.					
	Which sentences about colloids are correct?					
	Put ticks (✔) in the boxes next to the correct sentences.					
	Particles are mixed and dispersed through a liquid.					
	Solid particles dissolve in the liquid.					
	A colloid is a single compound.					
	Solid particles are suspended in a liquid.					
	[2	<u>']</u>				
	[Total: 4	١]				

Section C - Module P2

9 This question is about wind turbines.

Wind turbines are often located in groups on wind farms.

The energy of the wind can be transferred into electrical energy by the turbines.



The about the advantages and disadvantages of using wind turbines for producing electricity.
dvantages
sadvantages
[3

[Total: 3]

10	This	This question is about generation of electricity and the cost of using it.					
	(a)	Batteries produ	uce direct cui	rent.			
		A generator is	part of a pov	roduces alternat	oduces alternating current (AC).		
		Describe what	is meant by A	AC.			
							[1]
	(b)	Power stations are part of the electricity supply system.					
		The diagram s	hows the mai	n stages in the p	oroduction of elec	ctricity in a power station	١.
		stage A	st	age B	stage C	stage D	
		fuel burns (i) Describe	pro	eam is oduced	d during stage C	electricity transmitted using the national grid in the power station.	
		Describe of the second of the	why transform swer, write ab nat happens t	at stage D to chers are used. Sout o the voltage	nange the voltage).	[2]

(c) The electrical appliances in Robert's house use electricity.

In one week, the electrical appliances in Robert's house

- use 12 kilowatts of power
- are on for a total of 8 hours.

	one week.	
	The equations on page 2 may help you.	
	answer units of electrical energy	[2]
(ii)	The cost of a unit of electricity is 10 pence.	
	Calculate the cost of the electricity used in Robert's house in one week.	
	answer pence	 [1]

[Total: 8]

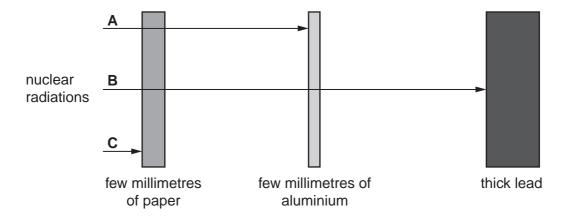
(i) Calculate the total number of units of electrical energy used by Robert's appliances in

11 This question is about **nuclear** radiation.

In a physics lesson, Pardeep's class are learning about nuclear radiation.

The teacher tells the class about how nuclear radiations penetrate substances.

The teacher does an experiment. The diagram shows the results.



Pardeep thinks that nuclear radiation **A** is **beta** radiation.

He does not think that **B** or **C** can be beta radiation.

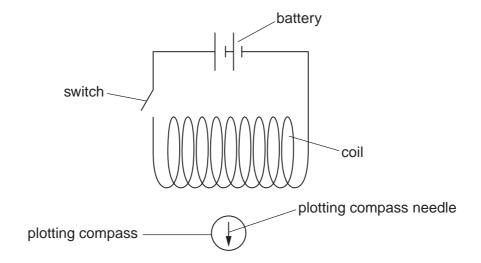
Explain why he is co	rrect.		

[Total: 3]

12 Kate is investigating magnetic fields.

She uses a coil and a plotting compass.

The coil is connected to a switch and a battery.



Kate closes the switch.

The plotting compass needle moves.
Explain why the plotting compass needle moves.

[Total: 2]

ay. [1]			
[1]			
[1]			
[2]			
[Total: 4]			
-			

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Q.1 diagram Data source: Wildfowl & Wetlands Trust, www.wwt.org.uk

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The Periodic Table of the Elements

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1	·			65 Zn zinc 30	112 Cd cadmium 48	201 Hg mercury 80	Elemei
1				63.5 Cu copper 29	108 Ag silver 47	197 Au gold 79	Rg roentgenium 111
1				59 Ni nickel 28	106 Pd palladium 46	195 Pt platinum 78	Ds darmstadtium 110
Ca Sc Ti V Cromium Fechenium				59 Co cobalt 27	103 Rh rhodium 45	192 Ir iridium 77	[268] Mt meitnerium 109
Ca Sc Ti V Cromium Fechenium	1 H hydrogen			56 Fe iron 26	101 Ru ruthenium 44	190 Os osmium 76	[277] Hs hassium 108
Partial				55 Mn manganese 25			[264] Bh bohrlum 107
Partial		mass ool number		52 Cr chromium 24	96 Mo molybdenum 42	184 W tungsten 74	Sg seaborgium 106
9 Be beryllium 4 4 24 Mg magnesium 12 2	Key	Key /e atomic i		51 V vanadium 23			[262] Db dubnium 105
9 Be beryllium 4 4 24 Mg magnestum 12 20 Ca catcum 20 88 Sr strontium 38 137 Ba bartum 56 [226] Ra radium 88		relati atc atomic		48 Ti titanium 22	91 Zr	178 Hf hafnium 72	[261] Rf rutherfordium 104
7 9 Li Be lithium 3 24 Na	·			Sc scandium 21	89 Y yttrium 39	139 La* lanthanum 57	[227] Ac* actinium 89
7 Li lithium 3 23 Na sodium 11 39 K potassium 19 85 Rb rubidium 37 133 Cs caesium 55 Fr francium 87	2	9 Be beryllium 4	24 Mg magnesium 12			137 Ba barium 56	[226] Ra radium 88
	_			39 K potassium 19	85 Rb rubidium 37	133 Cs caesium 55	Fr Fr francium 87

* The lanthanoids (atomic numbers 58-71) and the actinoids (atomic numbers 90-103) have been omitted.

The relative atomic masses of copper and chlorine have not been rounded to the nearest whole number.