

ALLIANCE

Mark scheme March 2003

# GCSE

## Science (Modular) Double Award and Chemistry (Modular)

### Module 05

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### **Metals: Foundation Tier**

Question	Key
One	1 – mercury
	2-iron
	3 – magnesium
	4 – potassium
Two	1 – conduct
	$2 - \operatorname{cut}$
	3 – corrode
	4 – melt
Three	1 – electrolysis
	2 – thermal decomposition
	3 – oxidation
	4 – neutralisation
Four	1 - S – iron ore, coke and limestone are put into the furnace
	2 - R – coke burns to form carbon dioxide
	3 - P – carbon dioxide reacts with coke to produce carbon monoxide
	4 - Q - carbon monoxide reacts with iron oxide to produce iron
Five	1 – metal M
	2 – metal N
	3 – metal L
	4 – metal K
0.	
Six	it bends and shapes easily
	it is a good conductor of electricity
0	· · · · · · · · · · · · · · · · · · ·
Seven	ammonia
	sodium hydroxide
<b>D</b> . 1	
Eight	8.1 – D, 8.2 – C, 8.3 – A, 8.4 – A
N 71	
Nine	9.1 - C, 9.2 - B, 9.3 - A, 9.4 - B
Ten	10.1 - A, 10.2 - A, 10.3 - B, 10.4 - B

### Metals: Higher Tier

Question	Key
One	1 – metal M
	2 – metal N
	3 – metal L
	4 – metal K
Two	1 – carbon monoxide
	2 – carbon dioxide
	3 – hydrogen
	4 – water
Three	ammonia
	sodium hydroxide
Four	R – sodium hydroxide and sulphuric acid react to produce the salt, sodium sulphate $S$ – sodium hydroxide solution is alkaline because it contains OH <sup>-</sup> (aq) ions
Five	5.1 - D, 5.2 - C, 5.3 - A, 5.4 - A
Six	6.1 - C, 6.2 - B, 6.3 - A, 6.4 - B
Seven	7.1 – A, 7.2 – A, 7.3 – B, 7.4 – B
Eight	8.1 – B, 8.2 – B, 8.3 – A, 8.4 – C
Nine	9.1 - C, 9.2 - B, 9.3 - A, 9.4 - C
Ten	10.1 - A, 10.2 - B, 10.3, -C, 10.4 - B