

Candidate Forename						Candidate Surname				
Centre Number						Candidate Number				

**OXFORD CAMBRIDGE AND RSA EXAMINATIONS
GENERAL CERTIFICATE OF SECONDARY EDUCATION**

J649

B482/01

APPLIED SCIENCE: DOUBLE AWARD

**Unit 2: Science for the needs of society
(Foundation Tier)**

**WEDNESDAY 9 JUNE 2010: Afternoon
DURATION: 1 hour**

SUITABLE FOR VISUALLY IMPAIRED CANDIDATES

**Candidates answer on the Question Paper
A calculator may be used for this paper**

OCR SUPPLIED MATERIALS:

None

OTHER MATERIALS REQUIRED:

**Pencil
Ruler (mm/cm)**

READ INSTRUCTIONS OVERLEAF

INSTRUCTIONS TO CANDIDATES

- Write your name clearly in capital letters, your Centre Number and Candidate Number in the boxes on the first page.
- Use black ink. Pencil may be used for graphs and diagrams only.
- Read each question carefully and make sure that you know what you have to do before starting your answer.
- Answer ALL the questions.
- Write your answer to each question in the space provided, however additional paper may be used if necessary.

INFORMATION FOR CANDIDATES

- The number of marks is given in brackets [] at the end of each question or part question.
- The total number of marks for this paper is 60.
- The marks allocated and the spaces provided are a good indication of the length of answers required.

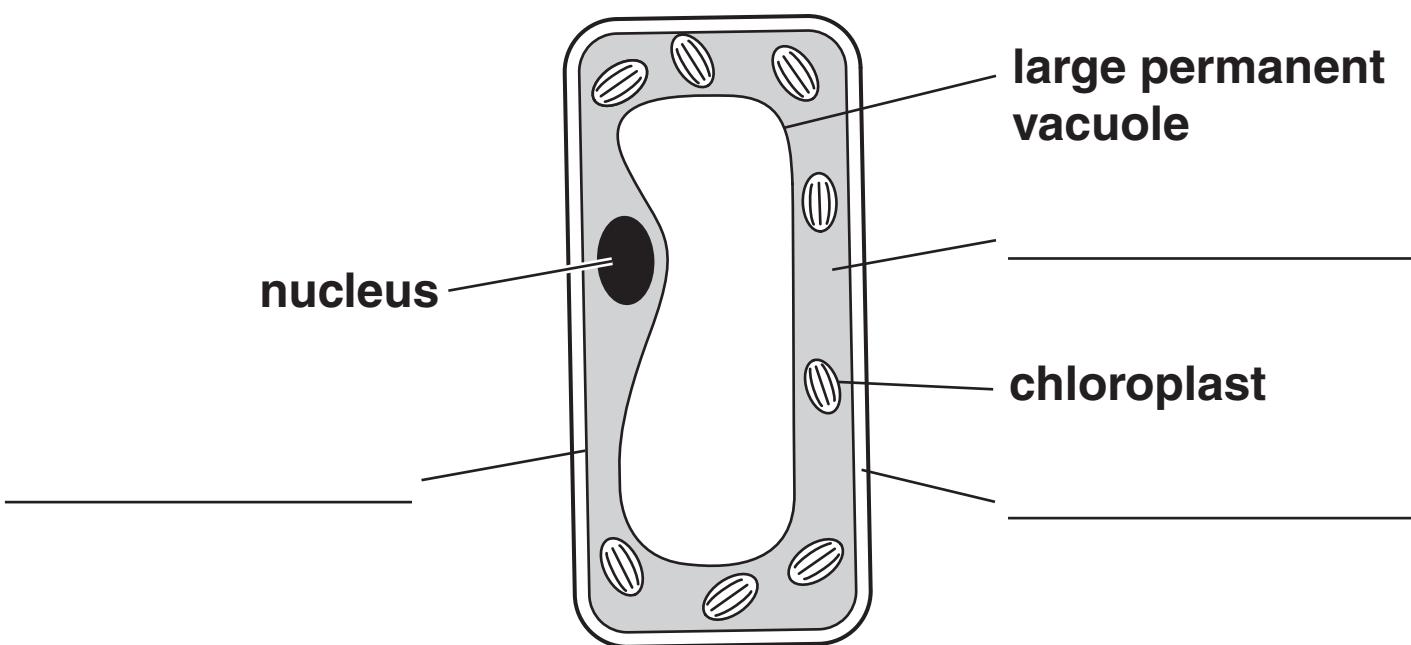
BLANK PAGE

Answer ALL the questions.

