

Mark Scheme 2825/04
June 2005

Mark Scheme	Unit Code 2825 / 04	Session June	Year 2005	Version Fourth draft (pre- standardisation)
Abbreviations, annotations and conventions used in the Mark Scheme	/ = alternative and acceptable answers for the same marking point ; = separates marking points NOT = answers which are not worthy of credit () = words which are not essential to gain credit _____ = (underlining) key words which must be used to gain credit ecf = error carried forward AW = alternative wording ora = or reverse argument			
Question	Expected Answers			Marks
1 (a)(i)	$\frac{4}{3}\pi r_0^3$ $\frac{4}{3}\pi r^3$ alone gets 0			1 [1]
(ii)	$A\frac{4}{3}\pi r_0^3$ allow ecf from (i)			1 [1]
(iii)	equates volumes: $\frac{4}{3}\pi r^3 = A\frac{4}{3}\pi r_0^3$ cancels ($\frac{4}{3}$) and π			1 1 [2]
(iv)	allow ecf giving $r^3 = Ar_0^3$ providing expression includes π graph: straight line through origin, drawn with ruler (2) attempted straight line through origin, drawn freehand gets 1/2 gradient = r_0^3 allow even if line was a curve (1)			3 [3]
(b)(i)	density = m/V $= 197 \times 1.67 \times 10^{-27} / (197 \times \frac{4}{3}\pi [1.4 \times 10^{-15}]^3)$ $= 1.5 \times 10^{17} \text{ kg m}^{-3}$ accept 1.4×10^{17} to 1.5×10^{17} uses 1.6×10^{-27} to give 1.39×10^{17} gets 1,0,1 = 2/3			1 1 1 [3]
(ii)	correct method: <i>either</i> correct ratio of densities <i>or</i> correct ratio of volumes stated clearly or clear from working; $m = \rho V$ so $\rho_n V_n = \rho_a V_a$ $V_n/V_a = \rho_a/\rho_n = 1.9 \times 10^4 / (1.5 \times 10^{17}) = 1.3 \times 10^{-13}$ so $1.3 \times 10^{-11} \%$ ans			1 1 [2]
				12

<p>2(a)(i)</p> <p><i>either</i> (mass / mass-energy / energy of separate nucleons) - (mass / mass-energy / energy of whole nucleus) or AW; <i>or</i> energy needed to separate / split / break apart neutrons and protons (completely); <i>or</i> energy released when separate nucleons / protons and neutrons combine to form nucleus;</p> <p>but NOT energy that binds / holds nucleus together NOT energy to break bonds between nucleons 'atoms' gets 0</p> <p>(ii)</p> <p><i>either</i> high binding energy (/ nucleon) means greater stability / less likely to fuse or fission <i>or</i> nuclides (tend to) move to / react towards the lowest potential energy/ highest binding energy (/nucleon) can be won in any of C, U, or Fe explanations or as separate statement;</p> <p>$^{12}_6\text{C}$ can undergo fusion; $^{235}_{92}\text{U}$ can undergo fission; $^{12}_6\text{C}$ and $^{235}_{92}\text{U}$ are both unstable gets 1/2</p> <p>$^{56}_{26}\text{Fe}$ is stable / does not experience fission or fusion;</p>	<p>1 [1]</p> <p>1</p> <p>1</p> <p>1</p> <p>1 [4]</p>	<p>[1]</p> <p>[4]</p>
<p>(b)(i)</p> <p><i>either</i> neutron that is at (thermal) equilibrium with medium / substance / material through which it is passing <i>or</i> neutron whose (kinetic) energy is equal / comparable / similar / to energy of atoms / molecules through which it is passing <i>or</i> slow moving neutron <i>or</i> neutron having low (kinetic) energy / energy of 1 - 10 eV;</p> <p>(ii)</p> <p>$^{235}_{92}\text{U} + ^1_0\text{n} \rightarrow ^{236}_{92}\text{U}$ '+ neutrino' gets 0</p> <p>(iii)</p> <p>$^{236}_{92}\text{U} \rightarrow ^{135}_{53}\text{I} + ^{95}_{39}\text{Y} + 6^1_0\text{n}$ accept $^{235}_{92}\text{U} + ^1_0\text{n} \rightarrow ^{135}_{53}\text{I} + ^{95}_{39}\text{Y} + 6^1_0\text{n}$ for 1/1</p> <p>(iv)</p> <p>$^{235}_{92}\text{U}$: 7.6 $^{135}_{53}\text{I}$: 8.4 $^{95}_{39}\text{Y}$: 8.6 MeV read from graph, nuclides identified with readings</p> <p>total BE: $^{235}_{92}\text{U}$: 7.6×235 (= 1786) $^{135}_{53}\text{I}$: 8.4×135 (= 1134) $^{95}_{39}\text{Y}$: 8.6×95 (= 817) three expressions</p> <p>so energy released = $1134 + 817 - 1786$ = 165 MeV</p> <p>$8.4 + 8.6 - 7.6 = 9.4$ MeV gets 1,0,1,0 = 2/4 uses 236 to get 157.4 MeV gets 3/4</p>	<p>1 [1]</p> <p>1 [1]</p> <p>1 [1]</p> <p>1</p> <p>1</p> <p>1 [4]</p> <p>1</p> <p>1 [4]</p>	<p>[1]</p> <p>[1]</p> <p>[1]</p> <p>[4]</p> <p>12</p>

