

New  
Specification



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Candidate Number

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General Certificate of Secondary Education  
2012

## Science: Physics

Unit P1

Higher Tier

[GPH12]



FRIDAY 15 JUNE, AFTERNOON

### TIME

1 hour 30 minutes.

### INSTRUCTIONS TO CANDIDATES

Write your Centre Number and Candidate Number in the spaces provided at the top of this page.  
Write your answers in the spaces provided in this question paper.  
Answer **all six** questions.

### INFORMATION FOR CANDIDATES

The total mark for this paper is 100.  
Figures in brackets printed down the right-hand side of pages indicate the marks awarded to each question or part question.  
Quality of written communication will be assessed in Questions **3(b)(ii)** and **5(b)**.

For Examiner's use only	
Question Number	Marks
1	
2	
3	
4	
5	
6	

<b>Total Marks</b>	
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7676.03R

- 1 (a) The diagram shows a solar panel. This is made up of a number of photocells.  
The photocells produce electricity directly from sunlight.  
Solar panels are placed on the roof of a house.



solar panel

© Zoonar / Thinkstock

On a cloudless summer day the solar energy shining on the panel every second is 6000 J. Of this amount 4800 J are reflected, the rest is converted to electricity.

- (i) Calculate the output electrical energy every second from the solar panel.  
**You are advised to show clearly how you get your answer.**

Output electrical energy = \_\_\_\_\_ J [1]

- (ii) Calculate the efficiency of the solar panel.  
**You are advised to show clearly how you get your answer.**

Efficiency = \_\_\_\_\_ [2]

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Marks	Remark

- (iii) On a certain summer day the panel generated electricity for 10 hours.  
Calculate the number of kilojoules generated on this day by the solar panel.  
**You are advised to show clearly how you get your answer.**

Number of kJ = \_\_\_\_\_ [2]

- (iv) State **one** advantage and **one** disadvantage of the use of the solar panel.

Advantage \_\_\_\_\_

\_\_\_\_\_

\_\_\_\_\_

Disadvantage \_\_\_\_\_

\_\_\_\_\_

\_\_\_\_\_ [2]

- (v) A family of four would use on average 54 000 kJ of electrical energy per day.  
State **two** things they could do to make up the difference between what the solar panel produces and what they need.

\_\_\_\_\_

\_\_\_\_\_

\_\_\_\_\_

\_\_\_\_\_

\_\_\_\_\_ [2]

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Marks Remark



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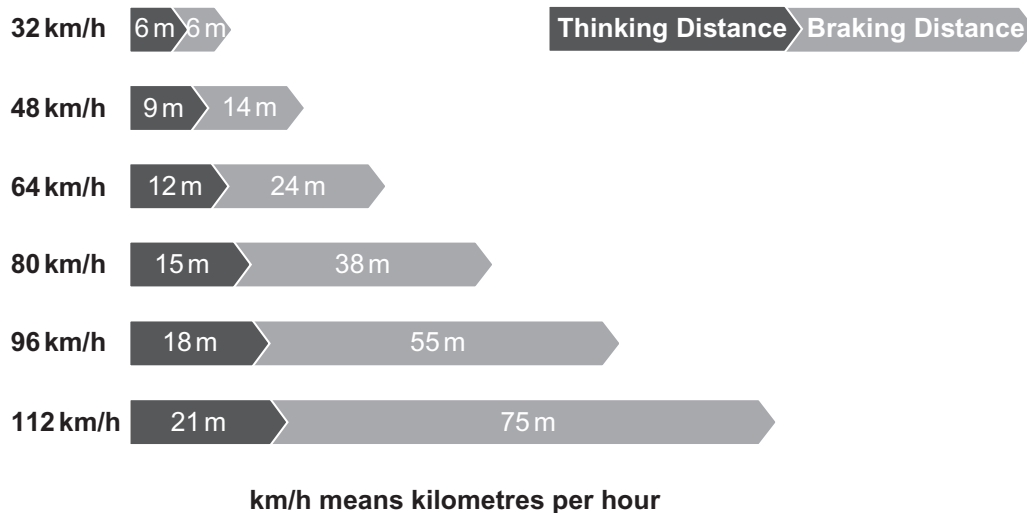
2 Speed limits in the Republic of Ireland are given in km/h.

