



Level 3 Certificate/Extended Certificate

APPLIED SCIENCE

ASC1/C

Unit 1 Key Concepts in Science
Section B – Chemistry

Mark scheme

January 2022

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2 2 1 A A S C 1 / C / M S

Mark schemes are prepared by the Lead Assessment Writer and considered, together with the relevant questions, by a panel of subject teachers. This mark scheme includes any amendments made at the standardisation events which all associates participate in and is the scheme which was used by them in this examination. The standardisation process ensures that the mark scheme covers the students' responses to questions and that every associate understands and applies it in the same correct way. As preparation for standardisation each associate analyses a number of students' scripts. Alternative answers not already covered by the mark scheme are discussed and legislated for. If, after the standardisation process, associates encounter unusual answers which have not been raised they are required to refer these to the Lead Examiner.

It must be stressed that a mark scheme is a working document, in many cases further developed and expanded on the basis of students' reactions to a particular paper. Assumptions about future mark schemes on the basis of one year's document should be avoided; whilst the guiding principles of assessment remain constant, details will change, depending on the content of a particular examination paper.

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Level of response marking instructions

Level of response mark schemes are broken down into levels, each of which has a descriptor. The descriptor for the level shows the average performance for the level. There are marks in each level.

Before you apply the mark scheme to a student's answer read through the answer and annotate it (as instructed) to show the qualities that are being looked for. You can then apply the mark scheme.

Step 1 Determine a level

Start at the lowest level of the mark scheme and use it as a ladder to see whether the answer meets the descriptor for that level. The descriptor for the level indicates the different qualities that might be seen in the student's answer for that level. If it meets the lowest level then go to the next one and decide if it meets this level, and so on, until you have a match between the level descriptor and the answer. With practice and familiarity you will find that for better answers you will be able to quickly skip through the lower levels of the mark scheme.

When assigning a level you should look at the overall quality of the answer and not look to pick holes in small and specific parts of the answer where the student has not performed quite as well as the rest. If the answer covers different aspects of different levels of the mark scheme you should use a best fit approach for defining the level and then use the variability of the response to help decide the mark within the level, ie if the response is predominantly level 2 with a small amount of level 3 material it would be placed in level 2 but be awarded a mark near the top of the level because of the level 3 content.

Step 2 Determine a mark

Once you have assigned a level you need to decide on the mark. The descriptors on how to allocate marks can help with this. The exemplar materials used during standardisation will help. There will be an answer in the standardising materials which will correspond with each level of the mark scheme. This answer will have been awarded a mark by the Lead Examiner. You can compare the student's answer with the example to determine if it is the same standard, better or worse than the example. You can then use this to allocate a mark for the answer based on the Lead Examiner's mark on the example.

You may well need to read back through the answer as you apply the mark scheme to clarify points and assure yourself that the level and the mark are appropriate.

Indicative content in the mark scheme is provided as a guide for examiners. It is not intended to be exhaustive and you must credit other valid points. Students do not have to cover all of the points mentioned in the Indicative content to reach the highest level of the mark scheme.

An answer which contains nothing of relevance to the question must be awarded no marks.

Question 1

Question	Answers	Extra information	Mark	AO/ Spec. Ref.
01.1	number of protons (in an atom)		1	AO1

Question	Answers	Extra information	Mark	AO/ Spec. Ref.
01.2	number of protons + number of neutrons		1	AO1

Question	Answers	Extra information	Mark	AO/ Spec. Ref.
01.3	highest energy (level) <u>electrons</u>	allow outer (shell) <u>electrons</u>	1	AO1
	are in the s (sub-)shell / level	if no other mark awarded allow 1 mark for element in group 1 or 2	1	AO1

Question	Answers	Extra information	Mark	AO/ Spec. Ref.
01.4	$(1s^2 2s^2) 2p^6 3s^2 3p^6 3d^8 4s^2$		1	AO2

Question	Answers	Extra information	Mark	AO/ Spec. Ref.
01.5	an electron is excited (from a lower energy level to a higher energy level)	Allow electron moves to a higher energy level	1	AO1
	and emit light of a certain frequency/wavelength/colour		1	AO1
	drop back to the lower energy level/ground state		1	AO1

Total Question 1		8
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Question 2

Question	Answers	Extra information	Mark	AO/ Spec. Ref.
02.1	all points correct		1	AO2
	anomalous point identified		1	AO3
	(curved) line of best fit not including the anomalous point		1	AO2

Question	Answers	Extra information	Mark	AO/ Spec. Ref.
02.2	number of shells increases or shielding increases		1	AO1
	(so) weaker attraction between the nucleus and the outer shell electrons		1	AO2

Question	Answers	Extra information	Mark	AO/ Spec. Ref.
02.3	$\frac{(79 \times 24 + 10 \times 25 + 11 \times 26)}{100}$		1	AO2
	= 24.32		1	AO2
	=24.3		1	AO2

Total Question 2		8
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Question 3

Question	Answers	Extra information	Mark	AO/ Spec. Ref.
03.1	the structure is giant or there are many bonds	do not accept giant metallic/ giant ionic/ carbon molecules	1	AO1
	each covalent bond		1	AO1
	is very strong or requires a lot of force to overcome		1	AO1

Question	Answers	Extra information	Mark	AO/ Spec. Ref.
03.2	all the electrons are in fixed places in bonds or no free/delocalised electrons		1	AO1

Total Question 3		4
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