

2006 HIGHER SCHOOL CERTIFICATE **EXAMINATION**

Physics

General Instructions

- Reading time 5 minutes
- Working time 3 hours
- Write using black or blue pen
- Draw diagrams using pencil
- Board-approved calculators may be used
- A data sheet, formulae sheets and Periodic Table are provided at the back of this paper
- Write your Centre Number and Student Number at the top of pages 13, 17, 19, 23, 25 and 27

Total marks - 100

(Section I

) Pages 2–28

75 marks

This section has two parts, Part A and Part B

Part A – 15 marks

- Attempt Questions 1–15
- Allow about 30 minutes for this part

Part B – 60 marks

- Attempt Questions 16–27
- Allow about 1 hour and 45 minutes for this part

Section II) Pages 29–44

25 marks

- Attempt ONE question from Questions 28–32
- Allow about 45 minutes for this section

Section I 75 marks

Part A – 15 marks Attempt Questions 1–15 Allow about 30 minutes for this part

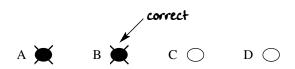
Use the multiple-choice answer sheet.

Select the alternative A, B, C or D that best answers the question. Fill in the response oval completely.

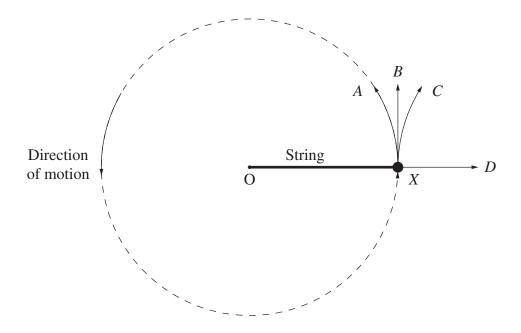
Sample: 2 + 4 = (A) 2 (B) 6 (C) 8 (D) 9 A \bigcirc B \bigcirc C \bigcirc D \bigcirc

If you think you have made a mistake, put a cross through the incorrect answer and fill in the new answer.

If you change your mind and have crossed out what you consider to be the correct answer, then indicate the correct answer by writing the word **correct** and drawing an arrow as follows.



- Given that G is the universal gravitational constant, and g is the magnitude of the acceleration due to gravity, which statement is true?
 - (A) The values of G and g depend on location.
 - (B) The values of G and g are independent of location.
 - (C) G is the same everywhere in the universe, but g is not.
 - (D) g is the same everywhere in the universe, but G is not.
- A mass attached to a length of string is moving in a circular path around a central point, O, on a flat, horizontal, frictionless table. This is depicted in the diagram below. The string breaks as the mass passes point *X*.



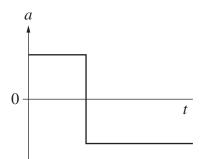
Which line best depicts the subsequent path of the mass?

- (A) Line A
- (B) Line B
- (C) Line C
- (D) Line D
- 3 What is the main reason why the Michelson-Morley experiment is considered important?
 - (A) It shows the existence of the aether.
 - (B) It suggests that light is an electromagnetic wave.
 - (C) It indicates that light can exhibit interference effects.
 - (D) It provides experimental support for the theory of relativity.

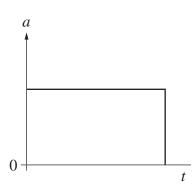
4 A stone is thrown horizontally from the top of a cliff and falls onto the beach below.

Which acceleration—time graph best describes the motion of the stone?

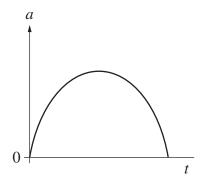
(A)



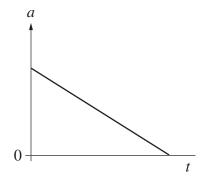
(B)



(C)



(D)

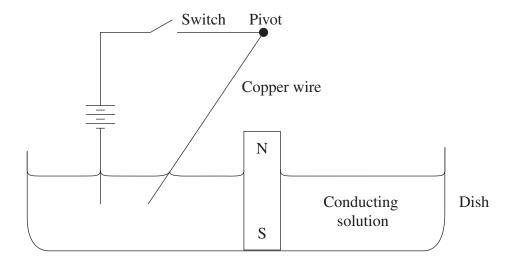


5 Two satellites, *X* and *Y*, are in circular orbits around Earth. Their masses are identical and their orbital radii are R and 16R, respectively.

What is the ratio of their orbital periods, T_X : T_Y ?

- (A) 1:4
- (B) 1:16
- (C) 1:32
- (D) 1:64

6 The diagram shows a magnet standing on the bottom of a dish filled with a conducting solution. A copper wire is suspended freely from a point above the magnet with its tip in the conducting solution. It is held in the position shown.

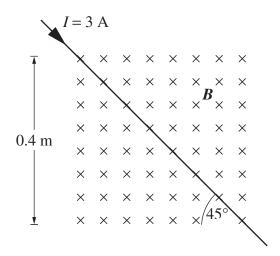


The switch is closed and the wire released.

Which of the following will be observed?

- (A) The wire will rotate about the magnet.
- (B) The wire will be attracted to the magnet.
- (C) The magnet will rotate about its vertical axis.
- (D) The solution in the dish will rotate about the magnet.

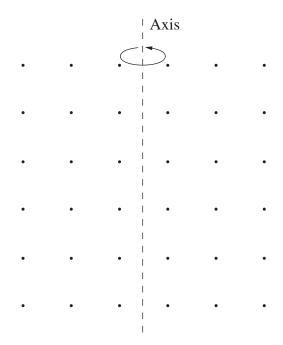
A current-carrying conductor passes through a square region of magnetic field, magnitude 0.5 T, as shown in the diagram. The magnetic field is directed into the page.



