

Examiners' Report/ Lead Examiner Feedback

November 2016

NQF BTEC Level 1/Level 2 Firsts in Applied Science

Unit 1: Principles of Science (20460E)

ALWAYS LEARNING PEARSON

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January 2016
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## General comments.

Students that did well this series, knew the meanings of key terms and could apply them to their answers to questions posed. They used good scientific language and were able to understand what was being asked for in the question and therefore apply their knowledge of the science well. The best students were able to apply scientific knowledge to new situations, write chemical formula and complete symbol equations.

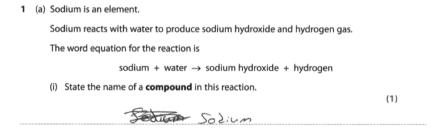
As in previous series, exam technique is still an issue for the weaker students; Centre's need to fully prepare Students for the exam by practicing exam technique, especially in relation to reading the question carefully and not repeating the stem of the question, also re-reading the question with the answer they have given to check that they have answered the question set.

## Feedback on specific questions.

Q1(a) (i) - In general students were able to answer this question well, with many knowing that either water or sodium hydroxide was a compound.

1	(a)	Sodium is an element.	
		Sodium reacts with water to produce sodium hydroxide and hydrogen gas.	
		The word equation for the reaction is	
		sodium + water $\rightarrow$ sodium hydroxide + hydrogen	
		(i) State the name of a <b>compound</b> in this reaction.	
			(1)
	S	dium hydroxide	

Students who got this wrong generally did so as they thought that sodium was a compound.



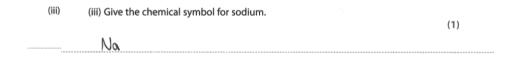
Q1 (a) (ii) - The majority of students were able to recall that the metals were found on the left hand side of the periodic table. In some cases, Students gave the number of a group that contained metals which was acceptable.

(ii) Sodium is a metal.

State where metals are found on the periodic table. (1)	
Left side & periodic table	
In some cases, it seems that students have been revising from a control periodic table and stated that the metals were found 'in the blue so Unfortunately, this and any other coloured sections referred to did any marks as it is not clear where on the periodic table the student referring to.	ection'. not gain
(ii) Sodium is a metal.  State where metals are found on the periodic table.	
Bue Sedion	(1)
Q1 (a)(iii). Many Students were able to correctly recall and write chemical symbol of sodium.	the

Q1 (a)(iii). chemical s

The students that did not know appeared to guess. A very common incorrect answer was 'So'.



In Q2(a)(i) and Q2(a)(ii) It was pleasing to see that most students knew the symbol that should be present on a container of a corrosive substance and that they knew a pH of 1 would be a strong acid.

In Q2(b) (i) few students were able to explain why farmers spread substances such as calcium carbonate on their fields by relating it to the presence of acidic soil and the need to neutralize this soil to score full marks.

(i) Explain why some	farmers spread cal	cium carbonate on th	neir fields.	
				(2)
2				
formers spreament	d Calcu	m Corbonat	e on t	heir fiel
is it news	alises t	re acidic	Soil	allow

In some cases, students did not use subject specific knowledge and tried to use their common knowledge stating that the calcium carbonate was added to help plant growth which was insufficient for credit.

(b) Calcium carbonate reacts with sulfuric acid to form calcium sulfate, carbon dioxide and water.
(i) Explain why some farmers spread calcium carbonate on their fields.
$\Gamma$ (2)
Farmers Spread Calcium Carbonate
on their fidds as the Calcium
Carbonate is used to make some
Crops grow.

Some students thought that the calcium carbonate stopped animals, birds or insects eating the crops. While others said it destroyed bacteria in the soil. Others said that this reaction supplied carbon dioxide and water for the crops to photosynthesize.

	m carbonate rea n dioxide and wa	icts with sulfuric ac ater.	cid to form calci	um sulfate,	
(i) Ex	plain why some	farmers spread cal	cium carbonate	on their fields.	(2)
	to.	Stol arm	us or ba	95 lating	Herc
	crofs				
	***************************************		***************************************		

Q2 (b) (ii) Students still find writing symbol equations difficult, it was found that more than in previous sessions were able to complete the equation for the reaction

