

Centre Number Candidate Number	
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MODIFIED LANGUAGE

INSTRUCTIONS TO CANDIDATES

- Write your name clearly in capital letters, your Centre Number and Candidate Number in the boxes above.
- Use black ink. Pencil may be used for graphs and diagrams only.
- Read each question carefully and make sure that you know what you have to do before starting your answer.
- Answer all the questions.

- Do **not** write in the bar codes.
- Write your answer to each question in the space provided, however additional paper may be used if necessary.

INFORMATION FOR CANDIDATES

- The number of marks is given in brackets [] at the end of each question or part question.
- The total number of marks for this paper is **42**.
- A list of physics equations is printed on page 2.
- The Periodic Table is printed on the back page.
- This document consists of **20** pages. Any blank pages are indicated.

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TWENTY FIRST CENTURY SCIENCE EQUATIONS

Useful Relationships

Explaining Motion

speed = $\frac{\text{distance travelled}}{\text{time taken}}$ momentum = mass × velocity change of momentum = resultant force × time for which it acts work done by a force = force × distance moved by the force change in energy = work done change in GPE = weight × vertical height difference kinetic energy = $\frac{1}{2}$ × mass × [velocity]²

Electric Circuits

resistance = $\frac{\text{voltage}}{\text{current}}$

$$\frac{V_{\rm p}}{V_{\rm s}} = \frac{N_{\rm p}}{N_{\rm s}}$$

energy transferred = power × time power = potential difference × current efficiency = $\frac{\text{energy usefully transferred}}{\text{total energy supplied}} \times 100\%$

The Wave Model of Radiation

wave speed = frequency \times wavelength

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Question 1 starts on page 4.

PLEASE DO NOT WRITE ON THIS PAGE

4

Answer all the questions.

1 Ben hurts his foot playing football. The hospital takes an X-ray image of his foot to see if any bones are broken.



(a) Ben asks his friends why X-rays can be used to make an image of his bones.



Who gives the best answer?

answer[1]

(b) Ben knows that exposure to X-rays can be dangerous. This is because X-ray photons carry a lot of energy and can damage cells.

Why do X-ray photons carry a lot of energy?

Put a tick (\checkmark) in the box next to the correct reason.

X-rays are invisible.

X-rays have a high frequency.

X-rays can travel through a vacuum.

(c) Here is an incomplete diagram of the electromagnetic spectrum.

Write X-ray in the correct place.

radio		infrared		ultraviolet		gamma	
(d) Com	lete the senter	nces. Choose v	words from the	e list.	1	[1]
	amplitude	soun	ld sp	eed	wavelength		
Infrar	ed and ultravio	let waves have	the same		through	empty space.	
Infrar	ed and ultravio	let waves mus	t have a differe	ent		[2	2]
						[Total:	5]

2 Jake is a security guard. He uses radio waves to communicate with his boss.



(a) The **amplitude** of the radio waves carries information about Jake's voice. Which other wave property can be used instead?

Put a (ring) around the answer.

frequency transvers	se speed	
---------------------	----------	--

[1]

(b) The radio waves carry digital information.

Which of these wave patterns, **A**, **B** or **C**, shows **digital** information which has been **amplitude modulated** onto a wave?



(c) Put a (ring) around the correct word to complete the sentences.

Jake uses his radio to call his boss.

The waves travel away from the radio to the **receiver transmitter**.

As the waves travel, their amplitude **decreases** increases stays the same.

The quality of the radio waves gets worse as **information noise** is added to them.

(d) Radio waves are transverse. What does this mean?

Put a tick (\checkmark) in the box next to the correct description.

The disturbance of the wave is ...

... in the same direction as the wave energy flow.

... in the opposite direction to the wave energy flow.

... at right angles to the direction of the wave energy flow.

[1]

[2]

3 (a) Micky plays dominoes in a lesson about waves.

Each domino has a **word** at the top and a **meaning** at the bottom. The meaning explains the word at the top of the next domino.

Each word must be placed below its correct meaning, as shown below.

longitudinal		
ber of waves per second from the source		meaning
frequency		word
distance from one crest to the next		
-	longitudinal ber of waves per second from the source frequency distance from one crest to the next	longitudinal ber of waves per second from the source frequency distance from one crest to the next

Here are the other three dominoes.