1 Kiara is a biologist. She works with plants.

(a) Kiara looks at a plant cell using a microscope.

(i) Complete the labels on the diagram of the plant cell.



[3]

(ii) Write down TWO parts of the plant cell that are NOT found in an animal cell.

[2]

(iii) Which part of the cell contains the genetic information?

[1]

- (iv) Which part of the cell absorbs light energy for photosynthesis?

[1]

- (b) Kiara selectively breeds the plants to improve them.

Complete the sentences that describe how Kiara selectively breeds the plants.

Choose from the words below.

CHARACTERISTICS

CHROMOSOMES

CROSS BREED

GROW

OFFSPRING

To breed a better plant, Kiara starts by selecting

plants with the best _____

She can then _____ the plants.

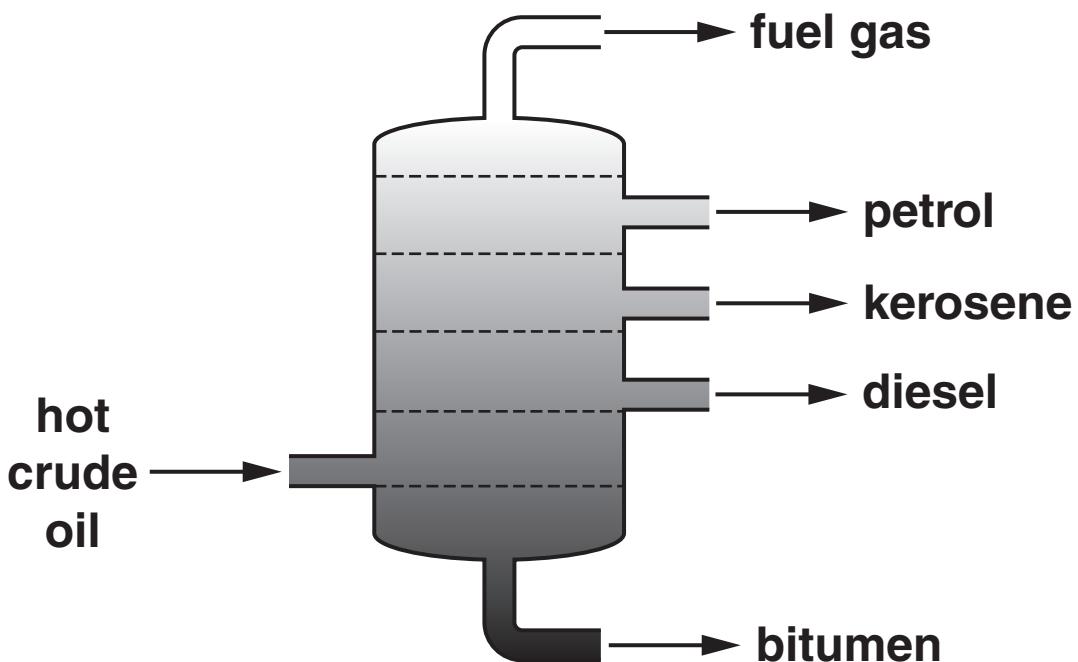
Kiara repeats the process after the plants have

produced _____

[3]

[Total: 10]

- 2 The diagram shows how crude oil is separated into useful chemicals.



(a) (i) What is the name for this process?

Put a **ring** around the correct answer.

COMBUSTION

FILTRATION

FRACTIONAL DISTILLATION

OXIDATION

[1]

(ii) Which of the following statements about this process are TRUE and which are FALSE?

Put a tick (✓) in one box in each row.

