

**GENERAL CERTIFICATE OF SECONDARY EDUCATION  
GATEWAY SCIENCE  
SCIENCE B**

**B622/01**

Unit 2 Modules B2 C2 P2 (Foundation Tier)



Candidates answer on the Question Paper  
A calculator may be used for this paper

**OCR Supplied Materials:**  
None

**Other Materials Required:**

- Pencil
- Ruler (cm/mm)

**Wednesday 16 June 2010  
Morning**

**Duration: 1 hour**



Candidate Forename		Candidate Surname	
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Centre Number							Candidate Number				
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**MODIFIED LANGUAGE**

**INSTRUCTIONS TO CANDIDATES**

- Write your name clearly in capital letters, your Centre Number and Candidate Number in the boxes above.
- Use black ink. Pencil may be used for graphs and diagrams only.
- Read each question carefully and make sure that you know what you have to do before starting your answer.
- Answer **all** the questions.
- Do **not** write in the bar codes.
- Write your answer to each question in the space provided. Additional paper may be used if necessary but you must clearly show your Candidate Number, Centre Number and question number(s).

**INFORMATION FOR CANDIDATES**

- The number of marks is given in brackets [ ] at the end of each question or part question.
- A list of physics equations is printed on page two.
- The Periodic Table is printed on the back page.
- The total number of marks for this paper is **60**.
- This document consists of **20** pages. Any blank pages are indicated.

**EQUATIONS**

$$\text{efficiency} = \frac{\text{useful energy output}}{\text{total energy input}}$$

$$\text{wave speed} = \text{frequency} \times \text{wavelength}$$

$$\text{power} = \text{voltage} \times \text{current}$$

$$\text{energy (kilowatt hours)} = \text{power (kW)} \times \text{time (h)}$$

Answer **all** the questions.

**Section A – Module B2**

- 1 Look at the picture.

It shows a landfill site being filled with household waste.



- (a) Britain needs more landfill sites than it did one hundred years ago.

Suggest **one** reason why.

..... [1]

- (b) Old quarries are often used for landfill sites.

The minerals taken from the quarries were used in buildings.

Minerals are an example of which type of resource?

Choose from the list.

**finite      fossil fuel      renewable**

answer ..... [1]

- (c) Old landfill sites can be turned into nature reserves.

This can help animals close to extinction.

What name do we use to describe animals close to extinction?

..... [1]

**[Total: 3]**

- 2 David and Linda investigate the animals in four different ponds.

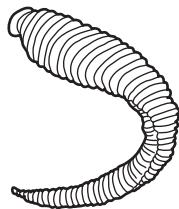
- (a) They want to find which animals are swimming in the ponds.

Put a (ring) around the best piece of equipment to use to collect swimming animals.

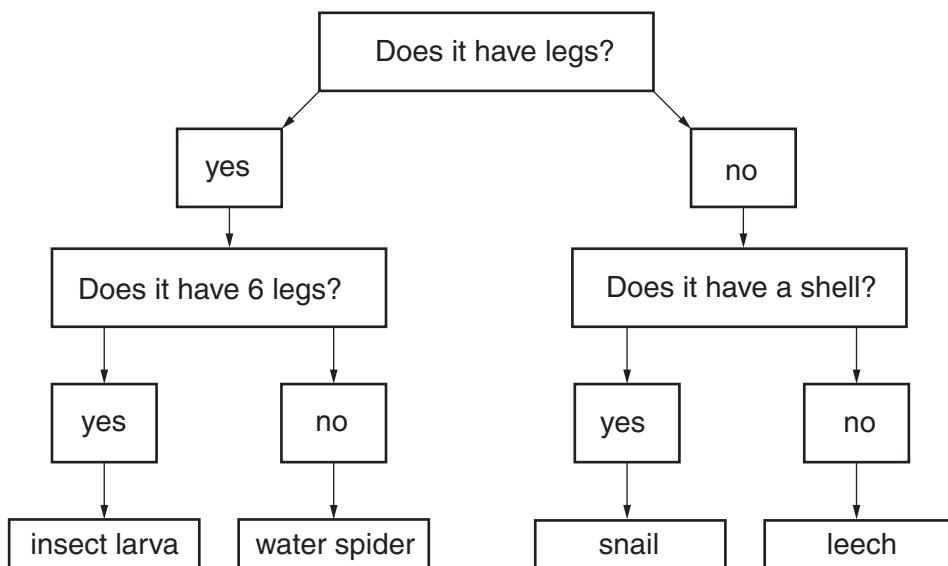
**net      pit-fall trap      pooter      quadrat**

[1]

- (b) Look at one of the animals they catch.