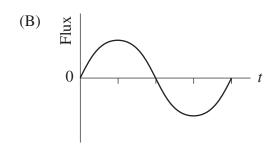
What is the magnitude of the magnetic force on the conductor?

- (A) 0.170 N
- (B) 0.424 N
- (C) 0.600 N
- (D) 0.849 N

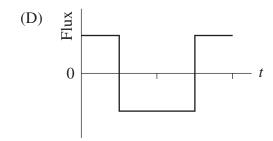
A square loop of wire, in a uniform magnetic field, is rotating at a constant rate about an axis as shown. The magnetic field is directed out of the plane of the page. At time t = 0 the plane of the loop is perpendicular to the magnetic field and side XY is moving out of the page.



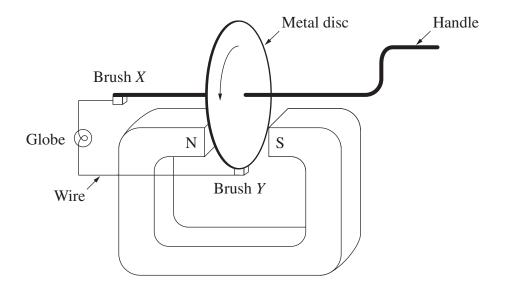
Which graph best represents the variation of the magnetic flux through the loop with time?



 $\begin{array}{c|c} (C) & \underset{H}{\times} \\ & 0 \end{array}$



9 Early electric generators were often very simple. A hand-operated version is depicted below.

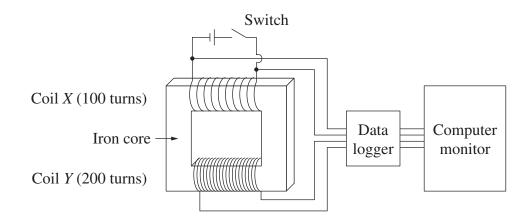


Brush *X* touches the metal axle and Brush *Y* touches the rim of the disc.

If the metal disc is rotated uniformly as shown, which statement about the current through the globe is correct?

- (A) No current flows.
- (B) A direct current flows from Y to X.
- (C) A direct current flows from X to Y.
- (D) An alternating current flows between *X* and *Y*.

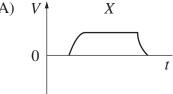
10 The apparatus shown is designed to investigate the operation of a transformer.

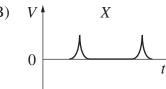


A student closes the switch for a short time, then opens it. The data logger records values of voltage for both coils for the duration of the investigation. The data logger software displays the results as a pair of voltage-time graphs on a computer monitor.

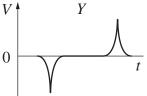
Which pair of graphs best depicts the student's results?

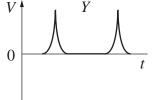
(A)



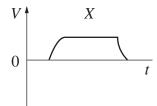


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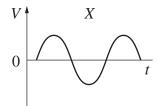




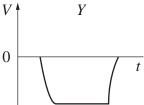
(C)

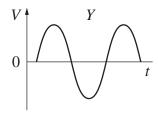


(D)



 V^{I}





11 Lawrence and William Bragg used X-rays to determine the crystal structure of materials.

Which property of waves was the basis of their technique?

- (A) Diffraction
- (B) Dispersion
- (C) Polarisation
- (D) Rarefaction
- 12 A charged non-magnetic particle is moving in a magnetic field.

What would NOT affect the magnetic force on the particle?

- (A) The strength of the magnetic field
- (B) The magnitude of the charge on the particle
- (C) The velocity component parallel to the magnetic field direction
- (D) The velocity component perpendicular to the magnetic field direction
- 13 The temperature of a metal is reduced.

Which statement correctly identifies the change in its electrical resistance and the reason for this change?

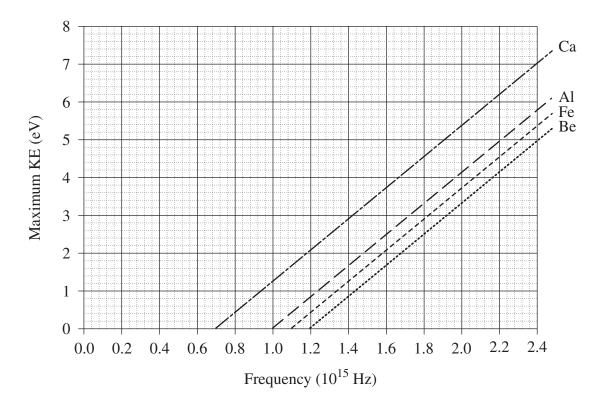
	Electrical resistance	Reason
(A)	Increases	Metal freezes
(B)	Decreases	More free electrons available
(C)	Increases	Electrons move more slowly
(D)	Decreases	Reduced metal lattice vibrations

14 A potential difference of 50 V is applied between two identical, parallel aluminium plates which are separated by a distance of 10 mm.

In order to double this electric field strength, which new arrangement should be used?

	Separation (mm)	Potential difference (V)	Plates
(A)	20	100	Aluminium
(B)	5	50	Perspex
(C)	10	100	Copper
(D)	20	50	Aluminium

When electromagnetic radiation shines on metals, photoelectrons may be emitted. The maximum kinetic energy of emitted photoelectrons is plotted against radiation frequency for four metals as shown in the graph.