TRUE FALSE

**All of the products shown
on the diagram are used as
fuels.**

**The chemicals produced
are fine, speciality
chemicals.**

**The main raw material is
non-renewable.**

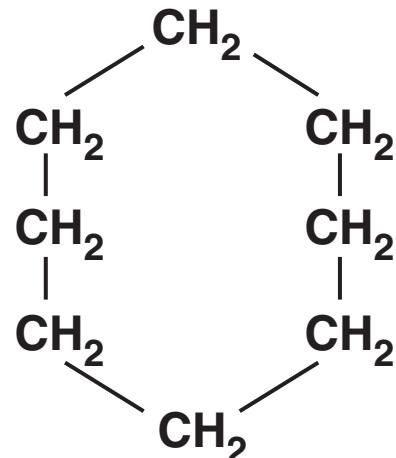
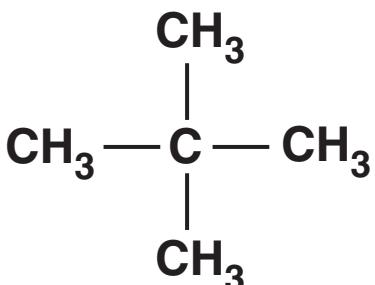
**All the boiling points of the
chemicals are the same.**

**The compounds named on
the diagram are organic
chemicals.**

[3]

(b) Some of the products of this process are used to make petrol.

The diagram shows the structure of two of the molecules in petrol.



(i) Give the NAMES of the two ELEMENTS that the molecules contain.

_____ and _____

[2]

(ii) Which of the following is the best description of petrol?

Put a tick (\checkmark) in one box next to the correct answer.

a pure compound

a pure element

a mixture of elements

a mixture of compounds

[1]

- (c) Some of the chemicals from crude oil are used to make new materials.

Which materials are made from chemicals in crude oil?

Put a **ring** around the two correct answers.

CHLORINE

COTTON

DYES

PLASTIC PACKAGING

SALTS

[2]

- (d) Crude oil is mined off the British coast near Aberdeen.

Some local people like having large mining operations near where they live.

Some local people do not.

Give one advantage and one disadvantage of living near a large mining operation.

advantage _____

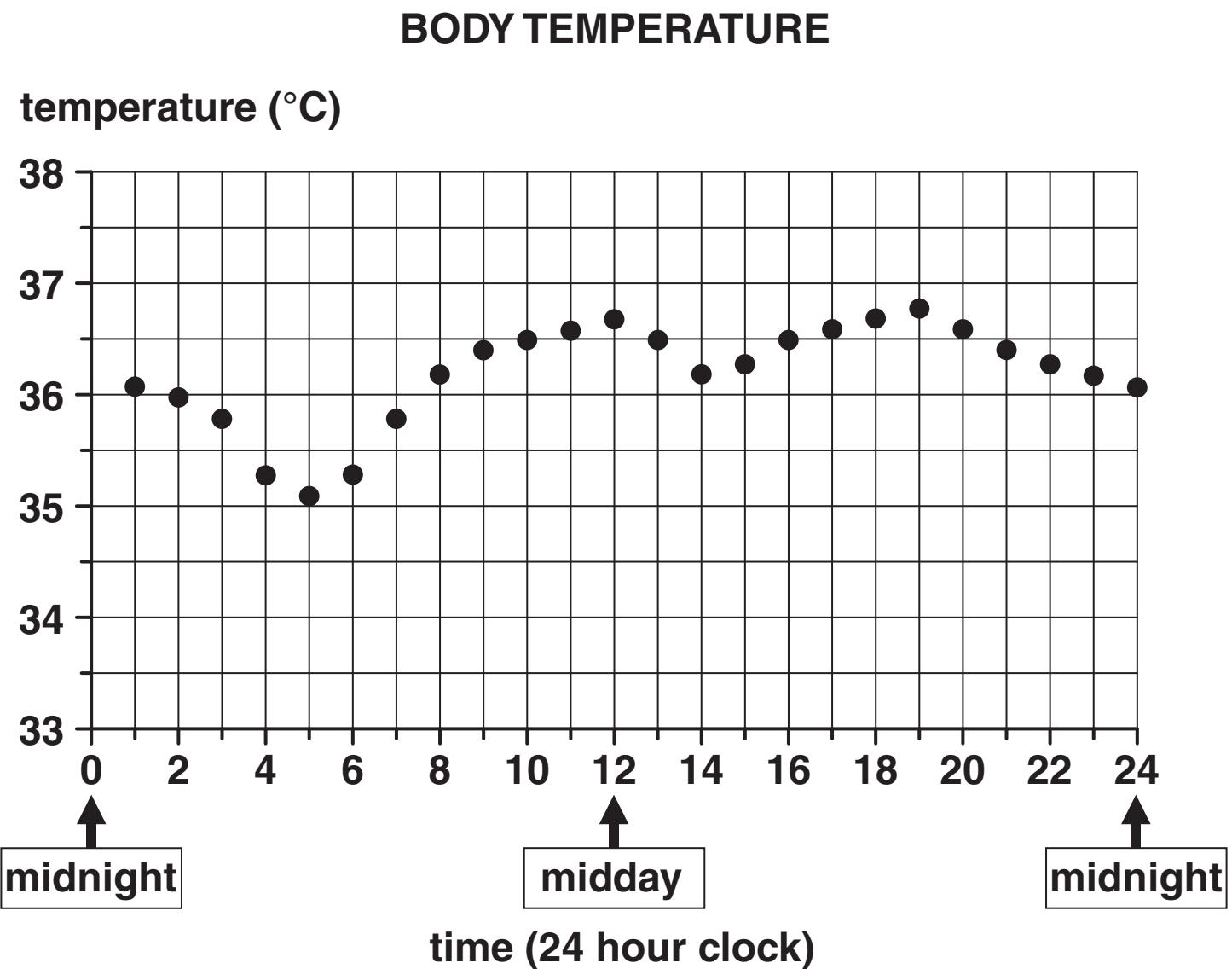
disadvantage _____ [2]

