

General Certificate of Secondary Education 2009–2010

Science: Single Award (Modular)

Materials and their Management Module 4

Foundation Tier

[GSC41]

FRIDAY 21 MAY 2010, MORNING

Centre Number		
71		

Candidate Number

41
GSC4

TIME

45 minutes.

INSTRUCTIONS TO CANDIDATES

Write your Centre Number and Candidate Number in the spaces provided at the top of this page. Write your answers in the spaces provided in this question paper. Answer **all six** questions.

INFORMATION FOR CANDIDATES

The total mark for this paper is 45. Figures in brackets printed down the right-hand side of pages indicate the marks awarded to each question or part question. A Data Leaflet, which includes a Periodic Table of the elements, is provided for you.



For Examiner's use only		
Question Number	Marks	
1		
2		
3		
4		
5		
6		
Total Marks		

electrical wire : bridges : bags :

wall insulation : floor tiles : shirts

Choose words from the list above to complete the table below. The first one has been done for you.

Material	Use
plastic	bags
ceramic	
copper	
fibre	
iron	

[4]

Examiner Only Marks Remark

(b) Many containers for liquids are made from glass.



What makes glass a good container for storing liquids? Tick (\checkmark) three correct boxes.

Glass is brittle	
Glass is strong	
Glass is lightweight	
Glass is unreactive	
Glass is transparent	
Glass has a low melting point	

[3]

) Many bottles are now made of pla
Give two reasons why plastic has
1
2
 Below are four fractions obtained to one use. One is already done.
Fraction
Naphtha
Refinery gas
Paraffin
Tar
 Name the process used to separate oil. Circle the correct answer.
filtration : chromatograph
) Methane (CH_4) is one of the gase
(i) Name the two elements prese Data Leaflet useful.
(ii) Give one harmful environme
(ii) Give one harmful environme
replaced glass to make I from crude oil. Match Use Lubricants Roads Gas cooke Chemicals Liquid fue e the different hydroca y : fractional dis as present in refinery g ent in methane. You make and

The bar chart below gives the percentage of different types of waste found Examiner Only Marks Rema in a town park. 40 30 Waste % 20 10 0 Glass Plastic Fruit Cans Paper Other waste Bottles Types of waste (a) (i) What percentage of waste is glass bottles? %[1] (ii) Suggest a material which you would expect to find included in 'other waste'. [1] (b) Paper bags are an example of a biodegradable material. (i) Explain fully the term biodegradable. [2] (ii) Give two environmental reasons why it is important that supermarkets use fewer plastic bags. 1._____ 2. [2]

3

Below are the four types of fingerprints. 4



5	(a)		tochromic and thermochromic paints are smart materials which r with a change in the surrounding environmental conditions.		Examine Marks	er Only Remark
		(i)	What environmental condition causes photochromic paint to change?			
				[1]		
		(ii)	Shown below is a baby's feeding bottle.			
			The bottle is made from thermochromic plastic which changes colour as temperature changes. Suggest how this may make the bottle safer to use.			
				[2]		
	(b)	Nar	notechnology involves the use of nano-sized particles.			
		(i)	Explain fully the meaning of nanotechnology .			
				[2]		
		(ii)	Give one use of nanotechnology.			
				[1]		
		(iii)	How many nanometres are in one metre? Circle the correct answer.			
			10^{-9} 10^9 10^3 10^{-3}	[1]		

Examiner Only

Re

- Reproduced with the permission of Fermanagh District Council (a) (i) Explain fully what is meant by the term hard water. [2] (ii) Name a compound that forms permanent hard water. [1] (b) Four samples of water (A, B, C and D) were tested for hardness. The results are shown below. Sample А В С D Volume of soap solution required 21 26 24 18 **before** boiling/cm³ Volume of soap solution required 11 12 24 13 after boiling/cm³ From the results which sample (A, B, C or D) has: (i) the hardest water? (ii) permanent hard water only? (iii) the greatest problem with kettle 'fur'? [3]
- 6 Many areas in Northern Ireland have hard water. Hardness is caused by certain salts of calcium and magnesium.

(c) Temporary hardness in water is caused by calcium hydrogenearbonate. Complete the word equation to show how temporary hardness is removed by boiling.

Calcium hydrogencarbonate \rightarrow water + _____ + ____



THIS IS THE END OF THE QUESTION PAPER

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