Electromagnetic radiation of wavelength 187 nm shines upon an unknown metal and the maximum kinetic energy of the photoelectrons is found to be 2.5 eV.

Based on this information, what is the unknown metal?

- (A) Al
- (B) Be
- (C) Ca
- (D) Fe

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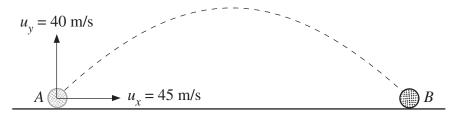
Physics							
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Section I (continued)							
Part B – 60 marks Attempt Questions 16–27 Allow about 1 hour and 45 minutes for this part				Stı	uden	t Nui	nbei
Answer the questions in the spaces provided.							
Show all relevant working in questions involving cal	cula	tions					
Ouestion 16 (6 marks)							

Please turn over

434 - 13 -

Question 16 (6 marks)

A projectile leaves the ground at point A with velocity components as shown in the diagram. It follows the path given by the dotted line and lands at point B.



(a)	State the horizontal component of the projectile's velocity when it lands.	1
(b)	Find the magnitude of the initial velocity of the projectile.	1
(c)	Calculate the maximum height attained by the projectile.	2
(d)	Calculate the range of the projectile, if it lands level with its starting position.	2

Question 17 (6 marks)	Marks
Parts of a space mission involve a spacecraft spending time in geostationary orbit, and then returning safely to Earth.	6
Analyse the forces acting on this spacecraft during these parts of the mission.	

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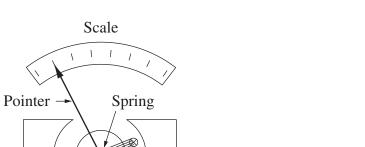
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Section I – Part B (continued)									
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Question 18 (3 marks)								1,1	
An object is stationary in space and located at a cartain planet. It is found that 1.0 MJ of work to a stationary point 20 000 km from the centre	needs to	be d							3
Calculate how much more work needs to be do point 80 000 km from the centre of the planet.	ne to mo	ove th	ie ob	ject	to a	statio	onary	I	
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435a - 17 -

Question 19 (3 marks)

The diagram shows the structure of a typical galvanometer.



Current-carrying coil Soft iron core

Describe how the galvanometer operates as an application of the motor effect.

2006 HIGHER SCHOOL CERTIFICATE EXAMINATION Physics							
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Section I – Part B (continued)							
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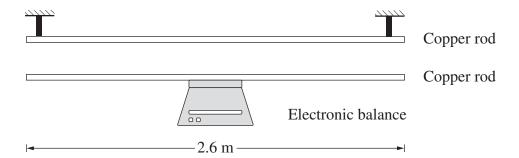
Question 20 (8 marks)

Please turn over

435b - 19 -

Question 20 (8 marks)

A balance was used to investigate the relationship between current and force. The balance was set up with one copper rod fixed to it and a second rod fixed above it, as shown in the diagram. Each rod was connected to a source of current. The diagram is not to scale.



The copper rods were rigid, each was 2.6 m long, and they were parallel. The current in the upper rod was kept constant at 50 A. Different currents were passed through the lower rod and the balance reading recorded for each current. The readings are given in the table below.

Current in lower rod (A)	Balance reading (kg)
2.8	0.5485
8.0	0.5480
12.2	0.5474
16.8	0.5470
20.0	0.5465

(a)	Identify the relative directions of the currents in both rods, and justify your answer.	2

Question 20 continues on page 21

Question 20 (continued)

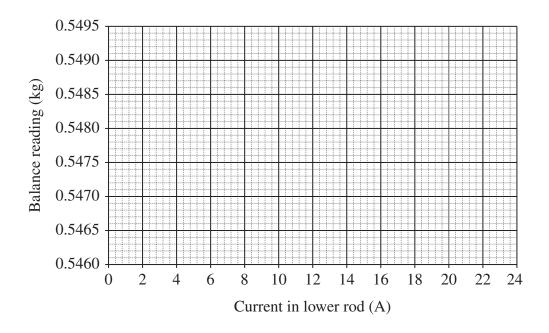
(d)

(b) Plot the data from the table onto the graph, using the scales and axes as indicated, and add the line of best fit (trend line).

2

1

3



(c)	Find the mass of the copper rod on the balance.

Calculate the distance between the two copper rods.

End of Question 20

Question 21 (6 marks)	Marks
Assess the impact on society and the environment of the potential applications of superconductors.	6

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	HIGHER SCHOOL CERTIFICATE EXAMINATIONS	N						entre	Nur	mber
Secti	ion I – Part B (continued)								INUI	
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									Ma	arks
Que	stion 22 (5 marks)									
	udent drops a bar magnet onto a large block net falls towards the copper, slowing down as				_					
(a)	Explain the physics responsible for this obs	ervation	1.							3
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		•••••	•••••		•••••	•••••				
(b)	Predict what will happen if the experiment cooled to approximately –50°C. Justify you	-	•		ith a	cop	per l	olock		2
		•••••	•••••		•••••	•••••	•••••			
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2006 HIGHER SCHOOL CERTIFICATE EXAMIN Physics	
	Centre Number
Section I – Part B (continued)	
	Student Number
Question 24 (3 marks)	Marks
Discuss the origins of unwanted heat producthese can be overcome.	etion in transformers and ways in which 3

Please turn over

436b - 25 -

Question 25 (6 marks)

