

OXFORD CAMBRIDGE AND RSA EXAMINATIONS
GCSE

A151/02

TWENTY FIRST CENTURY SCIENCE
ADDITIONAL SCIENCE A

Modules B4 C4 P4 (Higher Tier)

MONDAY 14 JANUARY 2013: Morning

DURATION: 1 hour
plus your additional time allowance

MODIFIED ENLARGED 18pt

| | | | |
|-------------------------------|--|------------------------------|--|
| Candidate forename | | Candidate surname | |
|-------------------------------|--|------------------------------|--|

| | | | | | | | | | | |
|--------------------------|--|--|--|--|--|-----------------------------|--|--|--|--|
| Centre number | | | | | | Candidate number | | | | |
|--------------------------|--|--|--|--|--|-----------------------------|--|--|--|--|

Candidates answer on the Question Paper.
A calculator may be used for this paper.

OCR SUPPLIED MATERIALS:

Periodic Table (inserted)

OTHER MATERIALS REQUIRED:

Pencil


Ruler (cm/mm)

READ INSTRUCTIONS OVERLEAF

INSTRUCTIONS TO CANDIDATES

- Write your name, centre number and candidate number in the boxes on the first page. Please write clearly and in capital letters.
- Use black ink. HB pencil may be used for graphs and diagrams only.
- Answer ALL the questions.
- Read each question carefully. Make sure you know what you have to do before starting your answer.
- Write your answer to each question in the space provided. Additional paper may be used if necessary but you must clearly show your candidate number, centre number and question number(s).

INFORMATION FOR CANDIDATES

- Your quality of written communication is assessed in questions marked with a pencil (.
- The number of marks is given in brackets [] at the end of each question or part question.
- A list of physics equations is printed on pages 4–5.
- An enlarged copy of the Periodic Table is inserted.
- The total number of marks for this paper is 60.

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TWENTY FIRST CENTURY SCIENCE EQUATIONS

USEFUL RELATIONSHIPS

THE EARTH IN THE UNIVERSE

$$\text{distance} = \text{wave speed} \times \text{time}$$

$$\text{wave speed} = \text{frequency} \times \text{wavelength}$$

SUSTAINABLE ENERGY

$$\text{energy transferred} = \text{power} \times \text{time}$$

$$\text{power} = \text{voltage} \times \text{current}$$

$$\text{efficiency} = \frac{\text{energy usefully transferred}}{\text{total energy supplied}} \times 100\%$$

EXPLAINING MOTION

$$\text{speed} = \frac{\text{distance travelled}}{\text{time taken}}$$

$$\text{acceleration} = \frac{\text{change in velocity}}{\text{time taken}}$$

$$\text{momentum} = \text{mass} \times \text{velocity}$$

$$\frac{\text{change of momentum}}{\text{force}} = \frac{\text{resultant}}{\text{force}} \times \frac{\text{time for which it acts}}{\text{it acts}}$$

$$\frac{\text{work done by a force}}{\text{force}} = \frac{\text{distance moved in the direction of the force}}{\text{direction of the force}}$$

$$\text{amount of energy transferred} = \text{work done}$$

$$\frac{\text{change in gravitational potential energy}}{\text{weight}} = \frac{\text{vertical height difference}}{\text{vertical height difference}}$$

$$\text{kinetic energy} = \frac{1}{2} \times \text{mass} \times [\text{velocity}]^2$$

ELECTRIC CIRCUITS

$$\text{power} = \text{voltage} \times \text{current}$$

$$\text{resistance} = \frac{\text{voltage}}{\text{current}}$$

$$\frac{\text{voltage across primary coil}}{\text{voltage across secondary coil}} = \frac{\text{number of turns in primary coil}}{\text{number of turns in secondary coil}}$$

