Please check the examination details below before entering your candidate information					
Candidate surname	Other names				
Pearson BTEC Level 1/Level 2 First Award	Learner Registration Number				
Tuesday 14 May	Tuesday 14 May 2019				
Afternoon (Time: 1 hour)	Paper Reference 20460E				
Applied Science Unit 1: Principles of Science					

Instructions

- Use **black** ink or ball-point pen.
- **Fill in the boxes** at the top of this page with your name, centre number and learner registration number.
- Answer **all** questions.
- Answer the questions in the spaces provided
 - there may be more space than you need.

Information

- The total mark for this paper is 54.
- The marks for **each** question are shown in brackets
 - use this as a guide as to how much time to spend on each question.

Advice

- Read each question carefully before you start to answer it.
- Try to answer every question.
- Check your answers if you have time at the end.

Turn over ▶



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Answer ALL questions. Write your answers in the spaces provided.

For multiple-choice questions put a cross in each correct box \boxtimes to indicate your answer. If you change your mind, put a line through the box \boxtimes and then put a cross in another box \boxtimes .

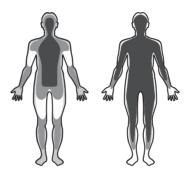
SECTION A: Biology

1 Different processes happen in the human body to keep the temperature constant.

Figure 1 shows two people.

Person A is too cold.

Person B is too warm.



person A person B

Figure 1

/_\	/:\	C:	*la = * la = .a .a = .a = :.a	نم برام ماما ماط	£ ^ + _	warm up their body.
(a)	(1)	Give one process	that happens in	the boay of	t person a to	warm up their body.
(~)	۱.,	cive one process	and mappens m		. pc.50	mann ap anen boay.

(ii) Give **two** processes that happen in the body of person B to cool down their body.

1______

(b) The processes used to keep the body temperature constant are involuntary responses.

State what is meant by the term **involuntary response**.

(1)

(1)

(Total for Question 1= 4 marks)

2 (a) Figure 2 shows a type of neurone.

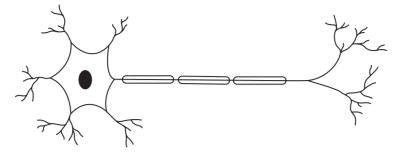


Figure 2

(i) Draw a line to the nucleus of the neurone.

(1)

(ii) State the function of the nucleus.

(1)

(iii) Draw a cross (x) on the axon of the neurone.

(1)

(iv) Name the system in the body that contains neurones.

(1)

(b) Figure 3 shows the gap between neurone A and neurone B.

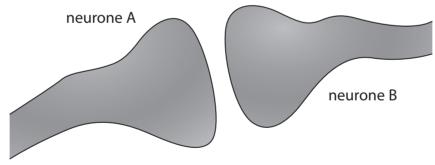


Figure 3

Complete the description to show how an impulse travels from neurone A to neurone B.

(2)

An electrical impulse travels along neurone A.

Neurone A releases which travel across the gap called the

.....

They attach to neurone B and an electrical impulse travels along neurone B.

(Total for Question 2 = 6 marks)

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3	(a)	A genotype can be heterozygous or homozygous. State what is meant by the term heterozygous .	
			(1)
	(b)	State what is meant by the term phenotype .	(1)
	(c)	The peppered moth can be either white speckled or black.	
		The black moth is caused by a genetic mutation.	
		In areas of high pollution, tree trunks are much darker because of high levels of soot in the atmosphere.	
		Figure 4 shows both types of moth on a tree trunk in an area with high levels of soc	ot.
		white speckled moth black r	noth
		Figure 4	
		Explain why, in areas with high levels of soot, this genetic mutation is beneficial to black moths.	
			(2)

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TOTAL FOR SECTION A = 18 MA	ON A = 18 MARKS		
(Total for Question 3 = 8 ma	rks)		
Tou may use a runnett square to support your answer.	(4)		
Explain why all the offspring pea plants produce yellow peas. You may use a Punnett square to support your answer.			
A parent pea plant with the genotype YY is crossed with a parent pea plant with the genotype yy.			
The allele for yellow peas is dominant.			
The allele for yellow peas is Y. The allele for green peas is y.			



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SECTION B: Chemistry

4 (a) Different substances have different hazard symbols.

Draw a line from each hazard symbol to its hazard.

(3)

hazard symbol







hazard

corrosive

explosive

flammable

harmful to the environment

irritant

oxidising

toxic

6

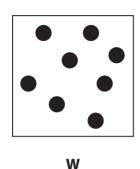


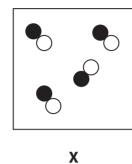
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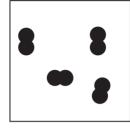
(b) Substances can be atoms or molecules.

They can be elements, compounds or mixtures.