Awaiting Copyright Clearance
Outline the roles of the deflection plates and the electrodes in the electron gun.
increased so that an electron gains a velocity of $0.60 c$, where c is the speed of light. The electron starts from rest at the cathode.
In a special investigation, the voltage between the cathode and the anode is increased so that an electron gains a velocity of 0.60 c, where c is the speed of light. The electron starts from rest at the cathode. Calculate the mass of this electron in the laboratory frame of reference.
increased so that an electron gains a velocity of $0.60 c$, where c is the speed of light. The electron starts from rest at the cathode.
increased so that an electron gains a velocity of 0.60 c, where c is the speed of light. The electron starts from rest at the cathode. Calculate the mass of this electron in the laboratory frame of reference. The distance between the anode and the screen, as measured in the electron's

2006 HIGHER SCHOOL CERTIFICATE EXAMINATION Physics									
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Section I – Part B (continued)									
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Question 26 (4 marks)								171	ui ii
Beginning in the late 19th century, observations radiation and the photoelectric effect led physicists light.		_					-		4
Use the above as an example to explain how scientist	sts test	t and	l val	idate	mod	dels.			
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437 - 27 -

Question 27 (4 marks)

	ücker was the first to observe cathode rays within gas discharge tubes. He inferred the rays were a form of electromagnetic radiation.	
(a)	Describe ONE subsequent observation that led other scientists to argue that cathode rays were charged particles.	2
(b)	Identify ONE potential hazard associated with performing discharge tube investigations, and outline ONE safe work practice which addresses this hazard.	2

2006 HIGHER SCHOOL CERTIFICATE EXAMINATION

Physics

Section II

25 marks Attempt ONE question from Questions 28–32 Allow about 45 minutes for this section

Answer the question in a writing booklet. Extra writing booklets are available.

Show all relevant working in questions involving calculations.

	Page	S
Question 28	Geophysics	4
Question 29	Medical Physics	6
Question 30	Astrophysics	8
Question 31	From Quanta to Quarks	0
Question 32	The Age of Silicon	4

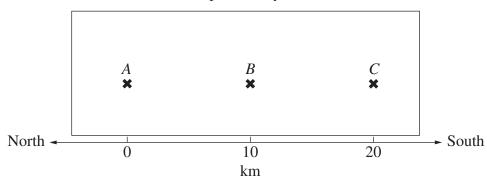
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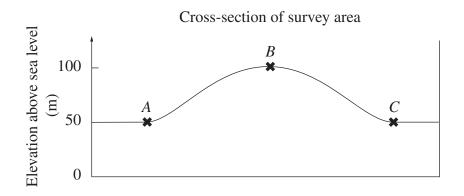
Question 28 — Geophysics (25 marks)

(a) Gravimetric surveys collect raw gravity data which must be further processed by the application of a number of corrections. The following data table, map, and cross-section are the result of a small-scale gravimetric survey.

Station	Gravitational reading (raw data) (mgal)
A (datum station)	979158.1
В	979161.6
C	979174.2

Map of survey area





Question 28 continues on page 31

Gravity data reduction involves the application of the following corrections:

Quantity	Correction	Application comment
Latitude correction	0.8 mgal/km	Subtract a correction for each kilometre south of datum
Free air correction	0.3 mgal/m	Add a correction for each elevation metre above sea-level
Bouguer correction	0.1 mgal/m	Subtract a correction for each metre above the datum station

The data for station A, the datum station, has been processed as below:

Correction or calculation item	Station A (datum)
Latitude correction	$-(0 \times 0.8)$
Free air correction	+(50 × 0.3)
Bouguer correction	$-(0 \times 0.1)$
Total correction	+15.0
Observed g at station	979158.1
Corrected g at station	979173.1
Corrected g at datum station	979173.1
Gravity anomaly at station (corrected <i>g</i> – corrected datum <i>g</i>)	0

- (i) Outline the role of the Bouguer correction in the reduction of gravity data.
- (ii) Using a table format and the information provided, reduce the gravity data for stations *B* and *C*, and draw an inference about the geology of the survey data.

Question 28 continues on page 32

4

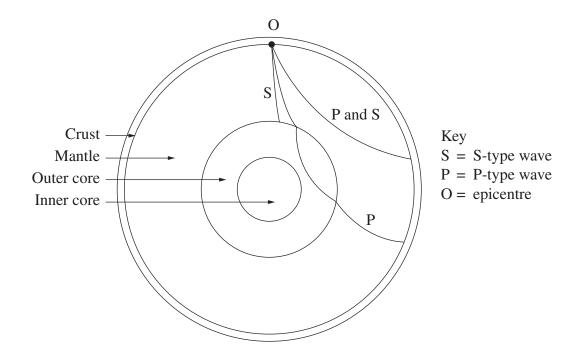
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(b) (i) During your study of geophysics you carried out a first-hand investigation to model the principles of reflection and refraction of seismic waves.

Describe how you ensured that the data obtained in your investigation were reliable.

(ii) The diagram shows the path of seismic waves through Earth.

Account for the curved path followed by all the waves, and justify why the S-type wave stops at the outer core and the P-type wave does not.

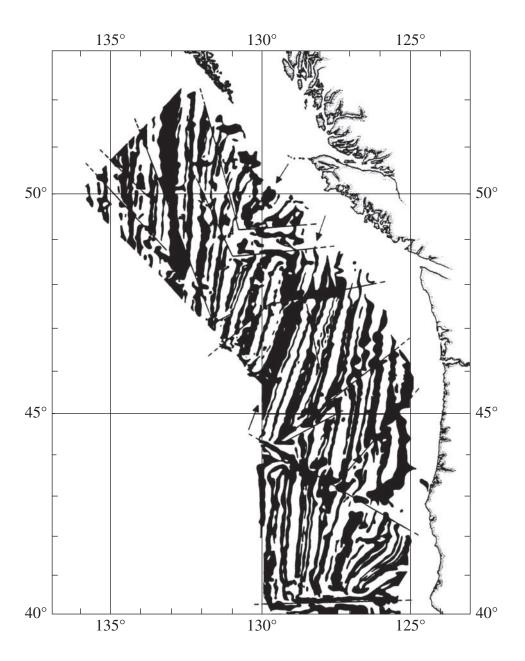


(c) 'Our understanding of Earth has changed as a result of developments in geophysics.'