RADIOACTIVE MATERIALS

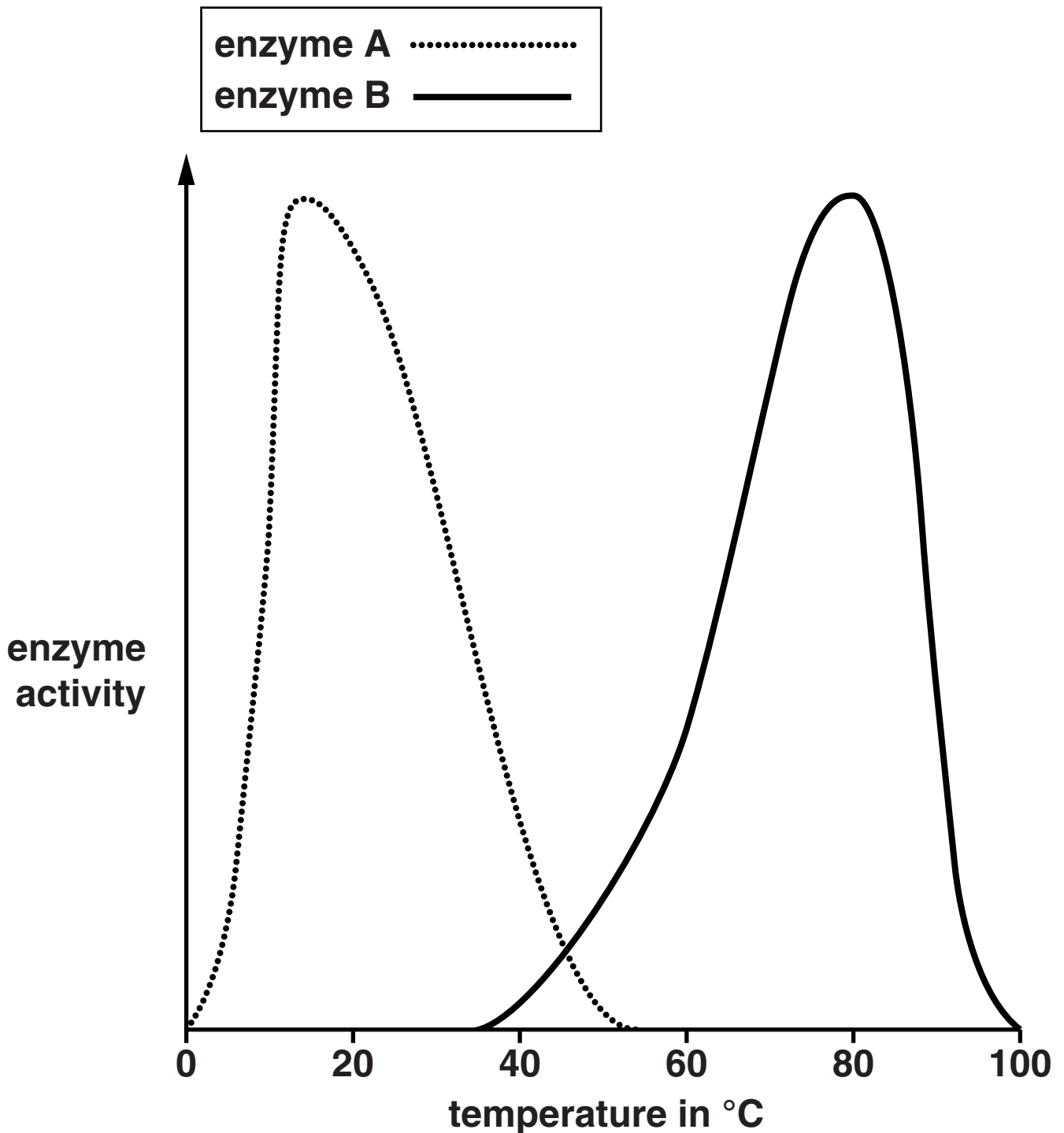
$$\text{energy} = \text{mass} \times [\text{speed of light in a vacuum}]^2$$

Answer ALL the questions.

