

Mark scheme November 2005

GCSE

Science: (Modular) Double Award and Chemistry

Module 05

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Metals : Foundation Tier

Question No.	KEY
One	1 - iron 2 - waste gases 3 - slag 4 - molten iron
Two	1 – platinum 2 – magnesium 3 – chromium 4 – copper
Three	1 – hydrogen 2 – oxygen 3 – carbon dioxide 4 – water
Four	1 - sodium 2 - mercury 3 - tungsten 4 - iron
Five	1 – metal M 2 – metal N 3 – metal L 4 – metal K
Six	it can be hammered into shape it is an unreactive metal
Seven	aluminium oxide is obtained from bauxite positive aluminium ions move to the negative electrode
Eight	8.1 - B, 8.2 - D, 8.3 - A, 8.4 - D
Nine	9.1 – D, 9.2 – C, 9.3 – A, 9.4 – C
Ten	10.1 – D, 10.2 – A, 10.3 – D, 10.4 – C

Metals : Higher Tier

KEY
1 – metal M
2 – metal N 3 – metal L
4 – metal K
1 – L
2 – M 3 – N
3 - N 4 - P
aluminium oxide is obtained from bauxite
positive aluminium ions move to the negative electrode
aluminium is more reactive than iron
iron oxide is reduced
5.1 – B, 5.2 – D, 5.3 – A, 5.4 – D
6.1 – D, 6.2 – C, 6.3 – A, 6.4 – C
7.1 – D, 7.2 – A, 7.3 – D, 7.4 – C
8.1 – B, 8.2 – C, 8.3 – A, 8.4 – B
9.1 – B, 9.2 – C, 9.3 – D, 9.4 – A
10.1 - C, 10.2 - C, 10.3, - D, 10.4 - C