

B622/01

GENERAL CERTIFICATE OF SECONDARY EDUCATION GATEWAY SCIENCE

SCIENCE B

Unit 2 Modules B2 C2 P2 (Foundation Tier)

FRIDAY 18 JANUARY 2008

Afternoon Time: 1 hour

Candidates answer on the question paper.

Additional materials (enclosed):

None

Calculators may be used. **Additional materials:** Pencil

Ruler (cm/mm)



Candidate Forename				Candidate Surname			
Centre Number				Candidate Number			

INSTRUCTIONS TO CANDIDATES

- Write your name in capital letters, your Centre Number and Candidate Number in the boxes above.
- Use blue or black ink. Pencil may be used for graphs and diagrams only.
- Read each question carefully and make sure that you know what you have to do before starting your answer.
- Answer all the questions.
- Do **not** write in the bar codes.
- Do not write outside the box bordering each page.
- Write your answer to each question in the space provided.

INFORMATION FOR CANDIDATES

- The number of marks for each question is given in brackets [] at the end of each question or part question.
- The total number of marks for this paper is 60.
- A list of physics equations is printed on page two.
- The Periodic Table is printed on the back page.

FOR EXAMINER'S USE						
Section	Max.	Mark				
Α	20					
В	20					
С	20					
TOTAL	60					

This document consists of 23 printed pages and 1 blank p	This	document	consists	of 23	printed	pages	and	1 blank	page
--	------	----------	----------	-------	---------	-------	-----	---------	------

SPA (NH/CGW) T47807/6

© OCR 2008 [K/103/4251]

OCR is an exempt Charity

[Turn over



2

EQUATIONS

efficiency = $\frac{\text{useful energy output}}{\text{total energy input}}$ wave speed = frequency × wavelength

power = voltage × current

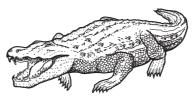
energy (kilowatt hours) = power (kW) × time (h)

Answer **all** the questions.

Section A – Module B2

1	(a)	Look at the list of resources that humans use.	
		Two of these resources are finite .	
		Put (rings) around the two finite resources.	
		fossil fuels	
		minerals	
		oxygen	
		water	
		wood	[2]
	(b)	When humans use resources they usually produce pollution.	,
		Why is the amount of pollution that humans produce increasing?	
			[1]
		[Total:	3]

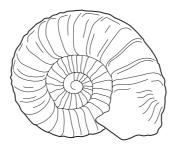
2 Look at the picture of a crocodile.



(a)	Crocodiles are predators.	
	Describe one feature, that you can see in the picture, that shows that crocodiles predators.	are
		.[1]
(b)	Look at the list of different animal groups.	
	Crocodiles belong to two of the groups.	
	Put (rings) around the two groups that crocodiles belong to.	
	amphibians	
	fish	
	invertebrates	
	mammals	
	reptiles	
	vertebrates	[2]
(c)	Crocodiles often rest with their mouths open.	[-]
(-)	Small birds sometimes go in and out of their mouths.	
	The birds are not harmed by the crocodiles.	
	The birds are not narmed by the crocodiles.	
	Suggest why crocodiles let these birds go in and out of their mouths.	
		[1]

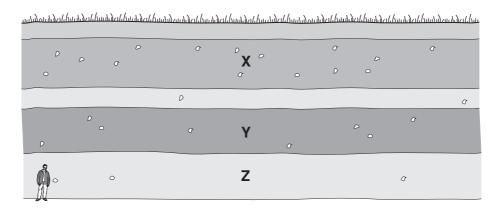
3 Chris is collecting fossils from a cliff.

The fossils are of extinct animals called ammonites.



an ammonite fossil

(a) Chris collects fossils from three different places, X, Y and Z, on the cliff.



Where are the oldest fossils found?

