

Mark scheme June 2005

GCSE

Chemistry (Modular)

Module 21

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Aqueous and Organic Chemistry: Foundation Tier

Question No.	KEY
One	1 – ammonium nitrate
	2 – oxygen
	3 – carbon dioxide
	4 – chlorine
Two	1 – solution
	2 – solvent
	3 – solute
	4 – solid
TO	1 T
Three	1-L
	2 – K
	3 - M
	4 – J
Four	1 – P
roui	$\begin{vmatrix} 1 - F \\ 2 - S \end{vmatrix}$
	$\begin{vmatrix} 2-3\\3-Q \end{vmatrix}$
	4 – R
Five	1 – anhydrous iron chloride
	2 – insoluble lead sulphate
	3 – potassium chloride
	4 – zinc sulphate
Six	heating oil is more expensive, per kg, than natural gas
	natural gas, when burned, is the cleanest fuel
Seven	a white precipitate forms when sodium carbonate solution is added to L
	the hardest water is L
Eight	8.1 – C, 8.2 – C, 8.3 – A, 8.4 – B
Nine	9.1 – B, 9.2 – A, 9.3 – B, 9.4 – B
Ten	10.1 – B, 10.2 – A, 10.3 – D, 10.4 – A

Aqueous and Organic Chemistry: Higher Tier

Question No.	KEY
One	 1 – anhydrous iron chloride 2 – insoluble lead sulphate 3 – potassium chloride 4 – zinc sulphate
	T Zine Sulphate
Two	1 – hydrogen chloride 2 – hydrogen cyanide 3 – silver chloride 4 – hydrogen
Three	a white precipitate forms when sodium carbonate solution is added to L the hardest water is L
Four	the diagram represents a molecule of polyvinvychloride polyvinylchloride will soften when it is heated and then harden as it cools
Five	5.1 – C, 5.2 – C, 5.3 – A, 5.4 – B
	5,612 5,616 13,611 2
Six	6.1 – B, 6.2 – A, 6.3 – B, 6.4 – B
Seven	7.1 – B, 7.2 – A, 7.3 – D, 7.4 – A
Eight	8.1 – A, 8.2 – A, 8.3 – A, 8.4 – D
Nine	9.1 – D, 9.2 – A, 9.3 – A, 9.4 – C
Ton	10.1 C 10.2 C 10.2 C 10.4 C
Ten	10.1 – C, 10.2 – C, 10.3, – C, 10.4 – C