3(a)	<p>energy = $V I t$ $= V \times (\text{area under } I-t \text{ graph})$ $= 1.2 \times 4 \times 10^6 \times (20 + 5)$ $= 1.2 \times 10^8 \text{ J}$</p> <p>no V gets 0/4 <i>except</i> if stated 'area under graph = charge' which gets 1/4 area calculation errors eg wrong triangle areas can get 3/4 omits 10^6 can get 3/4</p>	<p>1 1 1 1</p> <p>[4]</p>
(b)	<p>nuclei have (net) charge but atoms don't; nuclei would be deflected by B field / atoms are not;</p>	<p>1 1</p> <p>[2]</p>
(c)	<p>(momentum conservation: $m_H v_H = m_n v_n$ $m_H = 4 m_n$ so) $v_n = 4 v_H$</p> <p>$ke = \frac{1}{2} m v^2$</p> <p>ke of ${}^1_0n = \frac{1}{2} m (4 v_H)^2 = 8 m v_H^2$ } ke of ${}^4_2\text{He} = \frac{1}{2} \times 4 m v_H^2 = 2 m v_H^2$ } subs.</p> <p>so (ke of 1_0n) = 4 x (ke of He) (so 1_0n has 80%, ${}^4_2\text{He}$ has 20% of total ke)</p>	<p>1 1 1 1</p> <p>[4]</p> <p>10</p>
4(a)(i) (ii)	<p>energy = $30 + 10 \times 50 = 530 \text{ keV}$</p> <p>$\frac{1}{2} m v^2 = V e$ $\frac{1}{2} \times 1.67 \times 10^{-27} v^2 = 530 \times 10^3 \times 1.6 \times 10^{-19}$ $v = 1.01 \times 10^7 \text{ ms}^{-1}$ allow 1.0 or $1 \times 10^7 \text{ ms}^{-1}$</p> <p>omits 10^3 and gets 3.2×10^5 2/3 omits 1.6×10^{-19} and gets 2.5×10^{16} 1/3 omits 1.67×10^{-27} and gets 4.1×10^{-7} 1/3 ecf from (i): 500 keV to give 9.8×10^6 300 keV to give 7.6×10^6 40 keV to give 2.8×10^6 all can get 3/3</p>	<p>1</p> <p>[1]</p> <p>1 1 1</p> <p>[3]</p>