C amplitude	D wavelength	E speed
how far a wave goes in a second	height of a crest	energy flow parallel to wave disturbance

Write **C**, **D** or **E** in each blank domino so that each word is below its correct meaning. You may fill in the blank dominoes if it helps you work out the answers. (b) Which one of these actions could change the speed of a wave?

Put a tick (\checkmark) in the box next to the correct answer.

change the intensity of the wave change the amplitude of the wave change the frequency of the wave source change the medium that the wave passes through

(c) Alice talks to Micky about their work.



What do the sound waves transfer from Alice to Micky?

Put a (ring) around the correct answer.

atoms	energy	rays	wavelengths	
				[1]

[Total: 4]

[1]

4 Animals such as woodlice respond to changes in their environment.



This process follows a series of steps.

(a) Complete the sentences to explain these steps.

Choose words from the list.

	an action	an effector	a hormone	
	a motor r	neuron a i	receptor	
	a sensory neuron	a stimulus	a synapse	
	The change in the environment is	S		
	The change is detected by			
	Information is carried to the cent	ral nervous system	by	
	Information is carried from the ce	entral nervous syste	em by	
	The response is carried out by			
<i>(</i> ,)				[4]
(b)	Woodlice move away from light.			
	Which of these sentences explain	ns the reason for th	is response?	
	Put a tick (\checkmark) in the box next to the	he best answer.		
	They prefer dry conditions.			
	They are afraid of the light.			
	Their food source is always	in the dark.		
	Dark conditions are more fa	vourable for their su	ırvival.	
				[1]

(c) Which of these terms describes the behaviour of simple animals such as woodlice?

Put a tick (\checkmark) in the box next to the **best** answer.

a complex response	
a simple reflex	
a learned response	
a learned reflex	

[1]

[Total: 6]

5 (a) Label this diagram of a motor neuron.

Choose words from the list.



(b) What happens when a motor neuron is stimulated?

Put a tick (\checkmark) in the box next to the correct answer.

It gets shorter.

It moves to where it is needed.

It transmits an electrical impulse.

[1]

[Total: 4]

6 Baby Sam is learning to crawl.



The brain is responsible for memory and for learning.

(a) Draw a straight line from each function to its correct description.



[2]

(b) Babies learn new skills as they develop.

The sentences explain how this happens. They are in the wrong order.

- A If the experience is repeated the link becomes stronger.
- **B** It is easier for impulses to pass along the link.
- **C** When a baby tries something new, links between neurons are made in the brain.
- **D** The response gets easier to make.

Fill in the boxes to show the correct order. The first one has been done for you.

<i>C</i>	
----------	--

[2]

14

7 The pie chart shows the EU chemical industry sales for 2003.



Mary combines these sectors to show the sales of the three main areas of the chemical industry.

(a) Complete the table to show the total size of the three main areas.

area	total % sales
pharmaceuticals	
petrochemicals & bulk	
fine chemicals	

[2]

(b) The total sales of the industry are about 450 billion euros.

Using the information in the pie chart, put a ring around the total sales of the pharmaceuticals in billions of euros.



8 Foods such as fruit taste pleasant because they contain small amounts of acid.



(a) Sometimes, extra acid is added to foods.

Acids which can be added to food are given E numbers.

acid	E number
ethanoic acid	E260
citric acid	E330
tartaric acid	E334
hydrogen chloride	E507
sulfuric acid	E513

Give one E number for an acid which is solid when pure.

		answer			
	Give one E number for an acid which is a liquid when pure				
		answer			
	Give one E number for an acid which is a gas when pure.				
		answer[2]			
(b)	Acids in food show the normal reactions of an acid.				
	Write the labels acid, alkali, salt and water in the boxes. One has been done for you.				



9 Bobby reacts marble chips (calcium carbonate) with acid.

He measures the change in mass as the acid reacts.



(a) Why does he put cotton wool in the top of the flask?



Who gives the best answer?

answer[1]

(b) Bobby starts the reaction by adding some acid.

He adds extra acid part way through.



(iii) Put a (ring) around the mass of carbon dioxide given off by the end of the experiment.