[Total: 11]

3 Joe is in hospital.

A nurse attaches a sensor to Joe that automatically records his body temperature every hour.

The graph shows his temperature over 24 hours.



- (a) Give one advantage of using a machine to measure Joe's temperature rather than using a normal thermometer.**

[1]

(b) Joe sleeps from midnight until 5 o'clock.

(i) What happens to Joe's temperature while he sleeps?

[1]

(ii) Joe has a short sleep at midday.

Suggest what time Joe wakes up.

Explain your reasoning.

time _____

reason _____

[2]

(c) The nurse tells Joe that the average body temperature is usually 37°C .

(i) DRAW A LINE across the graph to show this temperature. [1]

(ii) How does Joe's temperature compare to the average?

[1]

(d) Joe notices that his body temperature increases when he moves around.

(i) What process in Joe's body produces most heat?

Put a **ring** around the correct answer.

CIRCULATION

RESPIRATION

DIGESTION

SWEATING

[1]

(ii) The nurse tells Joe that other changes happen in his body when he moves around.

Which of the following will INCREASE when Joe moves around?

Put a tick (**✓**) in the boxes next to the TWO correct answers.

number of blood cells

breathing rate

pulse rate

total volume of the blood

levels of salt in the blood

[2]

(e) The nurse tells Joe that he has an infection.

The infection is caused by a virus.

Suggest TWO ways that Joe could have caught the infection from another person.

1. _____

2. _____ [2]

[Total: 11]

- 4 A new type of electric car became available in the UK in 2005.**

Read the information about the electric car.

NEW ‘PLUG AND GO’ ELECTRIC CAR

New 2 seater electric cars could solve air pollution problems in cities. The cars can be charged at home or at pay-as-you-charge charging points. Typically, it can take from 30 minutes to a couple of hours to fully charge the car. In 2008 there were 60 garage charging points in the UK compared to thousands of garages selling petrol and diesel.

Car manufacturers have been working hard to find a replacement for petrol and diesel cars. One manufacturer is claiming 90% energy efficiency for one model of electric car.

The table shows how the cars compare to small petrol and diesel cars.

	cost to run (per mile)	top speed (miles per hour)	maximum distance before charging or filling (miles)
electric car	2p	50	48
petrol car	14p	130	630
diesel car	12p	100	800

- (a) The article says that the electric cars could ‘solve air pollution in cities’.

Which of the statements about air pollution from cars are TRUE and which are FALSE?

Put a tick (✓) in one box in each row.

TRUE FALSE

Petrol and diesel engines give off harmful gases.

No fuel is burnt in an electric car.

Air pollution from cars is worse in the countryside than in cities.

Electricity generation produces no harmful gases.

[2]

- (b) Eve works in a town. She has a petrol car but is thinking of changing to an electric car.

Use information from the article to give one advantage to Eve of changing to an electric car.

[1]

- (c) Liz works with Eve but drives long distances around the country as part of her work.

Explain why an electric car would NOT be suitable for Liz.

[2]

- (d) The manufacturers of one electric car claim that its energy efficiency is 90%.

Complete the Sankey diagram to show this information.

Choose from these words.