Use the key to identify the animal.



What did David and Linda catch? ..... [1]

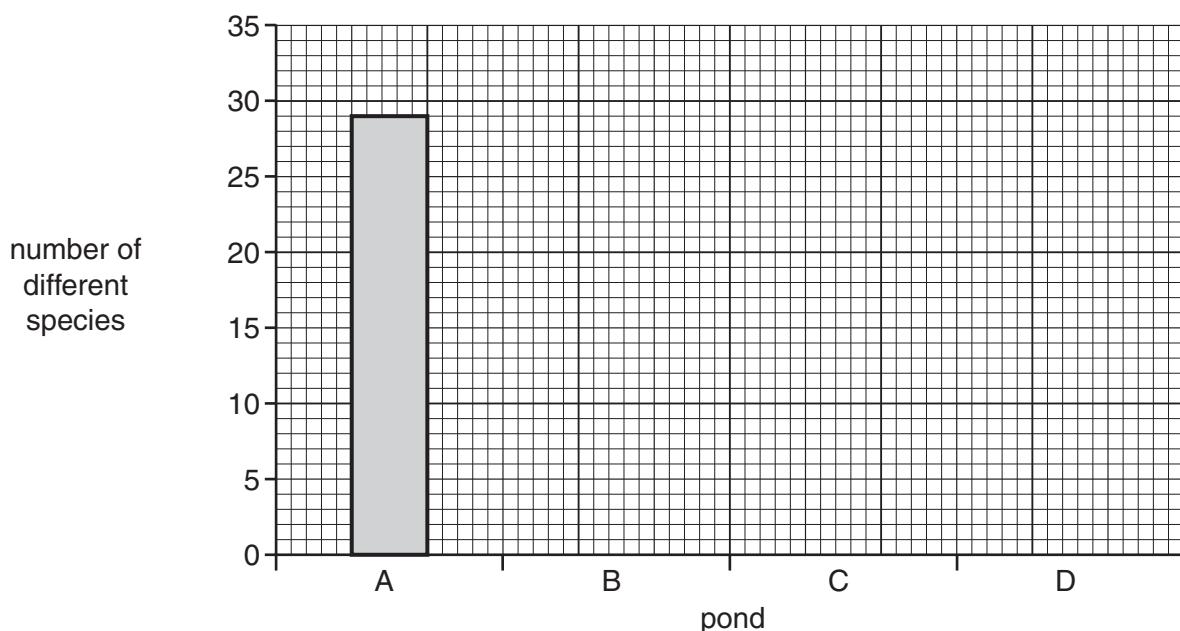
- (c) David and Linda count up the number of different species they collect.

The table shows their results.

pond	number of different species
A	
B	26
C	3
D	10

- (i) Use the data in the table to finish the bar chart.

[1]



- (ii) The result for pond A is missing from the table.

Look at the bar chart.

How many different species did they find in pond A? ..... [1]

- (iii) Which pond is likely to be the **most** polluted?

Choose from A, B, C or D. ....

Explain the reason for your answer. ....

[1]

- (d) All of the animals collected were invertebrates.

How are invertebrates different from vertebrates?