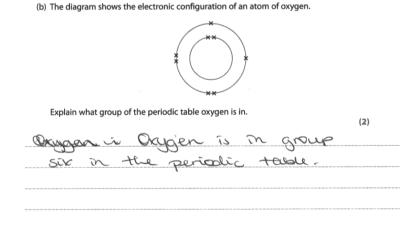
(ii) Complete the equation for the reaction between calcium carbonate and sulfuric acid. (2) 
$$\text{CaCO}_3 + \text{H}_2 \text{SO}_4 \rightarrow \underline{\qquad} \text{CaSO}_{4^*} + \underline{\qquad} \text{CO}_2 \underline{\qquad} + \underline{\qquad} \text{H}_2 \underline{\qquad}$$

Students that lost marks on this question did so because they wrote the word equation or did not follow the correct scientific conventions for writing formulae.

(ii) Complete the equation for the reaction between calcium carbonate and sulfuric acid.
 (2) CaCO<sub>3</sub> + H<sub>2</sub>SO<sub>4</sub> → CaH<sub>2</sub>SO<sub>4</sub> + Ca<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>CaH<sub>2</sub>C

Q3 (a)(i) a good proportion of students were able to recall the charge on the neutron and mass of a proton. Many students were able to relate the number of electrons on the outer shell of oxygen to the group in which it is found for part b, in some cases Students knew oxygen was in group 6 but failed to give a reason why and so did not score the second mark.

When students did not gain marks, it was often as they thought oxygen was in group 8 as it has 8 electrons or group 2 as it has 2 shells.



Some students did not give a group number at all and just said it was on the non-metal side of the Periodic Table or it was in the gases group because it is a gas.

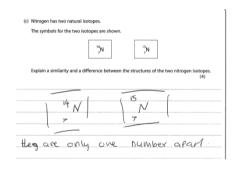
(b) The diagram shows the electronic configuration of an atom of oxygen.

	XX
E	explain what group of the periodic table oxygen is in. (2)
Oxe	ugen is in the garer group, because well its a gas.
• • •	vere able to gain all 4 marks for giving a concise e to explain a similarity and a difference between the
(0	c) Nitrogen has two natural isotopes.
	The symbols for the two isotopes are shown.
	14 N 15 N
	Explain a similarity and a difference between the structures of the two nitrogen isotopes.  (4)
	the two Nitrogen isotopes have
t	the two Nitragen isotopes have the some modern number of 7 vich means
t.l	nat the have the same number of
el	bections and protou, they have have a
olı	effect mass number because the
	the dont have the same
h	umber of neutrons.

However, many students lost marks as they were not specific in their answers. Some students were able to gain 2 marks by stating that the isotopes had the same atomic number but different mass number or same number of protons/electrons but different number of neutrons.

(c) Nitrogen has two natural isotopes.
The symbols for the two isotopes are shown.
14/N 15/N
Explain a similarity and a difference between the structures of the two nitrogen isotopes.
Nitrogen 14 and 15 are simplar as they have the Same atomic number. However they
the Same atomic number. However I they
have diggerent mass numbers. This is what
makes it an isotope.

However, it was quite common for students to get confused and say they had the same mass number but different atomic number or same number of neutrons but different numbers of protons/electrons. Weaker students simply stated what they could see, that they both had a 7 and one had 14 and one had 15 but did not relate this to anything specific.



Students found Q4 (a) (i) and Q4 (a) (ii) quite difficult with many thinking that the energy wasted was kinetic energy.

SECTION B: Physics	
4 Mark has a wind-up torch.	
	,
(a) Winding the handle will charge the battery in the torch.	* + 5 × 1
(i) State <b>one</b> type of energy wasted when winding the handle on the to	rch.
in State site type of energy wasted when what is the harder of the to	(1)
Kinetic Cnergy	***************************************

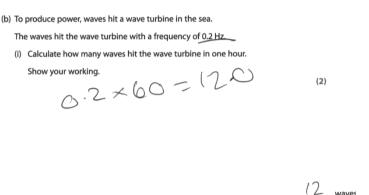
(ii) State the type of energy stored in the battery when the torch has been charged. (1)

