

Mark scheme November 2004

GCSE

Science: (Modular)
Single Award and Chemistry

Module 15

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Materials and Reactions: Foundation Tier

Question No.	KEY
One	1 – limestone
	2 – waste gases
	3 – molten iron
	4 – hot air
Two	1 – mixture
	2 – compound
	3 – molecule
	4 – element
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Three	1 – sodium chloride
	2 – hydrochloric acid
	3 – sulphuric acid
	4 – potassium hydroxide
Four	1 – argon
roui	2 – iron
	3 – carbon
	4 – potassium
	- pomosium
Five	1 – cement
	2 – concrete
	3 – carbon dioxide
	4 – slaked lime
Six	air
	concrete
Seven	columns of elements with similar properties are called Groups
	most of the elements are arranged in order of their relative atomic masses
Eight	8.1 – D, 8.2 – B, 8.3 – C, 8.4 – A
Nine	9.1 – C, 9.2 – C, 9.3 – C, 9.4 – D
Ten	10.1 – D, 10.2 – D, 10.3 – A, 10.4 – D

Materials and Reactions: Higher Tier

Question No.	KEY
One	1 – cement
	2 – concrete 3 – carbon dioxide
	4 – slaked lime
Two	1 – R
	$\begin{bmatrix} 2-S \\ 2 \end{bmatrix}$
	3 – P 4 – Q
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Three	columns of elements with similar properties are called Groups
	most of the elements are arranged in order of their relative atomic masses
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Four	calcium carbonate → calcium oxide + carbon dioxide
	decane $(C_{10}H_{22}) \rightarrow$ ethene +octane (C_8H_{18})
Five	5.1 – D, 5.2 – B, 5.3 – C, 5.4 – A
Six	6.1 – C, 6.2 – C, 6.3 – C, 6.4 – D
C	71 D 72 D 72 A 74 D
Seven	7.1 – D, 7.2 – D, 7.3 – A, 7.4 – D
Eight	8.1 – A, 8.2 – C, 8.3 – B, 8.4 – D
3	
Nine	9.1 - C, 9.2 - C, 9.3 - C, 9.4 - B
Ten	10.1 – C, 10.2 – A, 10.3, – A, 10.4 – B