0.4g	0.5 g	0.6 g	0.8 g	1.0 g	[1]
					[Total: 4]

10 Benzoic acid is added to foods as a preservative.

Benzoic acid forms solid crystals which must be purified before the acid is used.

(a) The table shows different methods of purifying chemicals, and information about when you would use each method.

Draw lines to link each **method** to the correct piece of **information**.

One has been done for you.



(b) Terry does a titration to check the purity of some benzoic acid.

He wants to use an indicator which changes colour when the solution becomes slightly alkaline.

	indicator	pH range for colour change
Α	gentian violet	0 to 2
В	methyl red	4 to 6
С	litmus	5 to 8
D	phenolphthalein	8 to 10
E	nitramine	11 to 13

(i) Which indicator, A, B, C, D, or E, should he use?

answer [1]

(ii) Terry does the titration five times.

He writes down his titration results.

titration number	1	2	3	4	5
volume of alkali in cm ³	25.9	25.1	25.0	25.0	25.0

Put a tick (\checkmark) in the box next to the result that he should use for his calculations.



[1]

[Total: 4]

END OF QUESTION PAPER



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The relative atomic masses of copper and chlorine have not been rounded to the nearest whole number.

* The lanthanoids (atomic numbers 58-71) and the actinoids (atomic numbers 90-103) have been omitted.

4 He ^{helium}	20 Ne ^{neon} 10	40 Ar ^{argon} 18	84 Kr krypton 36	131 Xe xenon 54	[222] Rn radon 86	t fully	
	19 F fluorine 9	35.5 Cl chlorine 17	80 Br ^{bromine} 35	127 I ^{iodine} 53	[210] At astatine 85	orted but no	
	16 O ^{oxygen} 8	32 S sultur 16	79 Se selenium 34	128 Te tellurium 52	[209] Po Polonium 84	ve been repo	
	14 N nitrogen 7	31 P phosphorus 15	75 As ^{arsenic} 33	122 Sb antimony 51	209 Bi bismuth 83	: 112-116 hav uthenticated	
	12 C carbon 6	28 Si 14	73 Ge germanium 32	119 Sn 50	207 Pb tead 82	mic numbers a	
	11 B boron 5	27 Al aluminium 13	70 Ga ^{gallium} 31	115 In indium 49	204 T1 thallium 81	nts with ato	
•			65 Zn ₂inc 30	112 Cd cadmium 48	201 Hg ^{mercury} 80	Eleme	
			63.5 Cu ^{copper} 29	108 Ag aliver 47	197 Au ^{gold} 79	[272] Rg 111	
			59 Ni 28	106 Pd Palladium 46	195 Pt Platinum 78	[271] Ds darmstadtium 110	
			59 Co ^{cobalt} 27	103 Rh ^{rhodium} 45	192 Ir 77	[268] Mt neitnerium 109	
1 H hydrogen 1			56 Fe ^{iron} 26	101 Ru ruthenium 44	190 Os ^{osmium} 76	[277] Hs hassium 108	
		_	55 Mn ^{manganese} 25	[98] Tc technetium 43	186 Re ^{rhenium} 75	[264] Bh ^{bohrium} 107	
	mass ool number		52 Cr chromium 24	96 Mo ^{molybdenum} 42	184 W tungsten 74	[266] Sg seaborgium 106	
ƙey	/e atomic mic symb ^{name} (proton) r	ve atomic mane (proton) r	51 V vanadium 23	93 Nb ^{niobium} 41	181 Ta tantalum 73	[262] Db ^{dubnium} 105	
	relati at o atomic		48 Ti titanium 22	91 Zr ^{zirconium} 40	178 Hf ^{hafnium} 72	[261] Rf rutherfordium 104	
		45 Sc scandium 21	89 Y 39	139 La* ^{lanthanum} 57	[227] Ac* ^{actinium} 89		
	9 Be ^{beryllium}	24 Mg 12	40 Ca calcium 20	88 Sr strontium 38	137 Ba ^{barium} 56	[226] Ra radium 88	
	7 Li ^{lithium} 3	23 Na sodium 11	39 K potassium 19	85 Rb rubidium 37	133 Cs caesium 55	[223] Fr ^{francium} 87	

The Periodic Table of the Elements

0

9

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4

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2

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