				Chantin	Ln	emica			
Q4(b) many stu missing parts of knew the missir Students simply they knew.	the eng par	electr ts bu ssed a	omagn it place	etic spe ed them ve other	ctru in t	ım. I :he in	n some correct	cases, storder. So	tudents ome
(10)				electromagne	tic spe	ectrum.			
				_					
	The diagram shows the electromagnetic spectrum.  Complete the diagram to show the missing parts of the electromagnetic spectrum.  (2)								
rac	io waves	microu	cour infra	wed visible	visible light ultraviolet X-rays gamma rays				
Com	- plete the	diagran	<b>n</b> to show th	e missing par	ts of tl	he electro	omagnetic sp		
					,			(2)	1
radio wa	ves b	wery	Eenergy	visible light	ultr	aviolet	X-rays	gamma rays	

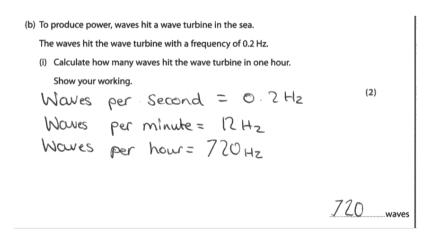
Q5 (a)(ii) 5, the majority of students were able to name one other renewable energy source with the majority giving solar or wind as their answer.

(ii) Name <b>one</b> other renewable energy source.	(1)
Solar energy	

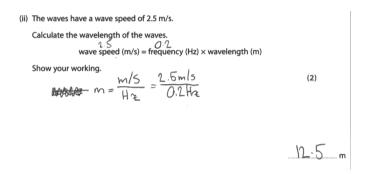
Q5 (b) students often scored 1 mark for calculating the number of waves per minute.



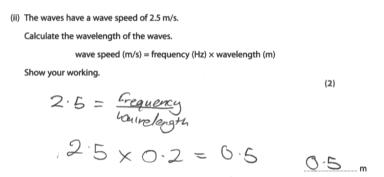
The higher ability students then multiplied by 60 again to get the amount of waves in an hour.



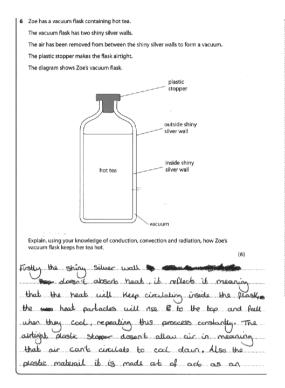
Students seemed to perform better in this second calculation in Q5 than in previous series, with many understanding how to calculate the wavelength of the wave.

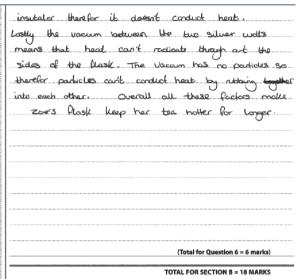


A common mistake that was observed was to multiply the 2 numbers together giving an answer of 0.5. Some did the division upside down giving an answer of 0.08.



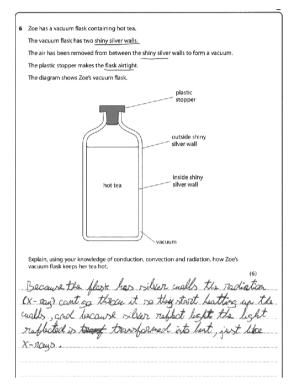
It was pleasing to see that many Students attempted Q6, although they did find it demanding. Only more able students were able to score full marks for explaining how the vacuum flask is able to keep the flask hot and considering ideas about conduction, convection and radiation.

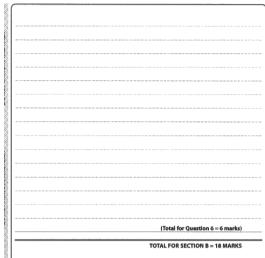




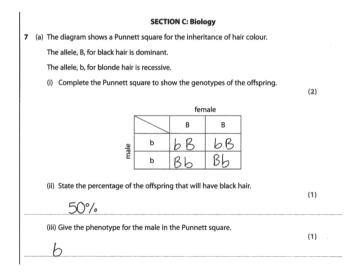
In this example, the student has an understanding that shiny silver wall reflects the heat which gains 1 mark. They state have some understanding that particles will circulate in the flask rising and falling as they cool - which although they have not described it as such, is the mechanism for convection which gained credit. They go on to state that the fact that there is an airtight lid, prevents the circulation from occurring as air cannot get in which gained further credit. Lastly they state that the plastic material of the lid is an insulator which can gained credit in the conduction section. There is another marking point that could be scored on the last section of the second page, stating that the vacuum has no particles so therefore particles cannot conduct, however this conduction marking point had already been scored for the plastic being an insulator so they were not awarded credit here.

