Surname

Centre Number Candidate Number

0

Other Names



New GCSE

4791/01

ADDITIONAL APPLIED SCIENCE UNIT 1: Science at Work in Applied Contexts FOUNDATION TIER

A.M. TUESDAY, 22 January 2013

l hour

For Examiner's use only			
Question	Maximum Mark	Mark Awarded	
1.	10		
2.	9		
3.	7		
4.	10		
5.	7		
6.	9		
7.	8		
Total	60		

ADDITIONAL MATERIALS

In addition to this paper you may require a calculator and a ruler.

INSTRUCTIONS TO CANDIDATES

Use black ink or black ball-point pen.

Write your name, centre number and candidate number in the spaces at the top of this page.

Answer all questions.

Write your answers in the spaces provided in this booklet.

INFORMATION FOR CANDIDATES

The number of marks is given in brackets at the end of each question or part-question.

You are reminded that assessment will take into account the quality of written communication used in your answer to question 7(b).

You are reminded to show all your working. Credit is given for correct working even when the final answer given is incorrect.

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Answer **all** the questions in the spaces provided.

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1. Amy visits the doctor because she always feels tired. The doctor takes a blood sample for analysis.

Some of the blood sample is put onto a microscope slide and looked at with a microscope. The blood sample contains red blood cells, white blood cells, platelets and plasma.



- (*a*) Which letter **A**, **B**, **C** or **D** shows the red blood cell? [1]
- (b) Here is a table about the components of blood and their functions. The first row has been completed for you. Complete the table. [2]

Component	Function
red blood cell	to carry oxygen
white blood cell	
	to help form blood clots

(c) Amy is found to be anaemic and is given iron tablets.

Give the chemical symbol for iron.

[1]

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4791 010003

Titration A Titration **B** Titration C Final burette reading 27.7 27.8 25.6 (cm^3) Initial burette reading 5.0 5.0 5.0 (cm^3) Titre (difference) 22.7 20.6 (cm^3) (i) Complete the table. [1] Give one reason why the manufacturer repeats the titration. (ii) [1] (iii) The manufacturer calculates the average titre from the titrations. Suggest which one of the titrations is not included in this calculation and give one reason why. [2] Titration Reason The manufacturer analyses another sample of tablets. They obtain an average titre *(e)* (i) of 22.6 cm³. Calculate the iron content of the tablets using the equation: [1] iron content (mg) average titre = 1.61 Tablets are rejected if the amount of iron is outside the range 13.90 - 14.10 mg. (ii) Give one reason whether or not this sample of tablets should be rejected. [1]

(d)The iron tablets are checked by the manufacturer by carrying out a titration.

The following table shows the titration results.

- Examiner only
- 2. A school group visits a farm to see how food is produced. The produce from the farm is sold in the onsite shop.
 - (a) In the shop they have various products that have been preserved. Complete the following sentences using words from the box below. [2]

	iree s	seven	ten	slows	speeds up	does not change	
Onic	ons can b	e preserv	ved by pick	cling with vi	negar. Vinegar h	as a pH of	
whic	h		1	the growth c	f bacteria.		
Whe raw 1	n serving meat cou	g in the sl nter to tl	hop, the as he cooked	ssistants hav meat count	e to wash their h er.	ands as they move from	ı th
(i)	Give on	e reason	why they	need to was	h their hands.		[
(ii)	Give on	e other p	recaution	that the shoj	o assistants take	when working with raw	an
Yogł	urt is on The pro	e of the	foods proo	duced on the	farm and sold i	n the shop.	tte
(1)	1 1		sequence		ionoo	teps below. Using the le	
	below, j	jut the st	leps in the	correct seq	lence.		[2
	below, j A.	Incuba	te for 3-6	correct seq	ience.		[2
	A. B.	Incuba Cool th	te for 3-6 l te milk to	correct sequences and the sequences and the sequences and the sequences and the sequences are sequence	ience.		[2
	A. B. C.	Incuba Cool th Heat th	te for 3-6 l te milk to te milk to	correct sequences hours 30°C 85°C to kill	contaminant or	ganisms	[2
	A. B. C. D.	Incuba Cool th Heat th Add the	te for 3-6 l te milk to te milk to te yoghurt	correct sequences hours 30°C 85°C to kill culture and	contaminant org	ganisms	[2
	A. B. C. D. One has	Incuba Cool th Heat th Add the s been do	te for 3-6 l te milk to te milk to te yoghurt one for you	correct sequences hours 30°C 85°C to kill culture and u.	contaminant org	ganisms	[2
	A. B. C. D. One has	Incuba Cool th Heat th Add the s been do	te for 3-6 l te milk to te milk to te yoghurt tone for you	correct sequences of the sequences of th	contaminant org	ganisms	[2
(ii)	A. B. C. D. One has	Incuba Cool th Heat th Add the s been do C –	te for 3-6 l te milk to te milk to te wilk to to yoghurt to for you	te'?	contaminant org	ganisms	[2

Litter	Litter 1	Litter 2
Average starting mass of piglet (kg)	7.9	7.8
Average mass gain each week (kg)	1.2	1.5

(d) Piglets are also reared on the farm. The farm is trialling feeding the piglets yoghurt.