- (i) Show clearly that a speed of 80 km/h is equal to a speed of 22.2 m/s.  
(Hint: 1 hour = 3600 seconds)

[3]

The **stopping distance** of a car is the **thinking distance** added to the **braking distance**. The thinking distance is the distance the car travels at constant speed before the driver reacts (reaction time) to a hazard on the road and applies the brakes. The braking distance is the distance the car travels from where the brakes were first applied to where the car stops.

The chart below shows the results of a study of stopping distance by an alert driver, on a dry day using a car with good tyres and brakes.



- (ii) Using the data from the chart that applies to a speed of 80 km/h calculate the thinking time (reaction time) of the driver.  
**You are advised to show clearly how you get your answer.**

Thinking time = \_\_\_\_\_ s [4]

Examiner Only	
Marks	Remark



- (vi) The speed of two cars following each other on a motorway is 112 km/h.  
Should the car in front suddenly brake it is advisable for the following car to leave a gap. The average length of a car is 4 m. Use the data from the chart to calculate the size of the required gap.  
**Give your answer in complete car lengths.**

Required gap = \_\_\_\_\_ car lengths [3]

- (vii) A car travelling at 112 km/h is equivalent to a speed of 31.1 m/s. Using the braking distance given in the chart calculate the deceleration of the car.  
**You are advised to show clearly how you get your answer.**

Deceleration = \_\_\_\_\_ m/s<sup>2</sup> [4]

- (viii) The car in part (vii) has a mass of 1500 kg. Calculate the required braking force to bring the car to rest in this distance.  
**You are advised to show clearly how you get your answer.**

Braking force = \_\_\_\_\_ N [2]

Examiner Only	
Marks	Remark

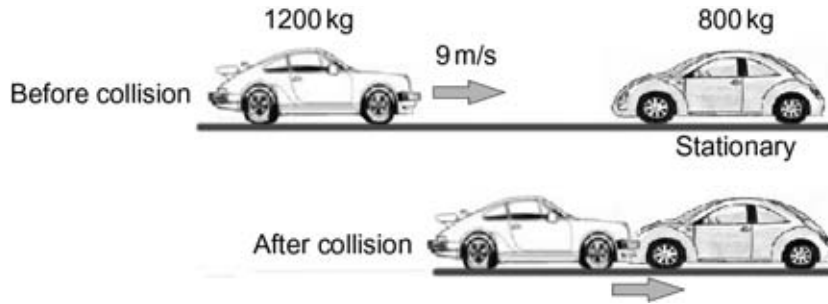


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- 3 (a) (i) State the Principle of Conservation of Momentum as it applies to collisions.

\_\_\_\_\_ [1]

A car of mass 1200 kg is travelling with a velocity of 9 m/s when it collides head-on with a stationary car of mass 800 kg. The two cars stick together as a result of the collision.



- (ii) Calculate the momentum of the 1200 kg car immediately before the collision.

**You are advised to show clearly how you get your answer.**

Momentum = \_\_\_\_\_ kg m/s [3]

- (iii) Calculate the velocity of the two cars after collision.  
Remember the two cars are stuck together after the collision.  
**You are advised to show clearly how you get your answer.**

Velocity of the combined cars after collision = \_\_\_\_\_ m/s [4]

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Marks	Remark

- (iv) Calculate the kinetic energy of the combined cars after the collision.

**You are advised to show clearly how you get your answer.**

Kinetic energy = \_\_\_\_\_ J [3]

- (b) The Transport Research Laboratory carries out research as to how to reduce the danger to drivers and passengers in road accidents.