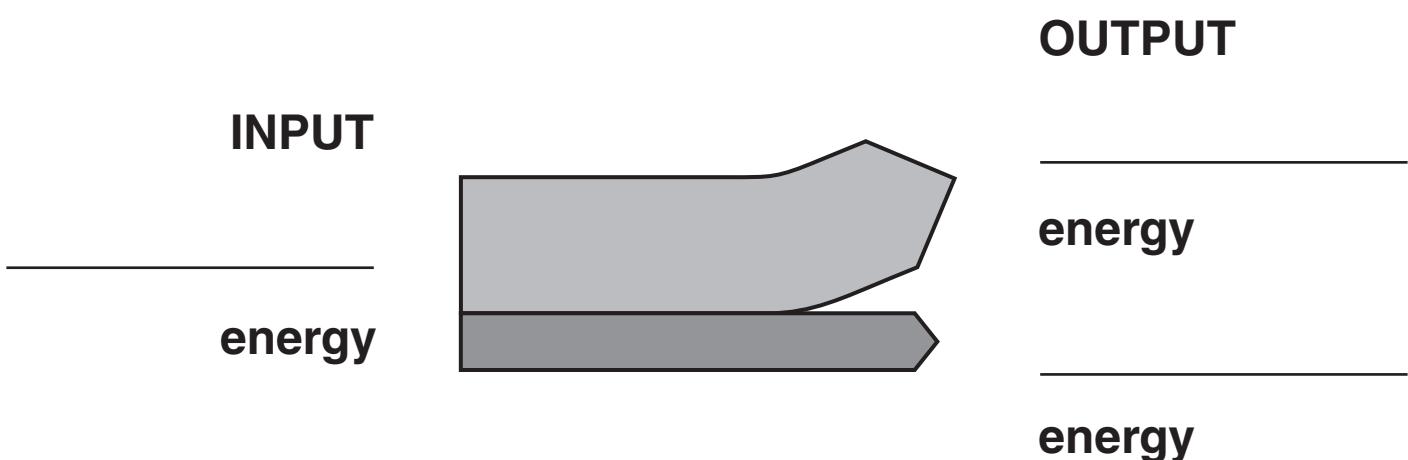
LIGHT

ELECTRICAL

HEAT

MOVEMENT

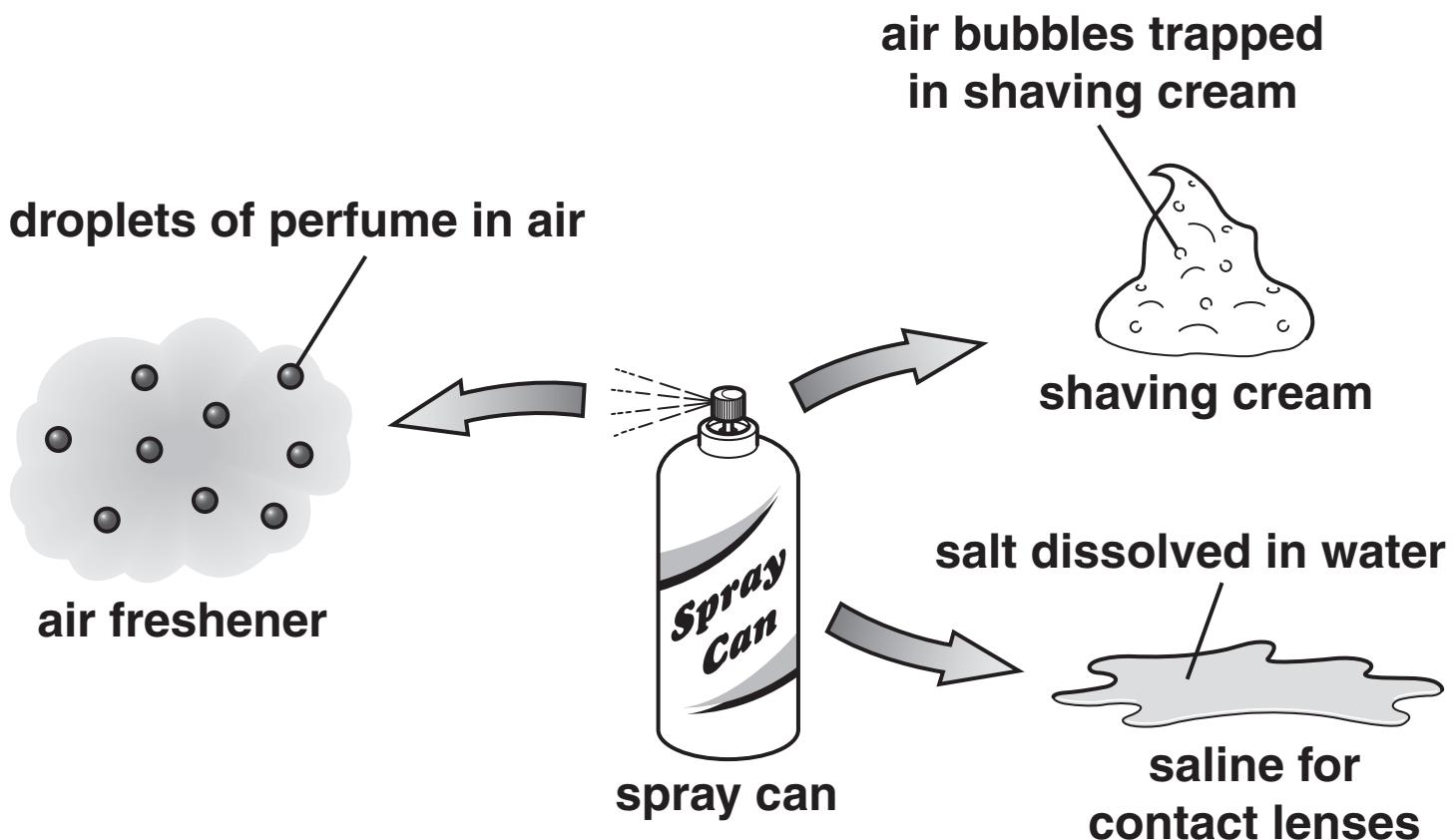
CHEMICAL



[3]

[Total: 8]

5 Some bathroom products are sold in spray cans.



(a) What type of mixture is saline?

