Write your name here	
Surname	Other names
Centre Number	Learner Registration Number
Pearson BTEC Level 1/Level 2 First Award	
Applied Science Unit 1: Principles of Science	
Friday 15 May 2015 – Morning	Paper Reference
Time: 1 hour	20460E
Time: Thous	20.002

Instructions

- Use **black** ink or ball-point pen.
- **Fill in the boxes** at the top of this page with your name, centre number and candidate number.
- Answer **all** questions.
- Answer the questions in the spaces provided
 - there may be more space than you need.

Information

- The total mark for this paper is 54.
- The marks for **each** question are shown in brackets
 - use this as a guide as to how much time to spend on each question.

Advice

- Read each question carefully before you start to answer it.
- Keep an eye on the time.
- Try to answer every question.
- Check your answers if you have time at the end.

Turn over ▶

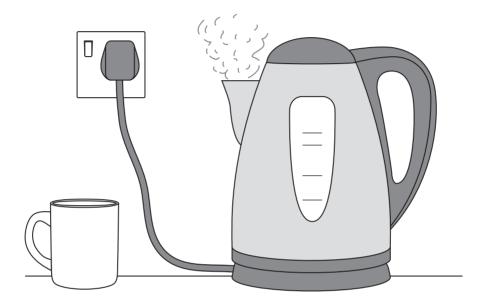
PEARSON

Some questions must be answered with a cross in a box \boxtimes . If you change your mind about an answer, put a line through the box \boxtimes and then put a cross in another box \boxtimes .

SECTION A: Physics

Answer ALL questions.

1 Jasmine boils water in this kettle to make a cup of tea.



(a) (i)	Name the type of energy used to power the kettle.	
		(1)

(ii) Name the type of useful energy produced when	the kettle is boiling.
	(1)

(iii) Name one type of energy wasted when the kettle is boiling.	
	(1)

(1)



(b) Jasmine holds a cup of tea and her hand becomes warm.

The cup transfers heat to her hand through conduction.



State **one** other type of heat transfer.

(1)

(Total for Question 1 = 4 marks)

2 Charlotte visits a hydroelectric power plant.



reservoir

(a) (i) Hydroelectricity is a form of renewable energy.

Give **one** other form of renewable energy.

(1)

(ii) Identify the type of energy stored in the water when it is in the reservoir.

(1)

(b) Charlotte makes a model of a hydroelectric power plant.

In 10 minutes her model produced 2700 joules of energy in total.

(i) Calculate the power produced by her model.

$$power(watts) = \frac{energy(joules)}{time(seconds)}$$

(2)

۱۸

(ii) The efficiency of Charlotte's model is 90%.

Calculate the useful energy supplied by the water in ten minutes.

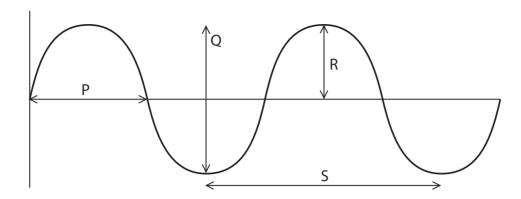
efficiency =
$$\frac{\text{useful energy}}{\text{total energy supplied}} \times 100\%$$

(2)

.....

(Total for Question 2 = 6 marks)

3 The diagram shows part of a wave.



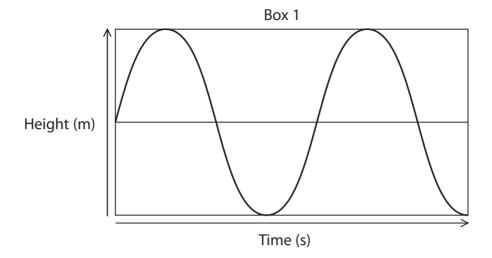
(a) Identify which arrow, P, Q, R, or S, shows the wavelength of the wave.

(1)

- B Q
- **D** S
- (b) Give the unit for frequency.

(1)

(c) The diagram in box 1 shows a wave.



In box 2 draw a wave that has the **same amplitude** but a **higher frequency** than the wave shown in box 1.