	Choose from: X Y Z	
	answer	[1]
(b)	The fossil only shows the ammonite shell.	
	Suggest why the rest of the animal did not fossilise.	
		[1]
(c)	Describe how fossils are formed from shells.	
		.[1]
(d)	Ammonites are now extinct.	
	What does the word extinct mean?	
		[1]

[Total: 4]

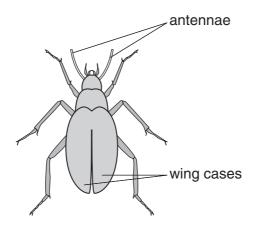
(a)	Plants make food by photosynthesis.
	Write about how plants make food by photosynthesis.
	In your answer include
	what plants use for photosynthesis
	what plants make in photosynthesis.
	[3]
(b)	Plants also respire.
	Explain why they respire.
	[1]
	[Total: 4]

- 5 Iain and Mary are investigating the animals and plants in the school playing field.
 - (a) They want to find out if any beetles are moving around the field at night.

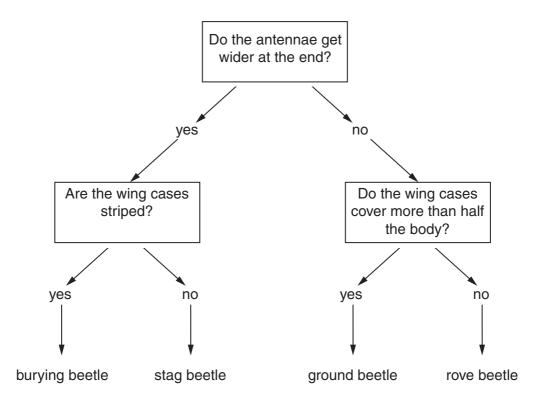
Put a (ring) around the best piece of equipment to use to catch beetles at night.

net pit-fall trap pooter [1]

(b) Look at one of the beetles lain and Mary catch.



Use the key to identify the beetle.



The type of beetle lain and Mary catch is a[1]

(c) Iain and Mary want to work out the number of dandelion plants in the playing field.

They use a quadrat to count the number of dandelion plants in different parts of the playing field.

The table shows their results.

quadrat	number of dandelions
1st	5
2nd	1
3rd	0
4th	2

Each quadrat has an area of 0.25 m².

The total area of the playing field is 20000 m².

Use this information to estimate the total number of dandelion plants in the playing field.

You are advised to show your working.

	[0]
estimated total number of dandelions =	1/31

[Total: 5]

Section B – Module C2

Look at the photograph of a car. 6

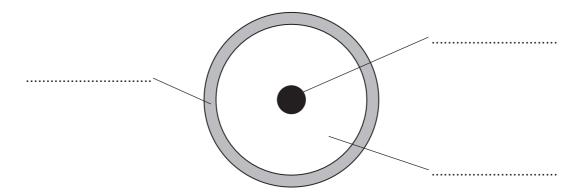


			© 15	tockphoto.com/ fornisiav	Stajouriar.					
(a)	Mar	ny different materials are used	d to make a	a car.						
	(i)	Steel is useful for making ca	r bodies.							
		Suggest why.								
						[1]				
	(ii)	Write down the name of ano	ther mate	rial that is used	when making cars.					
						[1]				
(b)	Ste	el is made from iron.								
	One disadvantage of using iron is that it rusts.									
	Wat	ter is needed for iron to rust.								
	Wri	te down the name of one othe	er substand	ce needed for i	on to rust.					
	Cho	pose from the list.								
		carbon dioxide	oil	nitrogen	oxygen					
	ans	wer				[1]				
(c)	Old	cars are taken to a scrap yar	d.							
	The	The materials in the car are recycled.								
	One	e advantage of recycling is to	reduce the	problems of di	sposal.					
	Wri	te down one other advantage	of recyclir	ng.						
						[1]				

[Total: 4]

7 This question is about the structure of the Earth.

Look at the diagram.



(a) Complete the labels on the	diagram.
--------------------------------	----------

Choose from the list.

core

crust

mantle

[2]

(b) The lithosphere is made up of tectonic plates.

Tectonic plates move slowly.

Tectonic plates meet at plate boundaries.

