



General Certificate of Secondary Education

Science B 4462 / Chemistry 4421

CHY1F Unit Chemistry 1

Mark Scheme

2007 examination - January series

Mark schemes are prepared by the Principal Examiner and considered, together with the relevant questions, by a panel of subject teachers. This mark scheme includes any amendments made at the standardisation meeting attended by all examiners and is the scheme which was used by them in this examination. The standardisation meeting ensures that the mark scheme covers the candidates' responses to questions and that every examiner understands and applies it in the same correct way. As preparation for the standardisation meeting each examiner analyses a number of candidates' scripts: alternative answers not already covered by the mark scheme are discussed at the meeting and legislated for. If, after this meeting, examiners encounter unusual answers which have not been discussed at the meeting they are required to refer these to the Principal Examiner.

It must be stressed that a mark scheme is a working document, in many cases further developed and expanded on the basis of candidates' reactions to a particular paper. Assumptions about future mark schemes on the basis of one year's document should be avoided; whilst the guiding principles of assessment remain constant, details will change, depending on the content of a particular examination paper.

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MARK SCHEME

Information to Examiners

1. General

The mark scheme for each question shows:

- the marks available for each part of the question
- the total marks available for the question
- the typical answer or answers which are expected
- extra information to help the Examiner make his or her judgement and help to delineate what is acceptable or not worthy of credit or, in discursive answers, to give an overview of the area in which a mark or marks may be awarded.

The extra information is aligned to the appropriate answer in the left-hand part of the mark scheme and should only be applied to that item in the mark scheme.

At the beginning of a part of a question a reminder may be given, for example: where consequential marking needs to be considered in a calculation; or the answer may be on the diagram or at a different place on the script.

In general the right hand side of the mark scheme is there to provide those extra details which confuse the main part of the mark scheme yet may be helpful in ensuring that marking is straightforward and consistent.

2. Emboldening

- 2.1** In a list of acceptable answers where more than one mark is available 'any **two** from' is used, with the number of marks emboldened. Each of the following lines is a potential mark.
- 2.2** A bold **and** is used to indicate that both parts of the answer are required to award the mark.
- 2.3** Alternative answers acceptable for a mark are indicated by the use of **or**. (Different terms in the mark scheme are shown by a / ; e.g. allow smooth / free movement.)

3. Marking points

3.1 Marking of lists

This applies to questions requiring a set number of responses, but for which candidates have provided extra responses. The general principle to be followed in such a situation is that 'right + wrong = wrong'.

Each error/contradiction negates each correct response. So, if the number of error/contradictions equals or exceeds the number of marks available for the question, no marks can be awarded.

However, responses considered to be neutral (indicated as * in example 1) are not penalised.

Example 1: What is the pH of an acidic solution? (1 mark)

Candidate	Response	Marks awarded
1	4,8	0
2	green, 5	0
3	red*, 5	1
4	red*, 8	0

Example 2: Name two planets in the solar system. (2 marks)

Candidate	Response	Marks awarded
1	Pluto, Mars, Moon	1
2	Pluto, Sun, Mars, Moon	0

3.2 Use of chemical symbols/formulae

If a candidate writes a chemical symbol/formula instead of a required chemical name, full credit can be given if the symbol/formula is correct and if, in the context of the question, such action is appropriate.

3.3 Marking procedure for calculations

Full marks can be given for a correct numerical answer, as shown in the column 'answers', without any working shown.

However if the answer is incorrect, mark(s) can be gained by correct substitution / working and this is shown in the 'extra information' column;

3.4 Interpretation of 'it'

Answers using the word 'it' should be given credit only if it is clear that the 'it' refers to the correct subject.

3.5 Errors carried forward

Any error in the answers to a structured question should be penalised once only.

Papers should be constructed in such a way that the number of times errors can be carried forward are kept to a minimum. Allowances for errors carried forward are most likely to be restricted to calculation questions and should be shown by the abbreviation e.c.f. in the marking scheme.

3.6 Phonetic spelling

The phonetic spelling of correct scientific terminology should be credited **unless** there is a possible confusion with another technical term.

3.7 Brackets

(.....) are used to indicate information which is not essential for the mark to be awarded but is included to help the examiner identify the sense of the answer required.

3.8 Unexpected Correct Answers not in the Mark Scheme

The Examiner should use professional judgement to award credit where a candidate has given an unexpected correct answer which is not covered by the mark scheme. The Examiner should consult with the Team Leader to confirm the judgement. The Team Leader should pass this answer on to the Principal Examiner with a view to informing all examiners.