- (i) In one experiment, a car with a momentum of 1350 kg m/s collides head on with a brick wall. The car comes to rest in 0.6 s. Calculate the average size of the force exerted on the car by the wall.

**You are advised to show clearly how you get your answer.**

Force on the car = \_\_\_\_\_ N [3]

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Marks	Remark

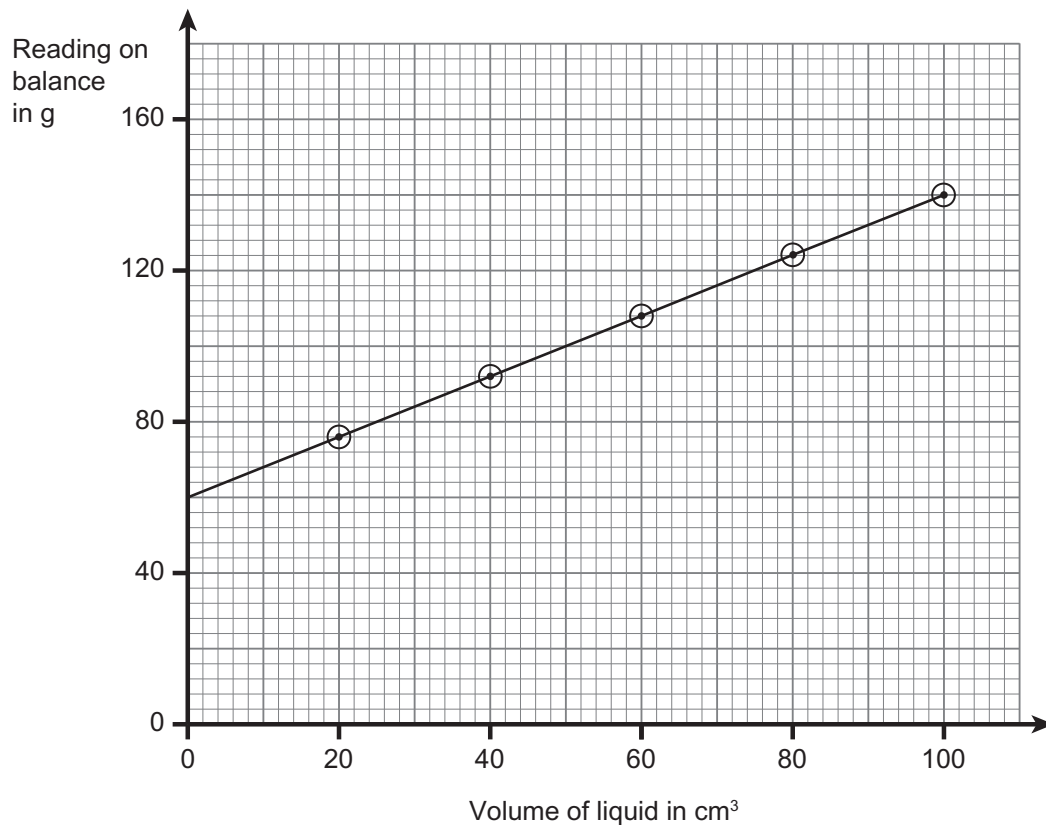


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- 4 Linda is given a measuring cylinder and an electronic balance. She places the empty measuring cylinder on the electronic balance and adds different volumes of liquid. Each time she measures the volume of the liquid she also records the reading on the electronic balance. She plots her results as a graph as shown below.



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- (i) What is the mass of the empty measuring cylinder?

\_\_\_\_\_ g [1]

Examiner Only	
Marks	Remark

The table shows four different liquids and their densities.

Liquid	Density in g/cm <sup>3</sup>
Petrol	0.7
Castor Oil	0.9
Water	1.0
Ethanol	0.8

- (ii) Using the data from the graph and your answer to part (i) identify the liquid Linda used.  
You **must** show clearly how you get your answer.