- 1 Corinne does an experiment using two different enzymes, A and B.**

She records the activity of each enzyme at different temperatures.

She plots her results on a graph.



(a) Both enzymes work on the same chemical.

One of the enzymes is from a bacterium that lives in hot springs at 80 °C. The other enzyme is from a bacterium that lives in the sea at 14 °C.

Corinne concludes that enzyme A comes from the bacterium that lives in the sea.

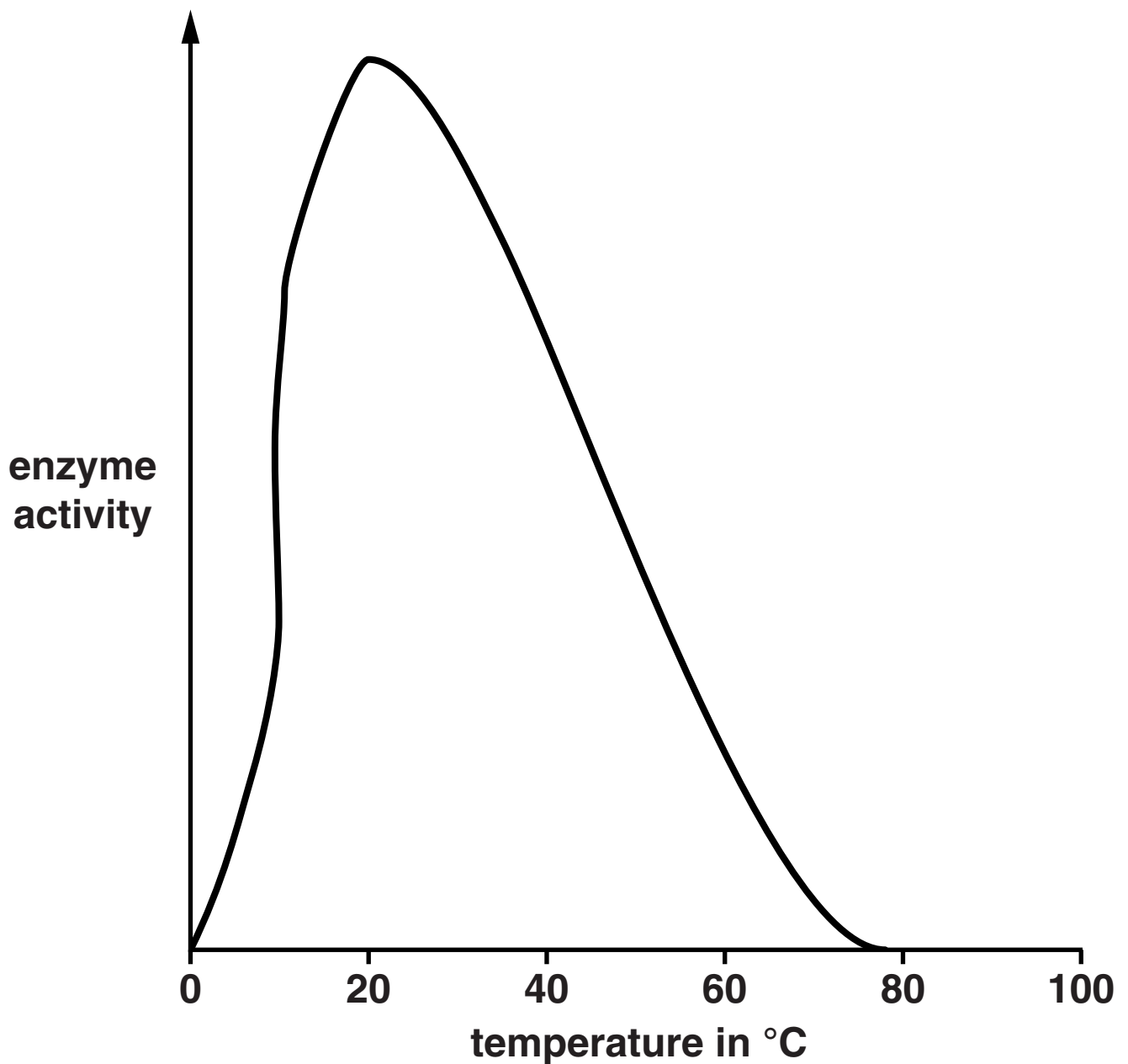
Explain why Corinne's conclusion is correct.