(2)



Radio waves	Microwaves	Infrared	Visible	Ultraviolet	X-rays	Gamma ray
Describ	e how radio wa	aves and gamn	na rays differ i	n their properti	ies and uses.	
		J	•			(4)
				(Total for Qu	uestion 3 = 8	marks)
			TC	TAL FOR SECT	ΓΙΟΝ A = 18	MARKS

SECTION B: Chemistry

Answer ALL questions.

- **4** Edward added hydrochloric acid to sodium carbonate. The reaction produced a gas.
 - (a) (i) State what Edward would have **seen** during the reaction.

(1)

(ii) Edward thinks that the gas produced is either carbon dioxide or hydrogen. Draw **one** line from each gas to the correct test for that gas.

(2)

Gas

Test for gas

turns universal indicator purple

carbon dioxide

lit splint gives a squeaky pop

turns limewater milky

hydrogen

relights a glowing splint

turns red litmus paper blue



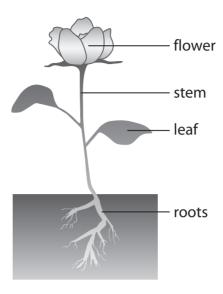
(iii) Give the formula for a molecule of carbon dioxide.	(1)
(iv) Give the formula for a molecule of hydrogen.	(1)
(b) Sodium carbonate is a base.	
Suggest a value for the pH of a sodium carbonate solution.	(1)
(Total for Question	n 4 = 6 marks)

The diagrams show the arrangement of particles in four different substances. В C D Use the diagrams to answer questions 5(a)(i) to 5(a)(iii). Each diagram can be used once, more than once or not at all. (a) (i) Identify which substance is an element. (1) (ii) Identify which substance is a mixture. (1) (iii) Identify which diagram could be of a molecule of water. (1) (b) Sodium chloride can be formed by the reaction of hydrochloric acid with sodium hydroxide. (i) State the name given to this type of reaction. (1) (ii) Write a word equation for the reaction. (2) (Total for Question 5 = 6 marks)

6 The diagram shows part of the periodic table. 1 2 5 6 0 7 19 F 35.5 Cl Explain why fluorine and chlorine are in the positions shown. Your answer should include ideas about the electronic configuration of the atoms periods and groups properties of the elements. (6) (Total for Question 6 = 6 marks) **TOTAL FOR SECTION B = 18 MARKS**

SECTION C: Biology Answer ALL questions. 7 (a) Human responses can be voluntary or involuntary. Give **one** example of a voluntary response. (1) (b) The human body maintains its temperature of 37 °C by homeostasis. (i) Give **one** way in which the human body will respond to being too cold. (1) (ii) Give **one** way in which the human body will respond to being too hot. (1) (c) Give **one** difference between the endocrine system and the nervous system. (1) (Total for Question 7 = 4 marks)

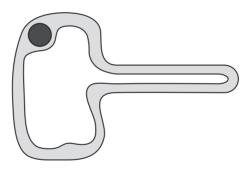
8 The diagram shows a plant.



(a) State **one** function of the leaf of the plant.

(1)

The diagram shows a root hair cell.



(b) (i) Draw a line to label the nucleus.

(1)

(ii) Name the component of the cell that allows entry and exit of substances.

(1)

(c) State **one** way that the root hair cell is adapted to absorb water and minerals from the soil.

(1)

(d) The xylem and phloem are present in the stem of t	he plant.
Explain how the xylem and phloem are adapted fo	
	(4)
Xylem	
Phloem	
	(Total for Question 8 = 8 marks)

9 Mr and Mrs Williams have two children, Hannah and Sam. Mr Williams has blue eyes (bb). Mrs Williams has brown eyes (Bb). Hannah has brown eyes. Sam has blue eyes. Male b b В Bb Bb Female b bb bb Use the Punnett square and your knowledge of genetics to explain why Hannah has brown eyes and Sam has blue eyes. (6) (Total for Question 9 = 6 marks) **TOTAL FOR SECTION C = 18 MARKS TOTAL FOR PAPER = 54 MARKS**