Liquid = \_\_\_\_\_ [4]

- (iii) She repeats the same procedure using the **same** measuring cylinder but using a **liquid of lower density** than the first liquid.  
On the grid draw the straight line she would expect to obtain. [2]

- (iv) Water has a density of 1.0 g/cm<sup>3</sup>. Calculate the reading Linda would obtain if she had placed 60 cm<sup>3</sup> of water in the measuring cylinder and placed it on the electronic balance.

Reading = \_\_\_\_\_ g [2]

- (v) Ice has a density of 0.9 g/cm<sup>3</sup> and water has density of 1.0 g/cm<sup>3</sup>.  
What does this tell you about the spacing of the molecules in the two states?

\_\_\_\_\_  
\_\_\_\_\_  
\_\_\_\_\_ [1]

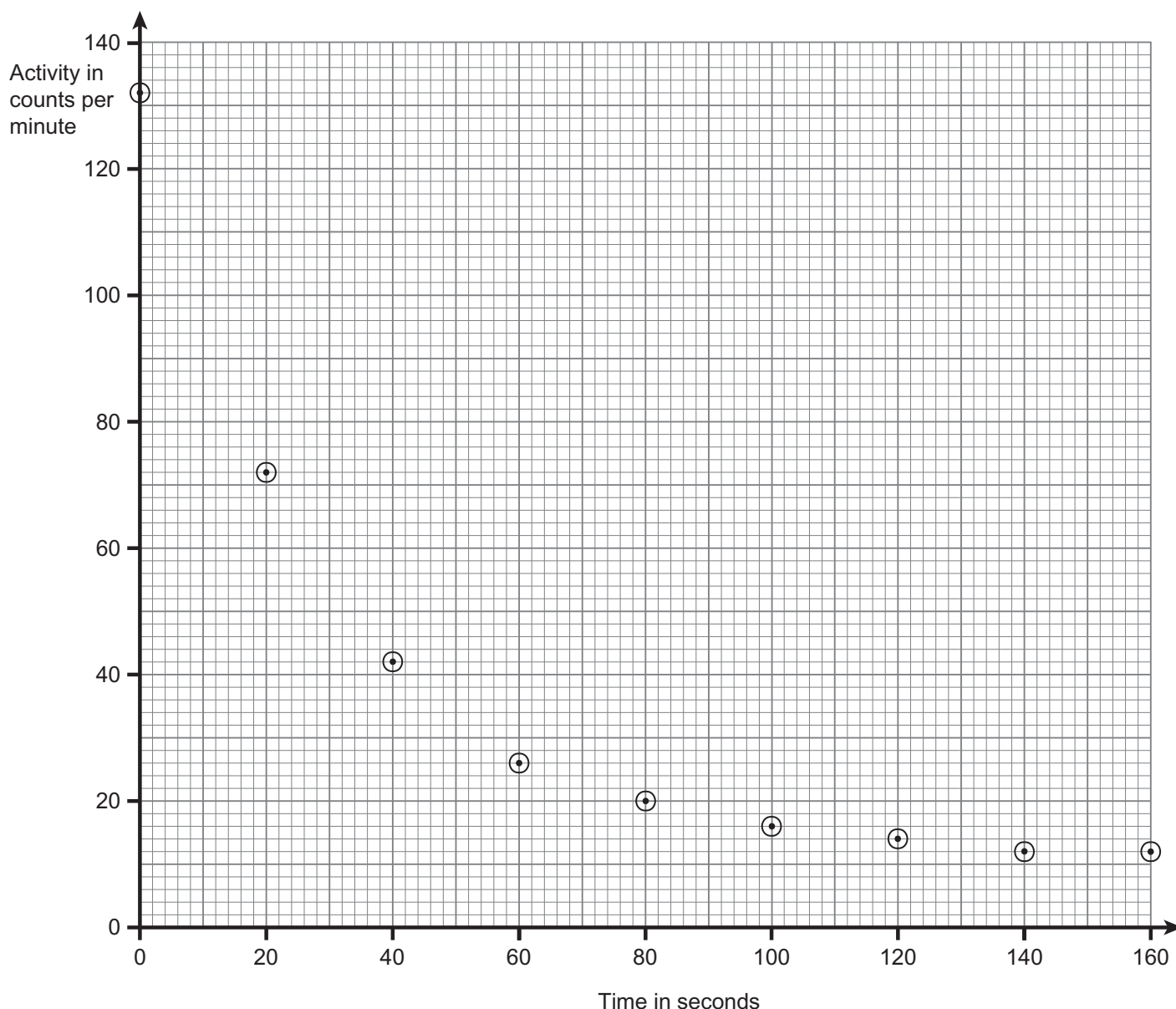
Examiner Only	
Marks	Remark







- 6 (a) When measuring the half-life of a radioactive substance, measurements of the activity were taken every 20 seconds. These measurements are shown plotted on the grid below.



- (i) Complete the graph by drawing the best fit curve. Use it to find the background activity.  
**Explain how you arrive at your answer.**

Background activity = \_\_\_\_\_ counts per minute

\_\_\_\_\_

\_\_\_\_\_ [3]

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Marks	Remark

- (ii) Using your value for the background activity, determine the activity due **only** to the radioactive substance at the **start of the experiment**.

Activity = \_\_\_\_\_ counts per minute [2]

- (b) Fission and fusion are nuclear reactions, which release large amounts of energy. The table below is intended to show a number of significant differences between the two reactions. Complete the table using the list of phrases/words below.