Question 1

	answers	extra information	mark
(a)	made of one sort of atom	accept it is in the periodic table accept it only has lithium atoms	1
(b)	nucleus labelled correctly		1
	electron labelled correctly		1
total			3

Question 2

	answers	extra information	mark
(a)	pressing		1
(b)(i)	water	accept hydrogen oxide / H ₂ O	1
(ii)	filtration or description of this process	accept decanting or similar description of a means of separating the seeds and skins from the liquids accept sieving do not accept distillation / evaporation	1
total			3

Question 3

	answers	extra information	mark
(a)	tungsten		1
	has the high(est) melting point	accept that metals other than tungsten are likely to melt	1
(b)	argon		1
	is an unreactive gas	accept that gases other than argon are reactive accept that argon is a noble gas or in Group 0	1
total			4

Question 4

	answers	extra information	mark
(a)	1 / one		1
	3 / three		1
(b)(i)	(wear safety) glasses / masks / gloves	accept do not handle hot objects / use tongs or accept other safety points such as, 'tie hair back'	1
(ii)	calcium oxide	accept quicklime	1
(iii)	carbon dioxide / CO ₂ is given off	accept a gas is given off	1
(c)	any two from: <ul style="list-style-type: none"> • conserves (saves) resources • avoids dumping / landfill • lowers energy requirements / save energy • stops quarries opening • less carbon dioxide / carbon emissions / greenhouse gases • lowers the price of glass • glass is non-biodegradable 	accept reduces waste do not accept speeds up the process do not accept cheaper process	2
(d)(i)	any two from: <ul style="list-style-type: none"> • dust / atmospheric pollution • noise • eyesore • destroys habitats • (extra) traffic • any other named pollution or description of pollution 		2

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	answers	extra information	mark
(ii)	any one from: <ul style="list-style-type: none">• employment• increases local trade / makes money• future development of quarry – recreation / park / lake / reservoir	do not accept cheaper houses / land / new roads	1
total			10

Question 5

	answers	extra information	mark
(a)(i)	poly(ethene)	accept polythene	1
(ii)	cracking		1
(iii)	hydrogen		1
(b)(i)	bar labelled 9		1
	bar drawn to correct height		1
(ii)	(boiling point) increases		1
(iii)	heat / evaporate (the crude oil)	accept separate by boiling point	1
	cool / condense (hydrocarbons at different temperatures)	accept smaller molecules go to top / larger molecules stay at bottom accept fractional distillation for two marks or distillation / fractionation for one mark	1

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	answers	extra information	mark
(c)	<p>yes</p> <p>any two from:</p> <ul style="list-style-type: none"> • because plastic does not biodegrade or running out of space for landfills or land cannot be used for a long time • it provides heat energy • which can be used to generate electricity / heat homes or greenhouses • any other advantage of burning • any other disadvantage of landfill <p>or</p> <p>no</p> <ul style="list-style-type: none"> • burning plastic produces carbon dioxide / carbon emissions / toxic gases • causes global warming / climate change / increase greenhouse effect / global dimming / acid rain • any other disadvantage of burning • any other advantage of landfill 	<p>accept landfill does not produce carbon dioxide / carbon emissions</p>	2
total			10

Question 6

	answers	extra information	mark
(a)	radioactivity (keeps the core hot)	accept half-life of radioactive elements has proved the Earth is older than 400 million years accept the Earth is not cooling do not accept fossil / rock evidence	1
(b)	any two from: <ul style="list-style-type: none"> the shapes of the two continents fit together (like a jigsaw) OWTTE the <u>same</u> type of rocks have been found the <u>same</u> fossils have been found rising magma rising through a gap under the Atlantic 	do not accept the continents / they are the same shape accept 'rocks match' accept 'fossils match'	2
(c)	any two from: <ul style="list-style-type: none"> earthquakes volcanoes idea of distance between America and Europe / Africa is increasing e.g. continental drift oceanic ridges formation of (new) mountain ranges formation of (new) islands magnetic stripes tsunamis 	accept seismic waves accept ocean floor spreading accept fold mountains or mountains increasing in height	2
total			5

Question 7

	answers	extra information	mark
(a)	oxygen and nitrogen		1
	20 – 21 % and 78 – 80%	accept any two correct responses in the correct space for one mark	1
(b)(i)	acid rain	accept toxic gas or consequence of acid rain	1
(ii)	idea of the removal or use of sulfur dioxide gas (from the waste gases)	do not accept remove sulfur from coal	1
(iii)	oxygen	accept O ₂	1
	water	accept H ₂ O accept hydrogen oxide / steam	1
(c)	any two from: <ul style="list-style-type: none"> • it's a 'greenhouse gas' or increase greenhouse effect • causes global warming or increase in the Earth's temperature • sea-levels rise or flooding • climate change • (polar) ice-caps melt • extension of deserts 	accept action of a 'greenhouse gas' mention of ozone / acid rain / global dimming = max 1 mark	2
(d)	idea trap / store / lock the carbon dioxide		1
	in the oil reservoir or under the sea <u>bed</u>	do not accept 'into the oil' / 'under the sea'	1
total			10