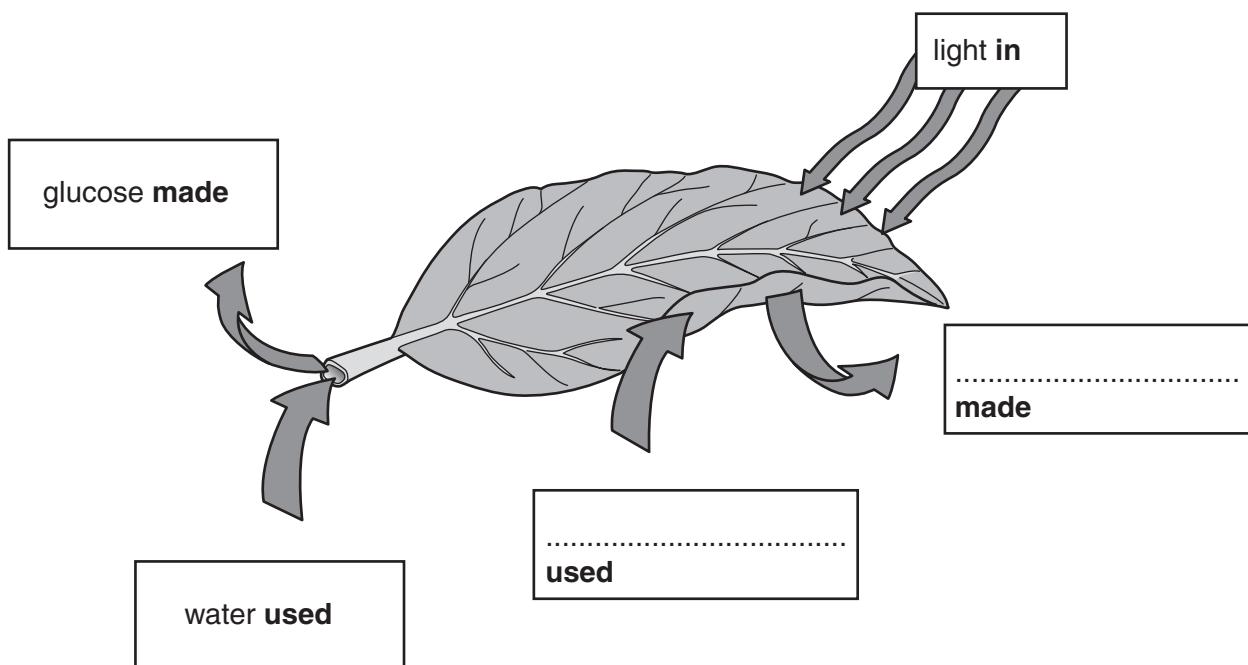
[1]

[Total: 6]

Turn over

- 3 Look at the diagram of a leaf.

It shows the substances involved in photosynthesis.



- (a) Finish the diagram to name:

- (i) The gas that is **used** in photosynthesis. [1]
- (ii) The gas that is **made** in photosynthesis. [1]

- (b) The rate of photosynthesis can be increased by giving the plant more light.

Write down **one other** way to increase the rate of photosynthesis.

..... [1]

- (c) Some of the glucose made by the plant can be used for energy.

The rest is changed into other substances for different uses.

Describe another use for glucose.

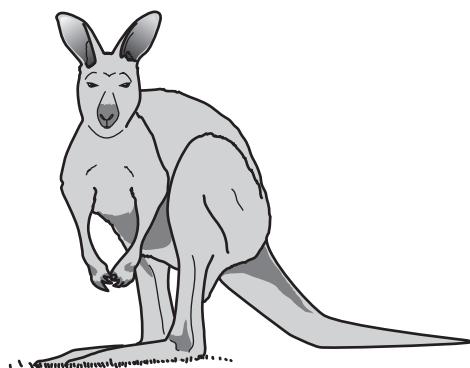
The substance it is changed into .....

What the new substance is used for .....

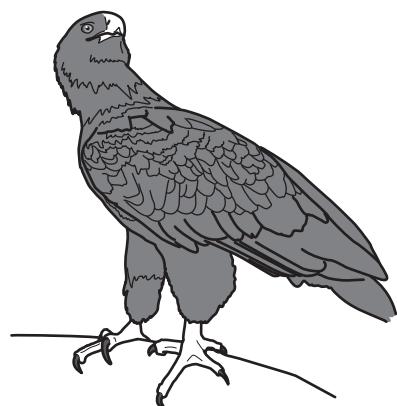
..... [2]

**[Total: 5]**

- 4 Look at the pictures of a kangaroo and a wedge-tailed eagle.



kangaroo



wedge-tailed eagle

- (a) The wedge-tailed eagles hunt kangaroos for food.

- (i) What term is used to describe animals that hunt prey for food?

..... [1]

- (ii) Describe how the kangaroo is adapted to avoid being caught as prey.

Use the picture to help you.

.....  
.....  
..... [2]

- (b) The population of kangaroos in an area goes up and down.

Explain why the population of eagles will also go up and down.