1. building of larger nuclei from small nuclei
2. the splitting of large nuclei
3. nuclear power station
4. requires very high temperatures to start
5. the sun
6. hydrogen
7. uranium
8. will start at room temperature

Write the number corresponding to the statement in the appropriate box in the table below.

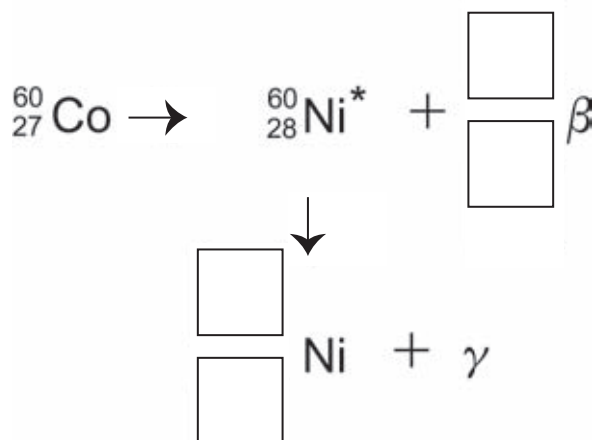
Nuclear Reaction	Fusion	Fission
Where the process can be found happening		
Fuel used		
Description of the reaction		
Conditions required to start		

[4]

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Marks	Remark

(c) Cobalt-60 is a beta emitter which decays to nickel.  
The nickel produced is a gamma emitter.

(i) Complete the decay equation below for the complete decay process by writing the correct values in the boxes.



[4]

After 15 years the measured activity of a cobalt-60 source is found to have fallen from 120 counts per minute to 15 counts per minute.

(ii) What is the half-life of cobalt-60?

**You are advised to show clearly how you get your answer.**

Half-life = \_\_\_\_\_ years [3]

(iii) State two possible uses of gamma radiation.

\_\_\_\_\_

\_\_\_\_\_

\_\_\_\_\_

\_\_\_\_\_ [2]

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(d) Describe how the electrons are arranged in the “Plum Pudding” model of the atom and the present Rutherford-Bohr model.

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[2]

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**THIS IS THE END OF THE QUESTION PAPER**

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