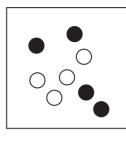
Figure 5 shows four substances W, X, Y and Z.







Y



Z

(1)

(1)

(1)

Figure 5

(i) Which substance is a mixture?

A substance W

substance X

substance **Y**

D substance **Z**

(ii) Which substance is a compound?

A substance W

substance X

substance **Y**

D substance **Z**

(iii) One of the substances could be oxygen.

Give the formula for a molecule of oxygen.

(Total for Question 4 = 6 marks)

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5	Magnesium is a metal. (a) Magnesium reacts with hydrochloric acid to produce magnesium chloride and hydrochloric the word equation for this reaction.	ogen. (1)
	+	(1)
	(iii) A student reacts a piece of magnesium ribbon with hydrochloric acid. State one observation the student would see as the magnesium reacts with the	acid. (1)
	(iv) The hydrogen produced can be tested using a lit splint. Give the positive result of this test.	(1)
	 (b) Magnesium has different isotopes. A sample of magnesium contains: 79% magnesium-24 10% magnesium-25 11% magnesium-26 Calculate the relative atomic mass of this sample of magnesium. 	(2)
	(Total for Question 5 = 6 marl	

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6 Figure 6 shows some information about aluminium.

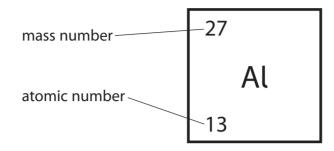


Figure 6

Describe, using the information in Figure 6, the structure of an atom of aluminium.

You may use diagrams to support your answer.

(6)

(Total for Question 6 = 6 marks)



TOTAL FOR SECTION B = 18 MARKS

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SECTION C: Physics

7 (a) Figure 7 shows a wave.

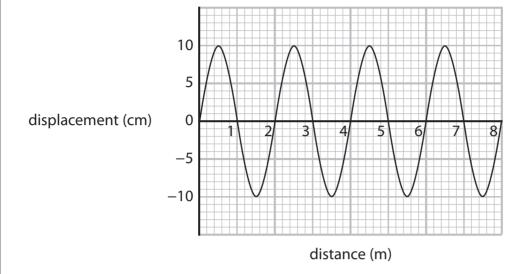


Figure 7

(i) Give the amplitude of the wave.

(1)

(ii) Give the wavelength of the wave.

(1)

(iii) Figure 8 shows a different wave.

Draw, on Figure 8, another wave with a higher frequency than the wave shown.

(1)

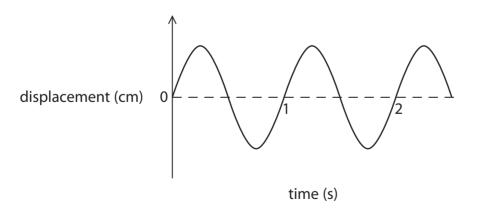


Figure 8



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(b) A wave has a wavelength of 3.0 m and a frequency of 10 Hz.

Calculate the wave speed.

wave speed (m/s) = wavelength (m)
$$\times$$
 frequency (Hz)

(1)

wave speed = m/s

(Total for Question 7 = 4 marks)

(a) Figure 9 shows a digital alarm clock.



Figure 9

(i) State **one** form of useful energy produced by the digital alarm clock.

(1)

(ii) State **one** form of wasted energy produced by the digital alarm clock.

(1)

(b) Figure 10 shows the energy transfers in a digital alarm clock.

total energy supplied

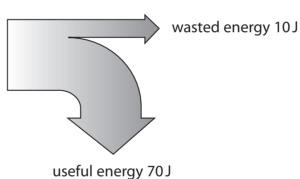


Figure 10

Calculate the efficiency of the digital alarm clock.

(2)

efficiency =
$$\frac{\text{useful energy}}{\text{total energy supplied}} \times 100\%$$

efficiency =



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(c) Figure 11 shows a double glazed window.

A double glazed window has two sheets of glass.

Air is trapped between the two sheets of glass.

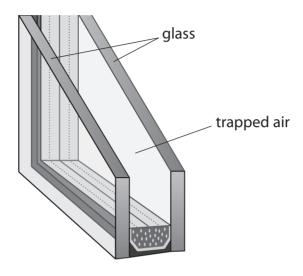


Figure 11

Explain how the double glazed window reduces heat loss from inside a house.

(4)

(Total for Question 8 = 8 marks)

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Figure 12 shows the electromagnetic spectrum.						
	magnetic spect ncreasing frequ		of a series of el	ectromagnetic	waves arrar	nged
increasing frequency						
radio waves	microwaves	infrared	visible light	ultraviolet	X-rays	gamma rays
			Figure 12			
Discuss the	uses and harm	oful effects of	two of the elec	tromagnetic wa	aves in the s	spectrum. (6)
						(0)

TOTAL FOR PAPER = 54 MARKS

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