.....  
.....  
..... [2]

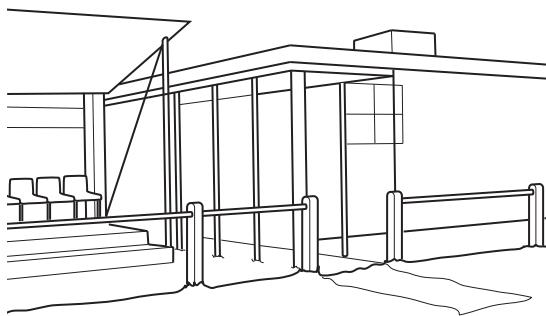
- (c) The eagle has to compete with other eagles for food.

Apart from food, write down **one other** thing that eagles compete for.

..... [1]  
**[Total: 6]**

**Section B – Module C2**

- 5 Look at the picture of a football clubhouse and barriers.



- (a) Write down the names of **two** construction materials used in making buildings.

1 .....

2 ..... [2]

- (b) The clubhouse has been painted.

Write down **two** reasons why the clubhouse has been painted.

1 .....

2 ..... [2]

- (c) The barriers have been painted with **phosphorescent** paint.

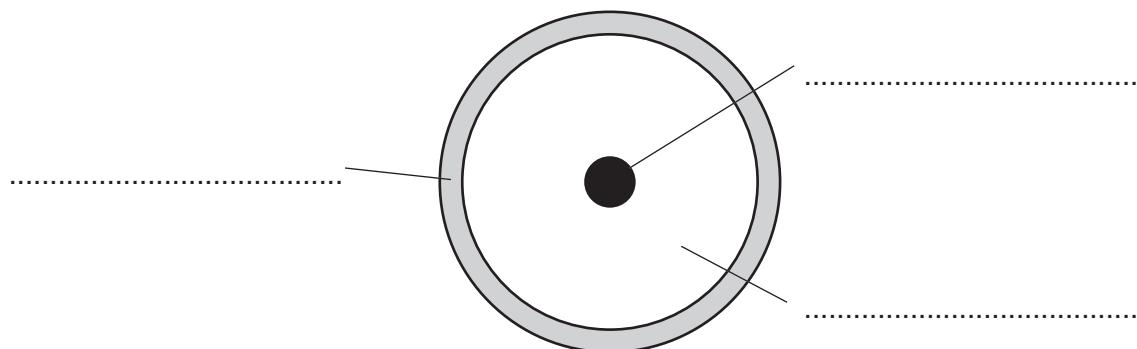
Describe one advantage of painting the barriers with phosphorescent paint.

.....  
..... [1]

[Total: 5]

- 6 This question is about the structure of the Earth.

(a) Look at the diagram of the Earth.



Complete the labels on the diagram.

Choose from the list.

**core**

**crust**

**mantle**

[2]

- (b) The outer layer of the Earth is made up of two types of tectonic plates.

One type of plate is oceanic.

Write down the name of the other **type** of plate.

..... [1]

- (c) The tectonic plates float **on top** of the mantle.

Explain why.

..... [1]

**[Total: 4]**

7 Clean air is a mixture of gases.

(a) Look at this list. It shows some of the gases found in clean air.

**argon**

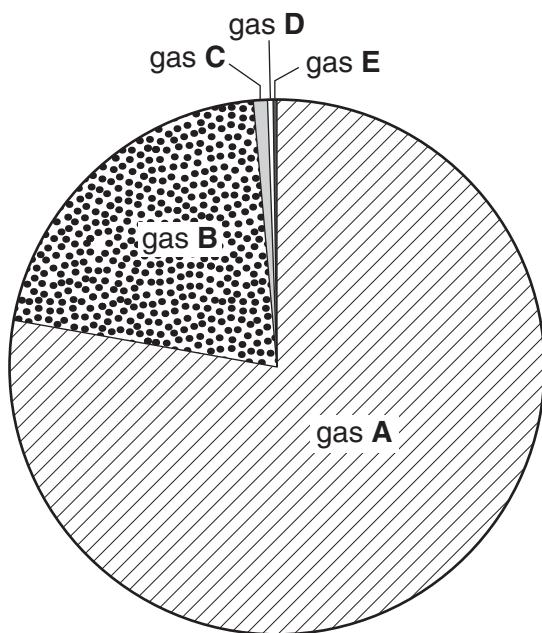
**carbon dioxide**

**nitrogen**

**oxygen**

**water vapour**

Look at this pie-chart. It gives information about the percentage of different gases in clean air.