[2]

(b) Corinne does the same experiment with a different enzyme, C.

This enzyme works on the same chemical as enzymes A and B.

She plots her results on a graph.



After the experiment Corinne extracts and cools enzyme C.

She repeats the experiment again with the same sample of enzyme and fresh substrate.

Describe and explain what the graph of enzyme activity will look like.

[2]

[TOTAL: 4]

2 Wheat plants are grown for food.

Scientists investigate how a wheat plant uses nitrogen.

Nitrogen fertiliser is added to the soil in which the wheat plant grows.

Scientists measure how much of this added nitrogen is present in the soil, roots and seeds over a four week period.

Here are their results.

| TIME (WEEKS) | % OF NITROGEN | | |
|-------------------------|----------------------|--------------|--------------|
| | SOIL | ROOTS | SEEDS |
| 1 | 100 | 0 | 0 |
| 2 | 85 | 15 | 0 |
| 3 | 80 | 15 | 5 |
| 4 | 75 | 15 | 10 |

- (a) Describe how the nitrogen gets from the fertiliser into nitrogen-containing molecules in the seeds.**



The quality of written communication will be assessed in your answer.

[6]

(b) The soil around the roots of the plant becomes waterlogged.

The way that the root cells release energy from glucose changes.

Complete the sentences.

Waterlogged soil contains less _____ .

The root cells will now start to release energy by the process of _____ .

This process also produces _____ and _____ .

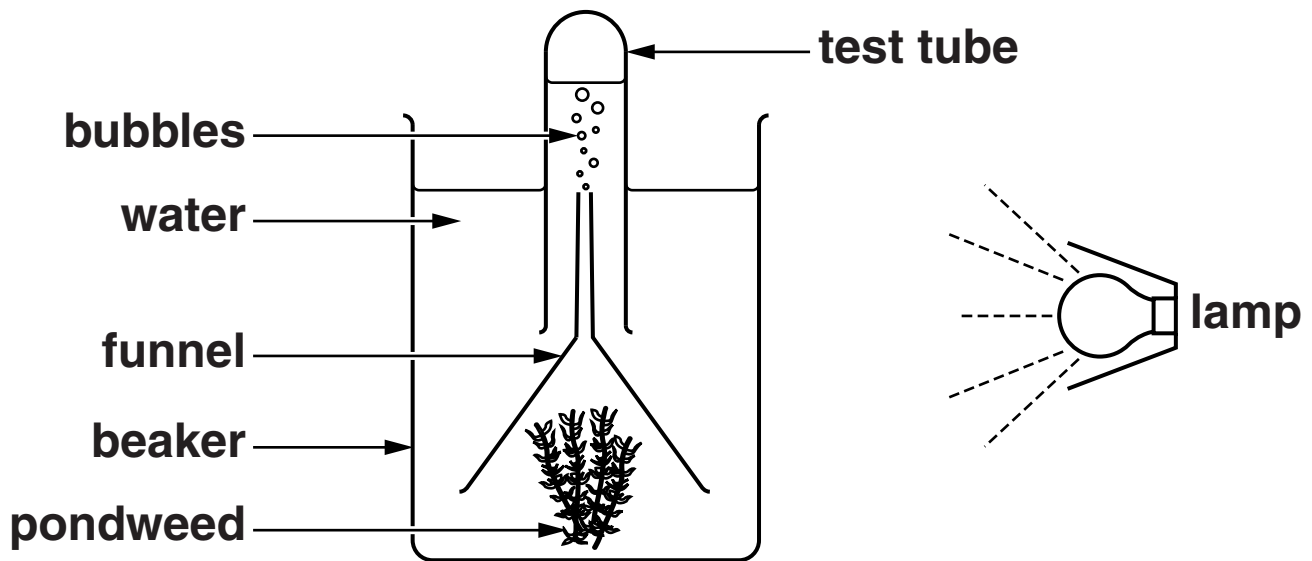
[2]

