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#### 2

### TWENTY FIRST CENTURY SCIENCE EQUATIONS

#### **Useful Relationships**

#### **Explaining Motion**

speed =  $\frac{\text{distance travelled}}{\text{time taken}}$ momentum = mass × velocity change of momentum = resultant force × time for which it acts work done by a force = force × distance moved by the force change in energy = work done change in GPE = weight × vertical height difference kinetic energy =  $\frac{1}{2}$  × mass × [velocity]<sup>2</sup>

#### **Electric Circuits**

resistance =  $\frac{\text{voltage}}{\text{current}}$   $\frac{V_p}{V_s} = \frac{N_p}{N_s}$ energy transferred = power × time power = potential difference × current efficiency =  $\frac{\text{energy usefully transferred}}{\text{total energy supplied}} \times 100\%$ 

### The Wave Model of Radiation

wave speed = frequency  $\times$  wavelength

3 BLANK PAGE

Question 1 starts on page 4

PLEASE DO NOT WRITE ON THIS PAGE

[Turn over

## **4** Answer **all** the questions.

**1** Jo uses a microwave oven to heat her dinner.



(a) These sentences are about the microwave oven.

Draw a straight line from the **start** of each sentence to its correct **end**.

| start                         |   | end                             |
|-------------------------------|---|---------------------------------|
| The microwaves interfere      |   | by the water in the food.       |
|                               | 7 |                                 |
| The microwaves are reflected  |   | when they pass through a gap.   |
|                               |   |                                 |
| The microwaves are diffracted |   | by the metal walls of the oven. |
|                               | - |                                 |

The microwaves are absorbed ...

... where they overlap with each other.

[3]

(b) This graph shows a microwave.



Which distance, A, B, C or D, is the wavelength of the microwave?

answer ..... [1]

[Total: 4]

2 Isobel uses a remote control to adjust her TV set.



(a) The remote control uses a beam of infrared to carry information to the TV set.

Finish the sentences by choosing the **best** words from this list.

# colour energy photons strength intensity momentum

| The                                | . of the beam is the energy it delivers per second. |
|------------------------------------|---|
| Its value depends on the number of | put into the beam each                              |
| second, as well as their           | [2]   |

(b) The infrared is modulated each time that Isobel presses a button on the remote control. The beam is pulsed on and off in a code, with a different code for each button.

Draw a straight line from the start of each sentence to its correct end.

| start                 | _ | end                            |
|-----------------------|---|--------------------------------|
| Each pulse represents |   | a 0 in the code.               |
|                       |   |                                |
|                       |   | a 1 in the code.               |
|                       |   |                                |
|                       |   | information as a digital code. |
|                       |   |                                |
|                       | ] |                                |

The pulsed infrared beam transfers  $\ldots$ 

[1]

... information as an analogue code.

(c) An LED is the source of the infrared in the remote control.



The LED is enclosed in a plastic lens.

(i) As the infrared leaves the plastic, most of it changes direction.

What is the name of this process?

Put a (ring) around the correct answer.

| diffraction | reflection | refraction | rotation |     |
|-------------|------------|------------|----------|-----|
|             |            |            |          | [1] |

(ii) Which one of these statements explains the change of direction?

Put a tick ( $\checkmark$ ) in the box next to the correct answer.

The infrared spreads out as it leaves the lens.
The infrared rotates against the plastic as it reflects off the air.
The infrared speeds up as it moves from the plastic into the air.
The infrared slows down as it moves from the plastic into the air.
[1]

[Total: 5]

**3** Jenny is a presenter for Radio CA.



- (a) Jenny sings into the microphone.
  - (i) The speed of sound waves in the studio is 340 m/s. Jenny sings a note of frequency 680 Hz. Which of these calculations gives the wavelength of her sound?

Put a (ring) around the correct answer.

| 680 | 680 x 340 | 340 |
|-----|-----------|-----|
| 340 | 000 ~ 040 | 680 |

(ii) Here are some statements about sound waves. Some of these statements are true. Some are false.

