

Mark scheme June 2004

GCSE

Science (Modular) **Double Award and Chemistry**

Module 05

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Metals: Foundation Tier

Question No.	KEY
One	1 – copper
	2 – aluminium
	3 – iron 4 – gold
	4 - goid
Two	1 – potassium chloride
	2 – nitric acid
	3 – copper oxide
	4 – zinc hydroxide
T1	
Three	1 – neutralisation 2 – corrosion
	3 – electrolysis
	4 – reduction
	1.00000000
Four	1 – P
	2 – Q
	3 - R
	4-S
Five	1 – metal Z
Tive	2 – metal Y
	3 – metal X
	4 – metal W
Six	it can be bent and shaped
	it is a strong metal
Seven	copper chloride is decomposed into its elements
Seven	copper ions move to the negative electrode
	topper tone move to me negative electrode
Eight	8.1 – D, 8.2 – B, 8.3 – B, 8.4 – C
Nine	9.1 – B, 9.2 – B, 9.3 – C, 9.4 – A
T	10.1 C 10.2 D 10.2 C 10.4 D
Ten	10.1 – C, 10.2 – D, 10.3 – C, 10.4 – B

Metals: Higher Tier

Question No.	KEY
One	1 – metal Z
	2 – metal Y
	3 – metal X
	4 – metal W
Two	1
1 WO	1 – copper 2 – copper oxide
	3 – oxygen
	4 – carbon monoxide
Three	copper chloride is decomposed into its elements
	copper ions move to the negative electrode
Four	at the end of electrolysis the negative electrode will be larger
	copper atoms from the impure electrode lose electrons to form copper ions
D.	
Five	5.1 – D, 5.2 – B, 5.3 – B, 5.4 – C
Six	(1 D (2 D (2 C (4 A
S1X	6.1 – B, 6.2 – B, 6.3 – C, 6.4 – A
Seven	7.1 – C, 7.2 – D, 7.3 – C, 7.4 – B
Beven	7.1 C, 7.2 D, 7.3 C, 7.4 D
Eight	8.1 – C, 8.2 – C, 8.3 – C, 8.4 – B
8	2, 2.2 2, 510 2, 51. 2
Nine	9.1 – A, 9.2 – D, 9.3 – B, 9.4 – B
Ten	10.1 – A, 10.2 – C, 10.3 – B, 10.4 – A