Discuss this statement with reference to the geophysical methods you have studied.

Question 28 continues on page 33

(d) (i) The map below shows a pattern of magnetic anomalies recorded in the rocks of the seabed. The stripes represent areas of positive and negative magnetic anomaly.



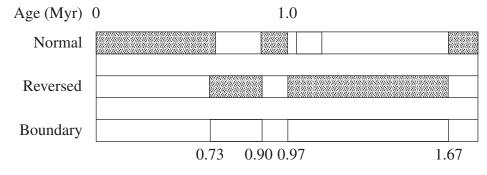
Explain the formation of these paleomagnetic patterns.

Question 28 continues on page 34

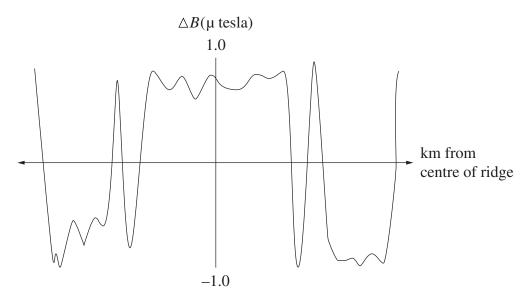
(ii) The spreading rate of an ocean can be calculated using paleomagnetic data as shown.

3

Use the information provided to calculate the spreading rate of the ocean, in units of mm/yr.



Magnetic polarity time scale



Magnetic anomaly profile across the mid-ocean ridge

This diagram is to scale.

1 mm = 1 km

End of Question 28

3

Question 29 — Medical Physics (25 marks)

(a) The acoustic impedance and density of a number of different types of body tissue, ultrasound gel and air are shown in the table:

Material	Acoustic impedance (kg m ⁻² s ⁻¹ × 10 ⁶)	Density (kg m ⁻³)
Fat	1.38	9.25×10^{2}
Skin	1.52	1.00×10^{3}
Ultrasound gel	1.54	1.01×10^{3}
Air	0.0004	1.3

- (i) Calculate the velocity of sound in fat tissue.
- (ii) Ultrasound gel is used to overcome the excessive reflection from the skin during a scan.

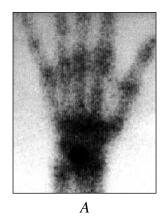
Explain why this is necessary, justifying your response with calculations.

- (b) During your study of Medical Physics you identified data sources, and gathered, processed and presented information to explain why MRI scans can be used to detect abnormalities in the body.
 - (i) Describe the criteria you would use to determine the reliability of a data source for this purpose.
 - (ii) Explain why MRI scans can be used to detect cancerous tissues. 3
- (c) Advances in our understanding of the electromagnetic spectrum have allowed scientists to investigate the human body in more detail.

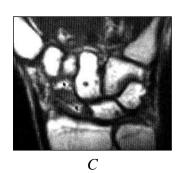
Assess the impact of these advances on the development of medical technologies.

Question 29 continues on page 36

Bones can be viewed with a range of different medical imaging techniques. The images shown below were obtained using three different techniques.







(i) The images shown above are an X-ray, a CAT scan and a bone scan (in no particular order).

Identify the images labelled *A*, *B* and *C*.

Compare the advantages and disadvantages of CAT scans to X-ray (ii) images.

3

3

1

Contrast the information provided by bone scans with that obtained by (iii) CAT scans and X-ray images.

End of Question 29

(i)

objects.

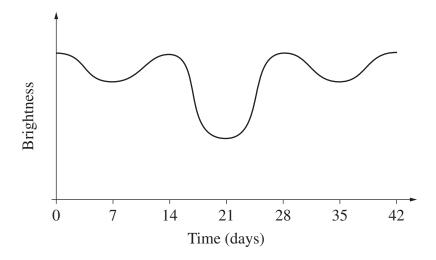
(c)

(a) (i) Describe the spectroscopic observations that would determine whether a particular star is really a binary star system.

2

(ii) The graph represents the variation in brightness of a binary star system.

3



Given that the mass of the system is determined to be 6×10^{32} kg, calculate the average distance between the stars within the system.

- (b) During your study of Astrophysics you performed a first-hand investigation into the spectra produced by different objects under different conditions.
- 2
- (ii) Explain how the absorption spectrum of a star is produced, and how it

Explain how you determined that the data you obtained were reliable.

- 4

can be used to determine the star's composition.

7

Assess the impact of technological advances on our understanding of the cosmos.

Astronomers employ a range of instruments and techniques to observe celestial

Question 30 continues on page 38

	Awaiting Copyright Clearance
(i)	Describe the reactions that occur in stars at the points marked A , B and C in its life cycle.
(ii)	Explain what type and mass of star is most likely to be formed at point A .
iii)	Compare the life cycle of a star that has a mass greater than 10 solar masses with the one depicted at point A .

End of Question 30

Question 31 — From Quanta to Quarks (25 marks)

(a) The Bohr picture of the atom explains the energy of the photons emitted when an electron falls from an initial orbit n_i to find an orbit of n_f .

