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| Surname | | | | | | Other Names | | | | | |
| Centre Number | | | | | | Candidate Number | | | | | |
| Candidate Signature | | | | | | | | | | | |

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General Certificate of Education
January 2002
Advanced Subsidiary Examination



PHYSICS (SPECIFICATION A)
Unit 1 Particles, Radiation and Quantum Phenomena

PA01

Monday 14 January 2002 Morning Session

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| <p>In addition to this paper you will require:</p> <ul style="list-style-type: none"> • a calculator; • a pencil and a ruler. |
|--|

| For Examiner's Use | | | |
|---------------------|------|--------|------|
| Number | Mark | Number | Mark |
| 1 | | | |
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| Total (Column 1) | → | | |
| Total (Column 2) | → | | |
| TOTAL | | | |
| Examiner's Initials | | | |

Time allowed: 1 hour 30 minutes

Instructions

- Use blue or black ink or ball-point pen.
- Fill in the boxes at the top of this page.
- Answer **all** questions in the spaces provided. All working must be shown.
- Do all rough work in this book. Cross through any work you do not want marked.

Information

- The maximum mark for this paper is 60.
- Mark allocations are shown in brackets.
- The paper carries 30% of the total marks for Physics Advanced Subsidiary and carries 15% of the total marks for Physics Advanced.
- A *Data Sheet* is provided on pages 3 and 4. You may wish to detach this perforated sheet at the start of the examination.
- You are expected to use a calculator where appropriate.
- In questions requiring description and explanation you will be assessed on your ability to use an appropriate form and style of writing, to organise relevant information clearly and coherently, and to use specialist vocabulary where appropriate. The degree of legibility of your handwriting and the level of accuracy of your spelling, punctuation and grammar will also be taken into account.

Data Sheet

- A perforated *Data Sheet* is provided as pages 3 and 4 of this question paper.
- This sheet may be useful for answering some of the questions in the examination.
- You may wish to detach this sheet before you begin work.

DATA SHEET

Turn over ▶

DATA SHEET

Answer **all** questions in the spaces provided

- 1 (a) The most abundant isotope of cobalt is represented by ${}^{59}_{27}\text{Co}$.

How many protons, neutrons and orbital electrons are there in a neutral atom of this element?

..... protons

..... neutrons

..... electrons

(2 marks)

- (b) How is the nuclide that has one less proton than the nickel nuclide, ${}^{61}_{28}\text{Ni}$, represented?

.....
(2 marks)

- (c) (i) The heaviest isotope of hydrogen, whose nucleon number is 3, is called tritium.
How is tritium represented?

.....

- (ii) Calculate the charge per unit mass, in C kg^{-1} , for a tritium nucleus.

.....

.....

.....

(3 marks)

TURN OVER FOR THE NEXT QUESTION

7

Turn over ►

- 2 (i) Calculate the energy, in J, of a photon of wavelength 4.50×10^{-7} m.

.....

.....

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.....

- (ii) Calculate the speed of an electron which has the same wavelength as the photon in part (i).

.....

.....

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(5 marks)

5

- 3 (a) When monochromatic light is incident on a metal plate, electrons are emitted only when the frequency of the light exceeds a certain threshold frequency.

Explain, in terms of energy, why this threshold frequency exists.

.....

.....

.....

.....

.....

(3 marks)

- (b) A sodium metal surface is illuminated with incident light of frequency 9.70×10^{14} Hz.
The maximum kinetic energy of an emitted electron is 2.49×10^{-19} J.

Calculate

- (i) the wavelength of the incident light,

.....

.....

- (ii) the energy, in J, of each incident photon,

.....

.....

- (iii) the work function, in J, of sodium,

.....

.....

- (iv) the work function, in eV, of sodium.

