

Mark scheme June 2006

GCSE

Science: (Modular)
Single Award

Module 15

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Materials & Reactions: Foundation Tier

Question No.	KEY
One	1 – hydrocarbons
	2 – fractions
	3 – molecules 4 – atoms
	+ - atoms
Two	1 – aqueous
	2 – acidic
	3 – alkaline
	4 – neutral
Three	1
Inree	1 – oxygen 2 – carbon dioxide
	3 – water (vapour)
	4 – sulphur dioxide
	·
Four	1 – copper
	2 – tin
	3 – sodium 4 – aluminium
	4 – aiumimum
Five	1 – limestone is quarried and powdered
	2 – limestone is heated in a kiln
	3 – carbon dioxide and quicklime are made
	4 – quicklime is reacted with water
a:	*
Six	L M
	IVI
Seven	Q
30.011	R
Eight	8.1 – A, 8.2 – D, 8.3 – C, 8.4 – D
Nine	9.1 – C, 9.2 – B, 9.3 – C, 9.4 – B
Ten	10.1 – B, 10.2 – B, 10.3 – B, 10.4 – B
1 011	10.1 - D, 10.2 - D, 10.3 - D, 10.4 - D

Materials & Reactions: Higher Tier

Question No.	KEY
One	1 – limestone is quarried and powdered
	2 – limestone is heated in a kiln
	3 – carbon dioxide and quicklime are made
	4 – quicklime is reacted with water
Two	1 – carbon monoxide
TWO	2 – copper
	3 – iron sulphate
	4 – carbon dioxide
Three	Q
	R
Four	it is a saturated hydrocarbon
	the carbon atoms are joined to the hydrogen atoms by single bonds
Five	5.1 – A, 5.2 – D, 5.3 – C, 5.4 – D
G:-	(1 C (2 D (2 C (4 D
Six	6.1 – C, 6.2 – B, 6.3 – C, 6.4 – B
Seven	7.1 – B, 7.2 – B 7.3 – B 7.4 – B
Seven	7.1 - B, 7.2 - B 7.5 - B 7.4 - B
Eight	8.1 – D, 8.2 – B, 8.3 – D, 8.4 – D
Digitt	0.1 2, 0.2 2, 0.0 2, 0.1 2
Nine	9.1 – D, 9.2 – B, 9.3 – A, 9.4 – A
Ten	10.1 – B, 10.2 – A, 10.3 – B, 10.4 – D