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Centre:

Test Series: 06 MARCH 2006 SERIES H

Module: 346021 AQUEOUS AND ORGANIC CHEMISTRY

**Candidate Number:** 

UCI:

**Candidate Name:** 

For completion by the Examination Invigilator. Please fill this oval if the candidate is absent:

## **FOUNDATION TIER**

Instructions on how to complete this answer sheet are given on the question paper. Please make sure you follow them carefully.

Questions ONE to FIVE: Choose one answer for each of the parts 1 to 4.

QUESTION ONE	1	2	3	4
condenses	0	0	0	0
dissolves	0	0	0	0
evaporates	0	0	0	0
filters	0	0	0	0
QUESTION THREE	1	2	3	4
strong acid	0	0	0	0
strong alkali	0	0	0	0
weak acid	0	0	0	0
weak alkali	0	0	0	0
QUESTION FIVE	1	2	3	4
chlorine	0	0	0	0
magnesium carbonate	0	0	0	0
nitric acid	0	0	0	0
zinc sulphate	0	0	0	0

air		-	0	0
catalysts	0	0	0	0
ethanol	0	0	0	0
sugar	0	0	0	0
QUESTION FOUR	1	2	3	4
calcium ions	0	0	0	0

QUESTION TWO

QUESTION FOUR	1	2	3	4
calcium ions	0	0	0	0
hydroxide ions	0	0	0	0
nitrate ions	0	0	0	0
sodium ions	0	0	0	0

Questions SIX and SEVEN: Choose two answers for each question.

	QUESTION SIX		
Р		0	
Q		0	
R		0	
S		0	
т		0	

QUESTION SEVEN		
de-ionised water is hard water	0	
some of the hardness is removed from Sample M when it is boiled	0	
the hardest water is Sample L	0	
when soap solution is first added to Sample K, a scum will form	0	
when water is boiled, all the hardness in the water is removed	0	

Questions EIGHT to TEN: Choose one answer for each of the parts 1 to 4.

C	UES A	TION B	EIGH <sup>*</sup> C	T D
8.1	0	0	0	0
8.2	0	0	0	0
8.3	0	0	0	0
8.4	0	0	0	0

	QUES A	STION B	NINE	D
9.1	0	0	0	0
9.2	0	0	0	0
9.3	0	0	0	0
9.4	0	0	0	0

	QUE:	STION B	TEN C	D
10.1	0	0	0	0
10.2	0	0	0	0
10.3	0	0	0	0
10.4	0	0	0	0

## **HIGHER TIER**

Instructions on how to complete this answer sheet are given on the question paper. Please make sure you follow them carefully.

Questions ONE and TWO: Choose one answer for each of the parts 1 to 4.

QUESTION ONE	1	2	3	4
chlorine	0	0	0	0
magnesium carbonate	0	0	0	0
nitric acid	0	0	0	0
zinc sulphate	0	0	0	0

QUESTION TWO	1	2	3	4
carbon dioxide	0	0	0	0
chloroethene	0	0	0	0
hydrogen chloride	0	0	0	0
hydrogen cyanide	0	0	0	0

Questions THREE and FOUR: Choose two answers for each question.

QUESTION THREE		1
de-ionised water is hard water	0	3
some of the hardness is removed from Sample M when it is boiled	0	
the hardest water is Sample L	0	-
when soap solution is first added to Sample K, a scum will form	0	
when water is boiled, all the hardness in the water is removed	0	

	QUESTION	FOUR	High ten deste	
				0
K				0
L				0
М				0
N				0

Questions FIVE to TEN: Choose one answer for each of the parts 1 to 4.

	QUES	STION B	FIVE	D
5.1	0	0	0	0
5.2	0	0	0	0
5.3	0	0	0	0
5.4	0	0	0	0

	OUE	STION	VISIX	
	A	В	C	D
6.1	0	0	0	0
6.2	0	0	0	0
6.3	0	0	0	0
6.4	0	0	0	0

QUESTION SEVEN						
	Α	В	С	D		
7.1	0	0	0	0		
7.2	0	0	0	0		
7.3	0	0	0	0		
7.4	0	0	0	0		

QUESTION EIGHT						
	Α	В	C	D		
8.1	0	0	0	0		
8.2	0	0	0	0		
8.3	0	0	0	0		
8.4	0	0	- 0	0		

	QUES A	STION B	NINE C	D
9.1	0	0	0	0
9.2	0	0	0	0
9.3	. 0	0	0	0
9.4	0	0	0	0)

	QUES A	STION B	TEN C	D
10.1	0	0	0	0
10.2	0	0	0	0
10.3	0	0	. 0	0
10.4	0	0	0	0

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