

GCSE

Additional Science A

General Certificate of Secondary Education A215/02

Unit 1: Modules B4, C4, P4

Mark Scheme for June 2010

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Qı	Question		Expected Answers		Marks	Additional Guidance
1	а		diffusion (1)		[1]	
	b		more oxygen than; less oxygen than; carbon dioxide;		[2]	correct pattern of rings for 2 marks one mistake for 1 mark a mistake is: a ring in the wrong place an extra ring a missing ring
	С		Molecules Chemical reactions Enzymes Cells change shape		[1]	more than one box ticked = 0 marks
			Total		[4]	

Qι	Question		Expected Answers		Additional Guidance
2	а		diffusion is movement (of particles) from high concentration to lower concentration/down concentration gradient (1) active transport moves (particles) from low to high concentration/against concentration gradient (1) active transport needs energy / diffusion does not need energy (1)	[3]	accept 'passive' for idea that diffusion does not require energy
	b	i	pituitary (1)	[1]	
		≡	greater volume and less dilute greater volume and more dilute Ecstasy decreased smaller volume and less dilute smaller volume and more dilute	[3]	four lines correct = 3 marks three lines correct = 2 marks two lines correct = 1 mark zero or one line correct = 0 marks
			Total	[7]	

Qı	ıestic	on Expected Answers	Marks	Additional Guidance
3	а	hypothalamus (1)	[1]	accept brain
				ignore CNS/ central nervous system
	b	any two from:	[2]	
		increases blood flow through surface capillaries;		accept more blood flow at surface of skin/near her skin
				do not accept capillaries move
		allowing (more) heat loss;		accept (more) energy loss
		by radiation;		ignore references to sweat and sweating
		Total	[3]	

4	а		Brenda (1)	[1]	
	b	i	one (1)	[1]	
		ii	any three from: pick up/ lose electrons at the electrode; (the compound is) ionic/made of ions; ions are charged; ions move;	[3]	do not accept spread for move
	С	i	some movement;	[2]	three columns correct = 2 marks
			it gets smaller;		two columns correct = 1 mark
			it produces bubbles;		zero or one column correct = 0 marks
		ij	lithium + water → lithium hydroxide + hydrogen (1)	[1]	accept reactants in either order
					accept products in either order
					accept symbol equation only if perfect
			Total	[8]	

Qι	Question		Expected Answers	Marks	Additional Guidance
5	а	i	g, aq, I	[2]	all three correct for 2 marks any two correct for1 mark must be correct symbols, not words not AQ or Aq or aQ accept L or G
		ii	OCI ⁻ (1)	[1]	•
	b	i	2Na + Cl ₂ → 2NaCl	[2]	all formulae correct = 1 mark correct balancing = 1 mark (allow correct multiples) e.g.2Na + $CI_2 \rightarrow 2CI$ Na scores 1 mark for balancing accept = for \rightarrow
		ii	highly reactive reactive less reactive unreactive	[1]	all three correct for 1 mark the pattern must not end on the bottom row of the table
			Total	[6]	
6	а		120 000 J / 120 kJ (2)	[2]	accept 300 000 J for 1 mark accept 420 000 J for 1 mark accept 180 000 J for 1 mark
					accept 200×600 or 600×200 for 1 mark
	b		friction/air resistance (1) transfers kinetic energy to heat energy / some of the work energy is converted to heat (1)	[2]	accept counter/opposing force ignore sound (energy) accept becomes heat
			Total	[4]	

Qu	Question		Expected Answers	Marks	Additional Guidance		
7	а		friction (1)	[2]	one mark for each correct word		
			reaction (1)				
	b		40N (1)	[1]			
	С	i	C (1)	[1]			
		=	The momentum constant. The reaction slows There must be a force The average speed less (1)	[1]			
			Total	[5]			
	Total [0]						
8	а		the same as in the opposite direction to	[1]	both must be correct for the mark		
	b		The velocity increases The reaction slows The weight momentum. (1) The total energy same (1) The counter force	[2]	one mark for each correct answer		
	С		20m (1)	[1]			
	d		A (1)	[1]			
			Total	[5]			

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