What can happen at a plate boundary when plates meet?

.....[1]

(c) There are two types of tectonic plate.

One type is a continental plate.

Write down the name of the other type of plate.

.....[1]

[Total: 4]

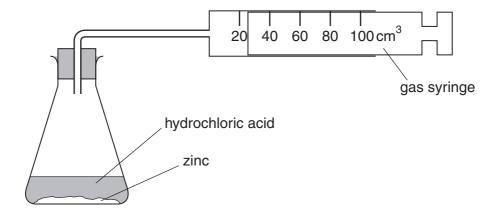
8 Phil and Ann investigate the reaction between zinc and hydrochloric acid.

Zinc chloride and hydrogen are made.

(a) Write the word equation for this reaction.

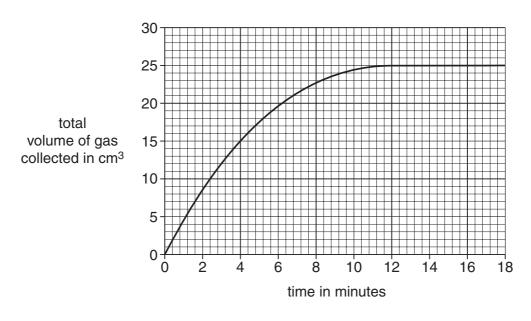


(b) The diagram shows the apparatus they use.



Look at the graph.

It shows their results when 1 g of zinc reacts with 20 cm³ of dilute hydrochloric acid.



- (i) How long does it take to make 15 cm³ of gas?[1]
- (ii) Some zinc is left at the end of the reaction.

Why does the reaction stop?

.....[1]

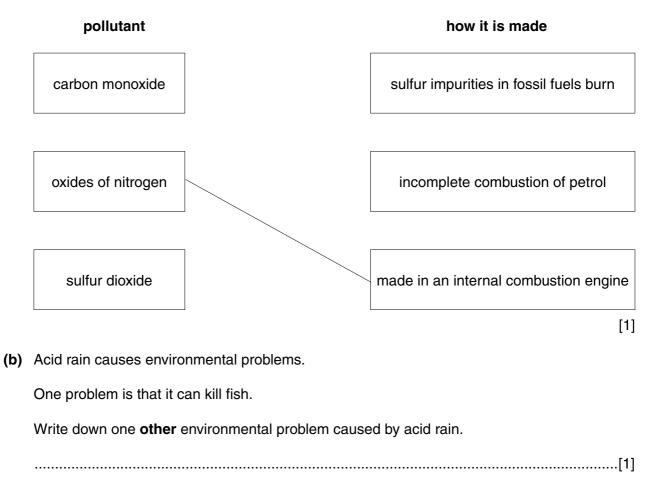
(iii) Phil and Ann want the reaction to go faster.

One way is to use a catalyst.	
Write about other ways they could make the reaction go faster.	
	.[3]
[Total	l: 6]

- **9** This question is about pollutants found in the air.
 - (a) Link each pollutant to how it is made.

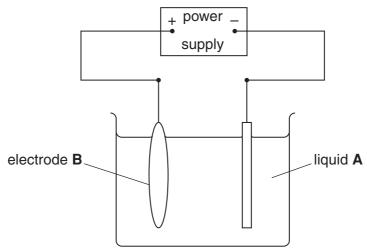
Draw two straight lines.

One has been done for you.



[Total: 2]

- 10 This question is about copper.
 - (a) Look at the diagram. It shows the apparatus needed to purify impure copper.



