

# The Periodic Table of Elements

1	2	3	4	5	6	7	0	
7 <b>Li</b> lithium 3	9 <b>Be</b> beryllium 4	11 <b>Na</b> sodium 11	12 <b>C</b> carbon 6	13 <b>Al</b> aluminium 13	14 <b>N</b> nitrogen 7	15 <b>O</b> oxygen 8	16 <b>F</b> fluorine 9	17 <b>Ne</b> neon 10
23 <b>Na</b> sodium 11	24 <b>Mg</b> magnesium 12	27 <b>Al</b> aluminium 13	28 <b>Si</b> silicon 14	31 <b>P</b> phosphorus 15	32 <b>S</b> sulfur 16	35.5 <b>Cl</b> chlorine 17	40 <b>Ar</b> argon 18	
39 <b>K</b> potassium 19	40 <b>Ca</b> calcium 20	45 <b>Sc</b> scandium 21	48 <b>Ti</b> titanium 22	51 <b>V</b> vanadium 23	52 <b>Cr</b> chromium 24	55 <b>Mn</b> manganese 25	56 <b>Fe</b> iron 26	
85 <b>Rb</b> rubidium 37	88 <b>Sr</b> strontium 38	89 <b>Y</b> yttrium 39	91 <b>Zr</b> zirconium 40	93 <b>Nb</b> niobium 41	96 <b>Mo</b> molybdenum 42	[97] <b>Tc</b> technetium 43	101 <b>Ru</b> ruthenium 44	
133 <b>Cs</b> caesium 55	137 <b>Ba</b> barium 56	139 <b>La*</b> lanthanum 57	178 <b>Hf</b> hafnium 72	181 <b>Ta</b> tantalum 73	184 <b>W</b> tungsten 74	186 <b>Re</b> rhenium 75	190 <b>Os</b> osmium 76	
[223] <b>Fr</b> francium 87	[226] <b>Ra</b> radium 88	[227] <b>Ac*</b> actinium 89	[267] <b>Rf</b> rutherfordium 104	[270] <b>Db</b> dubnium 105	[269] <b>Sg</b> seaborgium 106	[270] <b>Bh</b> bohrium 107	[270] <b>Hs</b> hassium 108	
				103 <b>Rh</b> rhodium 45	106 <b>Pd</b> palladium 46	108 <b>Ag</b> silver 47	112 <b>Cd</b> cadmium 48	
				109 <b>Mt</b> meitnerium 109	110 <b>Ds</b> darmstadtium 110	111 <b>Rg</b> roentgenium 111	112 <b>Cn</b> copernicium 112	
				192 <b>Ir</b> iridium 77	195 <b>Pt</b> platinum 78	197 <b>Au</b> gold 79	201 <b>Hg</b> mercury 80	
				199 <b>Ho</b> holmium 67	200 <b>Hf</b> hafnium 68	200.9 <b>Tm</b> thulium 69	201 <b>Pb</b> lead 82	
				201 <b>Er</b> erbium 68	202 <b>Tm</b> thulium 69	203 <b>Yb</b> ytterbium 70	207 <b>Pb</b> lead 82	
				208 <b>Tl</b> thallium 81	208.98 <b>Pb</b> lead 82	209 <b>Bi</b> bismuth 83	209 <b>Po</b> polonium 84	
				209 <b>Po</b> polonium 84	210 <b>At</b> astatine 85	210 <b>Rn</b> radon 86	210 <b>Rn</b> radon 86	
				210 <b>Rn</b> radon 86	210 <b>At</b> astatine 85	210 <b>Rn</b> radon 86	210 <b>Rn</b> radon 86	
				210 <b>Rn</b> radon 86	210 <b>At</b> astatine 85	210 <b>Rn</b> radon 86	210 <b>Rn</b> radon 86	

1 <b>H</b> hydrogen 1
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relative atomic mass atomic symbol name atomic (proton) number
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Key

\* The Lanthanides (atomic numbers 58 – 71) and the Actinides (atomic numbers 90 – 103) have been omitted.  
Relative atomic masses for **Cu** and **Cl** have not been rounded to the nearest whole number.