The understating of heat transfers is an area that does not seem well understood by Students with many showing clear misconceptions in their answers.





In general Students did well in the first two parts of the biology section in Q7. The majority were able to correctly complete the Punnett square to show the genotypes of the offspring. Of those, most were able to then state the percentage of the offspring that would have black hair. Some Students confused themselves and by swapping the alleles Bb and bB in the same Punnett square, whilst this meant that full marks could still be awarded for part ai, it did mean that they found it hard to give the correct answer for aii and gave the answer 50% rather than 100%. Part ciii confused Students with very few understanding the key term phenotype and many stating bb or b as their answer rather than giving the hair colour of blonde.



Only the more able students were able to score all four marks on this section, as in this case.

	SECTI	ON C: Biol	ogy		
7 (a) The diagram shows a Punnet	tt square fo	r the inher	itance of ha	ir colour.	
The allele, B, for black hair is	dominant.				
The allele, b, for blonde hair	is recessive	·.			
(i) Complete the Punnett sq	uare to sho	ow the gen	otypes of th	ne offspring.	(2)
					(2)
		fen	nale	1	
		В	В		
male	b	Вь	Вь		
Ĕ	b	Вь	Вь		
(ii) State the percentage of t	he offsprin	g that will l	have black l	nair.	
100%					(1)
(iii) Give the phenotype for t	he male in	the Punnet	t square.		(4)
Blonde bair					(1)
nonde nan					***************************************

The majority of students were able to give both organs in the central nervous system in Q8 (a).

8 (a) Homeostasis involves both the central nervous system and the peripheral ner	vous system.
Name the ${f two}$ organs that make up the central nervous system.	(2)
organ 1 Brain	
organ 2 Spinal Lord	

In some cases, students were lazy with their response and gave an answer as spine rather than spinal cord and so could not gain credit.

8 (a) Homeostasis involves both the central nervous system and the peripheral nervo	us system.
Name the <b>two</b> organs that make up the central nervous system.	(2)
organ 1 Spine Orain	***************************************
organ 2 Spine	

Q8 (b) many Students were able to state that temperature is a condition in the body that is controlled by homeostasis.

(b) Blood glucose concentration is one condition in the body that is controlled by homeostasi		
Name <b>two</b> other conditions in the body that are controlled by homeostasis.	(2)	
condition 1 & internal temperature of body		
condition 2 horwords	444444444444444444444444444444444444444	

However, a notable number repeated the stem of the question and stated that blood glucose as an answer and many others copied the key terms from the following question.

(b) Blood glucose concentration is one condition in the body that is controlled by hom	neostasis.
Name two other conditions in the body that are controlled by homeostasis.	
	(2)
condition 1 Hose bloca Mgar	
condition 2 NOTIVEJ	
	I
(b) Blood glucose concentration is one condition in the body that is controlled by h	nomeostasis.
Name <b>two</b> other conditions in the body that are controlled by homeostasis.	
	(2)
condition 1 in Sulin	······································
condition 2 & glucagan	»»

Fewer students attempted Q8 (c) than the other longer response questions. Those that did attempt the question usually knew that insulin and glucagon were used to control blood sugar levels but answers were often vague or confused. Many Students thought insulin raised blood sugar and glucagon lowered it. Glucagon, glycogen and glucose were terms that were often used incorrect in their answers. Often Students did not read the stem of the question carefully and talked about diabetics having to inject insulin rather than answering the question posed.

(c) Explain the role of insulin and glucagon in homeostasis. (4)	(c) Explain the role of insulin and glucagon in homeostasis. (4)
insulin controls the amount of glucose in the body	Insulin and glucagon is what
	controls your blood sugar and
	diabetes also wasomer whether
	your sugar level is up or down
	as you can get diabetes for
	two reasons.