Put a **ring** around the correct answer.

SUSPENSION

SOLUTION

SOLVENT

SOLVATION

EMULSION

[1]

- (b) Complete the table to show the continuous and dispersed phases in shaving cream and air freshener.

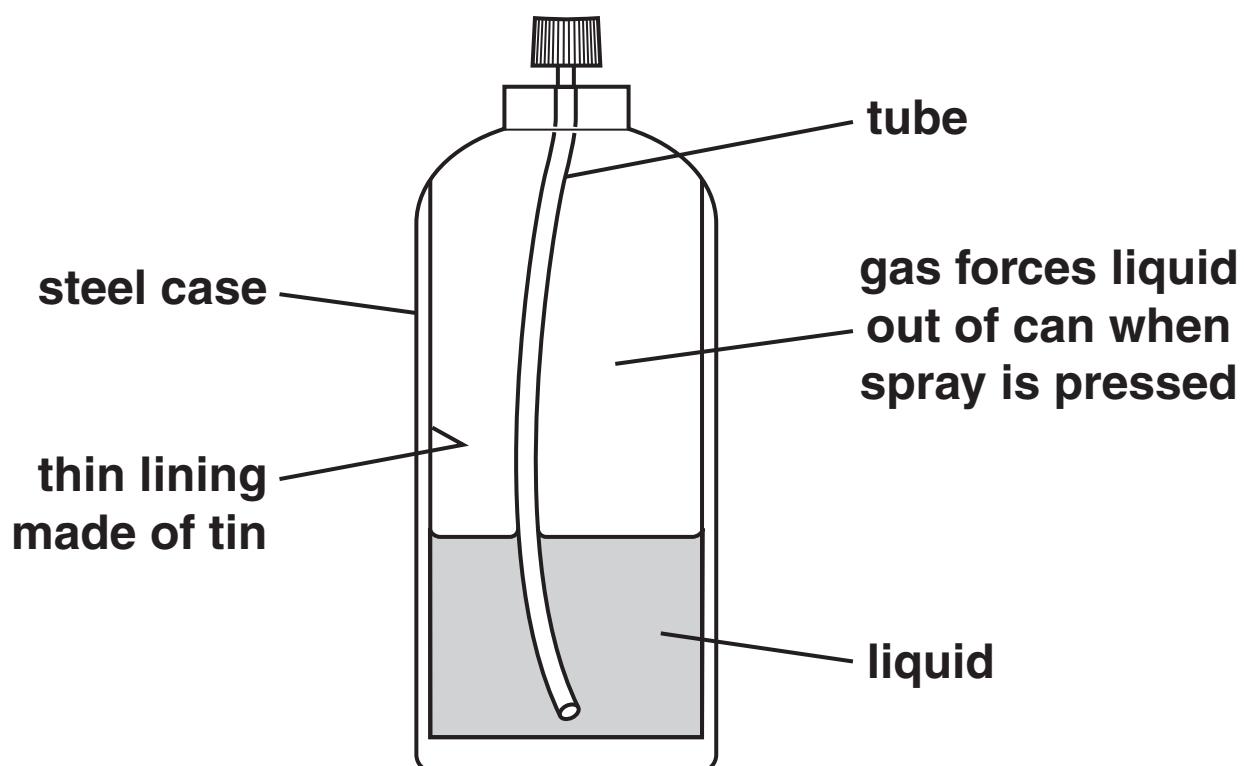
Use these words.

