

OXFORD CAMBRIDGE AND RSA EXAMINATIONS

Advanced Subsidiary GCE

CHEMISTRY

2813/03/PLAN

Practical Examination 1 (Part A – Planning Exercise)

For issue on or after: **14 NOVEMBER 2005**

Candidate Name	Centre Number	Candidate Number												
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TIME The Plan must be handed in by the deadline given by your teacher.

INSTRUCTIONS TO CANDIDATES

- Write your name in the space above.
- Write your Centre number and Candidate number in the boxes above.
- Attach this page to the front of your Plan.

INFORMATION FOR CANDIDATES

- In this Planning Exercise, you will be assessed on the Experimental and Investigative Skill:
Skill P Planning
- You will be awarded marks for the quality of your written communication.
- Detailed notes for guidance are given overleaf.

FOR EXAMINER'S USE		
	Max.	Mark
Planning	16	

Authentication by teacher

I declare that, to the best of my knowledge, the work submitted is that of the candidate concerned. I have provided details on my report form for the Practical Test of any assistance given.

Signature Date

This paper consists of 3 printed pages and 1 blank page.

Notes for guidance

- 1 Your Plan should have a clear and helpful structure and should be illustrated by diagrams, tables, charts, graphs etc. as appropriate. Remember that these can often be used to replace words in the text. Diagrams should be relevant to the content of your Plan and positioned appropriately. Labels on diagrams, flow charts or tables should be clear and concise. Large blocks of text should be included in the word count.
- 2 You should take care to use technical and scientific terms correctly and to write in clear and correct English.
- 3 Your Plan should be hand-written or word-processed on A4 paper, which should have a hole punched at the top left-hand corner. Pages should be numbered and should have a clear margin on the right hand side. You should write (or print) on one side of the paper only and each sheet should be marked with your Centre number and Candidate number.
- 4 You should show that you have consulted an appropriate range and variety of sources. At the end of your Plan you should list clearly the sources you have used. You should refer to these references in your Plan where appropriate. Where you have incorporated material which has been copied directly from a source such as a book or the Internet, this must be acknowledged in your Plan and details included in the references at the end. However, it should be noted that the inclusion of copied material will not in itself gain credit. The list of references should not be included in the word count.
- 5 Your Plan should be based on the use of standard equipment, apparatus, chemicals and other materials available in a school or college science laboratory.
- 6 Your Plan should be 500–800 words. A Plan that is in excess of 800 words is likely to have poor structure and unselective choice of material, so that full credit may not be available. You should indicate the number of words in the margin of the Plan at approximately 100 word intervals.
- 7 When you have finished, tie the pages loosely together (or use a treasury tag), with this sheet on the top, so that the pages turn over freely. Your Centre will give you the date by which your Plan must be handed in.

NOTICE TO CANDIDATE

The work you submit for assessment must be your own.

If you copy from someone else or allow another candidate to copy from you, or if you cheat in any other way, you may be disqualified from at least the subject concerned.

- 1 Any help or information you have received from people other than your subject teacher(s) must be clearly identified in the work itself.
- 2 Any books, information leaflets or other material (e.g. videos, software packages or information from the Internet) which you have used to help you complete this work must be clearly acknowledged in the work itself. To present material copied from books or other sources without acknowledgement will be regarded as deliberate deception.

Declaration by candidate

I have read and understood the **Notice to Candidate** (above). I have produced the work without any help from other people apart from that which I have declared in the work itself. I have acknowledged all source materials in the work itself.

Candidate's signature: Date:

Planning Exercise

You are provided with the following task.

To determine experimentally the relative atomic mass of strontium.

Strontium metal reacts with water producing aqueous strontium hydroxide and a gas.

The relative atomic mass of strontium, which is approximately 90, can be determined experimentally:

- (a) by measurement of the gas,
- (b) by titration of the strontium hydroxide produced.

Plan and describe how you would carry out **both** methods accurately.

[16 marks]

Your plan should include the following:

- relevant chemical knowledge and understanding from the AS part of your chemistry course, including balanced chemical equations;
- detailed practical methods, including a labelled diagram or diagrams, which give full instructions about how to obtain accurate results;
- the quantities of chemicals you would use in the first experiment, (a), and calculations to show how you decided these quantities;
- a specimen calculation to show how the relative atomic mass of strontium would be calculated from the titration in (b);
- a risk assessment and safety precautions.

Any quotations direct from the work of others should be acknowledged by quotation marks, with page references, and the sources should be included in the bibliography.

Quality of Written Communication will be assessed in your work for Skill P.

[14]

Quality of Written Communication [2]

[Total: 16]

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