What is the name of gas A?

Choose from the list.

answer ..... [1]

- (b) Sulfur dioxide and oxides of nitrogen are common pollutants found in dirty air.

These gases cause acid rain.

Write about the problems caused by acid rain.

.....  
.....  
.....

[2]

- (c) Carbon monoxide and oxides of nitrogen are found in the exhaust gases of cars.

These gases pollute the air.

Cars are fitted with catalytic converters.

Look at the word equation. It shows a reaction in a catalytic converter.



How does a catalytic converter reduce air pollution?

The word equation may help you.

.....  
.....

[1]

[Total: 4]

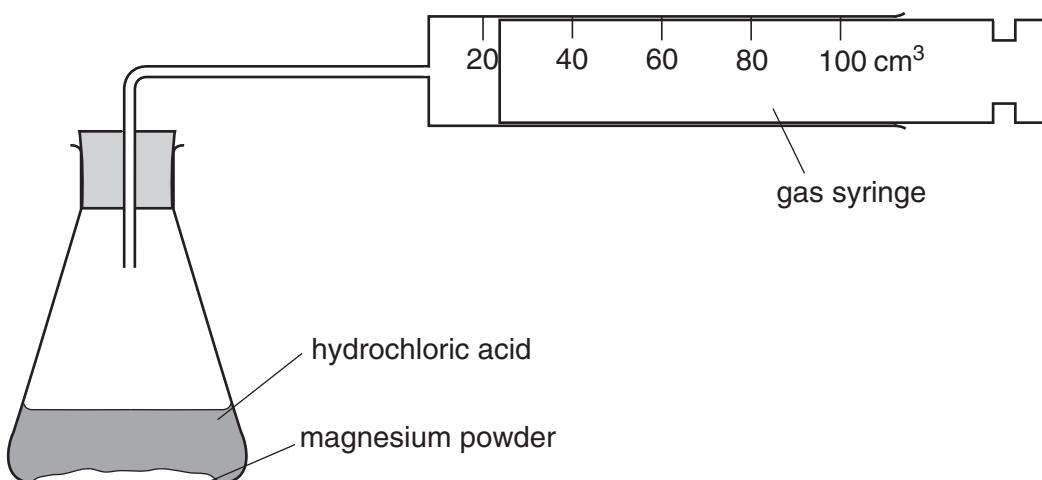
- 8 Ryan and Naomi investigate the reaction between magnesium and hydrochloric acid.

Magnesium chloride and hydrogen are made.

- (a) Write the **word** equation for this reaction.

..... [1]

- (b) The diagram shows the apparatus they use.



Look at the table.

It shows their results when 0.1 g of magnesium reacts with hydrochloric acid.

time in seconds	total volume of gas in syringe in $\text{cm}^3$
0	0
20	50
40	80
60	90
80	100
100	100

- (i) At what time does the reaction finish?

answer ..... seconds

[1]

- (ii) Complete the sentence.

The reaction is fastest between ..... seconds and ..... seconds. [1]

[Total: 3]

- 9 This question is about the properties of metals.

The table lists data for properties of some metals.

metal	density in g/cm <sup>3</sup>	relative strength	relative electrical conductivity
aluminium	2.7	70	40
copper	8.9	220	64
iron	7.9	210	11
tin	7.3	10	7
zinc	7.3	140	18

- (a) Write down the name of the metal with the **highest** electrical conductivity.

Choose from the table.

..... [1]

- (b) Some of the properties of aluminium and iron are different.

Write down **two** ways in which they are different.

You may use the data in the table to help you.

1 .....

2 ..... [2]

- (c) Copper and zinc make the alloy brass.

Write down **one use** for brass.

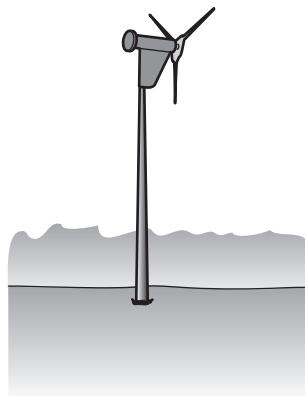
..... [1]

[Total: 4]

## Section C – Module P2

10 Look at the picture of a wind turbine.

It provides energy for a field studies centre on Exmoor.



(a) (i) Finish this sentence by choosing the **best** word from this list.