(i)	What is the name of the process used to purify impure copper?	
	Use the diagram to help you.	
		[1]
(ii)	Write down the name of liquid A .	
	Choose from the list.	
	copper sulfate solution	
	paraffin	
	dilute sulfuric acid	
	water	
	answer[[1]
(iii)	Write down the name of electrode B .	
	Choose from the list.	
	impure copper anode	
	impure copper cathode	
	pure copper anode	
	pure copper cathode	
		r4 1

(b)	Copper can be used to make alloys .
	Write down the name of one alloy.
	Choose from the list.
	brass
	iron
	mercury
	zinc
	answer[1]

[Total: 4]

16 BLANK PAGE

PLEASE DO NOT WRITE ON THIS PAGE

Section C - Module P2

11 This question is about nuclear radiation.



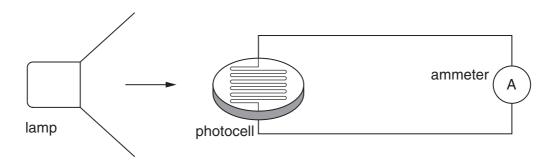
Michael is learning about the types of nuclear radiation.

(a)	His teacher tells him that there are three types of nuclear radiation.	
	Alpha is one type.	
	What are the other two types?	
	1	
	2	[2]
(b)	Michael finds out from the internet that nuclear radiation can be harmful.	
	Describe one way in which nuclear radiation can be harmful.	
		[1]
(c)	Michael's teacher, Mr Whitehead, shows his class some properties of alpha radiation.	
	Mr Whitehead uses the radioactive material safely .	
	Suggest one safety precaution he takes.	
		[1]
	[Tota	al: 4]

12 This question is about photocells.

Olivia investigates photocells.

Look at the diagram.



(a) Complete the following sentences about the photocell.

Choose from the list.

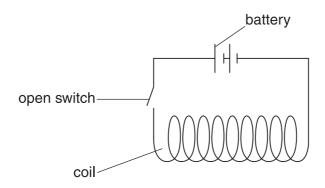
	current	electricity	light	
	The	from the lamp fal	ls on the photocell.	
	This is transferred into		by the photocell.	
	The ammeter shows that a		flows around the circuit.	[2]
(b)	Olivia covers half of the photocell	with a piece of pape	er.	
	What happens to the reading on t	the ammeter?		
				[1]

(c)	Olivia adds a voltmeter to the circuit to measure the voltage across the photocell.
	Here are her results.
	ammeter reading = 0.1 A
	voltmeter reading = 2.5V
	Calculate the power output of the photocell.
	The equations on page 2 may help you.
	answer units [3]
	[Total: 6]

13 This question is about magnetism and generating electric currents.

Sathvir makes a coil of wire and connects it to a battery.

Look at the diagram.



(a)	(i)	Sathvir	closes	the	switch.	A	current	flows	throug	h t	the	coi	I.
-----	-----	---------	--------	-----	---------	---	---------	-------	--------	-----	-----	-----	----

He puts iron filings near the coil. They are affected by a force.

What has been created around the coil?

.....[1]

(ii) Sathvir uses a piece of equipment instead of the iron filings.

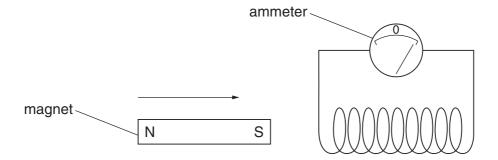
This shows the **direction** of the force.

What is the name of this piece of equipment?

.....[1]

(b) Sathvir replaces the battery with an ammeter.

Look at the diagram.



(i) The North and South ends of the magnet are called the [1]

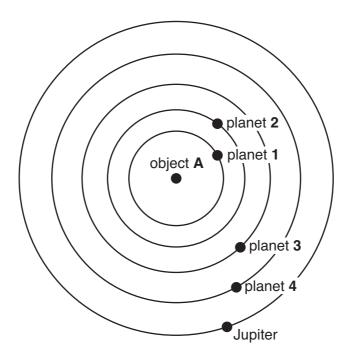
(ii) He then moves a magnet towards the coil.

A current flows in the coil. The ammeter shows a reading.
How could Sathvir make a bigger current flow in the coil?
Describe two things he could change to make a bigger current flow.
[2]
[Total: 5]

14 This question is about our Solar System.

Look at the diagram.

It shows **some** of the planets in our Solar System.



(a)	The planets	orbit around	object A.

What is the name of object A?