Write T in the box next to each true statement and F in the box next to each false one.

| The disturbance of a sound wave                            | T (true)<br>or<br>F (false) |
|--|-----------------------------|
| and its energy flow are in the same direction.             |                             |
| increases in amplitude as the sound gets louder.           |                             |
| is at right angles to the wave's direction of energy flow. |                             |
|  | [1]                         |

[1]

- (b) Information in the sound wave modulates the radio wave from the transmitter.
  - (i) Which diagram represents frequency modulation of the radio wave?
    - Put a (ring) around the correct diagram.

[1]

(ii) Complete these sentences. Choose words from this list.

# regular random amplified analogue amplitude modulated wavelength

| The frequency modulated radio wave is an  | signal.                 |
|---|-------------------------|
| As it moves away from the transmitter its | decreases.              |
| The radio wave also picks up              | signals called noise.   |
| This is still present when the signal is  | at the receiver.<br>[2] |

[Total: 5]

- 4 Jane has some copper. She uses this to make copper sulfate.
  - (a) Jane uses one reaction from the first list and one from the second list.

Draw one straight line from the correct first reaction to the correct second reaction.



(b) The copper sulfate Jane makes is not pure. She uses these four steps to purify the copper sulfate.

They are in the wrong order.

- A drying
- **B** filtration
- **C** dissolving
- **D** crystallisation

Fill in the boxes to show the right order. The first one has been done for you.



[2]

[Total: 4]

5 Michael reacts magnesium with sulfuric acid.

 $Mg + H_2SO_4 \rightarrow MgSO_4 + H_2$ 

- (a) Use relative atomic masses from the Periodic Table on the back page of this booklet to answer the following questions.
  - (i) What mass of hydrogen is produced when 24g of magnesium react with an excess of sulfuric acid?

mass of hydrogen = ..... g [1]

(ii) What is the relative formula mass of magnesium sulfate, MgSO<sub>4</sub>?

relative formula mass = .....[1]

(iii) What mass of magnesium sulfate is produced when 3g of magnesium react with an excess of sulfuric acid?

mass of magnesium sulfate = ...... g [1]

(b) Michael works out that his reaction should produce 8g of magnesium sulfate. In fact it only produces 2g.

Put a (ring) around the percentage yield for this reaction.

| 16% | 25%    | 40% | 60% | [1] |
|-----|--------|-----|-----|-----|
|     | == , = |     |     |     |

(c) Michael reacts magnesium with an excess of sulfuric acid at 20 °C. He measures the volume of hydrogen gas given off at intervals of time. He repeats the experiment five times, changing one of the conditions used each time. He plots a graph for each reaction, A, B, C, D and E. The line for Michael's first experiment at 20 °C is marked A.