**conduction**

**convection**

**radiation**

The Sun causes ..... currents in the air. This produces a wind. [1]

(ii) Finish these sentences by choosing the **best** words from this list.

**chemical**

**electrical**

**kinetic**

**thermal**

The wind has ..... energy.

The wind turbine transfers this into ..... energy. [2]

- (b) (i) The Sun is a renewable energy source.

Other renewable energy sources are burned to produce energy.

Write down the name of one of these **renewable** energy sources.

..... [1]

- (ii) Fossil fuels are non-renewable energy sources.

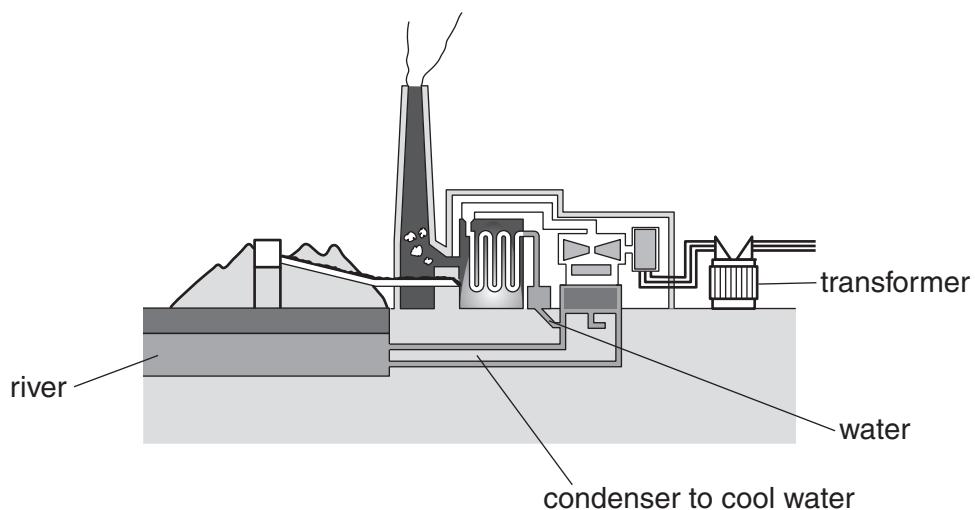
Write down the name of one **fossil fuel**.

..... [1]

[Total: 5]

- 11 Most of our electricity is generated in power stations.

Look at the diagram of a power station.



- (a) Describe how electricity is **generated** in the power station.

.....  
.....  
.....  
.....

[3]

- (b) The transformer increases the voltage of the electricity.

It is then transmitted around the country.

Why is electricity transmitted at high voltages?

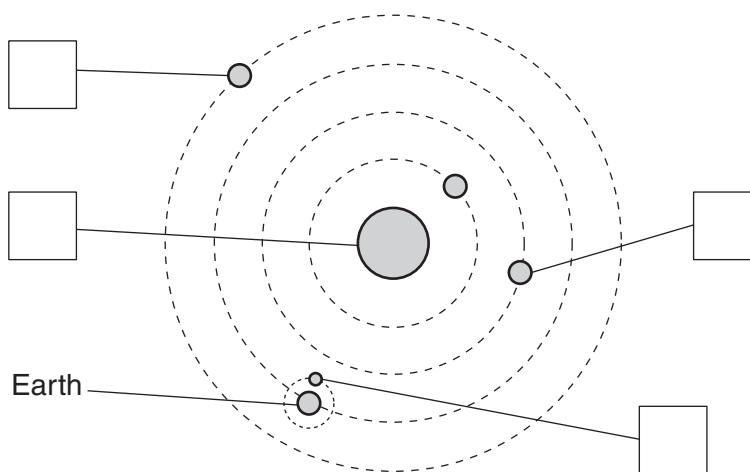
.....  
.....

[1]

[Total: 4]

- 12 Look at the diagram.

It represents part of our Solar System.



not to scale

- (a) Write the letter **S** in one box that shows the Sun. [1]
- (b) Write the letter **P** in one box that shows a planet. [1]
- (c) Write the letter **M** in one box that shows the Moon. [1]
- (d) Some spacecraft are sent into space unmanned.

Some spacecraft carry astronauts.

Write down two things that astronauts **must** have in a spacecraft.

- 1 .....
- 2 ..... [2]

[Total: 5]

13 Scientists think that the Universe started with an explosion.

- (a) What is this explosion known as?