The various energies depend upon the quantity $\left(\frac{1}{n_f^2} - \frac{1}{n_i^2}\right)$ and the values for this term are given in the table.

Table of values of
$$\left(\frac{1}{n_f^2} - \frac{1}{n_i^2}\right)$$
 for values of n_i and n_f from 1 to 6

n_f n_i	1	2	3	4	5	6
1	_	0.7500	0.8889	0.9375	0.9600	0.9722
2		_	0.1389	0.1875	0.2100	0.2222
3	_	_	_	0.0486	0.0711	0.0833
4	Ι	-	_	_	0.0225	0.0347
5	_	_	_	_	_	0.0122
6	I	ı	-	_	_	ı

- (i) Identify the physical reason for about one-half of the table appearing blank.
- (ii) Calculate the energy of the photon emitted when an electron falls from $n_i = 4$ to $n_f = 3$.
- (b) You have gathered, processed and analysed information related to the development of atomic theory.
 - (i) Describe how you ensured that the information you gathered was reliable.
 - (ii) The atomic theory changed as a result of the contributions of both Heisenberg and Pauli.

Analyse how the work of both scientists modified the atomic theory at that time.

Question 31 continues on page 40

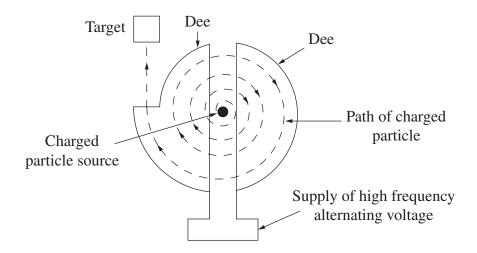
(c) Australia has a large supply of uranium which may be used in fission reactors to create energy. The equation describes the relevant transmutation reaction:

7

$$^{235}_{92}\mathrm{U} \, + \, ^{1}_{0}\mathrm{n} \, \, \rightarrow \, \, ^{141}_{56}\mathrm{Ba} \, + \, ^{92}_{36}\mathrm{Kr} \, + \, 3 ^{\, 1}_{0}\mathrm{n}$$

Analyse how the process described in this equation has been developed into a technology which produces a sustained and controlled amount of energy.

(d) The cyclotron, invented in 1932, accelerates charged particles to a very high speed. The diagram shows the basic design of a cyclotron. The Dees provide a strong magnetic field into the plane of the page.



(i) Explain the physical principles involved in the design of the cyclotron.

3

2

(ii) Account for the use of the cyclotron (or other accelerator) in the development of our understanding of matter.

2

(iii) Quarks are an important part of the Standard Model of Matter. The table shows the six types of quark and their charge (in units of e, the charge on an electron).

Quark	Up	Down	Strange	Charm	Bottom	Тор
Charge	$+\frac{2}{3}e$	$-\frac{1}{3}e$	$-\frac{1}{3}e$	$+\frac{2}{3}e$	$-\frac{1}{3}e$	$+\frac{2}{3}e$

Identify the quark composition of the proton and the neutron.

- (a) Lately, traffic control authorities in NSW have been replacing the incandescent bulbs in traffic lights with arrays of light emitting diodes (LEDs).
 - (i) Describe the structure and operation of an LED.

(ii) Explain why LEDs are preferable to ordinary light bulbs when used in traffic lights.

3

(b) (i) During your study of the Age of Silicon you identified data sources, and gathered, processed and analysed information to outline the rapid development of electronics.

3

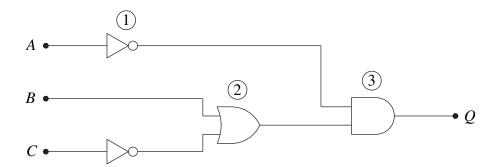
Describe the criteria you would use to determine the reliability of a data source for this purpose.

(ii) Circuits containing logic gates are an integral part of modern electronics.

An example of such a circuit is given below.

3

7



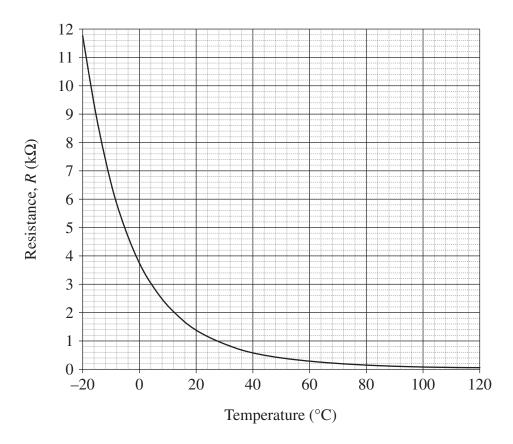
Identify each gate labelled \bigcirc , \bigcirc and \bigcirc and, using a truth table, determine the value of Q when A=0, B=1 and C=1.

(c) Computer developments since World War II (1945) have been characterised by increasing performance at decreasing cost. This trend may or may not continue in the future.

Assess this statement.

Question 32 continues on page 42

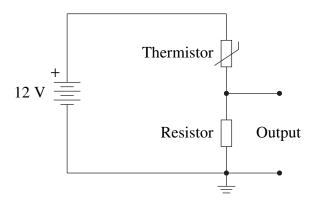
(d) The resistance of a thermistor as a function of temperature is as shown.



(i) Deduce the sign of the temperature coefficient of the thermistor. Explain your answer.

