Centre Number			Candidate Number		
Surname					
Other Names					
Candidate Signature					



General Certificate of Secondary Education Higher Tier June 2014

Additional Applied Science

AAS1HP



Thursday 15 May 2014 9.00 am to 10.00 am

H

For this paper you must have:

- a ruler
- a calculator
- the Equations Sheet (enclosed).

Time allowed

• 1 hour

Instructions

- Use black ink or black ball-point pen.
- Fill in the boxes at the top of this page.
- Answer all questions.
- You must answer the questions in the spaces provided. Do not write outside the box around each page or on blank pages.
- Do all rough work in this book. Cross through any work you do not want to be marked.

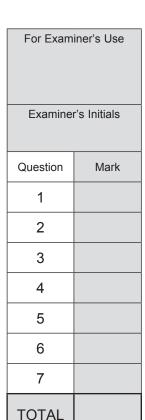
Information

- The marks for questions are shown in brackets.
- The maximum mark for this paper is 60.
- You are expected to use a calculator where appropriate.
- You are reminded of the need for good English and clear presentation in your answers.
- Question 3 should be answered in continuous prose.
 In this question you will be marked on your ability to:
 - use good English
 - organise information clearly
 - use specialist vocabulary where appropriate.

Advice

• In all calculations, show clearly how you work out your answer.





Answer all questions in the spaces provided.

Table 1 shows some of the characteristics of three varieties of tomato plants,A, B and C, and the tomatoes they produce.

Table 1

Characteristic	Tomato plant A	Tomato plant B	Tomato plant C
Average number of tomatoes per plant	25	50	35
Resistance of the plant to disease	none	none	resistant
Height of plant in m	2.5	2.0	1.5
Average diameter of tomatoes in cm	6	8	6

1 (a) (i) An agricultural scientist decides to selectively breed tomato plants B and C to produce a new variety of tomato plant with better characteristics.

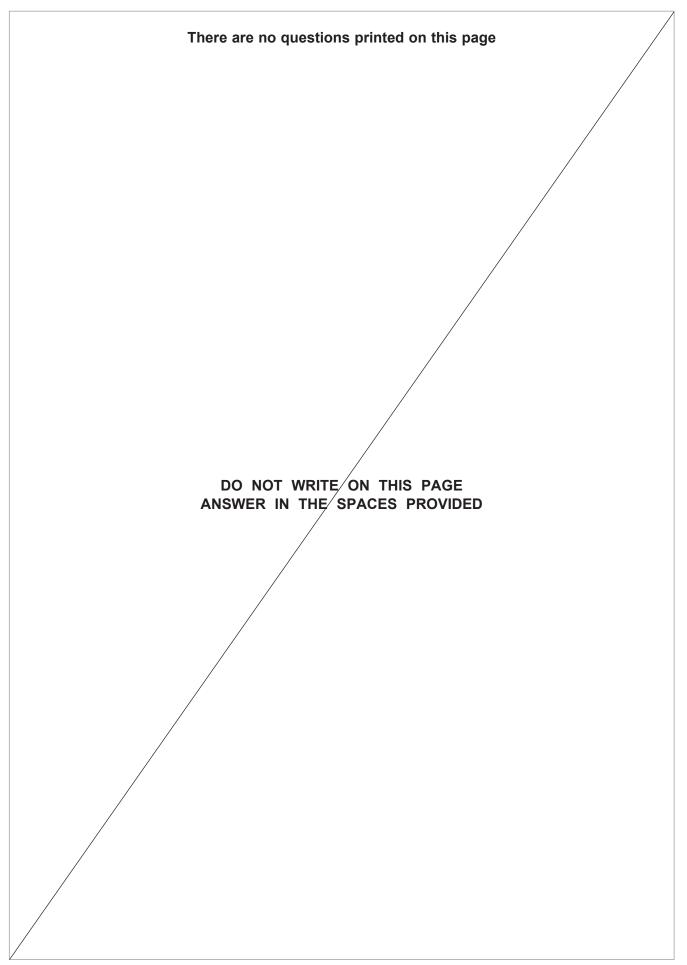
Suggest three reasons for choosing tomato plants B and C .	[3 marks]
1	
2	
3	