|      |              |                         |                                   |                                       |                       | 13                             |                 |                   |            |           |                |           |
|------|--------------|-------------------------|-----------------------------------|---------------------------------------|-----------------------|--------------------------------|-----------------|-------------------|------------|-----------|----------------|-----------|
|      | (i)          | He carrie<br>Which lin  | s out o<br>e, <b>B</b> , <b>C</b> | ne reacti<br>, <b>D</b> or <b>E</b> , | ion at 40<br>shows    | )°C.<br>this react             | tion?           |                   |            |           |                |           |
|      |              | Put a rir               | ng arc                            | ound the                              | correct a             | answer.                        |                 |                   |            |           |                |           |
|      |              |                         |                                   | В                                     | С                     |                                | D               | E                 |            |           |                | [1]       |
|      | (ii)         | In one rea<br>Which lin | action<br>e, <b>B</b> , <b>C</b>  | he uses ;<br>, <b>D</b> or <b>E</b> , | more ma<br>shows      | agnesium<br>this react         | n.<br>tion?     |                   |            |           |                |           |
|      |              | Put a rir               | ng arc                            | ound the                              | correct               | answer.                        |                 |                   |            |           |                |           |
|      |              |                         |                                   | В                                     | С                     |                                | D               | E                 |            |           |                | [1]       |
| (    | iii)         | In one rea<br>Which lin | action<br>e, <b>B</b> , <b>C</b>  | he uses t<br>, <b>D</b> or <b>E</b> , | the sam<br>shows      | e mass c<br>this react         | of mag<br>tion? | nesium, bu        | t in large | er piece: | S.             |           |
|      |              | Put a rir               | ng arc                            | ound the                              | correct a             | answer.                        |                 |                   |            |           |                |           |
|      |              |                         |                                   | В                                     | С                     |                                | D               | E                 |            |           |                | [1]       |
|      |              |                         |                                   |                                       |                       |                                |                 |                   |            |           | [              | Total: 7] |
| Dave | e rea        | icts coppe              | er carbo                          | onate wit                             | h nitric a            | acid.                          |                 |                   |            |           |                |           |
| (a)  | The          | reaction p              | oroduc                            | es carbo                              | n dioxide             | e, water a                     | and an          | other produ       | uct.       |           |                |           |
|      | Give         | e the name              | e and f                           | ormula o                              | f the oth             | er produ                       | ct.             |                   |            |           |                |           |
|      |              |                         |                                   |                                       |                       |                                |                 | name              |            |           |                |           |
|      |              |                         |                                   |                                       |                       |                                |                 | formula           |            |           |                | [2]       |
| (b)  | He t<br>This | hen reacts<br>produces  | s magr<br>s magn                  | nesium ca<br>esium ch                 | arbonate<br>Noride, N | e with hyo<br>⁄IgC <i>l</i> ₂. | drochle         | oric acid.        |            |           |                |           |
|      | Bala         | ance this e             | quatio                            | n for the                             | reaction              | ı.                             |                 |                   |            |           |                |           |
|      | ſ            | MgCO <sub>3</sub>       | +                                 | HC1                                   | $\rightarrow$         | H <sub>2</sub> O               | +               | MgCl <sub>2</sub> | +          | CC        | ) <sub>2</sub> | [1]       |
|      |              |                         |                                   |                                       |                       |                                |                 |                   |            |           | [              | iotal: 3] |

6

Charlie carries out an experiment using woodlice.
 He puts 20 woodlice into the centre of a petri dish so that they can move freely into four chambers,
 A, B, C and D. Each chamber has different conditions.



After five minutes, Charlie counts the woodlice in each chamber. He records his results in a table.

| chamber | chamber conditions | number of woodlice |
|---------|--------------------|--------------------|
| Α       | light and dry      | 1                  |
| В       | dark and dry       | 6                  |
| С       | light and moist    | 4                  |
| D       | dark and moist     | 9                  |

(a) What is the percentage of woodlice found in chamber D?

Put a (ring) around the correct answer.

| 9% | 20% | 45% | 90% |
|----|-----|-----|-----|
|    |     |     |     |

[1]

(b) Put a tick (✓) in the box next to each statement which could be used to explain the behaviour of the woodlice in the experiment.

| Woodlice are attracted to light.                       |     |
|--|-----|
| Woodlice are attracted to shade more than to moisture. |     |
| Woodlice avoid too much moisture.                      |     |
| Woodlice may dry out easily.                           |     |
|  | [2] |

(c) The behaviour pattern of the woodlice shown in this experiment is a reflex action.

Finish the sentence. Choose a word from this list.

|             | complex            | involuntary | voluntary  |            |
|-------------|--------------------|-------------|------------|------------|
| Simple refl | exes produce rapid |             | responses. | [1]        |
|             |                    |             |            | [Total: 4] |

- 8 This question is about synapses between sensory and motor neurons.
  - (a) Here are the steps which take place at a synapse as an impulse passes from a sensory neuron to a motor neuron.

They are in the wrong order.

- **A** chemical released into the synapse
- **B** chemical diffuses across the synapse
- C motor neuron transmits an impulse
- D sensory neuron transmits an impulse
- E chemical binds to the receptor molecules

Fill in the boxes to show the right order. The first one has been done for you.

[3]

(b) The drug ecstasy blocks the removal of the synapse chemical serotonin.