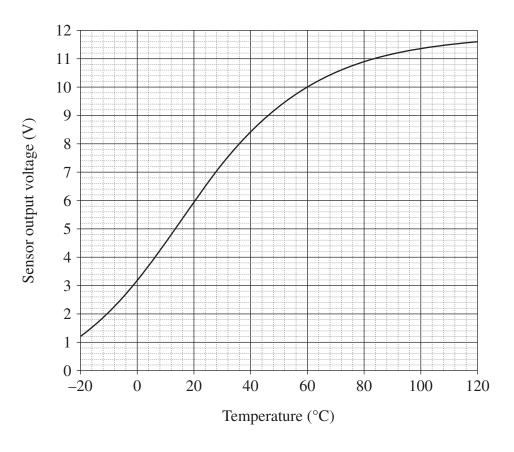
Question 32 continues on page 43

(ii) A potential divider sensor circuit using this thermistor and a fixed resistor is built as shown.



A chamber temperature of $+20^{\circ}$ C is required. On connecting a resistor, the sensor output voltage as a function of temperature is measured and plotted as below.

Determine the approximate sensitivity of the potential divider sensor in the temperature range 6°C to 20°C.



Question 32 continues on page 44

The output of the potential divider sensor is to be converted to give a -10.0 V signal at 20°C.

Design a suitable feedback amplifier to achieve this conversion. Base your circuit on an operational amplifier and assume a $\pm 15\,\mathrm{V}$ power supply. Show, using a fully labelled diagram, how your circuit should be connected to the temperature sensor.

End of paper

2006 HIGHER SCHOOL CERTIFICATE EXAMINATION

Physics

DATA SHEET

Charge on electron, q_e	$-1.602 \times 10^{-19} \mathrm{C}$
---------------------------	-------------------------------------

Mass of electron,
$$m_{\rho}$$
 9.109 × 10⁻³¹ kg

Mass of neutron,
$$m_n$$
 1.675 × 10⁻²⁷ kg

Mass of proton,
$$m_p$$
 1.673 × 10⁻²⁷ kg

Earth's gravitational acceleration,
$$g$$
 9.8 m s⁻²

Speed of light,
$$c$$
 3.00 × 10⁸ m s⁻¹

Magnetic force constant,
$$\left(k \equiv \frac{\mu_0}{2\pi}\right)$$
 $2.0 \times 10^{-7} \text{ N A}^{-2}$

Universal gravitational constant,
$$G$$
 6.67 × 10⁻¹¹ N m² kg⁻²

Mass of Earth
$$6.0 \times 10^{24} \text{ kg}$$

Planck constant,
$$h$$
 6.626 × 10⁻³⁴ J s

Rydberg constant, *R* (hydrogen)
$$1.097 \times 10^7 \text{ m}^{-1}$$

Atomic mass unit,
$$u$$
 1.661 × 10⁻²⁷ kg

931.5 MeV/
$$c^2$$

$$1 \text{ eV}$$
 $1.602 \times 10^{-19} \text{ J}$

Density of water,
$$\rho$$
 1.00 × 10³ kg m⁻³

Specific heat capacity of water
$$4.18 \times 10^3 \text{ J kg}^{-1} \text{ K}^{-1}$$

FORMULAE SHEET

$$v = f\lambda$$

$$I \propto \frac{1}{d^2}$$

$$\frac{v_1}{v_2} = \frac{\sin i}{\sin r}$$

$$E = \frac{F}{q}$$

$$R = \frac{V}{I}$$

$$P = VI$$

Energy = VIt

$$v_{\rm av} = \frac{\Delta r}{\Delta t}$$

$$a_{\rm av} = \frac{\Delta v}{\Delta t}$$
 therefore $a_{\rm av} = \frac{v - u}{t}$