..... [1]

- (b) What is happening to the size of the Universe?

..... [1]

- (c) At night, we can see a large number of stars.



- (i) Why can we **see** stars?

Put a tick (✓) in the box next to the correct answer.

Stars are closer to us than the Sun.

Stars give off their own light.

Stars reflect light from the Moon.

Stars reflect light from the Sun.

[1]

- (ii) Stars have a finite life and will eventually die.

How do stars **start** their life?

.....

[1]

**[Total: 4]**

- 14 This symbol is used to warn about dangerous radioactivity.



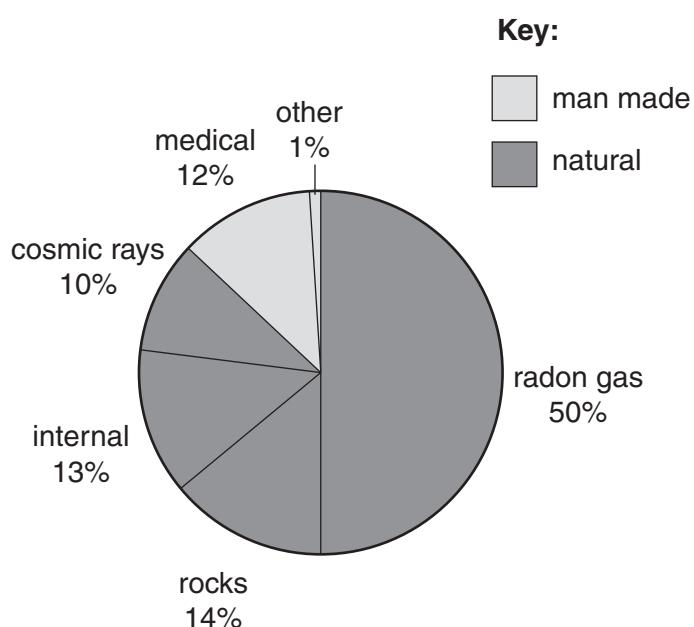
- (a) Nuclear radiation can also be **useful**.

Write down one example of how nuclear radiation is **useful**.

..... [1]

- (b) Background radiation is always around us.

The pie chart shows the sources of background radiation and their percentages.



What percentage of background radiation is natural?

answer ..... %

[1]

**[Total: 2]**

**END OF QUESTION PAPER**

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# The Periodic Table of the Elements

	1	2	3	4	5	6	7	0				
	7 Li lithium 3	9 Be beryllium 4	11 B boron 5	12 C carbon 6	14 N nitrogen 7	16 O oxygen 8	19 F fluorine 9	20 Ne neon 10				
Key	<table border="1"> <tr> <td>relative atomic mass</td><td>atomic symbol</td><td>name</td><td>atomic (proton) number</td></tr> </table>								relative atomic mass	atomic symbol	name	atomic (proton) number
relative atomic mass	atomic symbol	name	atomic (proton) number									
39 K potassium 19	40 Ca calcium 20	45 Sc scandium 21	48 Ti titanium 22	51 V vanadium 23	52 Cr chromium 24	55 Mn manganese 25	56 Fe iron 26	59 Co cobalt 27				
85 Rb rubidium 37	88 Sr strontium 38	89 Y yttrium 39	91 Nb niobium 41	93 Zr zirconium 40	96 Mo molybdenum 42	[98] Tc technetium 43	101 Ru ruthenium 44	103 Rh rhodium 45				
133 Cs caesium 55	137 Ba barium 56	139 La* lanthanum 57	178 Hf hafnium 72	181 Ta tantalum 73	184 W tungsten 74	186 Re rhenium 75	190 Os osmium 76	192 Ir iridium 77				
[223] Fr francium 87	[226] Ra radium 88	[227] Ac* actinium 89	[261] Rf rutherfordium 104	[262] Db dubnium 105	[266] Sg seaborgium 106	[264] Bh bohrium 107	[277] Hs hassium 108	[268] Mt meitnerium 109				
						[271] Ds darmstadtium 110	[272] Rg roentgenium 111					

Elements with atomic numbers 112-116 have been reported but not fully authenticated

\* The lanthanoids (atomic numbers 58-71) and the actinoids (atomic numbers 90-103) have been omitted.

The relative atomic masses of copper and chlorine have not been rounded to the nearest whole number.