How will this affect the amount of serotonin in the synapse gap between two neurons?

Put a tick ( $\checkmark$ ) in the box next to the correct answer.

| same amount of serotonin      |  |
|-------------------------------|--|
| increased amount of serotonin |  |
| decreased amount of serotonin |  |

[1]

(c) Synapse chemicals, like serotonin, are recognised by a specific receptor molecule found on one side of the synapse.

How does this affect the transmission of nerve impulses?

Put a tick ( $\checkmark$ ) in the box next to the correct answer.

The strength of the nerve impulse is increased.
The nerve impulses can only travel in one direction.
The speed of the nerve impulse transmission is increased.
[1]

[Total: 5]

9 The Russian scientist Ivan Pavlov is famous for his work with learned behaviour in dogs.

His experiments included the following steps.

- A dog salivates when given food.
- A bell is rung each time the dog is fed.
- After some time, the bell is rung without giving the dog food.
- The dog salivates when it hears the bell.
- (a) What is the function of each step?Draw a straight line from each step to its correct function.



(b) Which part of the dog's brain is involved in learned behaviour patterns?

Put a (ring) around the correct answer.

| hypothalamus |                                    | pituitary gland                    | medulla                | cerebral cortex  | r <b>tex</b> |  |
|--------------|------------------------------------|------------------------------------|------------------------|------------------|--------------|--|
|              |                                    |                                    |                        |                  | [1]          |  |
| (c)          | Which of the follow                | wing types of behaviour are        | e learned?             |                  |              |  |
|              | Put a tick ( $\checkmark$ ) in the | ne box next to <b>each</b> correct | answer.                |                  |              |  |
|              | Some bacter                        | ia can swim towards sourc          | es of food.            |                  |              |  |
|              | Some birds r                       | nay avoid eating caterpillar       | s with warning colours | S.               |              |  |
|              | Houseflies fly                     | rapidly away if they detec         | t any sign of moveme   | nt.              |              |  |
|              | Snails draw i                      | nto their shells if they deteo     | ct any sign of moveme  | ent.             |              |  |
|              | Goldfish may                       | swim to the front of their ta      | ank when people walk   | up to feed them. |              |  |
|              |                                    |                                    |                        |                  | [2]          |  |

(d) Three friends discuss different ways of explaining what happens in the human brain when we learn.



Jim Hannah Harry [1]

[Total: 5]

END OF QUESTION PAPER

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The Periodic Table of the Elements

