

ALLIANCE

Mark scheme March 2004



## Science (Modular) Single Award

## Module 15

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## Materials and Reactions: Foundation Tier

Question No.	KEY
One	1 – carbon 2 – oxygen 3 – water (vapour) 4 – sulphur
Two	1 – potassium 2 – iron (a transition element) 3 – carbon 4 – hydrogen
Three	1 - compounds 2 - elements 3 - atoms 4 - molecules
Four	$1 - \mathbf{L}$ $2 - \mathbf{J}$ $3 - \mathbf{K}$ $4 - \mathbf{M}$
Five	$ \begin{array}{c} 1 - \mathbf{W} \\ 2 - \mathbf{X} \\ 3 - \mathbf{Y} \\ 4 - \mathbf{Z} \end{array} $
Six	all Group 2 elements have similar properties vertical columns of elements are called Groups
Seven	decane has the largest molecules ethane and butane are gases at 20 °C
Eight	8.1 – B, 8.2 – C, 8.3 – C, 8.4 – A
Nine	9.1 - C, 9.2 - D, 9.3 - A, 9.4 - C
Ten	10.1 – A, 10.2 – A, 10.3 – D, 10.4 – B

## Materials and Reactions: Higher Tier

Question No.	КЕҮ
One	1 – W
	$2 - \mathbf{X}$ $3 - \mathbf{Y}$
	3 - 1 $4 - \mathbf{Z}$
Two	1 – ammonia
	2 – hydrogen
	3 – copper oxide 4 – ammonium sulphate
Three	decane has the largest molecules
	ethane and butane are gases at 20 °C
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Four	iron rusts less quickly when attached to a more reactive metal iron rusts more quickly when attached to copper
	Ton fusis more querkly when attached to copper
Five	5.1 - B, 5.2 - C, 5.3 - C, 5.4 - A
Six	6.1 - C, 6.2 - D, 6.3 - A, 6.4 - C
Seven	7.1 – A, 7.2 – A, 7.3 – D, 7.4 – B
Seven	7.1 - A, 7.2 - A, 7.5 - D, 7.7 - D
Eight	8.1 - D, 8.2 - A, 8.3 - A, 8.4 - D
Nine	9.1 – A, 9.2 – D, 9.3 – A, 9.4 – B
Ten	10.1 C 10.2 P 10.3 A 10.4 D
1 611	10.1 – C, 10.2 – B, 10.3, – A, 10.4 – D