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<p>(b)(i)</p> <p>(ii)</p>	<p>$4.3 \times 10^8 \text{ m s}^{-1}$ is greater than speed of light; nothing can travel faster than light;</p> <p>protons accelerate so travel greater distance in same time (1)</p> <p>positrons (much) less massive than protons; (1)</p> <p>so same energy means greater speed (for positrons); (1)</p> <p>positron speeds approach speed of light; (1)</p> <p>cannot exceed it, so speed becomes (approx.) constant; (1)</p> <p>- aware positrons are affected by relativity can get 1/2 of these marks</p> <p>electrodes of equal length means (positron) speeds are constant; (1) any 3</p>	<p>1 1 [2]</p> <p>3 [3]</p>
<p>(c)</p>	<p>rest energy + kinetic energy = $2hf$ omits rest energy or ke 1/2</p> <p>$2 \times (9.11 \times 10^{-31}) (3 \times 10^8)^2 + 650 \times 10^3 \times 1.6 \times 10^{-19} = 2 \times 6.63 \times 10^{-34} f$</p> <p>($1.64 \times 10^{-13} + 1.04 \times 10^{-13} = 13.3 \times 10^{-34} f$) $f = 2.02 \times 10^{20} \text{ Hz}$</p> <p>omits rest energy and gets $7.82 \times 10^{19} \text{ Hz}$ 2/4 omits kinetic energy and gets $1.23 \times 10^{20} \text{ Hz}$ 2/4 any further error (-1) each, to zero</p>	<p>2</p> <p>1</p> <p>1 [4]</p> <p>13</p>
<p>5(a)</p>	<p>sketch showing: 2 dees + alternating source connected;</p> <p>(uniform) magnetic field, in region of dees and normal to diagram;</p> <p>path - continuous circular loops of increasing radius entering both dees at every revolution with direction of travel shown eg by arrow;</p>	<p>1</p> <p>1</p> <p>1 [3]</p>

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(b)	<p><i>either</i> magnetic field exerts a force (on particle) <i>or</i> force from Fleming's left hand rule ;</p> <p><i>either</i> force (always) at right angles to (direction of) motion <i>or</i> force has no component along direction of motion / changes only <i>direction</i> of motion / doesn't change speed; NOT direction of travel perpendicular to magnetic field</p> <p>(so) acts as a centripetal force / experiences force towards centre of circle;</p>	<p>1</p> <p>1</p> <p>1 [3]</p>
(c)	<p>magnetic force = mass x acceleration $BQv = mv^2/R$ <i>or</i> $F = BQv$ (1) and $F = mv^2/R$ (1)</p> <p>$v = BQR/m$</p> <p>time for one orbit $T = 2\pi R/v$ $= 2\pi m/(BQ)$</p> <p>frequency $f = 1/T$ <i>or</i> $T = 1/f$ $= BQ/(2\pi m)$</p> <p>$T = \pi R/v$ to give $f = BQ/(\pi m)$ can get 2,0,1,1 = 4/5</p>	<p>2</p> <p>1</p> <p>1</p> <p>1 [5]</p> <p>11</p>
6(a)	<p>baryon: two examples proton; (1) neutron; (1) 3 particles quoted, including one wrong gets 1/2 only</p> <p>quark composition: proton uud; (1) neutron udd; (1)</p> <p>(aware consists of 3 quarks, unspecified, gets 1/2)</p> <p>stability: proton stable inside (stable) nucleus; (1) proton possible decay / half life = 10^{32} years when free; (1) allow any half life $> 10^{30}$ years</p> <p>neutron stable inside (stable) nucleus; (1) neutron half life = 10/15 minutes when free; (1)</p> <p style="text-align: right;">any 6</p>	<p>6</p>

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(b)	<p>lepton: two examples: electron; (1) positron; (1) neutrino; (1) any 2 (2) (allow muon, tauon) 3 particles including one wrong gets 1 only</p> <p>composition: fundamental (- no quark components); (1)</p> <p>forces: weak force / interaction; (1) electron / positron - (also) electromagnetic / electrostatic force; (1)</p> <p>where found: electron - in atom, outside nucleus or in β^- decay ; (1) positron (rarely) emerging from (high mass) radioisotopes / in β^+ decay / accelerating-colliding machines; (1) neutrino - travelling in space eg from Sun or emitted (with electron / positron) in beta decay ; (1)</p> <p>allow ONCE 'resulting from high energy particle collisions'</p>	<p>6 [12]</p> <p>any 6</p> <p>12</p>
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