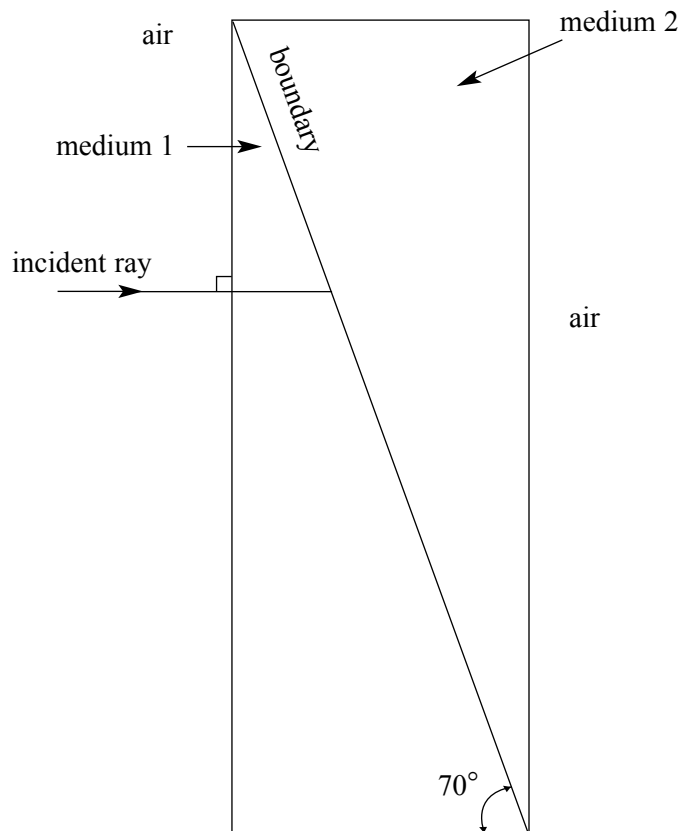
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(7 marks)

10

- 4 Two prisms made from different glass are placed in perfect contact to form a rectangular block surrounded by air as shown. Medium 1 has a smaller refractive index than medium 2.



- (a) A ray of light in air is incident normally on medium 1 as shown. At the boundary between medium 1 and medium 2 some light is transmitted and the remainder reflected.
- Sketch, without calculation, the path followed by the refracted ray as it enters medium 2 and then emerges into the air.
 - Sketch, without calculation, the path followed by the reflected ray showing it emerging from medium 1 into the air.

(4 marks)

(b) The refractive index of medium 1 is 1.40 and that of medium 2 is 1.60.

(i) Give the angle of incidence at the boundary between medium 1 and medium 2.

.....

(ii) Calculate the angle of refraction at this boundary.

.....

.....

.....

.....

.....

(4 marks)

(c) Calculate the critical angle for a ray passing from medium 2 into the air.

.....

.....

.....

.....

.....

.....

(2 marks)

10

TURN OVER FOR THE NEXT QUESTION

Turn over ▶

- 5 (a) An electron and a positron travelling with equal speeds meet head-on and annihilate, releasing $2.0 \times 10^6 \text{ eV}$ of energy. This energy is shared equally by two photons.

Calculate the energy, in J, of each photon.

.....

.....

(1 mark)

- (b) Complete the tables by writing **true** or **false** in the right hand column.

The first table has been completed as an example.

| | |
|--|-------|
| a neutron is a fundamental particle | false |
| an antineutron is not a fundamental particle | true |

| | |
|--|--|
| a neutron is a stable particle | |
| an antineutron is an unstable particle | |

| | |
|---|--|
| a neutron has a rest mass of $1.67 \times 10^{-27} \text{ kg}$ | |
| a antineutron has a rest mass of $-1.67 \times 10^{-27} \text{ kg}$ | |

| | |
|---|--|
| a neutron has no charge | |
| an antineutron has a charge of $1.60 \times 10^{-19} \text{ C}$ | |

(3 marks)

- (c) (i) The exchange particle responsible for the weak interaction is either a W boson or a Z boson.

