

GCSE

Additional Science A

General Certificate of Secondary Education

Unit A218/02: Ideas in Context (Higher Tier)

Mark Scheme for January 2013

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This mark scheme is published as an aid to teachers and students, to indicate the requirements of the examination. It shows the basis on which marks were awarded by examiners. It does not indicate the details of the discussions which took place at an examiners' meeting before marking commenced.

All examiners are instructed that alternative correct answers and unexpected approaches in candidates' scripts must be given marks that fairly reflect the relevant knowledge and skills demonstrated.

Mark schemes should be read in conjunction with the published question papers and the report on the examination.

OCR will not enter into any discussion or correspondence in connection with this mark scheme.

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Annotations

Used in the detailed Mark Scheme:

Annotation	Meaning				
/	alternative and acceptable answers for the same marking point				
(1)	separates marking points				
not/reject	answers which are not worthy of credit				
ignore	statements which are irrelevant - applies to neutral answers				
allow/accept	answers that can be accepted				
(words)	words which are not essential to gain credit				
<u>words</u>	underlined words must be present in answer to score a mark				
ecf	error carried forward				
AW/owtte	credit alternative wording / or words to that effect				
ORA	or reverse argument				

Available in scoris to annotate scripts:

Annotation	Meaning			
?	indicate uncertainty or ambiguity			
BOD	benefit of doubt			
CON	contradiction			
×	incorrect response			
ECF	error carried forward			
	draw attention to particular part of candidate's response			
	draw attention to particular part of candidate's response			
~~	draw attention to particular part of candidate's response			

Annotation	Meaning
NBOD	no benefit of doubt
R	reject
✓	correct response
3	draw attention to particular part of candidate's response
Λ	information omitted

Subject-specific Marking Instructions

- a. Accept any clear, unambiguous response (including mis-spellings of scientific terms if they are *phonetically* correct, but always check the guidance column for exclusions).
- b. Crossed out answers should be considered only if no other response has been made. When marking crossed out responses, accept correct answers which are clear and unambiguous.

e.g. for a one-mark question where ticks in the third <u>and</u> fourth boxes are required for the mark:

		₹
		væ [≥]
*	\checkmark	\checkmark
*	₹	\checkmark
This would be worth 1 mark.	This would be worth 0 marks.	This would be worth 1 mark.

c. The list principle:

If a list of responses greater than the number requested is given, work through the list from the beginning. Award one mark for each correct response, ignore any neutral response, and deduct one mark for any incorrect response, e.g. one which has an error of science. If the number of incorrect responses is equal to or greater than the number of correct responses, no marks are awarded. A neutral response is correct but irrelevant to the question.

d. Marking method for tick-box questions:

If there is a set of boxes, some of which should be ticked and others left empty, then judge the entire set of boxes.

If there is at least one tick, ignore crosses and other markings. If there are no ticks, accept clear, unambiguous indications, e.g. shading or crosses. Credit should be given according to the instructions given in the guidance column for the question. If more boxes are ticked than there are correct answers, then deduct one mark for each additional tick. Candidates cannot score less than zero marks.

e.g. if a question requires candidates to identify cities in England:

Edinburgh	
Manchester	
Paris	
Southampton	

the second and fourth boxes should have ticks (or other clear indication of choice) and the first and third should be blank (or have indication of choice crossed out).

Edinburgh			✓			✓	✓	✓	✓	
Manchester	✓	×	✓	✓	✓				✓	
Paris				✓	✓		✓	✓	✓	
Southampton	✓	×		✓		✓	✓		✓	
Score:	2	2	1	1	1	1	0	0	0	NR

- e. For answers marked by levels of response:
 - i. Read through the whole answer from start to finish
 - ii. Decide the level that best fits the answer match the quality of the answer to the closest level descriptor
 - iii. **To determine the mark within the level**, consider the following:

Descriptor	Award mark
A good match to the level descriptor	The higher mark in the level
Just matches the level descriptor	The lower mark in the level

iv. Use the L1, L2, L3 annotations in Scoris to show your decision; do not use ticks.

Quality of Written Communication skills assessed in 6-mark extended writing questions include:

- appropriate use of correct scientific terms
- spelling, punctuation and grammar
- developing a structured, persuasive argument
- selecting and using evidence to support an argument
- considering different sides of a debate in a balanced way
- logical sequencing.

