

Centre:

Test Series: 06 MARCH 2006 SERIES H

Module: 346005 METALS

Candidate Number:

UCI:

Candidate Name:

For completion by the Examination Invigilator. Please fill this oval if the candidate is absent:

FOUNDATION TIER

Instructions on how to complete this answer sheet are given on the question paper. Please make sure you follow them carefully.

copper

Questions ONE to FIVE: Choose one answer for each of the parts 1 to 4.

Questions ONE to FIVE: Choose o	ne ansv	ver	tor	eac
QUESTION ONE	1	2	3	4
aluminium	0	0	0	0
copper	0	0	0	0
iron	0	0	0	0
mercury	0	0	0	0
QUESTION THREE	1	2	3	4
chromium	0	0	0	0
copper	0	0	0	0
gold	0	0	0	0
magnesium	0	0	0	0
QUESTION FIVE	1	2	3	4
metal W	0	0	0	0
metal X	0	0	0	0
metal Y	0	0	0	0
metal Z	0	0	0	0

QUESTION TWO	1	2	3	4
a base	0	0	0	0
a salt	0	0	0	0
an acid	0	0	0	0
an alkali	0	0	0	.0
QUESTION FOUR	1	2	3	4
bromine	0	0	0	0
carbon	0	0	0	0
chlorine	0	0	0	0

Questions SIX and SEVEN: Choose two answers for each question.

QUESTION SIX all the metals are in Group 1 and the central block

argon is in Group 5

iron and platinum are transition metals

metal alloys are in Group 3

more than 3/4 of the elements are metals

QUEUTION OIX			
ntral block		0	
		0	
		0	
		0	
		0	

QUESTION SEVEN

it does not go rusty

it has a shiny surface

it is a fairly expensive metal

it is a good conductor of electricity

it is easy to bend and shape

Questions EIGHT to TEN: Choose one answer for each of the parts 1 to 4.

C	T D			
8.1	0	0	0	0
8.2	0	0	0	0
8.3	0	0	0	0
8.4	0	0	0	0

	A	STION B	C	D
9.1	0	0	0	0
9.2	0	0	0	0
9.3	0	0	0	0
9.4	0	0	0	0

	QUE:	STION B	C	D
10.1	0	0	0	0
10.2	0	0	0	0
10.3	0	0	0	0
10.4	0	0	0	0

HIGHER TIER

Instructions on how to complete this answer sheet are given on the question paper. Please make sure you follow them carefully.

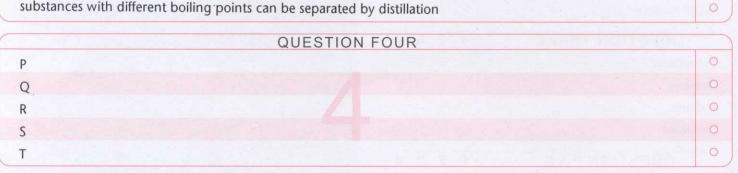
Questions ONE and TWO: Choose one answer for each of the parts 1 to 4.

_	decire of the difference of the other	CHIL	24401	101	Ou
	QUESTION ONE	1	2	3	4
	biodegrade	0	0	0	0
	combine	0	0	0	0
	condense	0	0	0	0
	neutralise	0	0	0	0

the parts 1 to 4.				
QUESTION TWO	1	2	3	4
ammonia	0	0	0	0
carbon dioxide	0	0	0	0
ethene	0	0	0	0
oxygen	0	0	0	0

Questions THREE and FOUR: Choose two answers for each question.

QUESTION THREE
air is a compound
carbon dioxide is an element
concrete is a mixture
crude oil is a mixture of several elements
substances with different boiling points can be separated by distillation



Questions FIVE to TEN: Choose one answer for each of the parts 1 to 4.

QUESTION FIVE A B C D							
5.1	0	0	0	0			
5.2	0	0	0	0			
5.3	0	0	0	0			
5.4	0	0	0	0)			

QUESTION SIX A B C D								
6.1	0	0	0	0				
6.2	O	0	0	. 0				
6.3	0	0	0	0				
6.4	0	0	0	0				

Q	UES	TION S	SEVE	N
	Α	В	C	D
7.1	0	0	0	0
7.2	0	0	0	0
7.3	0	0	0	0
7.4	0	0	0	0

C	UES A	TION	EIGH C	T D
8.1	0	0	0	0
8.2	0	0	0	0
8.3	0	0	0	0
8.4	0	0	0	0

	QUES A	TION B	NINE C	D
9.1	0	0	0	0
9.2	0	0	0	0
9.3	0	0	0	0
9.4	0	0	0	0

	QUES A	STION B	TEN C	D
10.1	0	0	0	0
10.2	0	0	0	0
10.3	0	0	0	0
10.4	0	0	0	0

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