Unfortunately, in this case the student has not answered the question carefully, the question posed asks what the role of insulin and Glucagon is. This learner has stated when insulin is released so does not answer the question about the role of insulin and so does not gain credit. The same applies for the glucagon. If they had stated after each sentence so that the blood glucose is lowered (for insulin) raised (for Glucagon) then marks could have been awarded.

(c) E	xplain the	role of insu	ılin and glu	cagon in hor	neostasis.			4.54
	after	you	lat	your	level	s Of	sugo	(4) V
	goes	wp	and	Tusu	lu i	s kle	аны	*
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	guc	aguu	10	rca a r	COI	***************************************		
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This final example shows an excellent response that scored full marks.

(c) Explain the ro	le of insulin and g	lucagon in h	omeosta	sis.	(4)
Insulin is	released	Ьз	the	pancreas	to lower
the blood o	glucose (	evels	when	chey g	et too
high. The	insulin t	TURNS	gluco	se into	glycogen a
stores it.	Glucagon	does	the	opposite	and is
released o	when the	blood	Suga	ur levelo	are too l
and need	to be	raise	d.	It turns	glycogen
back in	to go	cose.			
				Total for Questi	on 8 = 8 marks)

Q9 was the second of the two 6 markers with the levelled based markscheme. As in Q6 many Students attempted the question but this time often with more success than in Q6. Some good distinction level answers were seen.

Students that did less well, did so as they repeated what was in the stem of the question and therefore scored 0. Many Students were able to pick up one or two marks as pass level for understanding some simple biology such as the evaporation of water from the leaves or the fact that the xylem carried water up the stem. Better Students developed these ideas and mentioned transpiration or talked about stomata and guard cells opening and closing to release water. However, many Students went on to talk about photosynthesis which was not relevant to this question. In this example the learner was awarded merit level with 4 marks. The learner has made clear points from the three areas. They have stated that root hairs absorb the water and that the water is carried up through the xylem They have stated that water is lost to the air by transpiration and this involves the stomata losing water. This was found sufficient to gain the 4

Explain how water is transported from the soil through the roots and stem and is then lost to the air through the leaves.

(6)

Small hours on the roots about the water from the soil

the water sish then carried by the explem the up the soil

the glucose that is absorbed to its carried through the stem by phloem tubes to help feed the plant.

The plants uses what it needly wants and looses water through transpiration to the stan by cause step to out what is the stem in the leaves step to out what the stan is the leaves step to out what the stan is the leaves step to out what the stan is the leaves step to out.

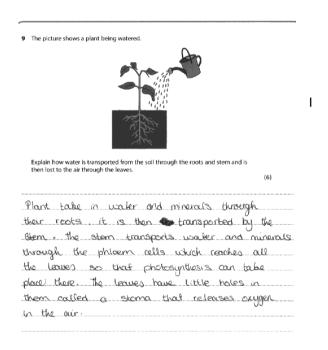
marks.

the Stomata controls wo	hat goes in and allows all the
excess water the plane whough the leaves.	
	Referencied 19 10 Section Medical Text 12 Texturally Section 19 to 18 Advantagement of the section 19 to 19 Section 19 to 19 Section
	(Total for Question 9 = 6 marks)

In this example the learner was awarded pass level with 2 marks. The first sentence is a repeat of the stem so gets no credit, learner should be reminded that repeated information from the stem of the question will gain no credit. The answer then goes on to describe the xylem carrying water from the roots to the stem which is incorrect and so is ignored. The last sentence some information regarding water being evaporated and its movement by transpiration, this was found to be sufficient for a pass level with 2 marks.

1		
	9 The picture shows a plant being watered.	
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	118	
	Explain how water is transported from the soil through the roots and stem and is then lost to the air through the leaves.	
	Firstly the roots absorb the water from	
	the surrounding soil and stores them	
	the plant. Then the zylen transports the w	
	up the roots to the stem and st	
	60 the leaves. The mater is then ex	
	off the leak through a process o	(Total for Question 9 = 6 marks)
	transpiration.	
ı		out increment the factors and in
		III Develop Distributed in the

At first glance, in this last example, this answer looks as though it should gain credit. However, the first sentence is really a repeat of the stem. Any comments regarding the phloem are ignored as they are not relevant to the question. Students should be taught to select from the information that they know only write down what is relevant to the question at the time. In the last sentence, the learner mentions stoma, however this is not linked to losing water so is not creditworthy.







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