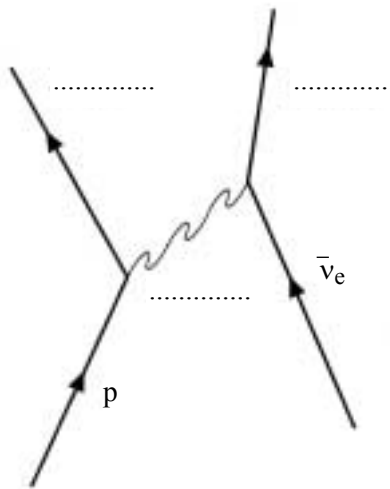
Give the name of another exchange particle and the interaction for which it is responsible.

exchange particle

interaction

- (ii) An antineutrino may be detected via its weak interaction with a proton as shown in the Feynman diagram.

Complete the labelling on the diagram.



(5 marks)

9

TURN OVER FOR THE NEXT QUESTION

Turn over ►

6 (a) (i) How do hadrons differ from all other subatomic particles?

.....
.....

(ii) Give the quark composition of the following particles.

neutron

neutral pion

(iii) Classify the following as either leptons, baryons or mesons.

kaon

muon

(5 marks)

(b) Which is the most stable baryon?

.....

(1 mark)

(c) This table may be useful in answering the questions which follow.

| particle | baryon number | lepton number | strangeness |
|---------------|---------------|---------------|-------------|
| π^- | 0 | 0 | 0 |
| p | 1 | 0 | 0 |
| \bar{p} | -1 | 0 | 0 |
| e^- | 0 | 1 | 0 |
| e^+ | 0 | -1 | 0 |
| $\bar{\nu}_e$ | 0 | -1 | 0 |

The particle X, which is a strange particle, decays in the following way:

$$X \rightarrow \pi^- + p$$

(i) State whether X is a meson, a baryon or a lepton.

.....

(ii) Use conservation laws to decide whether each of the following decays of the π^- is possible. Give a reason for your answer.

(A) $\pi^- \rightarrow e^+ + \nu_e$

Is this decay possible?

reason

(B) $\pi^- \rightarrow \bar{p} + e^- + e^+$

Is this decay possible?

reason

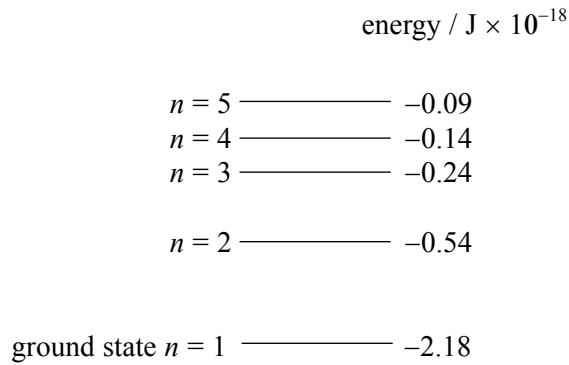
(5 marks)

TURN OVER FOR THE NEXT QUESTION

11

Turn over ►

- 7 The lowest energy levels of a hydrogen atom are represented in the diagram below, which is **not** to scale.



- (i) Describe what happens when a hydrogen atom is ionised.

.....

.....

- (ii) State the minimum amount of energy, in J, required to ionise a hydrogen atom from its ground state.

.....

- (iii) A hydrogen atom excited to the $n = 3$ energy level may emit either a single photon or two photons in returning to the ground state.

Describe what happens to the electron in each case.

.....

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.....

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.....

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.....

.....

- (iv) Use the diagram opposite to identify the transition which produces a photon of energy $2.09 \times 10^{-18} \text{ J}$.

.....

- (v) Calculate the frequency of an emitted photon due to a transition from level $n = 2$ to the ground state.

.....

.....

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(8 marks)

8

END OF QUESTIONS

THERE ARE NO QUESTIONS PRINTED ON THIS PAGE