1 (a) (ii)	Describe the process of selectively breeding tomato plants B and C to produce plants with the best characteristics.
	[3 marks]
1 (b)	The agricultural scientist recommends using a herbicide. Explain how using a herbicide could increase the yield of tomatoes.
	[2 marks]

Turn over for the next question



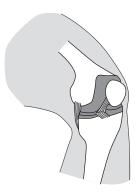




2 A footballer is recovering from a cartilage injury in her knee.

Figure 1 shows a diagram of a knee joint.

Figure 1



2 (a) (i)	Draw an arrow on Figure 1 to show the cartilage.	[1 mark]
2 (a) (ii)	State two functions of the cartilage in a knee joint.	[2 marks]
	1	
	2	

Question 2 continues on the next page



2 (b) To strengthen her leg muscles, the footballer was asked to stretch a length of elastic rope, as shown in **Figure 2**.

Figure 2



The force needed to extend the elastic rope was investigated.

The results are shown in **Table 2**.

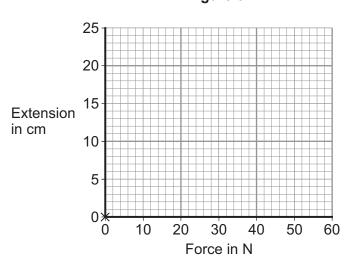
Table 2

Force in N	Extension in cm
0	0
10	4
20	8
30	15
40	16
50	20

2 (b) (i) On the grid provided (Figure 3), plot the results. The first point has been done for you.
Draw a line of best fit.

[3 marks]

Figure 3



2 (b) (ii) Use Figure 3 to find the force needed to extend the elastic rope by 18 cm.

[1 mark]

Force = N

2 (b) (iii) Does the elastic rope obey Hooke's Law?

Explain your answer.

[2 marks]

9

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Turn over for the next question



3	In this question you will be assessed on using good English, organising information clearly and using specialist terms where appropriate.
	Iron rusts when in contact with salty water.
	A student suggests that increasing the amount of salt in the water would cause iron to rust more quickly.
	Design an experiment to investigate whether the student is correct.
	You are provided with normal laboratory apparatus, salt and a mixture of iron nails. [6 marks]
	Extra space

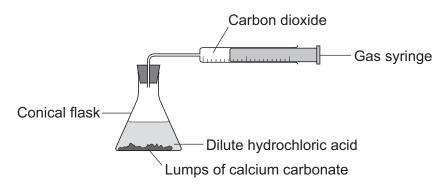


Turn over for the next question	



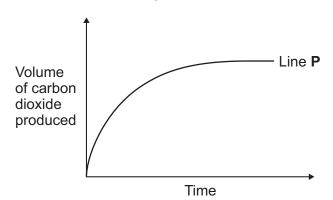
A student investigates the rate of reaction between calcium carbonate and dilute hydrochloric acid, as shown in **Figure 4**.

Figure 4



Line **P** on **Figure 5** shows a graph of the student's results.

Figure 5



4 (a)	Describe what happens to the rate of the reaction during the experiment.				



4 (b) Line Q on Figure 6 shows the student's results when the reaction was repeated under different conditions, but with the same concentration of hydrochloric acid.

Volume of carbon dioxide produced

Time

	Suggest one condition that has changed and explain your answer. [2 marks]
4 (c)	Another way to change the rate of this reaction is to increase the concentration of the hydrochloric acid.
	Explain what happens to the rate of reaction when the concentration of the hydrochloric acid is increased.
	Use ideas about collisions between the reacting particles in your answer. [3 marks]