You may use each word once, more than once or not at all.

	<u>SOLID</u>	<u>LIQUID</u>	<u>GAS</u>
	continuous phase	dispersed phase	
shaving cream			
air freshener			

[3]

- (c) The diagram shows the structure of a spray can.



The metals from the spray can can be recycled.

- (i) Give TWO advantages of recycling the metals from spray cans.

[2]

- (ii) Give one reason why it is difficult to recycle the metals from spray cans.

[1]

- (d) The spray can contains a gas to force the contents of the can to spray out.

The table shows some information about gases that were used in spray cans.

formula of gas	boiling point °C
$\text{CF}_3\text{CH}_2\text{F}$	16
CH_3OCH_3	19
CH_3CHF_2	19

- (i) What elements do ALL these gases contain?

[1]

- (ii) Spray cans that use these gases do not work as well if they are stored at 5°C.

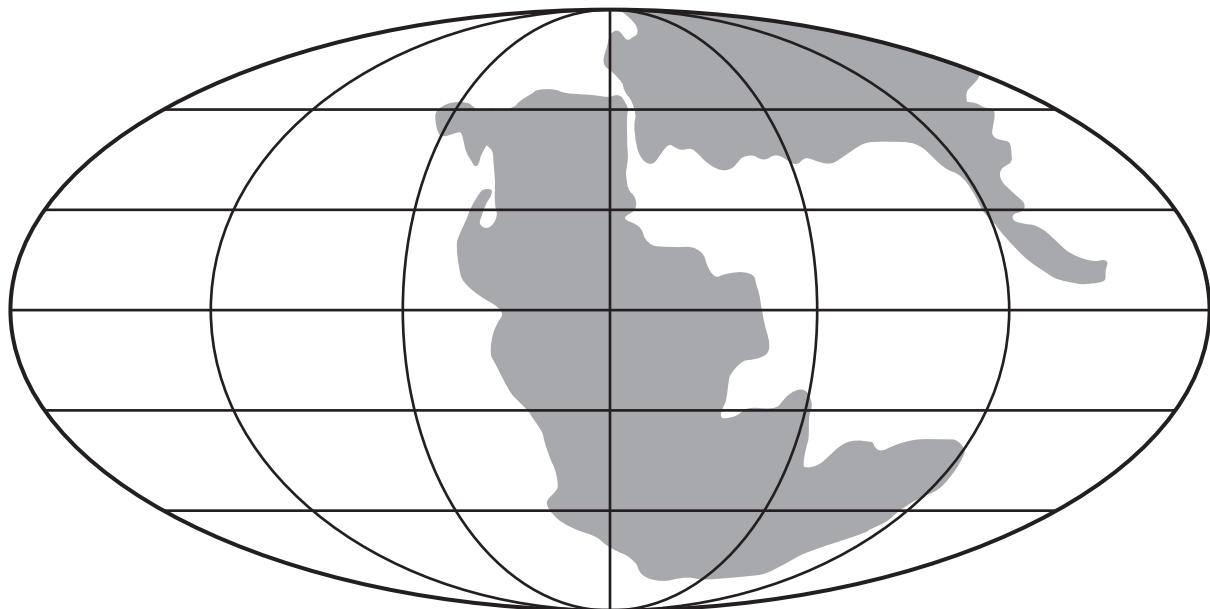
Explain why.