[TOTAL: 8]

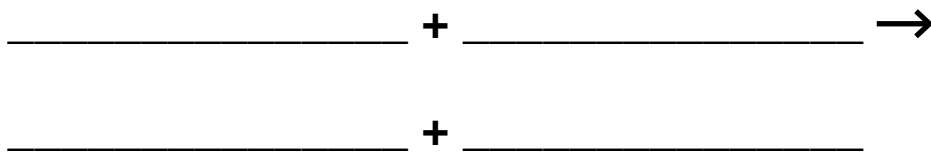
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TURN OVER FOR QUESTION 3

3 Anette does an experiment with pondweed.



(a) Write the symbol equation for photosynthesis.



[1]

(b) Anette changes the distance from the lamp to the pondweed and counts the bubbles produced in 5 minutes. She does this experiment four times at each distance.

During one set of readings she knocks over the lamp, but then puts it back in the same place.

Here are Anette's results.

| DISTANCE IN cm | NUMBER OF BUBBLES PRODUCED IN 5 MINUTES | | | |
|-------------------|--|-----------------|-----------------|-----------------|
| | EXPERIMENT 1 | EXPERIMENT 2 | EXPERIMENT 3 | EXPERIMENT 4 |
| 10 | 21 | 21 | 6 | 18 |
| 15 | 14 | 15 | 15 | 16 |
| 20 | 11 | 12 | 11 | 14 |
| 25 | 10 | 8 | 10 | 12 |

Anette looks at her results for the distance of 10 cm.

- (i) Using her results, what is the **NARROWEST RANGE** within which the true value of the number of bubbles lies?

answer _____ [1]

- (ii) Explain why this is different from the range of results recorded.

_____ [1]

(c) Anette repeats her experiment using a brighter bulb.

She finds that the most bubbles recorded on any experiment is still 21 bubbles.

Use ideas about photosynthesis to explain why she gets this result.

[2]

[TOTAL: 5]

4 Ryan investigates osmosis in pieces of raw potato.

He cuts six cylinders of potato, each with the same shape and mass.

He places each potato cylinder in a sugar solution.

Each solution contains the same sugar, but at a different concentration.

After 2 hours, he records the mass of each cylinder and calculates its percentage (%) change in mass.

Here are his results.

| CONCENTRATION OF SUGAR SOLUTION IN g/dm³ | PERCENTAGE (%) CHANGE IN MASS OF POTATO CYLINDER |
|--|---|
| 0 | +7 |
| 20 | +3 |
| 40 | +1 |
| 60 | |
| 80 | −4 |
| 100 | −6 |

(a) (i) Predict the value for the 60 g/dm³ sugar solution.

percentage change at 60 g/dm³ = _____ [1]

(ii) Suggest the concentration of cell contents.

concentration = _____ g/dm³ [1]

(b) Ryan suggests ways to get a better estimate of the concentration of the cell contents.

Put a tick (✓) in the box next to Ryan's best suggestion.

Record the change in mass in g instead of percentage change. ☐

Repeat the experiment using different sizes of potato cylinder. ☐

Repeat the experiment with concentrations greater than 100 g/dm^3 of sugar. ☐

Repeat each concentration and calculate the average percentage change in mass. ☐

Soak the potato in pure water before the experiment. ☐

[1]

[TOTAL: 3]

- 5 The chemical industry uses large amounts of chlorine. Some of this chlorine is transported across the country by lorry.

(a) The chlorine is carried in steel tanks.