Γ.	4
 · [Ц

(b) (i) The asteroid belt is between Jupiter and planet 4.

What is the name of planet 4?

.....[1]

(ii) What are asteroids mainly made of?

.....[1]

(c) (i) An asteroid is an example of a Near-Earth Object (NEO).

Name another example of a NEO.

.....[1]

(ii) What do scientists use to observe NEOs?

.....[1]

[Total: 5]

PLEASE DO NOT WRITE ON THIS PAGE

Copyright Acknowledgements:

Q6 photo © iStockphoto.com/Tomislav Stajduhar.

Permission to reproduce items where third-party owned material protected by copyright is included has been sought and cleared where possible. Every reasonable effort has been made by the publisher (OCR) to trace copyright holders, but if any items requiring clearance have unwittingly been included, the publisher will be pleased to make amends at the earliest possible opportunity.

OCR is part of the Cambridge Assessment Group. Cambridge Assessment is the brand name of University of Cambridge Local Examinations Syndicate (UCLES), which is itself a department of the University of Cambridge.

© OCR 2008

The Periodic Table of the Elements

0	He helium 2	20 Ne neon 10	40 Ar argon 18	84 Kr krypton 36	131 Xe xenon 54	[222] Rn radon 86	t fully
7		19 F fluorine 9	35.5 Cl chlorine 17	80 Br bromine 35	127 iodine 53	[210] At astatine 85	orted but no
9		16 0 oxygen 8	32 S sulfur 16	79 Selenium 34	128 Te tellurium 52	[209] Po polonium 84	ve been repo
2		14 N nitrogen 7	31 P phosphorus 15	75 As arsenic 33	122 Sb antimony 51	209 Bi bismuth 83	Elements with atomic numbers 112-116 have been reported but not fully authenticated
4		12 C carbon 6	28 Si silicon 14	73 Ge germanium 32	119 Sn tin 50	207 Pb lead 82	mic numbers a
3		11 B boron 5	27 Al aluminium 13	70 Ga gallium 31	115 In indium 49	204 T1 thallium 81	nts with ato
				65 Zn zinc 30	112 Cd cadmium 48	201 Hg mercury 80	Eleme
				63.5 Cu copper 29	108 Ag silver 47	197 Au gold 79	Rg roentgenium
				59 Ni nickel 28	106 Pd palladium 46	195 Pt platinum 78	[271] Ds darmstactium 110
				59 Co cobalt 27	103 Rh rhodium 45	192 Ir iridium 77	[268] Mt meitnerium 109
,	T hydrogen			56 Fe iron 26	101 Ru ruthenium 44	190 Os osmium 76	[277] Hs hassium 108
	_			55 Mn manganese 25	[98] Tc technetium 43	186 Re rhenium 75	[264] Bh bohrium 107
		mass ool number		52 Cr chromium 24	96 Mo molybdenum 42	184 W tungsten 74	[266]
	Key	relative atomic mass atomic symbol _{name} atomic (proton) number		51 V vanadium 23	93 Nb niobium 41	181 Ta tantalum 73	[262] Db dubnium 105
		relati at o		48 Ti titanium 22	91 Zr zirconium 40	178 Hf hafnium 72	[261] Rf rutherfordium 104
				45 Sc scandium 21	89 Y yttrium 39	139 La* lanthanum 57	[227] Ac* actinium 89
2		9 Be berytlium 4	24 Mg magnesium 12	40 Ca calcium 20	88 Sr strontium 38	137 Ba barium 56	[226] Ra radium 88
_		7 Li Utthium 3	23 Na sodium 11	39 K potassium 19	85 Rb rubidium 37	133 Cs caesium 55	[223] Fr francium 87
1 2			23 24 Na Mg sodium magnesium 11 12	40 Ca catcium 20	Sr strontium 38	137 Ba barium 56	[226] Ra radium 88

* The lanthanoids (atomic numbers 58-71) and the actinoids (atomic numbers 90-103) have been omitted.

The relative atomic masses of copper and chlorine have not been rounded to the nearest whole number