[2]

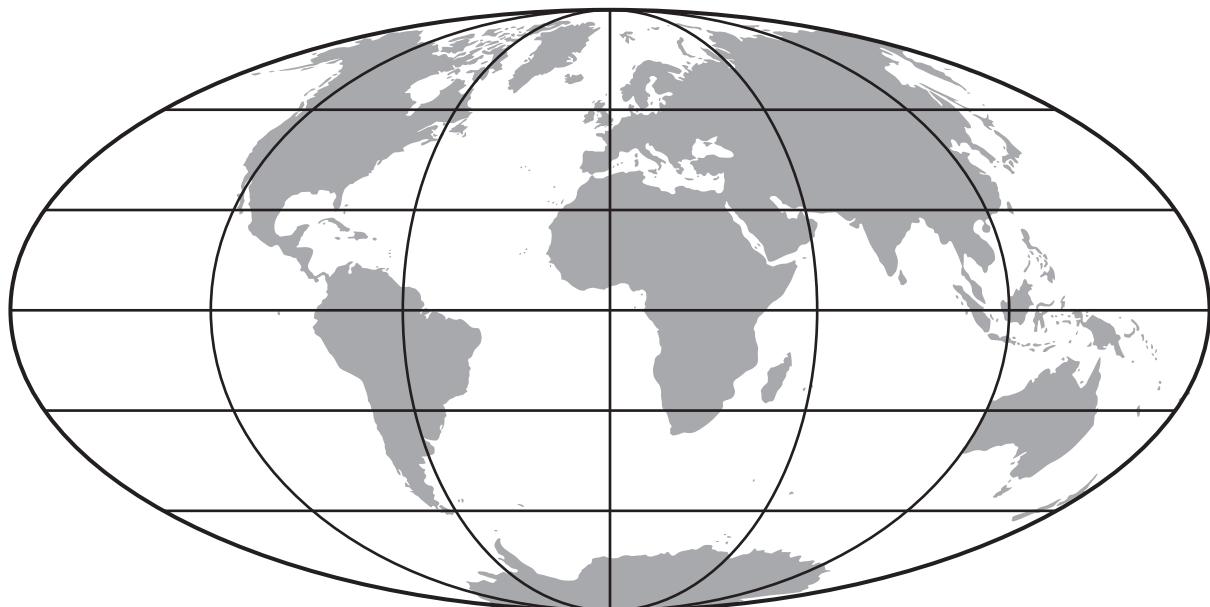
[Total: 10]

6 Scientists think the Earth was very different in the past.

The diagrams show the Earth 190 million years ago and today.



190 million years ago



today

- (a) (i) The arrangement of the continents has changed.**

Explain how the change has happened.

[2]

- (ii) The change in the continents is very slow.**

Other changes to the Earth's surface are caused by the same process.

Write down one other change caused by this process that is also very slow.

[1]

- (b) When life first developed on Earth the atmosphere was mostly carbon dioxide with very little oxygen.**

Now there is very little carbon dioxide and about 20% oxygen.

What process carried out by living things could have brought about this change?

[1]

(c) We could not survive without both oxygen and carbon dioxide in the atmosphere.

(i) Explain why oxygen is necessary.

[1]

(ii) Without carbon dioxide in the atmosphere, the climate of the Earth would not be suitable for life.

Explain why.

[2]

(d) The Earth was formed about 5 000 million years ago.

The Earth is part of the Solar System.

(i) What does the Solar System contain?

[2]

(ii) The Solar System is part of the Universe.

How do scientists think the Universe started?

[1]

[Total: 10]

END OF QUESTION PAPER



Copyright Information

OCR is committed to seeking permission to reproduce all third-party content that it uses in its assessment materials. OCR has attempted to identify and contact all copyright holders whose work is used in this paper. To avoid the issue of disclosure of answer-related information to candidates, all copyright acknowledgements are reproduced in the OCR Copyright Acknowledgements Booklet. This is produced for each series of examinations, is given to all schools that receive assessment material and is freely available to download from our public website (www.ocr.org.uk) after the live examination series.

If OCR has unwittingly failed to correctly acknowledge or clear any third-party content in this assessment material, OCR will be happy to correct its mistake at the earliest possible opportunity.

For queries or further information please contact the Copyright Team, First Floor, 9 Hills Road, Cambridge CB2 1GE.

OCR is part of the Cambridge Assessment Group; Cambridge Assessment is the brand name of University of Cambridge Local Examinations Syndicate (UCLES), which is itself a department of the University of Cambridge.