Steel is mainly iron.

The chlorine does not react with the tank unless there is a very hot fire.

At high temperatures chlorine reacts with iron to make small crystals of iron chloride, FeCl_3 .

- (i) Fill in the boxes to balance the equation for this reaction.



[1]

- (ii) The equation should also include state symbols.

Put a tick (✓) in the box next to each substance to show its state symbol at room temperature and pressure.

| | s | l | g | aq |
|---------------|---|---|---|----|
| chlorine | | | | |
| iron | | | | |
| iron chloride | | | | |

[1]

(b) Here is some information about one atom of chlorine.

| | |
|----------------------------------|-----------|
| atomic (proton) number | 17 |
| relative mass of the atom | 35 |

Use this information to work out the number of particles in the atom.

protons _____

neutrons _____

electrons _____

[2]

(c) Describe and explain what happens when a chlorine atom changes into a chloride ion.

_____ **[3]**

- (d) Chlorine reacts with sodium to make sodium chloride.**

The word equation for this reaction is

sodium + chlorine → sodium chloride

Write a balanced symbol equation for this reaction.

_____ **[2]**

- (e) Sodium chloride forms crystals.**

Describe the structure of a sodium chloride crystal.

_____ **[3]**

- (f) Sodium chloride crystals do not conduct electricity.**

Sodium chloride does conduct electricity when it is dissolved in water.

Explain why.

_____ **[2]**

[TOTAL: 14]

6 X, Y and Z are three elements in the Periodic Table.

| ELEMENT | X | Y | Z |
|-------------------------------------|----------------------------------|----------------------------------|----------------------------------|
| atomic (proton) number | less than 12 | 12 | more than 12 |
| melting point in °C | 1278 | 649 | 839 |
| density in g/cm³ | 1.85 | 1.74 | 1.54 |
| reaction with water | no reaction | slow | rapid |
| formula of chloride | XCl_2 | YCl_2 | ZCl_2 |
| formula of oxide | XO | YO | ZO |
| melting point of oxide in °C | 2550 | 2852 | 2554 |

**Jo thinks that X, Y and Z are in the same Group.
Ann thinks that they are not.**

Who is right?

Use evidence from the table to support your answer.

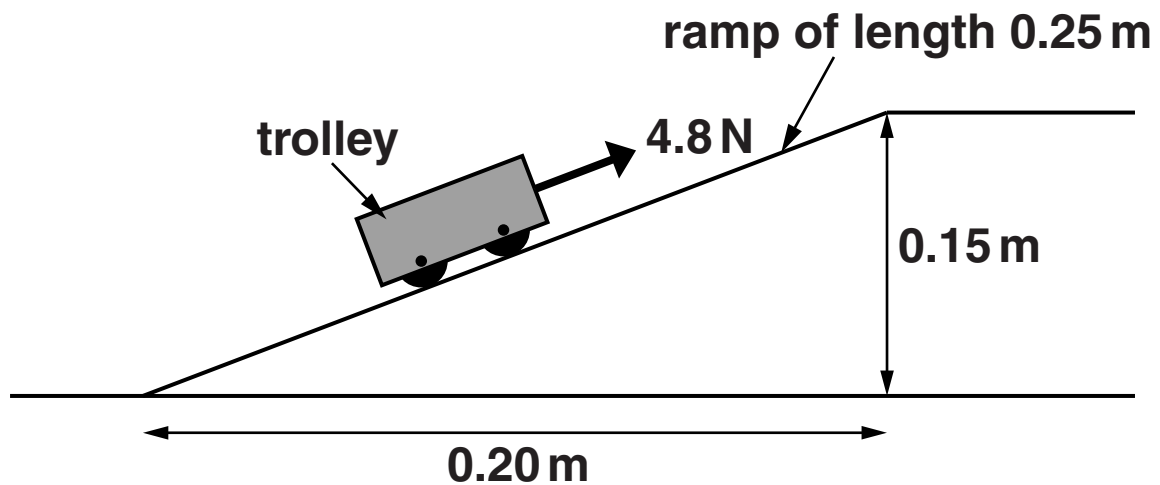


The quality of written communication will be assessed in your answer.