 $\Sigma F = ma$

$$F = \frac{mv^2}{r}$$

$$E_k = \frac{1}{2}mv^2$$

$$W = Fs$$

$$p = mv$$

Impulse = Ft

$$E_p = -G \frac{m_1 m_2}{r}$$

$$F = mg$$

$$v_x^2 = u_x^2$$

$$v = u + at$$

$$v_y^2 = u_y^2 + 2a_y \Delta y$$

$$\Delta x = u_x t$$

$$\Delta y = u_y t + \frac{1}{2} a_y t^2$$

$$\frac{r^3}{T^2} = \frac{GM}{4\pi^2}$$

$$F = \frac{Gm_1m_2}{d^2}$$

$$E=mc^2$$

$$l_{v} = l_{0} \sqrt{1 - \frac{v^{2}}{c^{2}}}$$

$$t_{v} = \frac{t_{0}}{\sqrt{1 - \frac{v^{2}}{c^{2}}}}$$

$$m_v = \frac{m_0}{\sqrt{1 - \frac{v^2}{c^2}}}$$

FORMULAE SHEET

$$\frac{F}{l} = k \frac{I_1 I_2}{d}$$

$$d = \frac{1}{p}$$

$$F = BIl \sin \theta$$

$$M = m - 5\log\left(\frac{d}{10}\right)$$

$$\tau = Fd$$

$$\frac{I_A}{I_B} = 100^{\left(m_B - m_A\right)/5}$$

$$\tau = nBIA\cos\theta$$

$$m_1 + m_2 = \frac{4\pi^2 r^3}{GT^2}$$

$$\frac{V_p}{V_s} = \frac{n_p}{n_s}$$

 $F = qvB\sin\theta$

$$\frac{1}{\lambda} = R \left(\frac{1}{n_f^2} - \frac{1}{n_i^2} \right)$$

$$E = \frac{V}{d}$$

$$\lambda = \frac{h}{mv}$$

$$E = hf$$

$$A_0 = \frac{V_{\text{out}}}{V_{\text{in}}}$$

$$c = f\lambda$$

$$\frac{V_{\text{out}}}{V_{\text{in}}} = -\frac{R_{\text{f}}}{R_{\text{i}}}$$

$$Z = \rho v$$

$$\frac{I_r}{I_0} = \frac{\left[Z_2 - Z_1\right]^2}{\left[Z_2 + Z_1\right]^2}$$

	2 He	4.003	попоп	10	Se	20.18	Neon	18	Ar	39.95	Argon	36	Kr	83.80	Krypton	54	Xe	131.3	Xenon	98	Kn	[222.0]	Radon			
				91	Ļ	19.00	Fluorine	17	ご	35.45	Chlorine	35	Br	79.90	Bromine	53	Ι	126.9	Iodine	85	At	[210.0]	Astatine			
				∞ ()	16.00	Oxygen	16	2	32.07	Sulfur	34	Se	78.96	Selenium	52	Te	127.6	Tellurium	84	Ро	[209.0]	Polonium			
				<u></u> ;	Z	14.01	Nitrogen	15	Ч	30.97	Phosphorus	33	As	74.92	Arsenic	51	Sb	121.8	Antimony	83	B1	209.0	Bismuth			
				90	<u>ن</u>	12.01	Carbon	14	2	28.09	Silicon	32	Ge	72.64	Germanium	50	Sn	118.7	Tin	82	Po	207.2	Lead			
				νç	В	10.81	Boron	13	AI	26.98	Aluminium	31	Сa	69.72	Gallium	49	In	114.8	Indium	81	II	204.4	Thallium			
PLTC												30	Zu	65.41	Zinc	48	Cq	112.4	Cadmium	08	ВН	200.6	Mercury			
FI FMFNTS					ment		ent					29	Cn	63.55	Copper	47	Ag	107.9	Silver	$6\dot{L}$	Au	197.0	Gold	111 Rg	[272]	Roentgenium
OF THE			_	-	Symbol of element		Name of element	1				28	Z	58.69	Nickel	46	Pd	106.4	Palladium	78	F.	195.1	Platinum	110 Ds	[271]	Darmstadtium
		KEY		79	Au	197.0	Gold					27	ပိ	58.93	Cobalt	45	Rh	102.9	Rhodium	ĹĹ	Ħ	192.2	Iridium	109 Mt	[568]	Meimerium
PERIONIC TAREE				Atomic Number		Atomic Weight						56	Fe	55.85	Iron	44	Ru	101.1	Ruthenium	9 <u>Z</u>	S	190.2	Osmium	108 Hs	[277]	Hassium
PFPIC				A		4						25	Mn	54.94	Manganese	43	$^{ m Lc}$	[97.91]	Technetium	75	Re	186.2	Rhenium	107 Bh	[264.1]	Bohrium
												24	Ċ	52.00	Chromium	42	Mo	95.94	Molybdenum	74	>	183.8	Tungsten	106 Sg	[266.1]	Seaborgium
																								105 Db	_	
												22	Ţ	47.87	Titanium	40	Zr	91.22	Zirconium	72	HĪ	178.5	Hafnium	104 Rf	[261.1]	Rutherfordium
			_																	57–71				89–103		Actinides
			4		Re	9.012	Beryllium	12	N N	24.31	Magnesium	20	Ca	40.08	Calcium	38	Sr	87.62	Strontium	56	Ва	137.3	Barium	88 Ra	[226.0]	Radium
	1 H	1.008 Hydrogen	11) diogoni	ω;	ī	6.941	Lithium	17	Na	22.99	Sodium	19	×	39.10	Potassium	37	Rb	85.47	Rubidium	55	S	132.9	Caesium	87 Fr	[223.0]	Francium
																						_				

57	28	59	9	61	62	63	2	65	99	29	89	69	70	71
La	రి	Pr	PN	Pm	Sm	Eu	рS	$^{\mathrm{Tp}}$	Dy	Но	Ē	Tm	Yb	Lu
138.9	140.1	140.9	144.2	[144.9]	150.4	152.0	157.3	158.9	162.5	164.9	167.3	168.9	173.0	175.0
Lanthanum	Cerium	Praseodymium	Neodymium	Promethium	Samarium	Europium	Gadolinium	Terbium	Dysprosium	Holmium	Erbium	Thulium	Ytterbium	Lutetium

Ľ	175.0	utetium			103	Lr	[262.1]	wrencium
Yp	73.0	terbium					[259.1]	
Tm							[258.1] [2	
								_
四四	167.	Erbiun			100	Fm	[257.1]	Fermin
Ho	164.9	Holmium			66	Es	[252.1]	Einsteinium
Dy	162.5	Dysprosium			86	Cţ	[251.1]	Californium
T.	158.9	Terbium			6	Bk	[247.1]	Berkelium
g	157.3	Gadolinium			96	Cm	[247.1]	Curium
E	152.0	Europium			95	Am	[243.1]	Americium
Sm	150.4	Samarium			94	Pu	[244.1]	Plutonium
Pm	[144.9]	Promethium			93	dZ	$[23\bar{7}.0]$	Neptunium
PZ	144.2	Neodymium			92	n	238.0	Uranium
Pr	140.9	Praseodymium			91	Pa	231.0	Protactinium
දුද	140.1	Cerium			06	Th	232.0	Thorium
Ľa	138.9	Lanthanum		Actinides	68	Ac	[227.0]	Actinium
			-	,				

Where the atomic weight is not known, the relative atomic mass of the most common radioactive isotope is shown in brackets. The atomic weights of Np and Tc are given for the isotopes 237 Np and 99 Tc.