Question 4 continues on the next page



4 (d)	The student writes a symbol equation for the reaction between calcium carbonate and dilute hydrochloric acid.	
	He calculates the relative formula masses of some of the compounds.	
	$CaCO_3$ + $2HCI$ \longrightarrow $CaCl_2$ + CO_2 + H_2O	
relative formul mass		
4 (d) (i)	Fill in the empty box to show the relative formula mass of calcium chloride.	
	The relative atomic mass of Ca = 40 and of Cl = 35.5. [1 mark]	
4 (d) (ii)	The student uses 5.0 g of calcium carbonate in his experiment.	
	What is the maximum mass of carbon dioxide that can be made from 5.0 g of calcium carbonate?	
	[2 marks]	
	Answer g	
4 (e)	The student collects much less carbon dioxide than he expects from 5.0 g of calcium carbonate.	
	Suggest one reason why. [1 mark]	





5 (a)	The oxygen supply to the muscles is very important for an athlete's success.
	Describe the function of the heart and lungs in providing oxygen to an athlete's muscles.
	[5 marks]
5 (b)	Athletes can respire aerobically or anaerobically during exercise.
	The equations show aerobic respiration and anaerobic respiration.
	Aerobic: $C_6H_{12}O_6 + 6O_2 \longrightarrow 6CO_2 + 6H_2O + 2900 \text{ kJ of energy}$
	Anaerobic: $C_6H_{12}O_6 \longrightarrow 2C_3H_6O_3 + 150 \text{ kJ of energy}$
5 (b) (i)	Use your knowledge of anaerobic respiration to suggest the name of the compound
	with the formula $C_3H_6O_3$ [1 mark]



5 (b) (ii)	Explain why anaerobic respiration releases less energy than aerobic respiration. [2 marks]
5 (c)	Glucose is also needed for an athlete's success.
	During a race, the glucose concentration in the blood becomes too low.
	Describe how the body raises the amount of glucose in the blood if the level of glucose is too low.
	[3 marks]

Turn over for the next question





	10
6 (a)	The properties and uses of a polymer depend on its structure.
	Figure 7 shows where each of three different forces of attraction can be found in a polymer.
	Force A represents the covalent bonds within the monomer.
	Force B represents the covalent bonds between monomers.
	Force C represents cross-links between the polymer chains.
	Figure 7
	Force B
	Polymer chain
	Polymer chain — Force C
	Which two statements describe changes that would increase the melting point of this polymer?
	Tick (✓) two boxes. [2 marks]
	make force A stronger
	make force B stronger
	make force C weaker
	introduce more cross-links
	make the polymer chains longer



6 (b)	In Table 3, which row, P, Q, R or S, describes a thermosetting polymer?
	Tick (✓) one box.

[1 mark]

Table 3

	Softened by heating	Cross-links between chains	Tick (√)
Р	yes	no	
Q	no	yes	
R	yes	yes	
s	no	no	

6 (c)	Explain why a thermoplastic polymer would not be suitable for making the case of a plastic kettle.
	[1 mark]

Turn over for the next question



7 Paint containers are usually made of steel. 7 (a) State **two** properties of steel which make it a suitable material for making paint containers. [2 marks] Property 2 Figure 8 shows a screwdriver being used to remove the lid from a paint container. 7 (b) Figure 8 Downward force 5 N Upward Screwdriver force Lid Perpendicular distance from pivot 20 cm Distance from pivot 0.4 cm 7 (b) (i) The turning effect (moment) produced by the screwdriver is applied to the lid of the paint container. Calculate the upward force on the lid. Give the correct unit in your answer. Use the Equations Sheet and the data from Figure 8 to help you answer the question. [3 marks]

7 (b) (ii)	The paint container is made of steel with a density of 7800 kg/m ³ .
	The mass of the empty paint container and lid is 0.39 kg.
	Calculate the volume of steel used to make the paint container and lid.
	Use the Equations Sheet to help you answer the question. [3 marks]
	Volume = m ³
7 (b) (iii)	Some paint containers are now made of a plastic material with a density of 920 kg/m ³ .
	Use your knowledge of materials to give one advantage and one disadvantage of using plastic instead of steel for paint containers.
	[2 marks]
	Advantage
	Disadvantage

END OF QUESTIONS



There are no questions printed on this page

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Question 2: Figure 2 © Thinkstock Question 7: Figure 8 © Getty

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