

Mark scheme June 2004

GCSE

Chemistry (Modular)

Module 21

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Question No.	КЕҮ
One	1 – water (vapour)
	$2 - \operatorname{carbon}$
	3 – carbon dioxide 4 – carbon monoxide
Two	1 – water
	2 – chlorine
	3 – ethanol
	4 – ammonium nitrate
Three	1 – crystallise
	2 – dissolve
	3 – evaporate 4 – condense
Four	1 – water is passed through filter beds
	2 – chlorine is added
	3 – to kill any bacteria
	4 – carbon dioxide is added under pressure
D'	
Five	1 – anhydrous iron chloride
	2 – insoluble lead sulphate 3 – soluble zinc sulphate
	4 – soluble sodium nitrate
Six	fish only survive if the water contains sufficient dissolved oxygen
	oxygen is carried around the body by the blood
Seven	ethanoic acid is a weak acid
	sodium hydroxide is fully ionised in water
Eight	8.1 - D, 8.2 - B, 8.3 - B, 8.4 - B
Nine	9.1 - A, 9.2 - A, 9.3 - C, 9.4 - B
Ten	10.1 - B, 10.2 - A, 10.3 - B, 10.4 - B

Aqueous & Organic Chemistry: Foundation Tier

Aqueous & Organic Chemistry: Higher Tier

Question No.	КЕҮ
One	1 – anhydrous iron chloride
	2 – insoluble lead sulphate
	3 – soluble zinc sulphate
	4 – soluble sodium nitrate
Two	1 – citric acid
1 WO	2 - methanoic acid
	3 – phosphoric acid
	4 – ethanoic acid
Three	ethanoic acid is a weak acid
	sodium hydroxide is fully ionised in water
Four	Р
	Q
Five	5.1 – D, 5.2 – B, 5.3 – B, 5.4 – B
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Six	6.1 – A, 6.2 – A, 6.3 – C, 6.4 – B
Seven	7.1 – B, 7.2 – A, 7.3 – B, 7.4 – B
Seven	7.1 - D, 7.2 - A, 7.3 - D, 7.4 - D
Eight	8.1 - D, 8.2 - A, 8.3 - B, 8.4 - B
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Nine	9.1 - A, 9.2 - D, 9.3 - B, 9.4 - B
Ten	10.1 - D, 10.2 - B, 10.3 - A, 10.4 - D