[6]

[TOTAL: 6]

- 7 Enid uses this apparatus to investigate the force needed to pull a trolley up a ramp.



Enid makes a prediction about the experiment.

The work done pulling the trolley up the ramp at a steady speed will always equal its change of gravitational potential energy.

(a) Put ticks (✓) in the boxes next to the TWO assumptions that she is making FOR THIS PREDICTION.

There is no friction to transfer energy as heat.

☐

The kinetic energy of the trolley does not change as it moves.

☐

The total energy of an isolated system always stays the same.

☐

The weight of the trolley decreases as it moves up the ramp.

☐

The force pulling the trolley along the ramp will increase its momentum.

☐

[1]

(b) Enid finds that a 4.8 N force is needed to pull a trolley of weight 6.0 N and mass 0.60 kg up the ramp.

She thinks that this agrees with her prediction. Is she correct? Justify your answer.

[3]

[TOTAL: 4]

- 8 Mike stands on some ice.
Every time he tries to start walking, he stays in the
same place.**

**Explain why it is easy to start walking across grass,
but difficult to start walking across ice.**



**The quality of written communication will be
assessed in your answer.**

[6]

[TOTAL: 6]

9 Clint does some research on bullet-proof vests.

- (a) In one experiment, he finds that a vest can stop a bullet of mass 15 g travelling at a speed of 420 m/s in just 3.0×10^{-5} s.**

$1 \text{ g} = 1.0 \times 10^{-3} \text{ kg}.$

- (i) What is the momentum change of the bullet, in kg m/s, as it stops?**

Put a ring around the correct value.

6.3 kg m/s

63 kg m/s

630 kg m/s

6300 kg m/s

[1]

- (ii) Calculate the force exerted on the vest by the bullet as it stops.**

force = _____ N [1]

(b) Clint investigates the use of foam materials in bullet-proof vests.

These vests have the same weight, but are much thicker.

(i) Explain how the thicker vest could be better for the wearer when it stops a bullet.

[2]

(ii) Clint decides to get his findings published. Five of these sentences describe the process of getting his work published. One of them is NOT part of the peer-review process.

- A Clint sends an article to the publishers.**
- B Clint uses the reports to amend the article.**
- C The publisher sends the article to some other scientists.**
- D The article is published in the scientific journal.**
- E The article is read critically and reports are written about it.**
- F The publisher pays a scientist to repeat Clint's experiments.**

Complete the grid to show the FIVE sentences in the correct order.

| | | | | |
|--|--|--|--|--|
| | | | | |
|--|--|--|--|--|

[2]

(iii) Here are some critical comments scientists make about Clint's work.

ALLAN
My experiments
gave similar
results last year.

BESS
Clint didn't do
enough repeat
measurements.

CARL
The vests that are
used at the moment
work just fine anyway.

DEBS
This discovery
should be kept
secret from
criminals.

Who gives a good scientific reason for NOT publishing Clint's work?

answer _____ [1]

[TOTAL: 7]

10 Here is some data for three different electric cars.

| NAME OF CAR | TOP SPEED IN m/s | ACCELERATING TIME IN s | TOTAL MASS IN kg |
|--------------------|-----------------------------|-----------------------------------|-----------------------------|
| CitiStroll | 25 | 10 | 200 |
| EasyShop | 15 | 5 | 400 |
| GoFar | 20 | 4 | 600 |

The ACCELERATING TIME is how long it takes for each car to reach its top speed from a standing start.

- (a) The manufacturers of GoFar claim that their car has the greatest acceleration.
Are they right? Justify your answer.**

[2]

- (b) Calculate the kinetic energy of a GoFar car at its top speed.**

kinetic energy = _____ J [1]

[TOTAL: 3]

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