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	Ques	tion	Answer	Marks	Guidance
1	(a)		any two from: sweating; breathing; faeces; tears;	1	two correct responses = 1 mark ignore respiration
	(b)	(i)	pituitary (gland)	1	
		(ii)	idea of change from steady state / change from constant internal environment; reversal of change / bring it back idea / acts to correct;	3	accept normal for 'steady' accept 'change in concentration of blood plasma/water (in the blood) / too much or too little water (in the blood); accept 'too high or too low' as AW for 'change'
			too little water in blood more ADH (produced) / too much water less ADH (produced);		
		(iii)	any two from alcohol reduces ADH production/ADH suppression; less water is (re)absorbed back (into the body) (at the kidney); more water is lost (in the urine)/greater volume of more dilute urine produced	2	

	Quest	ion			Answ	/er		Marks	Guidance
1	(c)			I	I I	atova tha		2	3 correct for 2 marks 2 correct = 1 mark
				always increases	always decreases	stays the same or decreases	stays the same		1 or 0 correct = 0 marks
			sugar				✓		
			urea		✓				
			salt			✓			
	/ IX		, .						
	(d)		(passive overall) movement of molecules/particles; (from a region of their) high concentration to (a region of their) low concentration;					2	ignore references to partially permeable membrane ignore concentration gradient
			1 mark f	or QWC				1	QWC = correct spelling of non-technical terms of two or fewer syllables for a response which engages with the question
	(e)		(10 x 84) ÷ 12 =; ents each we	nok)			2	70 = 2 marks
			70 (patie	TILS CAULI WE	J GN)				allow 7 for (1) mark
		Total						14	

Q	uesti	on	Answer	Marks	Guidance
2	(a)	(i)	at least one ray refracting correctly entering and leaving the prism (1) colours of spectrum in correct order i.e. red, orange, yellow, green, blue, indigo, violet - at least two (1) white light red light red light	2	accept straight lines drawn with out a rule accept a vertical list
		(ii)	both slow down (1) violet light slows down more than red light = 2 marks	2	reject 'refracts' as change of speed ecf from diagram
	(b)		the number of cycles/waves/oscillations/vibrations produced/passing a point) (1) per second (1)	2	
	(c)		Uses 0.03 x 10 ¹⁰ / 3x10 ⁸ / 300,000,000 (m/s);	3	correct answer (20m) scores 3 marks
			Speed of light ÷1.5x10 ⁷ / (wavelength =) <u>speed</u> ; frequency		
			20 (m);		allow ecf from incorrect speed of light
	(d)	(i)	the vibrations are at right angles to the direction of travel of the wave	1	allow information given in form of a diagram including the relative direction

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Q	Question		Answer	Marks	Guidance
2	(e)		both 300 000 km/s	1	allow both 300,000,000 m/s with units
	(f)		any two from: idea of addition / constructive interference; idea of subtraction / destructive interference; wave amplitudes are affected (by interference);	2	allow all marks on diagrams. Summing amplitudes would need to be correct by eye and may assume same scales accept 'waves cancel each other out'
			Total	13	

C	uesti	on	Answer	Marks	Guidance
3	3 (a)		hydrochloric acid AND 2HCI; (1) carbon dioxide AND CO ₂ ; (1)	3	
			water AND H ₂ O; (1)		
	(b)		any two from: potassium is not safe to react with water / too reactive; potassium oxide/carbonate/hydroxide are soluble; titration gives a neutral solution	2	
	(c)	(i)	any two from: add (magnesium carbonate) to (dilute) <u>sulfuric</u> acid; add too much / add excess (magnesium carbonate) idea; filter idea;	2	
		(ii)	any two from: evaporate; to remove some water; leave to crystallise; dry the crystals in a desiccator or oven;	2	accept heat / boil not to dryness / all water accept dry on filter paper
		(iii)	mass decreases; because a gas/carbon dioxide is made;	2	
		(iv)	120; 6.0 ecf	2	(ecf = 0.05 x incorrect RFM)
			Total	13	

OCR (Oxford Cambridge and RSA Examinations)
1 Hills Road
Cambridge
CB1 2EU

OCR Customer Contact Centre

Education and Learning

Telephone: 01223 553998 Facsimile: 01223 552627

Email: general.qualifications@ocr.org.uk

www.ocr.org.uk

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Telephone: 01223 552552 Facsimile: 01223 552553





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