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(i) How can you tell from the data in the table that litter **2** must have been fed yoghurt as well as milk from the mother? [1]

(ii) Calculate the average mass gain of a piglet from litter **2** in 4 weeks.

Average mass gain in 4 weeks = kg

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[1]

- Wellsprings mg/cm³ Calcium 1.0 Magnesium 0.2 Sodium 0.8 Potassium 0.1 2.4 Bicarbonate Chloride 1.2 Sulfate 1.0 Nitrate 1.1 Total dry residue 7.9 mg at 180°C pH at source 6.1 mg/cm³ 0.9 Calcium Magnesium 1.1 Sodium 0.1 Potassium 0.1 1.9 Bicarbonate
- 3. "Wellsprings" and "Vital H_2O " are companies producing bottled water. The data from the labels on their bottles is shown below.

Vital H₂O

0.9

0.0

0.1

7.3

5.4 mg

Chloride

Sulfate

Nitrate

at 180°C

pH at source

Total dry residue

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Using the data from the labels, state which water is acidic giving one reason for (a)(i) your answer. [1] [1]

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- (ii) Give one difference, other than pH, between the two bottles of water.
- (b)The table below shows analytical techniques that the companies could use to test their water.

Technique	Observation	Ions present	
	yellow flame	sodium	
Flame test	brick red flame	calcium	
	lilac flame	potassium	
Toot with codium hydroxide	white precipitate forms	calcium	
Test with sodium hydroxide	brown precipitate forms	iron(III)	
Test with barium chloride	white precipitate forms	sulfate	

- Which of these three techniques could not be used to test for the presence of the (i) metal ions in the water? [1]
- (ii) What observations would you expect to make when a sample of each water is tested with barium chloride? [2]

Wellsprings

Vital H₂O

Give one reason why the results of the flame test for 'Wellsprings' water might be (iii) difficult to interpret. [1]

The mass of the dry residue is measured with a balance. Suggest one reason why the (c)companies should regularly check the balance. [1]

Examiner The Jones family make a New Year resolution to get fit. They join a gym. (a)Each person is given an initial fitness assessment. Susan Jones, their 10 year old daughter, has a BMI of 17. Give one reason why this (i) information **may not** be useful to the fitness instructors. [1] Give one example, apart from BMI, of the basic information taken as part of the (ii) fitness assessment. [1] State why this information is kept confidential. (iii) [1]

Mrs Jones has an active job and is 'on her feet' for most of the day. Her training (b)programme includes using the following pieces of equipment.



(i) Which piece of equipment should Mrs Jones use for aerobic exercise? Give one reason for your choice. [2]

Equipment	
-----------	--

Reason

4.

only

	(ii)	Which piece of equipment should Mrs Jones use for building muscle mass ? Give one reason for your choice. [2] Equipment	Examiner only
(c)	(i)	Mr Jones is a taxi driver and does little exercise. Give one reason why his lifestyle could affect his health. [1]	
	(ii)	Give one short term effect on Mr Jones' body when he begins an exercise programme for the first time. [1]	
	(iii)	Give one long term effect on Mr Jones' heart when he exercises regularly. [1]	

The Wildlife Trust manages a number of reserves where they maintain natural habitats. 5.

A farm shares a boundary with a Wildlife Trust reserve. The plants are monitored to avoid the spread of species across the boundary (invasive species).

A quadrat can be used to sample the number of plants.



Sample A is from the boundary between the farmland and Wildlife Trust reserve.

Sample **B** is taken 10 metres from the boundary on the farmland.

(a) Below are two quadrat samples.











Rapeseed

Sample **B**

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	A			
	В			
(ii) -	The farmer gr reserve. Expla farmland.	ows rapeseed as a crop. Tea ain why the farmer needs	sel is a plant that grows in the to monitor the teasel plant	e Wildlife s on his [3]
iii)	Give one reas	on why there is a greater v	ariety of species on farms clo	ose to the
	reserve compa	red to farms further away.	· · ·	[1]

6. Gareth and Kevin entered a 1500 m race. Their performance during the race was measured. The time they take to complete each 300 m stage of the race was measured.

A distance-time graph is plotted below for Kevin.



(a)	Use	the graph to find the:		Examiner only
	(i)	time taken by Kevin to complete the race.	[1]	
		seconds		
	(ii)	fastest 300 m section (A, B, C, D or E) ran by Kevin.	[1]	

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(b) Gareth's performance in the same race is recorded in the table below.

GARETH			
Time (seconds)	Distance (metres)		
0	0		
60	300		
120	600		
180	900		
240	1200		
320	1500		

- (i) On the same graph, **plot** the performance of Gareth using the values shown in the table above. [3]
- (ii) Calculate the mean speed for **Gareth** during the first **900 m** using the equation: [2]

Mean speed = $\frac{\text{distance}}{\text{time}}$

Answer: m/s

(c) Use the data to explain how the fitness of Gareth compares with Kevin. [2]

1)	Gor-Tex [®] is a material often used for the jackets worn by cyclists in the rain. Gor-Tex [®] is
•)	described as a 'breathable' material. What is meant by the term ' <i>breathable</i> '? [2]
5)	Lycra [®] is used to make cyclist's shorts. Thinsulate [®] is used as the filling in their jackets. Explain why the structure and properties of Lycra [®] and Thinsulate [®] make them suitable for these uses. [6] <i>QWC</i>

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END OF PAPER