| 0 | 4<br>hettum<br>2 | 20<br>Neon<br>10                                    | 40<br>Ar<br><sup>argon</sup><br>18 | 84<br>Krypton<br>36                         | 131<br>Xe <sup>xenon</sup><br>54 | [222]<br><b>Rn</b><br>radon<br>86           | t fully  |
|---|------------------|---|------------------------------------|---|----------------------------------|---|--|
| ~ |                  | 19<br>fluorine<br>9                                 | 35.5<br>Cl<br>chlorine<br>17       | 80<br>Br<br><sup>bromine</sup><br>35        | 127<br>I<br>53                   | [210]<br>At<br>astatine<br>85               | orted but no                                   |
| 9 |                  | 16<br>O<br><sup>oxygen</sup><br>8                   | 32<br>S<br><sup>sulfur</sup><br>16 | 79<br>Se<br>34                              | 128<br>Te<br>tellurium<br>52     | [209]<br>Po<br>84                           | re been repo                                   |
| ß |                  | 14<br>N<br>nitrogen<br>7                            | 31<br>Phosphorus<br>15             | 75<br>As<br><sup>arsenic</sup><br>33        | 122<br>Sb<br>antimony<br>51      | 209<br>Bi<br>83                             | 112-116 hav                                    |
| 4 |                  | 12<br>C<br>carbon<br>6                              | 28<br>Si<br>14                     | 73<br><b>Ge</b><br>germanium<br>32          | 119<br>Sn<br>50                  | 207<br><b>Pb</b><br><sup>lead</sup><br>82   | mic numbers<br>a                               |
| m |                  | 11<br>B<br><sup>boron</sup><br>5                    | 27<br>Al<br>atuminium<br>13        | 70<br><b>Ga</b><br><sup>galtium</sup><br>31 | 115<br>In<br>indium<br>49        | 204<br>T1<br>thallium<br>81                 | nts with ato                                   |
|   |                  |   |                                    | 65<br>Zn<br><sup>zinc</sup><br>30           | 112<br>Cd<br>cadmium<br>48       | 201<br>Hg<br><sup>mercury</sup><br>80       | Eleme  |
|   |                  |   |                                    | 63.5<br>Cu<br><sup>copper</sup><br>29       | 108<br>Ag<br>silver<br>47        | 197<br><b>Au</b><br><sup>gold</sup><br>79   | [272]<br>Rg<br>111                             |
|   |                  |   |                                    | 59<br>Ni<br><sup>nickel</sup><br>28         | 106<br>Pd<br>Palladium<br>46     | 195<br>Pt<br><sup>platinum</sup><br>78      | [271]<br>Ds<br>damstadtum<br>110               |
|   |                  |   |                                    | 59<br>Co<br><sup>cobalt</sup><br>27         | 103<br>Rh<br>rhođium<br>45       | 192<br>Ir<br>77                             | [268]<br>Mt<br>109                             |
|   | hydrogen<br>1    |   |                                    | 56<br>Fe<br><sup>iron</sup><br>26           | 101<br>Ru<br>ruthenium<br>44     | 190<br><b>Os</b><br><sup>osmium</sup><br>76 | [277]<br>HS<br>hassium<br>108                  |
| L |                  |   |                                    | 55<br>Mn<br><sup>manganese</sup><br>25      | [98]<br>Tc<br>technetium<br>43   | 186<br>Re<br><sup>rhenium</sup><br>75       | [264]<br>Bh<br>bohrium<br>107                  |
|   |                  | mass<br>ool<br>number                               |                                    | 52<br>Cr<br>chromium<br>24                  | 96<br>Mo<br>Motybdenum<br>42     | 184<br>W<br>tungsten<br>74                  | [266]<br>Sg<br>seaborgium<br>106               |
|   | Key              | ve atomic<br><b>omic symt</b><br>name<br>(proton) r |                                    | 51<br>V<br>vanadium<br>23                   | 93<br><b>ND</b><br>41            | 181<br>Ta<br><sup>tantalum</sup><br>73      | [262]<br>Db<br>dubnium<br>105                  |
|   |                  | relati<br><b>at</b> o<br>atomic                     |                                    | 48<br>Ti<br>titanium<br>22                  | 91<br>Zr<br>zirconium<br>40      | 178<br>Hf<br><sup>hafnium</sup><br>72       | [261]<br>Rf<br><sup>nutherfordium</sup><br>104 |
|   |                  |   | -                                  | 45<br>Sc<br>scandium<br>21                  | 89<br>Y<br>39                    | 139<br>La*<br>Ianthanum<br>57               | [227]<br>Ac*<br><sup>actinium</sup><br>89      |
| 2 |                  | 9<br>Be<br>beryttium<br>4                           | 24<br>Mg<br>12                     | 40<br>Ca<br>calcium<br>20                   | 88<br>Sr<br>strontium<br>38      | 137<br>Ba<br><sup>barium</sup><br>56        | [226]<br><b>Ra</b><br>radium<br>88             |
| - |                  | 7<br>Li<br><sup>lithium</sup><br>3                  | 23<br>Na<br>11                     | 39<br>K<br>potassium<br>19                  | 85<br>Rb<br>rubidium<br>37       | 133<br>Cs<br>caesium<br>55                  | [223]<br>Fr<br>francium<br>87                  |
|   |                  |   |                                    |   |                                  |   |  |

\* The lanthanoids (atomic numbers 58-71) and the actinoids (atomic numbers 90-103) have been omitted.

The relative atomic masses of copper and chlorine have not been rounded to the nearest whole number.

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